

Key Words: Cesium, Illite
In situ remediation
Wetlands
Soil amendment
Environmental impact
Turbidity

**SHORT-TERM ENVIRONMENTAL IMPACTS OF ILLITE CLAYS
WHEN USED AS AN *IN SITU* METHOD FOR REMEDIATING ¹³⁷Cs-
CONTAMINATED WETLANDS (U)**

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This document was prepared in conjunction with work accomplished under Contract No. DE-AC09-96SR18500 with the U. S. Department of Energy.

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1.0. EXECUTIVE SUMMARY

Over 1,200 hectares (3,000 acres) of wetlands such as L-Lake; Par Pond; Ponds A, 2, 4, 5, and B; canals; and several creeks (e.g. Steel Creek, Lower Three Runs, and Pen Branch) on the Savannah River Site (SRS) are contaminated with 560 Ci of ^{137}Cs (Carlton et al., 1992). These environmentally sensitive wetlands pose a significant remediation challenge to the Department of Energy (DOE). A new technology is needed to avoid environmentally destructive remediation (Whicker et al., 2004). Current muck and truck technologies destroy the sensitive ecosystems, and may increase dose to workers.

Because of relatively low amount of clay and weak ^{137}Cs retention capabilities of kaolinite-dominant sediments on the SRS, ^{137}Cs has a greater bioavailability here than at any other DOE site. We have previously shown that naturally occurring illite minerals, with a high complexing capability for ^{137}Cs , can sequester ^{137}Cs and reduce its bioavailability when applied to ^{137}Cs -contaminated wetlands. Previous research showed that an *in situ* remediation method using illite minerals reduced ^{137}Cs concentrations in the water 25- to 30-fold, in aquatic plants 3- to 5-fold, and in fish 2- to 3-fold (Hinton et al. 1999, 2001, 2002).

During this funding period (Fy03) we re-sampled study sites in Pond A and R-Canal that had been treated in 2001 with illite clays. The data revealed that ^{137}Cs concentrations in water are still lower than untreated control sites, some 112 weeks later. This encouraging result indicates that the *in situ* remediation technique has a longer-term effectiveness than was previously reported. This positive finding occurred despite the study site experiencing a severe drought, as well as flooded conditions during the two-year period.

However, the application of large amounts of clay to the wetlands creates conditions of high turbidity, siltation, and sedimentation that could have environmental impacts of their own. Sedimentation and turbidity are common by-products of activities such as agriculture, forestry, mining, and road construction. Research reported herein examined the environmental impacts of the clay application in a pond habitat (R-Canal and Pond A), and a simulated stream habitat represented by an artificial stream at the Savannah River Ecology Laboratory's Aquatic Ecology Laboratory (AEL). We also began research on adapting the remediation technique originally designed for ponds and lakes, to stream environments.

Applying the illite clays caused water turbidity to increase by two to three orders-of-magnitude. Although water clarity improved quickly at an exponential rate, with a half time of 1.2 days, some 100 days were required for the high clay treatments to return to the turbidity levels of control sites. The environmental impact of the turbid water was assessed by examining other

water quality measurements, phytoplankton productivity, chlorophyll-a production, the abundance of macro-invertebrates, and fish mortality. No significant differences in pH, electrical conductivity, dissolved oxygen or oxidation-reduction potential were found among treatments. Electrical conductivity values would be expected to decrease if the illites were absorbing ionic solutes from the water column, such as K or NH₄. Thus, the clay application is not likely to decrease the availability of essential plant nutrients in treated waters. No significant differences were observed in phytoplankton productivity, chlorophyll-a levels. No difference in total numbers of macroinvertebrate insects was observed between controls and illite-treated limnocostracans. However, differences in species assemblages of macroinvertebrates were apparent, two years after the initial clay application, between controls and high clay application treatments. No difference was found between the low clay applications and controls. The divergence in species assemblages among treatments significantly increased after the second clay application. These results are very encouraging in that they suggest that although impacts did occur to the macroinvertebrate community, recovery was occurring and that the impacts are not permanent. When the illite was applied to flowing streams (raceways at the Aquatic Ecology Laboratory), water clarity returned within a matter of hours and fish mortality was not affected.

Analyses of particle settling and flocculation rates among three commercially available Todd Light Illites that differed in particle size and moisture content revealed the flexibility of the clay application method and the versatility of the Todd Light Illites to different remediation needs. Air-floated Todd Light Illites were found to disperse quickly throughout the water column and would be ideal for remediating lakes, or possibly long reaches of a contaminated stream. If, however, small localized contaminated hot spots within streams are the target of remediation, then the crude Todd Light Illite's more rapid settling characteristics would allow one to place the mineral specifically where needed, rather than broadly dispersing it within the stream. More work on field deployment methods for various contaminant dispersal scenarios would be useful.

Studies on the *in situ* application of illite minerals to ¹³⁷Cs-contaminated wetlands continue to produce positive results. Additional measurements are needed, however, to determine the long-term effectiveness of the technique, and the environmental impacts on parameters not measured in this study. When coupled to the earlier work of Hinton et al. (1999, 2001, 2002), the results reported herein support the use of illite applications to ¹³⁷Cs-contaminated wetlands on the SRS as an attractive alternative to more environmentally destructive remediation methods.

2.0. INTRODUCTION

2.1. Cesium Contaminated Wetlands on the SRS

Numerous aquatic systems on the Savannah River Site (SRS) such as L-Lake; Par Pond; Ponds A, 2, 4, 5 and B; several creeks (e.g., Steel Creek, Lower Three Runs, Fourmile Branch, and Pen Branch); as well as many kilometers of canals that connected the reactors to their cooling ponds are contaminated with ^{137}Cs , ^{90}Sr , and isotopes of Pu. Combined, these aquatic systems represent over 3,000 acres of wetlands.

Cesium-137 is often the radionuclide of primary concern in these systems because of its biological mobility and long half-life. Cesium's implication to human health was evident during the draw down of Par Pond. Exposure to ^{137}Cs comprised 95% of the lifetime fatal cancer risks to an adult farmer hypothetically living on the exposed sediments, while strontium's and plutonium's contribution to risks were 0.6 and 4.1%, respectively (Whicker et al., 1993).

Inventories of ^{137}Cs within contaminated wetlands on SRS have been estimated at 560 Ci (Carlton et al., 1992). The change in ^{137}Cs concentrations in site streams over time shows that peak releases occurred in the mid-1960s, with the highest median discharge of about 20 Ci released in 1964. Of the total ^{137}Cs activities in SRS streams, Steel Creek, Lower Three Runs, Fourmile Branch, and Pen Branch, contained 54%, 21%, 14%, and 11%, respectively (RAC, 1999).

The problems of contaminated aquatic systems and their associated risks are heightened on the SRS because ^{137}Cs is more biologically available here than at any other DOE facility. Two main reasons for the enhanced mobility and bioavailability of ^{137}Cs in the aquatic systems on the SRS are the low percentage of clays in SRS soils and the fact that the existing clays are dominated by kaolinite and iron oxides. Kaolinite (a 1:1 clay) forms weak bonds with ^{137}Cs . In contrast, 2:1 phyllosilicate minerals, such as illites and micas, form very strong bonds with ^{137}Cs . Illite minerals can sorb ^{137}Cs almost irreversibly, meaning once the ^{137}Cs sorbs to the mineral it is held and will not readily desorb. Interestingly, SRS sediments have some mica-like minerals (Ruhe and Matney, 1980), but they are largely ineffective in forming strong bonds with ^{137}Cs (Seaman et al., 2001), presumably because of lower particle unit charge, due to extensive weathering.

2.3. Remediation Challenge

All sites contaminated with radionuclides on the SRS fall under the Environmental Protection Agency's (EPA) Comprehensive Environmental Response, Compensation, and Liability Act (CERCLA) regulations and thus may represent a remediation challenge to DOE. Remediation of

wetlands is particularly problematic because current cleanup technology for aquatic systems is generally destructive to the ecosystem. If cleanup is required, then a likely scenario would entail draining the impoundment and either immobilizing the sediments, where 95% of the radionuclides reside (Whicker et al. 1990), with some form of grout, or hauling the sediments off to be buried. These methods would be expensive. Costs to remediate just the Old-R Discharge Canal have been estimated at over \$117,000,000 (ERD, 2002). Additionally, the traditional “muck and truck” methods would expose workers to inhalation doses, would destroy the associated ecosystems, and would likely invoke negative responses from stakeholders, South Carolina Department of Health and Environmental Control (SCDHEC), EPA, and National Environmental Policy Act (NEPA) regulators.

2.4. Application of Illite to Sequester ^{137}Cs in Contaminated Sediments

Because high Cs mobility on the SRS is due, in part, to a low adsorptive capability of the kaolinitic-dominated soils, we hypothesized that adding naturally occurring micaceous minerals to wetlands would sequester ^{137}Cs and reduce its bioavailability. We tested numerous minerals in the laboratory and found that the addition of illite minerals enhanced the adsorption and sequestration of ^{137}Cs , even in the presence of competing ions such as ammonium (Hinton et al., 1999). We then applied the concept to a field demonstration in Pond A and R-canal using limnocorrals (Hinton et al., 2002), and found that the addition of illite minerals significantly reduced ^{137}Cs concentrations in the water (**Fig. 1**), as well as reduced the bioavailability of ^{137}Cs in aquatic plants and fish (**Fig. 2**).

2.5. Environmental Impacts of Clay Application

The addition of illite can sequester much of the ^{137}Cs in contaminated wetlands on the SRS and reduce the bioavailability of ^{137}Cs (**Fig. 1 and Fig. 2**). However, the application of large amounts of clay to the wetlands creates conditions of high turbidity, siltation, and sedimentation that could have environmental impacts of their own. Sedimentation and turbidity are common by-products of activities such as agriculture, forestry, mining, and road construction (Henley et al., 2000). Increases in sedimentation and turbidity in aquatic ecosystems have been associated with impacts on phytoplankton (Grobbelar, 1985; Holz et al., 1997) and zooplankton, macrophyte biomass (Henley et al., 2000), and insects (Gray and Ward, 1982; Henley et al., 2000), fish (Newcombe and MacDonald, 1991), and filter-feeding bivalves (Ellis et al., 2002).

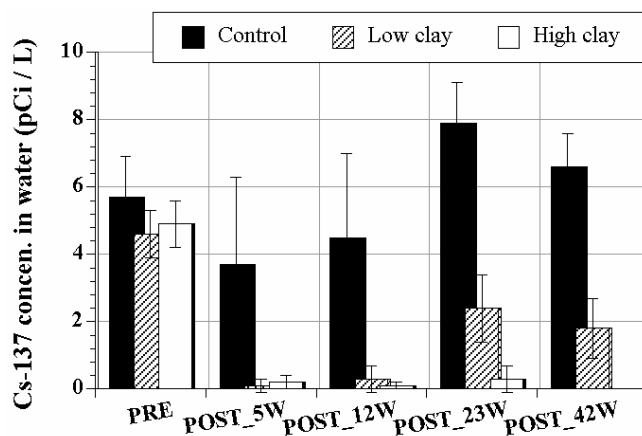


Figure 1. Mean (\pm SE) activity of ^{137}Cs in water samples collected from controls and limnecorrals treated with two levels of illite (0.25 cm and 2.5 cm of illite covering the pond sediments). Data show ^{137}Cs concentrations prior to the addition of the clay treatment (PRE) and the reduction on weeks 5, 12, 23, and 42 subsequent to the treatment. ^{137}Cs levels in the high clay treatment for week 42 were below limits of detection (from Hinton et al. 2002).

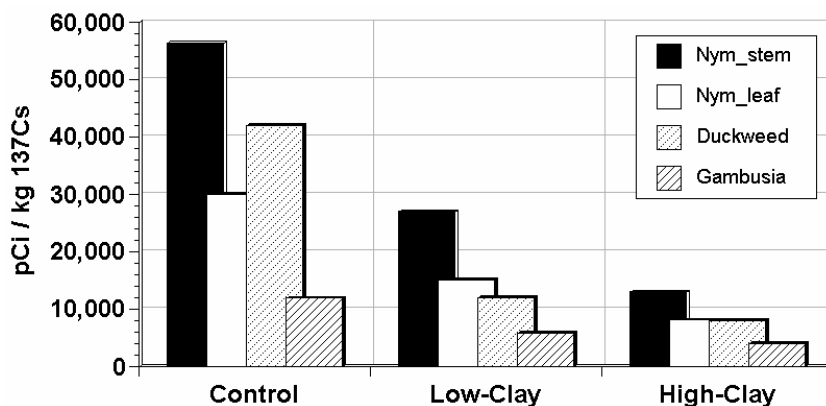


Figure 2. Reductions in mean ^{137}Cs concentrations in the stems and leaves of an aquatic plant species (*Nymphae odorata*), duckweed, and fish (*Gambusia*) following treatment of the water with two levels of illite (0.25 cm and 2.5 cm of illite covering the pond sediments). Results are from replicated limnecorrals placed in Pond A and R-canal (from Hinton et al. 2002).

Aquatic organisms are affected by increased turbidity, primarily from the limitation of light available to algae and macrophytes, and by reduced habitat due to siltation of small spaces. Phytoplankton productivity can suffer from light limitation, abundance of algae can decrease, and communities have been shown to shift from dominance by cyanophytes to flagellated chlorophytes under very turbid conditions (Holz et al., 1997). Decreases in algal productivity

tend to lower concentrations of dissolved oxygen, harming insects and other organisms. Siltation of periphyton can lower the quality of food for snails and herbivorous insects (Hann et al., 2001), and an increase in suspended sediments can lower food quality for filter feeders such as mussels and clams (Ellis et al., 2002; Sobral and Widdows, 2000). Detrimental effects on some species of fish and frogs have also been observed (Newcombe and MacDonald, 1991; Dupuis and Steventon, 1999; Gillespie, 2002). The severity of damage to a wetland ecosystem from suspended sediments, however, depends on the duration of exposure to conditions of high turbidity and on the organisms that inhabit the area, as well as the concentration of suspended sediment (Newcombe and MacDonald, 1991).

3.0. OBJECTIVES

We previously conducted successful laboratory proof-of-concept studies (Hinton et al, 1999 and 2001) and a moderate-sized field demonstration of the *in situ* remediation method in R-Canal and Pond A (Hinton et al., 2002). For the current funding period FY03, we re-sampled our study sites 112 weeks after the initial clay application to examine their longer term effectiveness at reducing concentrations of ^{137}Cs in the water. We then reapplied illite minerals to R-Canal and Pond A limnocorrals to evaluate the short-term environmental impacts of turbidity caused from the clay application. In anticipation of transferring this remediation technology from a lake application to flowing streams, we also performed short-term mesocosm stream experiments at the Aquatic Ecology Laboratory (AEL) to observe the dynamics of clay and water in artificial streams, and to test for impacts to fish.

Our previous test of candidate minerals demonstrated that Todd Light Illite minerals had excellent qualities for remediating ^{137}Cs -contaminated wetlands (Hinton et al. 1999, 2001, 2002). Todd Light is a brand name of illites that can be purchased from a commercial supplier (the Kentucky-Tennessee Clay Company in Nashville, TN). Todd Light Illites, however, come in five different particle sizes with different moisture contents. Particle size and moisture content influence mineral settling rates and could be particularly important to a stream application of the remediation technique. Therefore, three Todd Light Illites (air-floated, semi-dry and crude with moisture contents of 1- 3%, 8-12%, and 20-22%, respectively) were purchased for further laboratory testing.

Our specific objectives were to:

- 1) Evaluate longer-term effectiveness of the illite additions on reducing ^{137}Cs concentrations in water.

- 2) Examine the settling behavior and ^{137}Cs adsorption capacity of three Todd Light illites in the laboratory.
- 3) Assess the short-term effects of turbidity, caused from the illite application, on pond environments in R-Canal and Pond A.
- 4) Conduct artificial stream experiments at AEL to evaluate the effects of an illite application on fish mortality.
- 5) Choose an illite type based on chemical and physical characteristics of the mineral, effectiveness for ^{137}Cs removal, and environmental impacts.
- 6) Select a site for a full-sized, flowing stream, field-demonstration of the remediation technique on the SRS, based on ^{137}Cs concentrations, stream flow, selected stream dimensions, biodiversity, and stream accessibility.

4.0. MATERIALS AND METHODS

4.1. Settling Behavior and Cs Adsorption Capacity of Todd Light Illites

We compared the settling rates and flocculation behavior of three Todd Light Illites within beakers of static water (3g of each clay; 1.5 L water; duration 24 hrs). After 19 hours, the water was vigorously stirred with a glass rod for 10 seconds and the re-suspension and flocculation in the water were qualitatively noted.

The Todd Light Illites were also examined for settling characteristics using a batch beaker test. This study was performed in the laboratory by adding the illites to tap water in beakers. This experiment was conducted in a similar manner (20 g illites in 3.5 L water) to experiments described above except that the water was continuously mixed for 70 h by a Stir-pak[®] Laboratory Mixer (overhead stirrer, Cole-Parmer Instrument Company, Vernon Hills, IL) at a speed of 875 rpm. Suspension and flocculation of illites in the water were observed. The experiment was replicated three times.

The principal characteristic of importance to the remediation method is the mineral's ability to adsorb and retain ^{137}Cs . Previous experiments had shown that illites were very good in this characteristic (Hinton et al 1999, 2000); however, we were unsure how the three Todd Lights might differ. Therefore, adsorption experiments were conducted in 5-L beakers, each containing 20 g of minerals and 3.5 L of ^{137}Cs -contaminated water from Pond B. Each of the three Todd Light Illites was considered a separate treatment and each was replicated three times. All water samples were filtered ($< 0.45 \mu\text{m}$) immediately after collection from Pond B. The three Todd Light Illites were added to the contaminated Pond B water, and after 70 hours aliquots of water

were drawn and compared to control samples for ^{137}Cs activity. Stable element analysis was carried out on sub-samples.

Physicochemical properties of the different types of Todd Light illites were determined following methods of the Natural Resources Conservation Service (NRCS, 1996). Analyses were performed on air-dried colloid samples that had been crushed and passed through a 0.23-mm sieve. All analyses were conducted in triplicate. Cation exchange capacities (CEC) were determined by summation of exchangeable cations (Ca, Mg, Na, K, and Al) extracted with three overnight extractions of 0.05 M BaCl_2 . Elemental compositions of the BaCl_2 and dithionite-reduced extracts were determined by inductively coupled plasma – optical emission spectroscopy (ICP-OES). Organic carbon (OC) was determined using a Leco Carbon Analyzer, Model CR-12 (Leco Corp., St. Joseph, MI). Total sand, silt, and clay concentrations were determined by the micropipette method (Miller and Miller, 1987) in which soil particles were dispersed in a 0.5% (w/v) sodium metaphosphate and 0.1 M NaOH solution. Clay pH was measured in a 1:1 clay-water suspension.

4.2. Turbidity Impact Analyses within Limnocorrals

A field test was conducted in replicated limnocorrals that had been set-up in 2001. Limnocorrals are field-deployed mesocosms that isolate a column of water and its associated sediments from the surrounding water, but allow exchange between sediments, water, and air. They are made of flexible plastic with a floatation collar at the top that keeps the column's open-end floating on the water surface. Retention rings at mid-depth and at the bottom maintain the column's shape. The bottom is weighted down and inserted into the pond's sediments. Three meter diameter limnocorrals were used for this *in situ* remediation deployment.

Sixteen limnocorrals were randomly placed in R-Canal and Pond A during the second week of April 2001 (**Fig. 3**). Four of these limnocorrals were designated as controls receiving no clay, six received low-clay treatments of 27 kg (60 lbs) of illite, enough to cover the sediment with 0.25 cm illite, and six received high-clay treatments of 270 kg (600 lbs) of illite, enough to cover the sediment with 2.5 cm of illite. The application of illite for the first field test of the technique was performed in May of 2001. We re-applied illite to the existing limnocorrals on 3 July 2003. Prior to the second clay application, water samples were taken to characterize the longer-term ability of the clay application to reduce ^{137}Cs concentrations in the water. As was noted after the first application of the illite in 2001, the addition of the illite caused the water to become turbid (**Fig. 4**) for some two weeks. Indeed, this observation was the motivation for the current study of examining the environmental impacts of the turbidity.

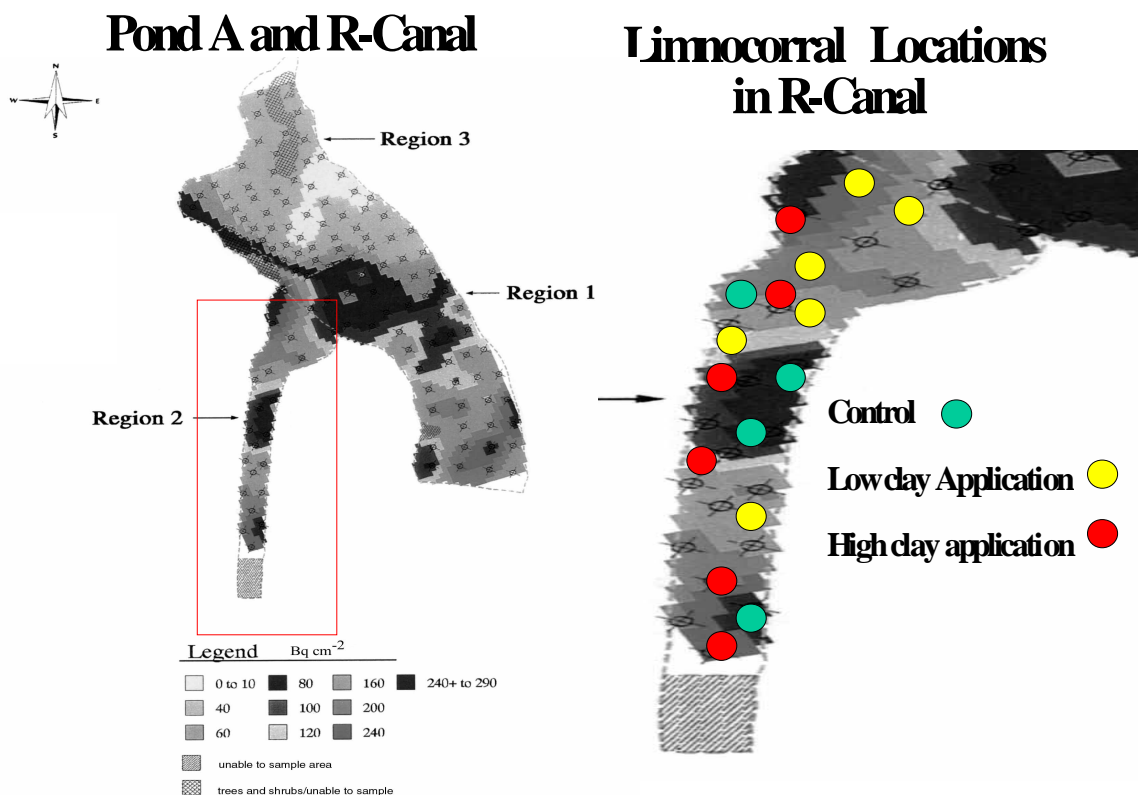


Figure 3. Randomized locations of limnocorrals placed in R-Canal and Pond A on the Savannah River Site. Six limnocorrals were treated with enough clay to cover 0.25 cm of the sediment surface (Low clay application), six received a treatment 10 times that amount (2.5 cm on the sediment surface: High clay application), and four limnocorrals received no clay and served as untreated controls.

4.2.1. Turbidity and Water Quality

Turbidity samples were taken 2 days prior to the clay application, and on days 0, 1, 4, 7, 12, 14, and 18 after the application, by filling 30-mL bottles with water collected at 5 cm depth. Turbidity values in NTU (Nephelometric Turbidity Units) were measured using a LaMotte 2020 Turbidimeter (LaMotte Co., Chestertown, MD). Water quality measurements were taken with a Quanta Water Quality MultiMeter (HydroLab Co., Austin, TX) two days before clay application and on days 4, 7, 12, 14, and 18 after clay application. Temperature ($^{\circ}\text{C}$), pH, dissolved oxygen (DO, mg L^{-1}), electrical conductivity ($\bullet \text{ s cm}^{-1}$), and oxidation-reduction potential (ORP, mV) were recorded.



Figure 4. *Turbid limnocorral in foreground after receiving a high clay treatment, and a control, untreated limnocorral in the background.*

4.2.2. Macroinvertebrates

On 9 June, 2003, 24 days prior to the second clay additions, 30 Hester-Dendy (H-D) multiplate samplers (**Fig. 5**) were placed in 12 of the 16 limnocorrals (4 replicates for each control, low clay, and high clay treatment) located in R-Canal and Pond A. Each sampler consisted of sixteen 7.6-cm square plates separated by approximately 0.3-1.0 cm (total surface area of 0.205 m²). Samplers were tied to the sides of the limnocorrals and hung in the water with their tops approximately 15 cm below the water surface. Fourteen pre-treatment samplers were collected 4 days prior to illite treatment, 30 June 2003, and sixteen H-D samplers were collected 7 days after clay application, on 10 July 2003. Samplers were collected by placing bags made of 400-µm mesh around the sampler in the water, then cutting the string and removing the bag and sampler together. The samplers were stored in coolers and taken to the laboratory until they were sorted for macroinvertebrates. Within three days of collection, the H-D samplers were disassembled, all parts placed in a pan, and all insects found were preserved in 4% formaldehyde solution. Insects were later identified under microscopic examination with the aid of reference materials (Brigham et al., 1982; Merritt and Cummins, 1984).



Figure 5. *Hester-Dendy (H-D) multiplate samplers were placed in 12 of the 16 limnocorrals (4 replicates for each control, low clay, and high clay treatment) located in R-Canal and Pond A to collect aquatic macroinvertebrates.*

Multiple Hester-Dendy samplers were placed in two limnocorrals, one control and one high clay treatment. Comparison of these samplers indicated variation among samplers within a limnocorral. Thus the true community organization would be more accurately characterized by multiple samplers. Though the data may be best viewed as exploratory it does provide valuable incite for the design of a more extensive study. The data for each individual sampler is presented for comparison, but to allow the most accurate characterization of the communities as possible, the multiple samplers were averaged before making comparisons among limnocorrals. This avoids weighting a single corral unduly while allowing use of all of the collected data. Numbers of insects collected in each limnocorral was \log_n transformed prior to analyses. Principal component analysis was employed to summarize the community data. Numbers of components to be examined was determined from their associated eigenvalues and scree plots. Factor scores saved from the PCA were used to assess changes in the species assemblages. Because of the small sample sizes that result if the data is dissected into subgroups, we used an ANOVA model that included clay status (BCA vs. ACA), treatment (control, LCA, HCA) and the clay status*treatment interaction term. Graphs comparing factor scores among the clay status or treatments were composed of the least squares means from this model.

4.2.3. *Phytoplankton Productivity*

Phytoplankton productivity was measured on 4, 12, and 17 days after treatment using the dissolved oxygen method (Wetzel and Likens, 1991; Clesceri et al., 1998). In each limnocorral, water was collected from 10 cm below the surface in three 300-mL glass bottles with ground glass stoppers. Dissolved oxygen levels in one bottle were recorded using a YSI 5100 dissolved oxygen probe (YSI Inc., Yellow Springs, OH), and the two remaining bottles, one clear and one covered in black electrical tape to exclude light, were incubated 10 cm below water surface for approximately 4 hours. Dissolved oxygen levels in light and dark bottles were recorded, and gross primary productivity, respiration, and net primary productivity calculated (Wetzel and Likens, 1991).

4.2.4. *Chlorophyll-a*

Water samples were taken from twelve limnocorrals for chlorophyll-a analysis 12 and 17 days after illite treatment. These grab samples were taken about 5 cm below the water surface in 500-mL dark plastic bottles. Samples were immediately stored in an ice cooler and taken to the laboratory. Water was filtered through glass fiber filters, and the filters were frozen until analysis was performed. The filters were then ground in acetone using a glass grinding tube and centrifuged for five minutes. The supernatant was transferred to quartz cuvettes, and absorbances at 665 and 750 nm were measured with SpectraMax Plus spectrophotometer (Molecular Devices, Sunnyvale, CA). The solution was acidified with 3 drops of 1N-HCl, and absorbances were measured again at the same two wavelengths. Phytoplankton cells contain chlorophyll-a and pheophytins, which are degradation products of chlorophyll-a. We were only interested in the amount of chlorophyll-a, thus we corrected for the amount of pheopigments in the sample. The acid turns all of the chlorophyll into pheopigments (it removes the Mg from the chlorophyll). Calculations based on these absorbances were made to determine the density of chlorophyll-a (Wetzel and Likens, 1991). Calculations were as follows:

$$\bullet \text{ g chlorophyll-a L}^{-1} = \frac{k * F * C * E}{V} \quad [\text{Eq. 1}]$$

where: k = absorption coefficient of chlorophyll-a = 11.0

F = factor to equate the reduction in absorbancy to initial chlorophyll concentration =
2.43

C = (A665b – A750b – (A665a – A750a); b is before acidification, a is after acidification

E = extract in mL

V = volume of water filtered

$$\bullet \text{ g pheophytins} = \frac{k * F * P * E}{V} \quad [\text{Eq. 2}]$$

where: P = 1.7 (A665a – A750a) – (A665b – A750b)

4.2.5. ¹³⁷Cs in Water

Water samples for ¹³⁷Cs analyses were collected 2 days before (1 July 2003) and 14 days after (17 July 2003) clay application from each of the 16 limnocorrals. Three-liter samples were taken about 15 cm below the water surface, filtered with 0.45 μ m Nitrocellulose filter paper, acidified with nitric acid and stabilized with CsCl. The samples were assayed for ¹³⁷Cs in 3 L Marinelli Beakers placed on a 70 X 90 mm² High-Purity Germanium (HPGE) detector (GAMMA Products Inc., Palos Hills, IL). Subsamples were analyzed for stable elements using ICP-MS. Other parameters, such as pH, electrical conductivity, temperature and dissolved oxygen were determined in the field using a HydroLab instrument.

4.3. Fish Mortality Study in an Artificial Flowing Stream

4.3.1. Study Organisms

To evaluate effects of the illite application on stream fish, two locally common species found in the middle Savannah River Basin were used, yellowfin shiners (*Notropis lutipinnis*) and tessellated darters (*Etheostoma olmstedi*). Yellowfin shiners, a relatively small species, are usually less than 65-mm standard length (SL, the distance from the tip of the snout to the base of the caudal fin). These shiners are often abundant in small to intermediate size tributary streams where sufficient gravel in the sediment is present to allow bluehead chubs (*Nocomis leptocephalus*) to build pebble mound nests upon which yellowfin shiners also spawn in an obligate spawning relationship (McAuliffe and Bennett, 1981; Wallin, 1989 and 1992). The benthic-spawning behavior of this species makes them susceptible to disturbance of stream sediments and thus a good test of the environmental impacts from the clay applications. Yellowfin shiners feed on insects and a variety of aquatic invertebrates and are an important

insectivorous predator (Sheldon and Meffe, 1983). The number of native insectivorous cyprinid species present at a site is often incorporated into stream environmental assessments (Karr et al., 1986; Paller, 1996).

The tessellated darter is a relatively small fish rarely exceeding 70-mm SL. Males defend nest sites that consist of a cleaned flat undersurface of submerged wood or stone with a saucer swept in the sediments below. Females are attracted to the nest and eggs are attached to the undersurface in a single layer (Gale and Deutsch, 1985; DeWoody et al., 2000). Upon hatching, the embryos drop into the depression. Tessellated darters are benthic feeders preying on a variety of aquatic insects, especially chironomid larvae (Sheldon and Meffe, 1993). Darters (Percidae, Etheostomatinae) are generally sensitive to environmental disturbance due to their benthic habits (Page, 1983). Consequently, darters are often incorporated into stream environmental assessments (Karr et al., 1986), including on the Savannah River Site (Paller, 1996; Paller et al., 2000).

4.3.2. Study Area and Fish Collection

Using a backpack electrofisher and a 3 mm mesh seine, fish were collected on 22-23 July, 14-15 August, and 3-5 September 2003. Two backpack electrofishers and two crews, covering opposite sides of the stream and moving upstream simultaneously, were used at all collection sites. Approximately 80 tessellated darters and 200 yellowfin shiners were obtained during each collection period from Fourmile Branch, a tributary of the Savannah River located entirely within the SRS (**Fig. 6**). Fourmile Branch, originating near the center of the SRS, runs approximately 24 km to the Savannah River. It is characterized as a blackwater, sandy bottom stream with a channel that alternates between swift runs and pools. Due to thermal effluent releases prior to 1986, Fourmile Branch has an early successional riparian vegetation structure forming a relatively open stream canopy and promoting abundant macrophyte growth (Fletcher et al., 2000). The 57 km² watershed of Fourmile Branch includes a number of SRS facilities, waste sites, and discharges (WSRC, 1998). Except for the extreme headwaters, most of Fourmile Branch is potentially influenced by SRS operations. A number of radioactive and non-radioactive constituents of concern have been detected in environmental media collected from Fourmile Branch (WSRC, 1998).

4.3.3. Samplings and Analyses in Artificial Flowing Streams

Three indoor continuous flow-through raceways (U shape; 35 cm width, 33 cm depth, with 10 m and 8 m lengths for outside and inside banks, respectively) at SREL's Aquatic Ecology

Laboratory were used for this study. Stream water was pumped from Upper Three Runs into a reservoir tank and subsequently released into the raceways. A filter system was employed to remove larger litter and organic matter. The filter system was a two-stage process. A screen was attached to the outlet pipe of the reservoir tank and a 3-mm mesh plastic screen was installed in the raceways. Water depths were maintained at 13-14 cm at the beginning of the raceways and 14-15 cm at the lower end, with an average velocity of 9 cm sec⁻¹ in each raceway. Water temperatures did not vary among raceways. Water temperatures varied slightly among replicates due to seasonal temperature changes: 22-23 °C in August and 20-21 °C in September. Turbidity samples were collected at 2, 3, 4, 5, 6, and 7 m of the raceways.

Upon return to the laboratory, fish were held in a 660-L continuous flow-through raceway until stocked into the experimental raceways. During the first replicate, fish were weighed in a tared beaker of water before being introduced into the experimental raceways (**Table 1**). This added holding time and handling proved too stressful, particularly to the yellowfin shiners that suffered high mortality. Therefore, the weighing procedure was discontinued for the remaining replicates and instead of all being held for three days as all fish were in the first replicate, tessellated darters and yellowfin shiners were held for approximately 24 hours and one to several hours, respectively, prior to stocking in replicates 2 and 3.

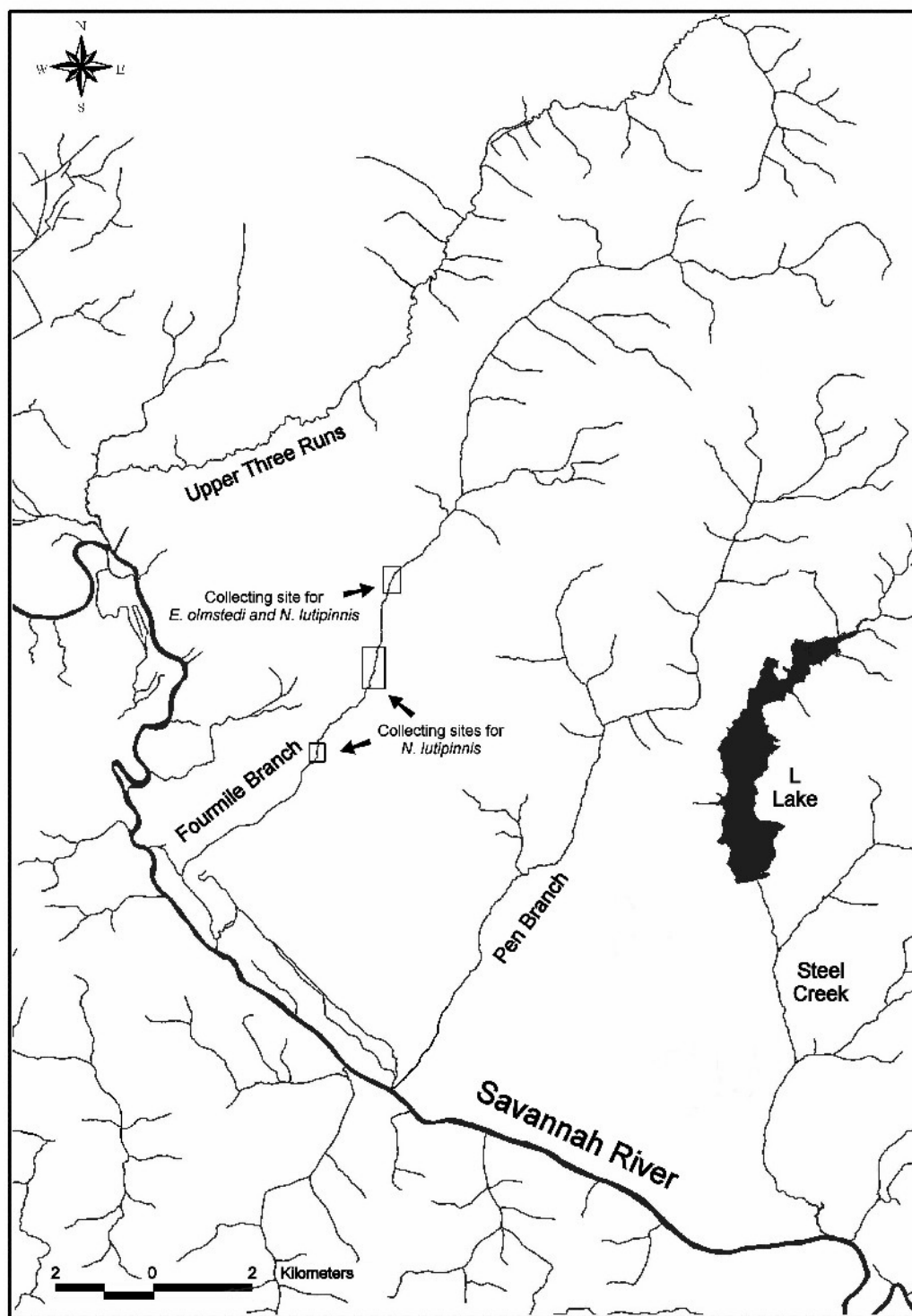


Figure 6. Map of the study area and three fish collection locations on Fourmile Branch.

Table 1. Combined body weight of 50 yellowfin shiners and 20 tessellated darters measured on 24 July 2003. Fish were weighed in a tared beaker of water before being introduced into the experimental raceways.

Fish	Total Body Weight (g)			
	Raceway 1	Raceway 2	Raceway 3	Avg \pm Stdev
Tessellated darter	53.0	48.5	50.0	50.5 \pm 2.3
Yellowfin shiner	96.5	106.0	95.4	99.3 \pm 5.8
Total	149.4	154.5	145.4	149.8 \pm 4.5

Following the holding period, fish were randomly placed into one of the three treatment raceways: untreated control, treated with air-floated illite, or treated with semi-dry illite. For each of the three replicates of the experiment, treatment condition was rotated among raceways. Each raceway was stocked with 50 yellowfin shiners and 20 tessellated darters. Stocked fish were allowed to stabilize in the raceway systems for three days prior to beginning the experiment. Any fish that died in the raceways during this stabilization period were replaced with fish from the holding tank. After the three-day acclimation period, 9.8 kg of air-floated and semi-dry illites were applied to raceways. The amount of illite applied was equivalent to that which can cover the sediment with 0.25 cm illite. Each replicate of the experiment was performed for 7 days. Thus, nine trials were conducted (2 types of illites and 1 control X 3 replicates). Turbidity was measured on the day clay was applied, and then on the 1st, 2nd, 4th and 7th day after application. Fish survival was measured on the 7th day.

To minimize disturbance of fish in the experimental raceways, curtains were hung around each raceway. Habituation to feeding area was avoided by reaching through the curtain and pouring the food into the first ½ of the raceway, at locations that varied each day. Raceway and holding tank fish were fed an average of 4.4 g per day of a 50% mixture of commercially available frozen brine shrimp and bloodworms (Chironomidae larvae) (San Francisco Bay Brand, Newark, CA). The amount of food provided to each raceway was calculated as approximately three percent of the combined body weight of all fish in the raceway (**Table 1**).

4.4. Survey of Potential SRS Locations for Future Field Deployment

4.4.1. Site Selection

The Integrator Operable Unit (IOU) concept represents an effort to understand the cumulative impacts and potential risks of hazardous waste sites within each of the watersheds on the SRS. These watersheds are referred to as Integrator Operable Units because the streams within each

watershed integrate the effects of all waste sites within the watershed. This integration results from the transport of soluble contaminants such as ^{137}Cs from any or all waste sites and discharges within the watershed to the stream by surface runoff or subsurface discharge. There are six IOUs on the SRS. Five of these correspond to the drainage basins of the five major streams on the SRS (Upper Three Runs, Fourmile Branch, Pen Branch, Steel Creek, and Lower Three Runs). The remaining IOU is the portion of the Savannah River and Savannah River Swamp associated with the SRS. All of these streams discharge into the Savannah River.

Using ArcView GIS (Geographic Information System) and a collection of spatial databases, four of the six IOUs (Lower Three Runs, Fourmile Branch, Steel Creek, and Pen Branch) were selected as potential sites for the clay remediation method based on road accessibility and over-flight count rate data of ^{137}Cs . Consequently sections of the stream were assessed as potential sites to conduct a larger-scale demonstration of the illite technology.

4.4.2. Site Assessment

A Global Positioning System (GPS) unit was used to navigate to a hot spot for ^{137}Cs on each stream. Velocity, depth (cm), and width (m) at three locations on each stream were measured for physical characteristic comparison. Each sampling location was separated by 20 m for a total stream segment length of 40 m (**Fig. 7**). Velocity (m sec^{-1}), with a flow meter (Model 2035, General Oceanics, Inc., Miami, FL), and depth were measured in the center and at each edge of the stream at each location. Where depth equaled or exceeded 0.6 m, two current velocity measurements were taken, one at 0.2 times the depth and one at 0.8 times the depth. Where the depth was less than 0.6 m, only one measurement was taken at 0.6 times the stream depth. Five replicate sediment samples were collected with a 3.8-cm diameter PVC tube to a depth of 10 cm at each location. At each location, one in the center of the stream and 4 more samples located 1 m upstream, 1 m downstream, 1 m to the right and 1 m to the left of the center sample were collected. The sediment samples were characterized for ^{137}Cs concentration and stable elements. For Pen Branch, 3 replicate sediment samples were collected at a side and center of the stream. All sediment samples were oven dried to a constant weight at 105°C and sieved at 2 mm.

Analyses of ^{137}Cs in sediments were performed on a Packard Autogamma detector (MINAXI γ Auto-Gamma[®] 5000 Series, Packard, Downers Grove, IL) with a well-type NaI (sodium iodine) crystal (7.6 x 8.3 cm with a 1.5 % relative efficiency). Additional analyses of sediments were performed on a 70 x 90 mm² HPGE detector. Cs-137 standards, traceable to the National Institute of Standards and Technology, were assayed in the same physical geometry as the

samples. Standard QA/QC methods were followed (i.e. use of controls, replicates and blanks for analytical work).

Water samples were also collected in the middle of location 2 in each stream (**Fig. 7**) and analyzed for water quality and ^{137}Cs concentration in the same way as described in section 4.2.

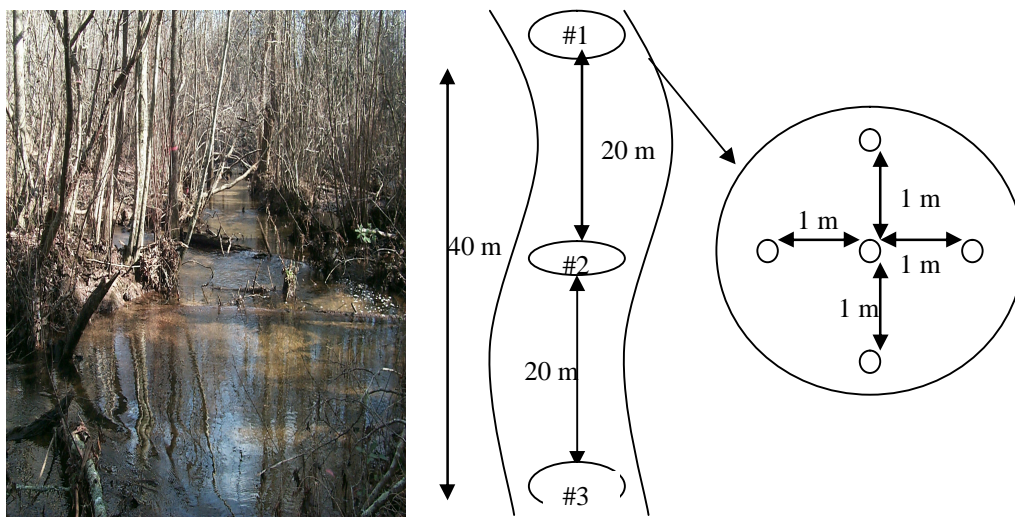


Figure 7. Sampling method for site evaluation of clay application. Five replicate sediment samples were taken at every location, one in the center of the stream, one 1 m upstream and downstream of the center and 1 m to each side of the center.

4.5. Statistical Analyses

The experimental units in this research were the replicated limnocorrals in R-Canal and Pond A, and the replicated raceways in the artificial stream experiments. Thus sample sizes for statistical analyses were the number of limnocorrals or raceways, not the total number of water, macroinvertebrate, fish, turbidity, productivity, or chlorophyll-a samples taken. Data from samplers were combined prior to analysis for macroinvertebrates. Differences between treatments (controls, low clay, and high clay applications) were compared using Analysis of Variance (ANOVA) and General Linear Models (GLMs) of SAS (SAS Institute, Inc., Cary, NC) (SAS, 1999). Statistical analyses for fish mortality used the same SAS software.

5.0. RESULTS AND DISCUSSION

5.1. Settling Behavior and Cs Adsorption Capacity of Todd Light Illites

Analyses of the three Todd Light Illites indicated that surface area, OC, pH, and texture were similar ($p > 0.05$), and that Al, Ca, K, Mg, and CEC differed among them (**Table 2**).

Table 2. Selected physicochemical properties of illites used in the study (mean \pm Stdev; n = 3).

Property ^a	Type of Illite		
	Air-floated	Semi-dry	Crude
pH	4.2 \pm 0.2a [†]	4.7 \pm 0.2a	4.4 \pm 0.1a
OC (%)	1.09 \pm 0.01a	1.07 \pm 0.02a	1.06 \pm 0.02a
CEC (meq 100g ⁻¹)	7.68 \pm 0.03a	6.16 \pm 0.07b	4.92 \pm 0.07c
EC (\bullet S cm ⁻¹)	155.0 \pm 2.1a	111.0 \pm 1.3a	119.1 \pm 2.4a
Surface Area (m ² g ⁻¹)	31.3 \pm 0.5a	28.6 \pm 0.6a	32.2 \pm 1.2a
Clay (%)	55.6 \pm 0.1a	57.4 \pm 4.3a	56.3 \pm 2.0a
Sand (%)	34.8 \pm 0.2a	30.7 \pm 4.6a	35.2 \pm 2.9a
Silt (%)	9.6 \pm 0.1a	11.9 \pm 0.3a	8.5 \pm 0.9a
Ca (mg kg ⁻¹)	599 \pm 6.1a	489 \pm 6.4b	389 \pm 6.2c
K (mg kg ⁻¹)	36.7 \pm 0.5a	18.8 \pm 0.1b	11.0 \pm 0.5c
Mg (mg kg ⁻¹)	182 \pm 5.8a	146 \pm 5.4b	128 \pm 1.4c
Na (mg kg ⁻¹)	23.1 \pm 2.5a	13.4 \pm 0.3a	9.7 \pm 0.3b
Al (mg kg ⁻¹)	40.8 \pm 0.3a	32.9 \pm 3.0b	29.2 \pm 1.1c
Fe (mg kg ⁻¹)	18.9 \pm 0.3a	17.5 \pm 0.4a	25.8 \pm 0.9a

^a OC = organic carbon, CEC = cation exchange capacity, EC = electrical conductivity

[†] Within a property, means among the three types of illites followed by the same letter are not significantly different (p > 0.05).

Turbidity was measured at the end of the laboratory settling experiment with readings of 360 \pm 2.65, 208 \pm 7.09, and 140 \pm 1.53 NTU (mean \pm Stdev.) for air-floated, semi-dry, and crude treatment, respectively. As anticipated, the crude illite produced the lowest turbidity because it had the largest particle size (average size > 2 cm in diameter) and highest moisture content. Semi-dry and crude illites immediately flocculated and gradually re-suspended into the water column from the sediment when stirred. The lower CEC of the semi-dry and crude illites may have induced instability by weakening the repulsive forces between particles (Goldberg et al., 1990) and causing colloid flocculation. The water, after suspension, became darker as time increased (**Table 3**). The suspended particles stayed in the water throughout the experimental period, about 10 days.

Our initial expectation from this experiment was that the results would allow us to recommend a specific Todd Light Illite for use on the SRS. However, the settling characteristics are so different among the three Todd Light Illites examined that they actually add diversity to the remediation tool box, and recommending a single one would likely do disservice to what the

others offer. In other words, knowing the settling characteristics of these minerals now allows WSRC to choose a Todd Light Illite that best suits the site specific needs of that remediation project. If dispersing the mineral throughout the water column is desired, as might be needed if trying to reduce ^{137}Cs mobility in a lake or a long reach of stream, then the air-floated Todd Light Illite would be the best choice. However, if small localized hot spots are being targeted within a stream, then the large particle size and rapid settling characteristics of the crude Todd Light Illite would allow WSRC to control the application and place the mineral in specific locations. Knowledge of the different settling behaviors of the illite minerals provides greater flexibility and control during the remediation of diverse types of wetlands.

Table 3. Results of a particle settling and flocculation experiments using different types of Todd Light Illites (air-floated, semi-dry, and crude).

Characteristic	Air-floated	Semi-Dry	Crude
Water Cleared ^a	3 hours	Immediately	Immediately
Beaker Bottom Covered (%)	100%	70%	50%
Percolation	Slow	Middle	Fast
Water color after suspension	Dark	Light	Medium

^aMeasured by LaMotte 2020 Turbidimeter

Experiments were conducted to compare the three Todd Light Illites' ability to reduce ^{137}Cs concentrations in water. After 70 hours within ^{137}Cs -contaminated Pond B water, the air-floated, semi-dry, and crude Todd Light Illites had similar capacities to remove ^{137}Cs ($p > 0.05$; **Fig. 8**). Thus, choices of which Todd Light Illite to use for a specific remediation project should be based on settling characteristics of the three minerals, as describe above, rather than on their ^{137}Cs absorptive capabilities.

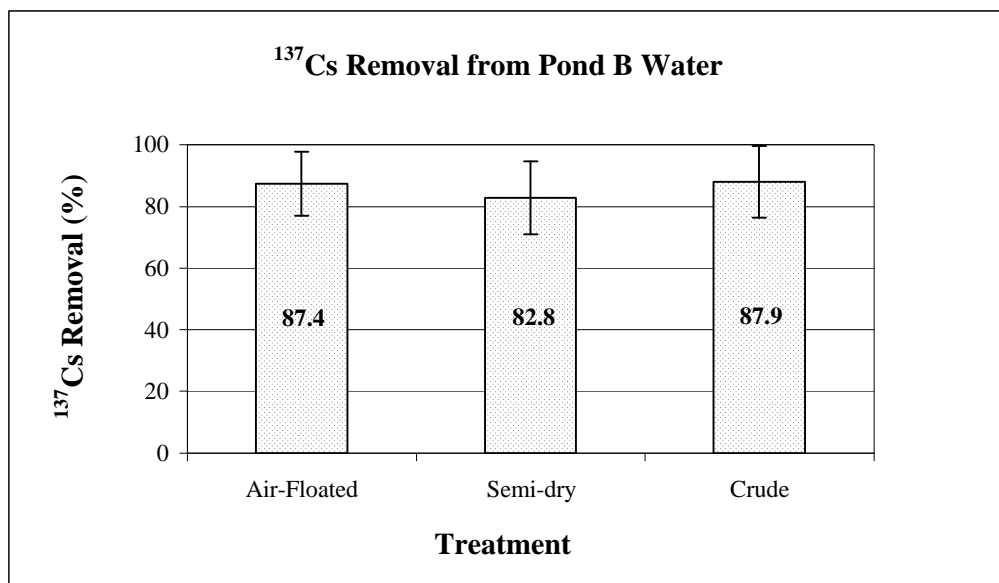


Figure 8. A comparison of mean (\pm St dev.) ¹³⁷Cs removal efficiencies of contaminated Pond B water samples treated with air-floated, semi-dry, and crude Todd Light Illites.

5.2. Limnocorral Turbidity Impact Analyses

5.2.1. ¹³⁷Cs Concentrations in Water: Short- and Longer-Term Effectiveness

Water samples were collected from the limnocorrals in Pond A and R-canal 2 days before the second clay application of illite (112 weeks after the first application), and analyzed to evaluate longer-term abilities of the illite application to reduce ¹³⁷Cs concentrations in the water. During the interim 112 weeks extreme conditions occurred, including both drought and flooding that caused the limnocorrals to dry up, as well as for water to flow over the floatation rings during high water-level conditions. One hundred and twelve weeks after the initial application of illites, average ¹³⁷Cs concentrations in the water of the limnocorrals that received the high clay application (HCA), low clay application (LCA), and controls were 3.14, 4.15, and 6.07 pCi L⁻¹, respectively. Thus, ¹³⁷Cs concentrations in the water of the HCA were still significantly lower than that of the controls ($p < 0.05$). Although mean concentrations of ¹³⁷Cs in the water of the LCA were lower than controls, the differences were not significant ($p > 0.05$; **Fig. 9**). The high application rate of illite resulted in reduced ¹³⁷Cs concentrations in the water for over two years, suggesting that it may be an effective measure for a longer period than previously evaluated.

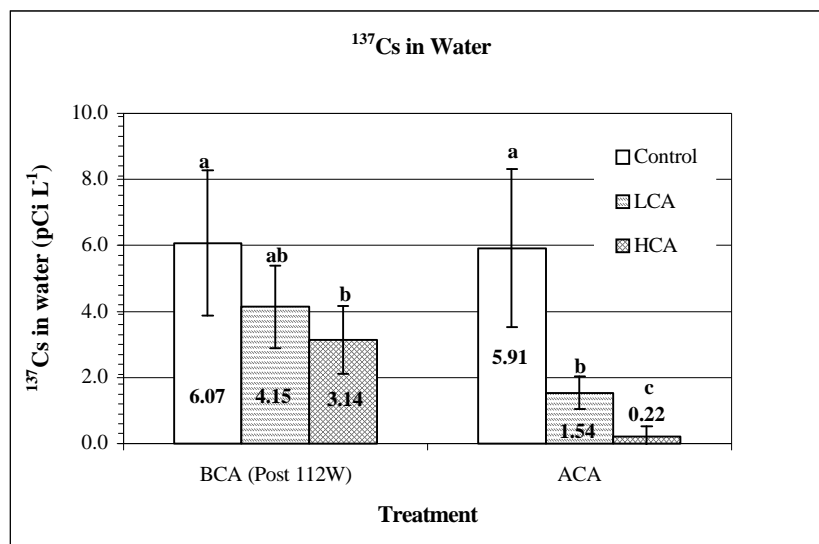


Figure 9. Mean (\pm Stdev.) activity of ^{137}Cs in water samples collected from control and treated limnocorrals (LCA: low clay application; HCA: high clay application). Data show samples taken 112 weeks after the original treatment (BCA; post 112W) and two weeks after a re-application of the illite treatments (ACA). Within time periods (BCA and ACA), treatment means followed by the same letter are not significantly different at $p > 0.05$.

After a second application of the illites to the limnocorrals in Pond A and R-canal, ^{137}Cs concentrations dropped significantly below controls in both treatments (**Fig. 9**), as was observed after the first application. Water samples were collected before (1 July 2003) and two weeks after air-floated Todd Light illites were added to each of the 16 limnocorrals. The clay material effectively sorbed ^{137}Cs out of the water column. Concentrations of ^{137}Cs were significantly reduced ($p < 0.05$) in the water column of the low and high clay-amended treatments. Analyses of water quality and stable element concentrations prior to the start of the experiment and after clay application suggested that adding the clay amendments did not significantly ($p > 0.05$) alter the water chemistry in the illite-treated limnocorrals (**Table 4**).

Table 4. Water quality measurements were taken once before clay application, July 1, and several times after clay application, on July 7, 10, 15, 17, and 21. Temperature, pH, dissolved oxygen (DO), electrical conductivity, and oxidation-reduction potential (ORP) were recorded.

Treatments: Control – no clay application; LCA – low clay application (0.25cm layer of clay on the bottom of the sediment); HCA – high clay application (2.5 cm layer of clay on the bottom of the sediment).

Treatment	Sample No.	Sampling Date					
		July 1	July 7	July 10	July 15	July 17	July 21
Temperature (°C)							
Control	LC1	25.47	28.06	27.60	26.16	27.03	25.24
	LC6	25.82	28.70	28.01	26.47	27.23	25.42
	LC9	25.86	28.08	28.02	26.55	27.28	25.50
	LC10	25.76	28.49	28.19	26.69	27.38	25.67
Treatment Avg ± Stdev		25.73 ± 0.18	28.33 ± 0.32	27.96 ± 0.25	26.47 ± 0.22	27.23 ± 0.15	25.46 ± 0.18
LCA	LC4	25.66	28.35	27.81	26.36	27.15	25.31
	LC8	25.71	27.77	27.96	26.59	27.25	25.52
	LC12	25.81	28.26	28.03	26.61	27.36	25.63
	LC13	25.78	29.20	28.20	26.76	27.36	25.63
	LC15	25.97	29.27	28.79	27.16	27.80	25.88
	LC16	25.99	28.36	28.40	26.98	27.67	25.83
Treatment Avg ± Stdev		25.82 ± 0.13	28.54 ± 0.58	28.20 ± 0.35	26.74 ± 0.29	27.43 ± 0.25	25.63 ± 0.21
HCA	LC2	25.56	27.48	27.54	26.12	27.01	25.20
	LC3	25.56	26.66	27.50	26.09	26.92	25.23
	LC5	25.70	27.27	27.7	26.23	27.05	25.40
	LC7	25.63	27.39	27.90	26.50	27.17	25.49
	LC11	25.82	27.41	27.81	26.46	27.24	25.56
	LC14	25.89	27.22	27.86	26.49	27.25	25.56
Treatment Avg ± Stdev		25.69 ± 0.14	27.24 ± 0.30	27.72 ± 0.17	26.32 ± 0.19	27.11 ± 0.13	25.41 ± 0.16

Treatment	Sample No.	Sampling Date					
		July 1	July 7	July 10	July 15	July 17	July 21
pH							
Control	LC1	7.27	6.62	6.50	7.08	7.17	9.15
	LC6	7.32	7.01	6.10	6.52	6.70	7.50
	LC9	7.24	6.27	6.04	6.30	6.52	6.67
	LC10	6.75	5.66	5.49	5.82	6.08	6.86
		7.15	6.39	6.03	6.43	6.62	7.55
Treatment Avg ± Stdev		± 0.27	± 0.57	± 0.42	± 0.52	± 0.45	± 1.13
LCA	LC4	7.16	5.96	6.08	6.71	6.73	8.13
	LC8	7.08	6.33	5.97	6.31	6.53	7.00
	LC12	6.62	5.73	5.52	5.76	6.25	6.03
	LC13	6.86	5.79	5.76	5.97	6.61	6.25
	LC15	6.55	5.75	6.01	5.70	6.06	5.90
	LC16	6.33	5.53	5.58	5.54	5.90	5.67

Treatment Avg \pm Stdev		6.77 \pm 0.32	5.85 \pm 0.27	5.82 \pm 0.24	6.00 \pm 0.44	6.35 \pm 0.33	6.50 \pm 0.92
HCA	LC2	7.43	6.30	6.07	7.20	7.29	9.69
	LC3	7.28	5.88	5.60	6.53	6.37	8.29
	LC5	7.18	5.49	5.49	6.51	6.34	8.17
	LC7	6.83	6.34	5.32	5.94	5.86	7.05
	LC11	6.35	5.33	5.17	5.65	6.21	5.96
	LC14	6.62	5.99	5.36	5.51	6.13	5.97
Treatment Avg \pm Stdev		\pm 0.42	\pm 0.41	\pm 0.31	\pm 0.64	\pm 0.49	\pm 1.47

		<u>DO (mg L⁻¹)</u>					
Control	LC1	3.78	4.87	4.27	2.04	3.31	1.09
	LC6	6.94	8.41	6.37	4.19	5.36	2.21
	LC9	6.76	5.50	4.35	0.68	0.91	0.34
	LC10	2.11	1.80	2.29	1.03	1.67	0.37
Treatment Avg \pm Stdev		\pm 4.90 \pm 2.36	\pm 5.15 \pm 2.71	\pm 4.32 \pm 1.67	\pm 1.99 \pm 1.58	\pm 2.81 \pm 1.97	\pm 1.00 \pm 0.88
LCA	LC4	2.03	4.03	3.02	1.18	2.74	0.52
	LC8	5.00	5.52	5.12	2.71	3.36	1.77
	LC12	2.44	3.92	3.39	2.24	2.80	1.46
	LC13	4.64	5.88	5.24	4.27	4.79	2.81
	LC15	4.65	3.20	2.49	1.77	2.36	1.30
	LC16	1.97	2.32	2.50	1.82	1.74	1.09
Treatment Avg \pm Stdev		\pm 3.46 \pm 1.45	\pm 4.15 \pm 1.36	\pm 3.63 \pm 1.25	\pm 2.33 \pm 1.08	\pm 2.97 \pm 1.04	\pm 1.49 \pm 0.77
HCA	LC2	1.53	0.27	1.16	0.65	2.29	1.50
	LC3	4.93	4.36	4.68	3.99	4.49	4.61
	LC5	5.97	4.30	4.49	3.62	4.26	4.53
	LC7	4.60	3.93	4.95	3.29	4.25	3.00
	LC11	2.98	2.80	2.68	1.83	2.77	1.97
	LC14	5.09	3.85	4.72	3.40	4.61	4.70
Treatment Avg \pm Stdev		\pm 4.18 \pm 1.63	\pm 3.25 \pm 1.56	\pm 3.78 \pm 1.53	\pm 2.80 \pm 1.28	\pm 3.78 \pm 0.99	\pm 3.39 \pm 1.43

Treatment	Sample No.	Sampling Date					
		July 1	July 7	July 10	July 15	July 17	July 21
<u>Electrical Conductivity (\bullet s cm⁻¹)</u>							
Control	LC1	54	40	44	49	19	49
	LC6	22	22	22	23	22	24
	LC9	26	31	33	37	36	40
	LC10	24	27	26	24	23	26
Treatment Avg \pm Stdev		31.5 \pm 15.1	30.0 \pm 7.6	31.3 \pm 9.6	33.3 \pm 12.3	25.0 \pm 7.5	34.8 \pm 11.9
LCA	LC4	28	29	30	36	35	37
	LC8	27	26	26	29	29	30
	LC12	24	27	26	26	24	25
	LC13	39	25	29	35	35	39
	LC15	27	28	31	31	29	27
	LC16	29	28	26	29	30	31

Treatment Avg ± Stdev		29.0 ± 5.2	27.2 ± 1.5	28.0 ± 2.3	31.0 ± 3.9	30.3 ± 4.2	31.5 ± 5.5
HCA	LC2	35	35	36	43	44	46
	LC3	24	20	20	21	20	21
	LC5	33	30	31	33	33	33
	LC7	27	27	26	27	27	27
	LC11	25	22	23	23	22	22
	LC14	26	23	24	25	25	25
Treatment Avg ± Stdev		28.3 ± 4.6	26.2 ± 5.6	26.7 ± 5.9	28.7 ± 8.1	28.5 ± 8.8	29.0 ± 9.4
ORP (mV)							
Control	LC1	371	312	500	540	538	531
	LC6	382	356	510	574	558	535
	LC9	388	373	436	540	526	534
	LC10	390	384	440	548	537	500
Treatment Avg ± Stdev		383 ± 9	356 ± 32	472 ± 39	551 ± 16	540 ± 13	525 ± 17
LCA	LC4	386	350	508	563	550	556
	LC8	394	378	427	572	529	552
	LC12	399	387	458	569	548	552
	LC13	379	383	459	564	549	541
	LC15	363	387	454	550	533	551
	LC16	404	340	441	552	517	528
Treatment Avg ± Stdev		388 ± 15	371 ± 21	458 ± 28	562 ± 14	538 ± 14	547 ± 10
HCA	LC2	373	151	494	455	528	504
	LC3	378	353	525	584	575	572
	LC5	386	382	507	562	563	574
	LC7	402	399	528	590	566	545
	LC11	401	389	455	567	546	547
	LC14	385	378	460	572	547	556
Treatment Avg ± Stdev		388 ± 12	342 ± 95	495 ± 32	555 ± 50	554 ± 17	550 ± 26

5.2.2. Turbidity and Water Quality

Turbidity following the clay application increased two to three orders of magnitude in the low and high clay treated limnocorrals. Initial concentrations of clay averaged approximately 7,700 mg L⁻¹ and 77,000 mg L⁻¹ for low- and high-clay treatments, respectively. In spite of these high values, the relatively rapid settling of clay may have lessened its impacts on ecosystem processes. Turbidity in the high and low clay-treatment limnocorrals decreased exponentially after treatment (**Fig. 10**). The rate of turbidity decrease was modeled with the following equations:

$$\text{high clay treatment: } y = 3,227 (e^{-0.56x}), R^2=0.95,$$

$$\text{low clay treatment: } y = 184 (e^{-0.58x}), R^2=0.90,$$

where y = turbidity (NTU), x days following treatment.

Turbidity values recorded one day after clay application were less than half of initial values. The model equations presented above agree with the field observations in that the halftime for clearing turbidity is calculated to be 1.2 days. Field observations indicated that the low clay treatments had cleared to levels close to those of the controls after 16 days. Model estimates of the time to reach background turbidity levels in the high clay treatments were 115 days, although field observations from the clay application in 2001 suggest a more rapid clearance rate.

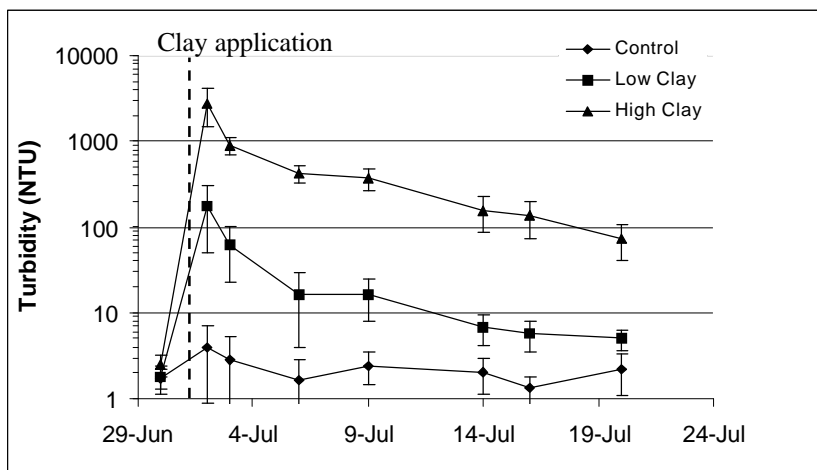


Figure 10. Changes in turbidity (NTU; mean \pm Stdev) are shown from before treatment (29 June, 2003) to nineteen days after treatment. Turbidity decreased exponentially following application.

No significant differences ($p > 0.05$) in pH, electrical conductivity, dissolved oxygen, or oxidation-reduction potential were found among treatments on any sampling date (**Table 4**). Electrical conductivity values would be expected to decrease if illite were absorbing ionic solutes from the water column in addition to ^{137}Cs . Since there is no decrease in electrical conductivity in low or high clay treatments, it seems that clay application will not likely decrease the availability of essential plant nutrients in treated waters, such as K^+ or NH_4^+ , nor did it increase solute concentrations that may have been contributed from the processed mineral.

The water temperature of high clay limnocorrals was significantly lower ($p < 0.05$) than the water temperature of both control and low clay treatments on the first sampling date (July 7) after treatment (**Fig. 11**). On all other sampling dates no significant differences ($p > 0.05$) in water temperature between any treatments were found. A slight decrease in temperature in high clay treated limnocorrals was most likely due to reduced light penetration because of the very turbid water.

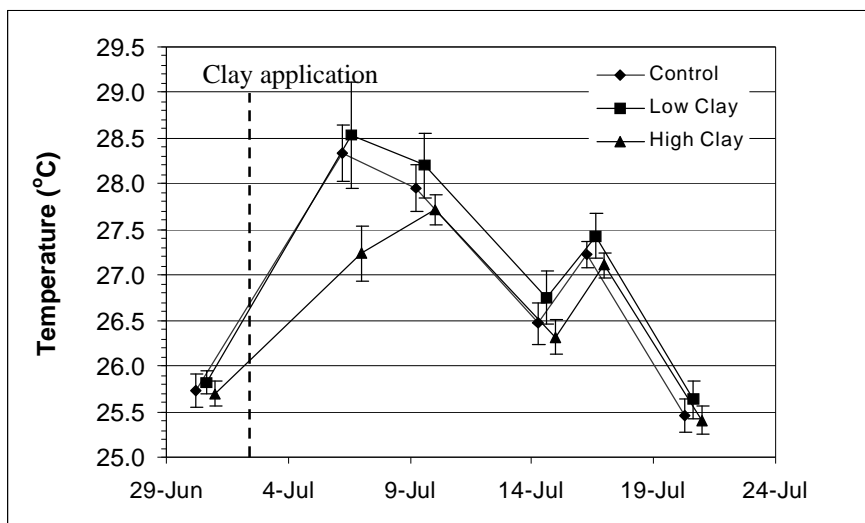


Figure 11. Temperature changes (mean \pm Stdev.) of limnocorrals from June 29 through July 21. Temperature of high clay amended limnocorrals drops below those levels of control and low clay amended limnocorrals briefly after clay application.

5.2.3. Macroinvertebrates, Phytoplankton Productivity, and Chlorophyll-a

No significant differences ($p > 0.05$) in chlorophyll-a levels (**Table 5**) were found between clay treatments, nor were chlorophyll-a levels correlated with turbidity for any treatment. Phytoplankton productivity and respiration levels were not significantly different ($p > 0.05$) among treatments (**Table 6**), and productivity levels were not correlated with turbidity. Large variations among replicates within treatments for both chlorophyll-a and productivity measurements complicated interpretation of these data, and suggests that more replicates were needed.

The five most common families of aquatic insects were Coenagrionidae (Odonata, Zygoptera), Libellulidae (Odonata, Anisoptera), Baetidae (Ephemeroptera), Caenidae (Ephemeroptera), and Chironomidae (Diptera) (**Fig. 12**). Occasional Coleopterans, Megalopterans, and Hemipterans were found, along with other families of Odonates and Dipterans. Total numbers of insects (all families included) found in a Hester-Dendy sampler were analyzed to compare the aquatic insect abundance among treatments (**Table 7**). No significant difference ($p > 0.05$) in abundances was found among treatments. Although the changes within a treatment are highly variable and not significant, mean changes in insect abundance were higher in the clay application treatments (**Table 7**).

Table 5. Water samples were taken for chlorophyll-a analysis on July 21. Treatments: Control – no clay application; LCA – low clay application (0.25 cm layer of clay on the bottom of the sediments); HCA – high clay application (2.5 cm layer of clay on the bottom of the sediments).

Treatment	Sample No.	Chlorophyll-a (\bullet g L ⁻¹)		
		Each Sample	Average	Stdev
Control	LC1	0.00	2.51	4.58
	LC6	9.36		
	LC9	0.67		
	LC10	0.00		
LCA	LC4	8.69	4.01	4.55
	LC8	9.36		
	LC13	4.01		
	LC15	0.00		
	LC16	0.67		
HCA	LC3	4.01	2.79	3.46
	LC5	0.00		
	LC7	7.13		
	LC11	0.00		

Table 6. Gross primary productivity, net primary productivity, and respiration of phytoplankton were measured on July 7, 15, and 21. Treatments: Control – no clay application; LCA – low clay application (0.25 cm layer of clay on the bottom of the sediments); HCA – high clay application (2.5 cm layer of clay on the bottom of the sediments).

Treatment	Avg. \pm Stdev.
<u>Gross Primary Productivity (mg C m⁻³ h⁻¹)</u>	
Control	62.6 \pm 46.5
LCA	41.2 \pm 23.3
HCA	81.3 \pm 20.4
<u>Net Primary Productivity (mg C m⁻³ h⁻¹)</u>	
Control	13.2 \pm 53.6
LCA	4.13 \pm 46.1
HCA	39.9 \pm 41.3
<u>Respiration (mg C m⁻³ h⁻¹)</u>	
Control	49.3 \pm 17.5
LCA	40.2 \pm 28.6
HCA	38.4 \pm 15.4

Principal component analyses produced 2 components that explained 63 percent of the variance (**Table 8**). Factor loadings listed in **Table 8** indicate which taxa were correlated to each component. Abundance of caenid mayflies and libellulid dragonflies were positively correlated to the first component while baetid mayflies were negatively correlated. Chironomids, coenagrionid damselflies and to a lesser extent the libellulid dragonflies loaded positively on the second principal component (PC-II; **Table 8**). In **Figure 13** each polygon encloses all of the points for each of the three treatments, before (**Fig. 13A**) and after (**Fig. 13B**) the second clay treatment. The position and shape of the polygons illustrate that species assemblages shifted after the second clay treatment and differed among treatments. Analysis of Variance of the factor scores of PC-I confirmed that species assemblages were different before and after the second clay treatment and also differed among treatments (**Table 9**). The response of the assemblage was consistent in that not only was the relative abundance of caenid mayflies and libellulid dragonflies higher with baetid mayflies lower after the second clay treatment (ACA), but this pattern was enhanced by larger amounts of clay being added (**Fig. 14**).

Unlike PC-I, mean PC-II factor scores did not differ before and after the second clay treatment or among treatments. The scores for PC-II were generally highly variable resulting in only 17 % of the variance being explained by the ANOVA model (**Table 9**). It is interesting to note that while means did not differ among treatments, both within the BCA samples and in the ACA, standard deviations were lower in the controls than in the low or high clay treatments respectively (BCA: 0.22, 0.76, 1.20; ACA; 0.54, 0.76, 1.8). The higher variability may indicate reduced stability in the species assemblages after clay treatment.

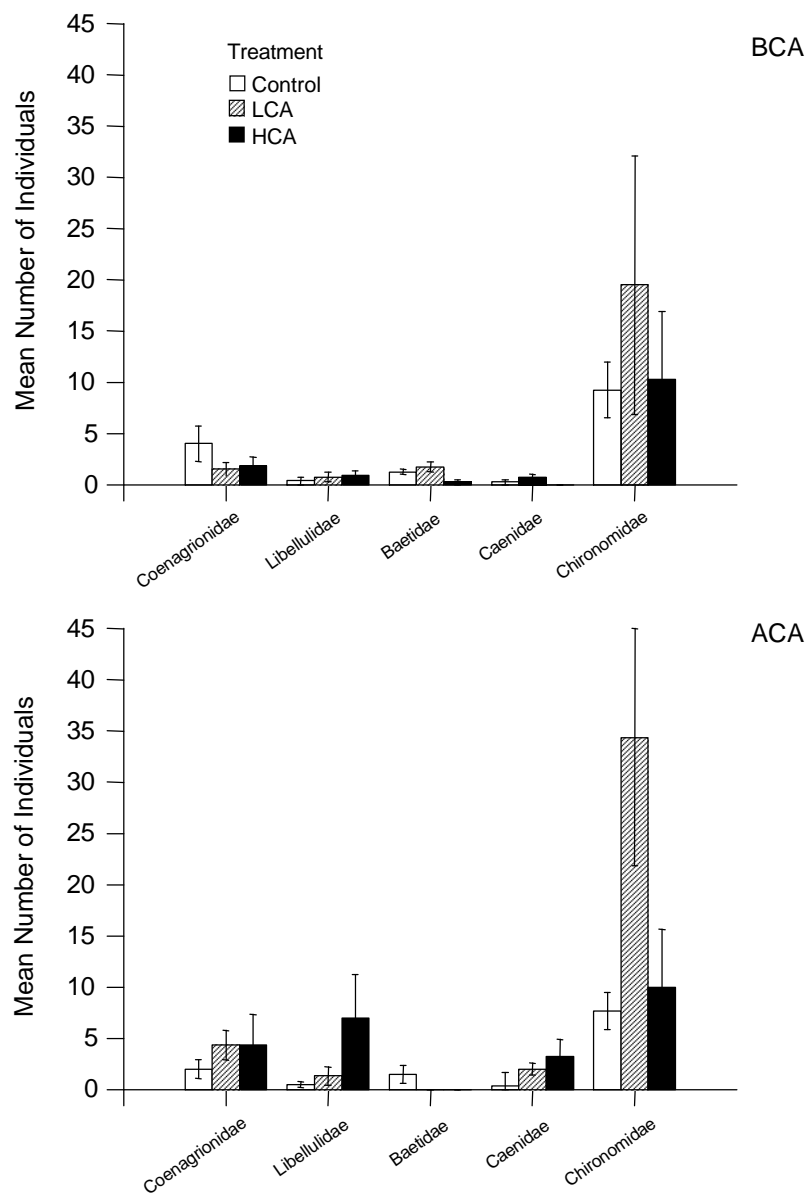


Figure 12. Mean numbers of individuals of the of aquatic insect families most commonly found in the Hester-Dendy samplers. One year after first application (BCA) and seven days after the second application (ACA). Error bars ± 1 SE.

Table 7. Abundance of aquatic insects was measured before (BCA) and after the second clay application (ACA). Total numbers of insects and number of insects belonging to the most common taxa are presented. Taxonomic richness measured as the total number of families and orders of insects found in the Hester-Dendy samplers. Means and standard deviations are for the samplers; see Figure 12 for limnocoral means.

Treatments: Control – no clay application; LCA – low clay application (0.25cm layer of clay on the bottom of the sediment); HCA – high clay application (2.5 cm layer of clay on the bottom of the sediment).

Treatment Sample No.		Number of Insects									
		Family	Order	Total Insect	Coenagrionidae	Libellulidae	Baetidae	Caenidae	Chironomidae	Odonata	Ephemeroptera
BCA											
Control	LC1	3	3	16	9	0	2	0	5	9	2
	LC6-1	3	2	14	5	2	0	0	7	7	0
	LC6-2	5	4	12	1	1	2	2	5	2	4
	LC9	3	3	19	1	0	1	0	17	1	1
	LC10	3	3	14	3	0	1	0	9	3	1
Treatment Avg ±		3.4	3.0	15.0	3.8	0.6	1.2	0.4	8.6	4.4	1.6
Stdev		± 0.9	± 0.7	± 2.7	± 3.4	± 0.9	± 0.8	± 0.9	± 5.0	± 3.4	± 1.5
LCA	LC4	4	3	13	2	0	1	1	9	2	2
	LC8	3	3	60	1	0	2	0	57	1	2
	LC12	5	4	16	0	1	3	1	10	1	4
	LC15	5	4	11	3	2	1	1	2	5	2
Treatment Avg ±		4.3	3.5	25.0	1.5	0.8	1.8	0.8	19.5	2.3	2.5
Stdev		± 1.0	± 0.6	± 23.4	± 1.3	± 1.0	± 1.0	± 0.5	± 25.3	± 1.9	± 1.0
HCA	LC2	3	2	8	0	0	1	0	6	0	1
	LC3-1	3	2	39	7	3	0	0	29	10	0
	LC3-2	2	2	32	0	0	0	0	31	0	0
	LC5	3	2	9	3	2	0	0	4	5	0
	LC7	2	2	2	1	0	0	0	1	1	0
Treatment Avg ±		2.6	2.0	18.0	2.2	1.0	0.2	0.0	14.2	3.2	0.2
Stdev		± 0.6	± 0.0	± 16.4	± 3.0	± 1.4	± 0.5	± 0.0	± 14.6	± 4.3	± 0.5

Treatment Sample No.		Number of Insects									
		Family	Order	Total Insect	Coenagrionidae	Libellulidae	Baetidae	Caenidae	Chironomidae	Odonata	Ephemeroptera
		<u>ACA</u>									
Control	LC1	5	5	8	1	0	1	0	4	1	1
	LC6-1	6	5	38	1	1	0	1	17	2	1
	LC6-2	5	4	17	4	1	0	3	7	5	3
	LC6-3	5	4	34	4	1	0	12	14	5	12
	LC9	5	4	14	4	1	1	0	7	5	1
	LC10	3	3	13	0	0	4	0	7	0	4
Treatment Avg ±		4.8	4.2	20.7	2.3	0.7	1.0	2.7	9.3	3.0	3.7
Stdev		± 1.0	± 0.8	± 12.3	± 1.9	± 0.5	± 1.6	± 4.7	± 5.0	± 2.3	± 4.3
LCA	LC4	4	3	24	4	3	0	3	14	7	3
	LC8	4	3	69	7	1	0	1	57	8	1
	LC15	3	3	36	2	0	0	2	32	2	2
Treatment Avg ±		3.7	3.0	43.0	4.3	1.3	0.0	2.0	34.3	5.7	2.0
Stdev		± 0.6	± 0.0	± 23.3	± 2.5	± 1.5	± 0.0	± 1.0	± 21.6	± 3.2	± 1.0
HCA	LC2	5	4	15	1	0	0	7	5	1	7
	LC3-1	4	2	46	16	12	0	0	16	28	0
	LC3-2	4	3	82	8	24	0	3	47	32	3
	LC3-3	4	3	48	16	15	0	0	15	31	0
	LC5	5	4	28	1	11	0	5	9	12	5
	LC7	2	2	3	2	0	0	0	0	2	0
Treatment Avg ±		4.0	3.0	37.0	7.3	10.3	0.0	2.5	15.3	17.7	2.5
Stdev		± 1.1	± 0.9	± 28.1	± 7.2	± 9.2	± 0.0	± 3.0	± 16.7	± 14.5	± 3.0
<u>Index by Changes between BCA and ACA (ACA-BCA)</u>											
Control		1.4	1.2	5.7	-1.5	0.1	-0.2	2.3	0.7	-1.4	2.1
LCA		-0.6	-0.5	18.0	2.8	0.6	-1.8	1.3	14.8	3.4	-0.5
HCA		1.4	1.0	19.0	5.1	9.3	-0.2	2.5	1.1	14.5	2.3

Table 8. Loadings of abundance of the 5 aquatic insect families collected in the Hester-Dendy samplers on the first 2 principal components.

	PC-I	PC-II
Baetidae	-0.836	-0.087
Caenidae	0.756	0.005
Libellulidae	0.626	0.471
Chironomidae	-0.104	0.835
Coenagrionidae	0.363	0.662
Eigenvalues	2.153	1.015
Percent total variance explained	36	27

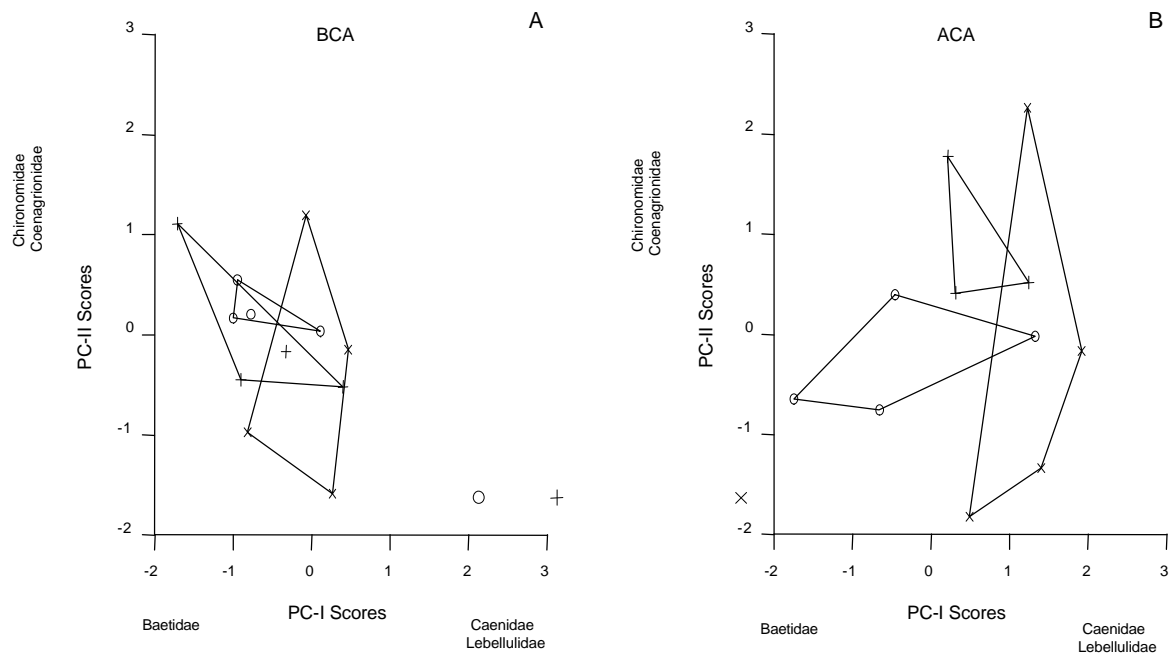


Figure 13. Comparison of factor scores of the first and second principal components. Polygons enclose all points for each treatment: control, low clay application, high clay application. A) BCA-before second clay application; B) ACA-after second clay application.

Table 9. Results of ANOVA of the effects of clay status (ACA vs. BCA) and treatment on each of the two principle components. Aquatic insect families with loadings of absolute value greater than 0.4 are listed beside the component heading.

Source	df	MS	F	p
PC-I (Caenidae, Libellulidae, - Baetidae)				R ² =0.51
Clay Status	1	4.924	7.825	0.01
Treatment	2	2.558	4.064	0.04
Treatment*Clay Status	2	0.647	1.029	0.38
Error	17	0.629		
PC-II (Chironomidae, Coenagrionidae, Libellulidae)				R ² =0.17
Clay Status	1	0.177	0.165	0.69
Treatment	2	1.102	1.029	0.38
Treatment*Clay Status	2	0.910	0.850	0.45
Error	17	1.071		

It is important to note that in PC-I, treatments were more similar before the second treatment (**Fig. 14B**) than after (**Fig. 14C**). While recovery does not appear complete, the more widely overlapping polygons in **Figure 13A** (BCA) indicates more similar assemblages than the more separated polygons in **Figure 13B** (ACA). Thus communities may be on a trajectory to recover from the disturbance.

Caenid mayflies are sprawlers, a functional group that generally lives on top of the substrate, and some taxa (e.g. *Caenis*) frequently inhabit littoral sediments (Merritt and Cummins 1984). Similarly libellid dragonflies are sprawlers or climbers and may live on the bottom sediments or climb aquatic vegetation. It is interesting that these two families increased their prevalence on the samplers in response to the clay application. Because the benthos was not sampled during the experiment, we do not know the abundance of these organisms before or after the treatment. However, it is possible that some bottom dwelling caenids and libellulids relocated from the bottom sediments to suspended samplers to avoid clay deposition. This behavior would also be consistent with an increase in abundance on the samplers in the clay application treatments. In streams, behavioral avoidance by aquatic insects of suspended solids has been noted (Ward 1992) so avoidance of sediment deposition when possible seems plausible. Coenagrionid damselflies

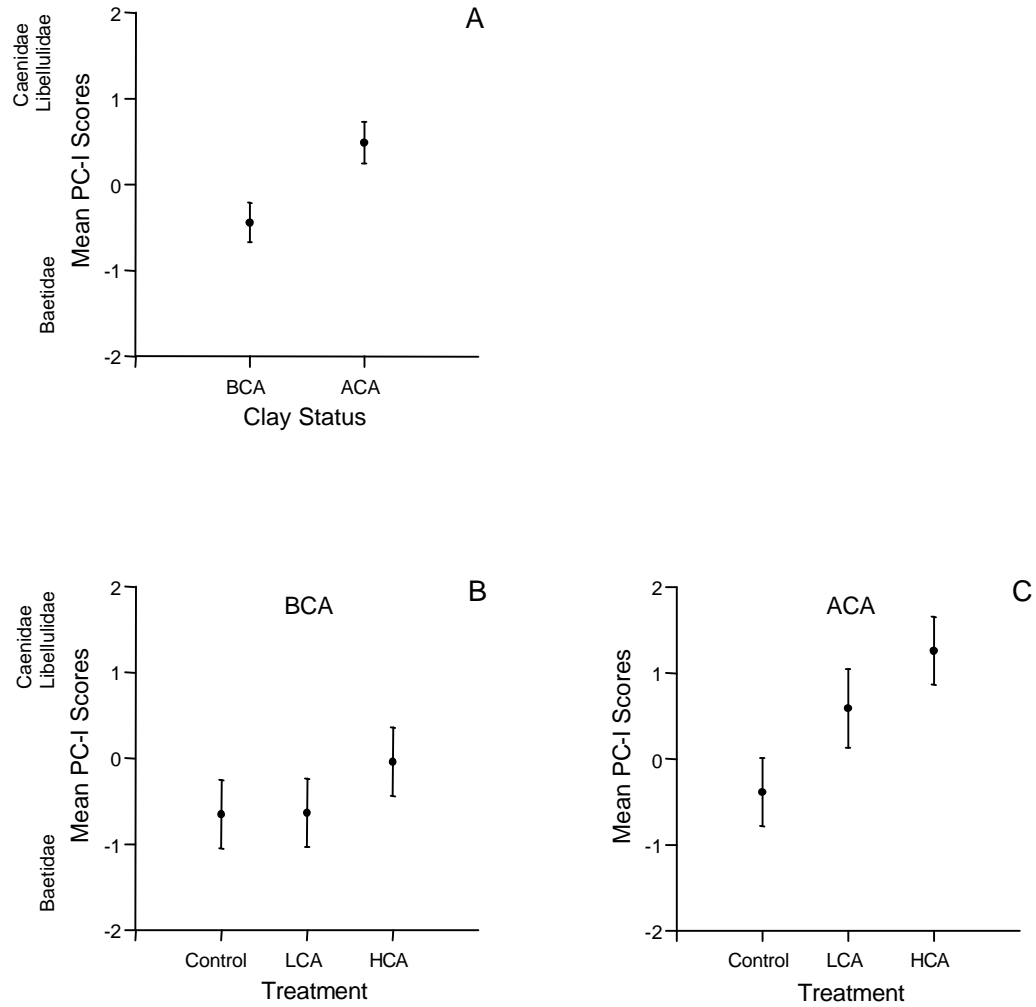


Figure 14. Response of PC-I scores to clay treatment. Least squares means saved from ANOVA ± 1 standard error. A) mean scores before (BCA) and after (ACA) the second clay application with the effects of treatment controlled. B) Mean scores across treatments before the second clay application: no clay applied (control), low clay application (LCA) and high clay application (HCA). C) Mean scores across treatments after the second clay treatment.

that are typically climbers and frequently live amongst dense vegetation of still waters (Merritt and Cummins 1984) may have been less affected by the sedimentation. Baetid mayflies swim with fish-like body undulations to submerged objects or aquatic vegetation to which they cling. Baetids were in low numbers prior to disturbance, but decreased or possibly disappeared after clay application. Chironomids as a family are a very diverse group and more difficult to generalize.

In general hanging a single sampler near the water surface may have characterized the phytophyllous insect population better than the benthos. Because of the potential impact of covering the bottom with sediment and possible complex behavioral responses of aquatic insects that may have affected sampling, more extensive sampling of the benthos and multiple water levels will be conducted in future studies.

The lack of impact on dissolved oxygen is consistent with the lack of change in algal abundance or productivity and in future studies this could be confirmed with comprehensive analyses of diel cycles of dissolved oxygen concentrations. Over longer periods of time changes in algal populations are responsible for cascading effects on higher trophic levels in aquatic ecosystems because of resulting decreases in dissolved oxygen and reduced food availability and quality (McCormick and Cairns, 1997). Without changes in algal abundance or productivity, therefore, lack of a more severe impact on water quality and insect community becomes less surprising. Composition of algal populations was not examined, and overall abundance can sometimes mask changes in community structure (Beklioglu and Moss, 1996). High turbidity could shift limitations on algal productivity from nutrients to light availability (Holz et al., 1997), and resultant changes in phytoplankton community structure could be examined as another indicator of the effects of clay treatments. Benthic algae, not examined in this study, is often an important component of freshwater food webs, particularly in shallow clear lakes (Hecky and Hesslein 1995) like Pond A. Siltation of surfaces can limit algal growth and decrease the proportion of its organic content (Graham, 1990). Food quality for consumers of periphyton is reduced even when there is no decrease in total periphyton abundance (Hann et al., 2001). Snails or tadpoles might be good indicators of a reduction in periphyton quality, and filter-feeding bivalves such as freshwater mussels and clams would make good subjects to test whether clay decreases the food quality of phytoplankton. In the current study, snail abundance did decline on the Hester-Dendy samplers in response to clay application.

In undamaged freshwater ecosystems, any given family of insects normally makes up only 3-23% of the total insect community (Hart and Fuller, 1974). Control and pre-treatment insect samples from the limnocorrals showed that Chironomids and Coenagrionids often constituted

greater percentages of the population than this. In addition, some taxa of the dominant families found in the samplers are known to exhibit high tolerances to low water quality, including low dissolved oxygen. Coenagrionids have been shown to maintain higher populations in low water quality (Brigham et al., 1982); Coenagrionid and Libellulid nymphs both need to hide from prey to be able to hunt, and low levels of increased siltation may help them do this. Chironomids, and Baetid and Caenid mayflies are mostly detritivores or scrapers with a few predaceous chironomids, but some taxa are tolerant to low water quality. Variation in the effects of suspended sediment on macroinvertebrates is also often attributed to the history of the exposed area (Lenat et al., 1981; Gray and Ward, 1982). The initial insect communities in R-Canal may have already been indicative of poor quality waters, and the increased turbidity may have had a weaker effect than it would have had on a more pristine area.

5.3. Fish Mortality Study in Artificial Flowing Streams

The effects of turbidity on fish mortality were studied. Effects of turbidity have been previously grouped into three categories: (1) Lethal effects – lethal effects kill individual fish, cause population reductions, or damage the capacity of the ecosystem to produce fish, (2) Sub-lethal effects – sub-lethal effects injure the tissues or physiology of the organism, but not enough to cause death, (3) Behavioral effects – behavioral effects change activity patterns and/or detrimentally alter fish behavior. In this study, we only examined lethality.

Three continuous water-flowing raceways were used for this study and the turbidity prior to clay application was examined as a background level (**Fig. 13**). The raceways used pumped water from Upper Three Runs and litter, and organic matter in the stream water caused the pre-treated turbidity. Turbidity values (average value of three raceways) decreased exponentially with time and stabilized between 7-8 NTUs, twelve hours after the system was turned on. Based on this pretreatment characterization of the raceways we turned on the flowing system for 12 hours before fish were introduced.

Turbidity values for air-floated, semi-dry, and control treated raceways were measured before, during and after clay application (**Table 8**). Samples taken two hours after the application for air-floated treated raceways and one hour after for semi-dry treated raceways showed that turbidity had decreased to levels close to those of the controls ($p > 0.05$). Such a rapid reduction in turbidity would likely have diminished impacts to fish.

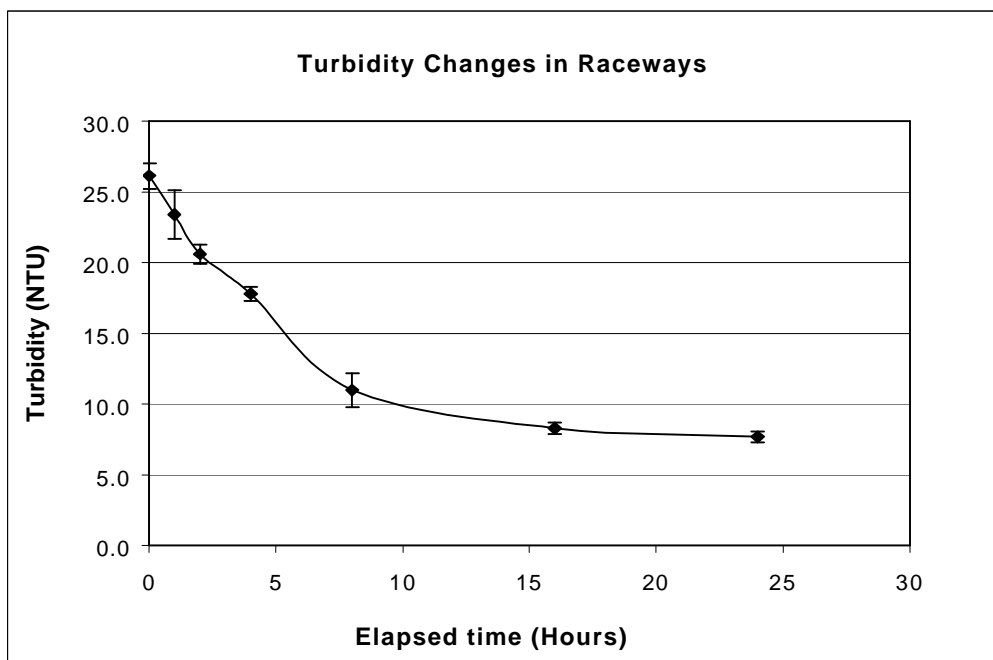


Figure 13. *Turbidity (mean ± Stdev.; n=3) was measured in the raceways as a function of elapsed time after introducing Upper Three Runs water and prior to clay treatment. Turbidity decreased exponentially with time ($y = 22 (e^{-0.05x})$; $R^2=0.88$).*

Fish mortality was tested by introducing 50 yellowfin shiners and 20 tessellated darters into the control, air-floated, and semi-dry illite treated raceways (**Fig. 14**). After seven days, no significant ($p > 0.05$) differences in fish mortality occurred in control and illite treated raceways. Although fish mortality for yellowfin shiners was 4.0 and 2.7% for air-floated and semi-dry illite treated raceways, respectively, these differences were not statistically significant and fell within the range of mortality observed in the control raceway (6.7%). Variables other than illite that likely contributed to fish mortality include stress of handling, life stage, physiological condition, water temperature, and disease. All 20 tessellated darters in each of the three replicates survived throughout the entire seven-day experiments. It is worthy to note that in preliminary experiments, when small refuges from the clay occurred in a raceway, many fish (both dusky shiners and tessellated darters) often remained in these areas, indicating a behavior avoidance of the clay. Such refuges were minimized in the three replicates. In the treatment of a real stream, many fish will likely avoid the clay treatment by migrating upstream, downstream or into any available side channels

Table 8. Changes in turbidity (mean \pm Stdev.) for air-floated, semi-dry, and control treated raceways were measured before (BCA), during (DCA) and after clay application. Turbidity values were measured 1, 2, 4, 8, 24, 48, 96, and 168 hours after the clay application and at 2, 3, 4, 5, 6, and 7 m of raceways.

Time	Sample Collection					
	2m	3m	4m	5m	6m	7m
<u>Air-floated Clay</u>						
BCA	4.34 \pm 0.49	4.17 \pm 0.43	3.86 \pm 0.22	4.26 \pm 0.90	4.09 \pm 0.57	4.34 \pm 0.49
DCA	525 \pm 250	314 \pm 28.5	232 \pm 38.3	371 \pm 145	364 \pm 68.1	332 \pm 125
1hr	21.1 \pm 13.8	18.9 \pm 12.0	20.3 \pm 14.1	24.9 \pm 15.2	28.3 \pm 18.1	20.8 \pm 8.9
2hr	9.57 \pm 2.02	11.0 \pm 1.68	23.4 \pm 5.16	10.2 \pm 4.71	11.2 \pm 2.26	10.7 \pm 2.14
4hr	5.58 \pm 1.45	5.71 \pm 0.27	6.16 \pm 1.47	6.16 \pm 0.04	7.19 \pm 0.01	7.40 \pm 0.19
8hr	4.41 \pm 0.02	6.27 \pm 0.87	7.00 \pm 1.40	8.80 \pm 4.01	9.45 \pm 4.32	9.97 \pm 5.22
24hr	4.26 \pm 0.06	4.63 \pm 0.24	5.56 \pm 0.03	5.07 \pm 0.30	4.58 \pm 0.14	4.93 \pm 0.65
48hr	4.60 \pm 1.35	5.92 \pm 3.10	5.98 \pm 3.02	6.21 \pm 3.69	6.95 \pm 4.71	6.48 \pm 3.89
96hr	4.17 \pm 0.36	4.04 \pm 0.06	4.57 \pm 0.21	4.05 \pm 0.18	4.03 \pm 0.20	3.71 \pm 0.29
168hr	5.76 \pm 0.91	5.33 \pm 0.83	5.12 \pm 1.00	5.36 \pm 0.89	5.89 \pm 1.01	5.86 \pm 0.95
<u>Semi-dry Clay</u>						
BCA	5.90 \pm 2.27	4.50 \pm 0.76	4.52 \pm 0.65	4.66 \pm 0.84	5.55 \pm 1.39	5.22 \pm 0.61
DCA	72.1 \pm 49.5	52.3 \pm 32.9	50.7 \pm 19.5	81.3 \pm 56.7	75.2 \pm 53.1	73.7 \pm 48.6
1hr	5.94 \pm 1.21	5.68 \pm 0.80	5.35 \pm 0.80	7.39 \pm 1.78	10.0 \pm 6.39	6.56 \pm 1.10
2hr	7.53 \pm 1.62	5.32 \pm 0.47	6.11 \pm 0.31	8.93 \pm 2.44	7.62 \pm 1.59	9.35 \pm 2.78
4hr	4.87 \pm 1.41	4.43 \pm 0.88	4.66 \pm 1.71	4.69 \pm 1.48	5.22 \pm 2.48	4.75 \pm 1.61
8hr	4.21 \pm 0.54	4.41 \pm 1.54	3.77 \pm 0.15	5.05 \pm 1.72	5.25 \pm 2.23	4.60 \pm 1.10
24hr	4.00 \pm 0.57	4.00 \pm 0.67	3.67 \pm 0.05	3.49 \pm 0.41	3.54 \pm 0.26	3.81 \pm 0.90
48hr	4.28 \pm 0.77	4.12 \pm 0.94	4.45 \pm 1.24	4.45 \pm 1.68	4.96 \pm 2.15	4.33 \pm 1.17
96hr	3.82 \pm 0.53	3.54 \pm 0.39	3.71 \pm 0.16	3.59 \pm 0.27	3.65 \pm 0.15	3.78 \pm 0.79
168hr	5.14 \pm 1.50	4.51 \pm 1.33	4.36 \pm 0.73	5.24 \pm 1.10	4.58 \pm 0.99	4.57 \pm 1.20
<u>Control</u>						
BCA	4.52 \pm 0.43	4.41 \pm 0.57	4.09 \pm 0.19	4.85 \pm 0.82	4.79 \pm 0.69	4.59 \pm 0.45
DCA	4.12 \pm 0.34	3.94 \pm 0.12	3.87 \pm 0.35	4.01 \pm 0.55	4.18 \pm 0.27	4.10 \pm 0.28
1hr	3.72 \pm 0.31	3.57 \pm 0.08	3.50 \pm 0.20	3.65 \pm 0.23	3.86 \pm 0.39	3.39 \pm 0.21
2hr	4.45 \pm 0.18	4.25 \pm 0.27	4.17 \pm 0.40	4.89 \pm 0.50	4.65 \pm 0.39	4.67 \pm 0.59
4hr	5.89 \pm 0.36	6.26 \pm 0.39	5.93 \pm 0.37	6.21 \pm 0.26	5.53 \pm 0.31	6.12 \pm 0.51
8hr	4.52 \pm 0.43	4.41 \pm 0.57	4.09 \pm 0.19	4.85 \pm 0.82	4.79 \pm 0.69	4.59 \pm 0.45
24hr	5.06 \pm 1.18	5.39 \pm 1.58	5.22 \pm 1.49	5.45 \pm 1.56	5.35 \pm 1.41	5.33 \pm 1.70
48hr	3.72 \pm 0.31	3.57 \pm 0.08	3.50 \pm 0.20	3.65 \pm 0.23	3.86 \pm 0.39	3.39 \pm 0.21
96hr	4.45 \pm 0.18	4.25 \pm 0.27	4.17 \pm 0.40	4.89 \pm 0.50	4.65 \pm 0.39	4.67 \pm 0.59
168hr	5.06 \pm 1.18	5.39 \pm 1.58	5.22 \pm 1.49	5.45 \pm 1.56	5.35 \pm 1.41	5.33 \pm 1.70

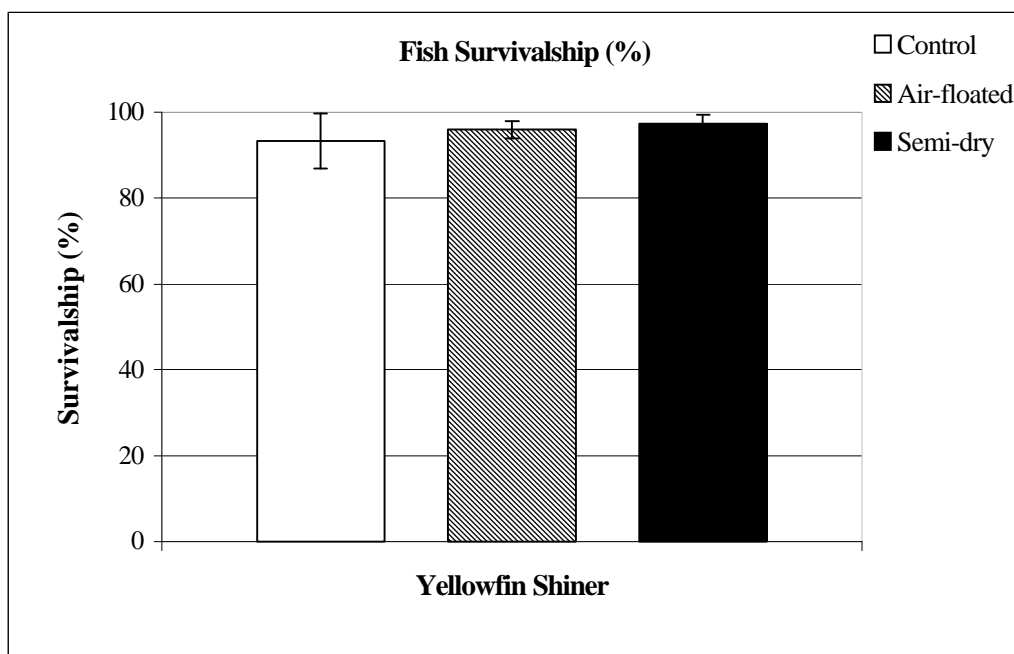


Figure 14. Mean (\pm Stdev.) survivorship of yellowfin shiners collected from raceways. No significant differences ($p > 0.05$) exist among treatments for the fish survivorship.

5.4. Survey of Potential SRS Locations for Future Field Deployment

Selection of the best field site for clay application was based on physical, chemical, and biological characteristics of the streams, as well as accessibility (**Table 9**). We evaluated these parameters and the results are shown on a scale of 1-4 (1: being the best and 4: being the worst). Physical parameters such as water velocity, depth, and width of streams were measured (**Table 10**). Bio-abundance was visually monitored and indicated how many fish were in streams with the potential to evaluate the environmental impacts of the illite application.

The ^{137}Cs levels in sediment and water represented chemical characteristics (**Fig. 15**). Since our main objective was to reduce the biological availability of ^{137}Cs in sensitive wetland environments, detectable concentration of ^{137}Cs in streams was critical. Ranking of the streams by their ^{137}Cs concentrations in sediments and water were: Fourmile Branch > Lower Three Runs > Pen Branch • Steel Creek. There were no significant differences ($p > 0.05$) in ^{137}Cs concentrations in water between the middle and edge of the streams at any single sample location. Based on the results of the site evaluation, Fourmile Branch was suggested as a potential field site for illite remediation.

Table 9. Evaluation of potential sites for illite application. Results showed on a scale of 1-4 (1: being the best and 4: being the worst).

Parameter	Fourmile Branch	Lower Three Runs	Pen Branch	Steel Creek
Accessibility ^a	1	3	1	4
Bio-abundance ^b	2	3	1	3
¹³⁷ Cs concentration ^c	1	2	3	3
Velocity/Width/Depth	1	3	2	3
Total score	5	11	7	13
Site Selection	1	3	2	4

^abased on relative ease to access the site to conduct a field demonstration

^bbased on number of fish

^cbased on ¹³⁷Cs concentrations in water and sediments of the streams

Table 10. Velocity, depth, and width (mean \pm Stdev.) at 3 locations on each stream were measured for comparison.

Potential Site	Velocity (m sec ⁻¹)	Depth (cm)	Width (m)
Fourmile Branch	0.71 \pm 0.13	22.0 \pm 4.00	3.10 \pm 0.75
Lower Three Runs	1.00 \pm 0.37	22.8 \pm 5.23	8.87 \pm 0.40
Pen Branch	0.32 \pm 0.12	27.4 \pm 3.21	23.0 \pm 3.21
Steel Creek	0.74 \pm 0.22	25.6 \pm 8.55	5.47 \pm 1.45

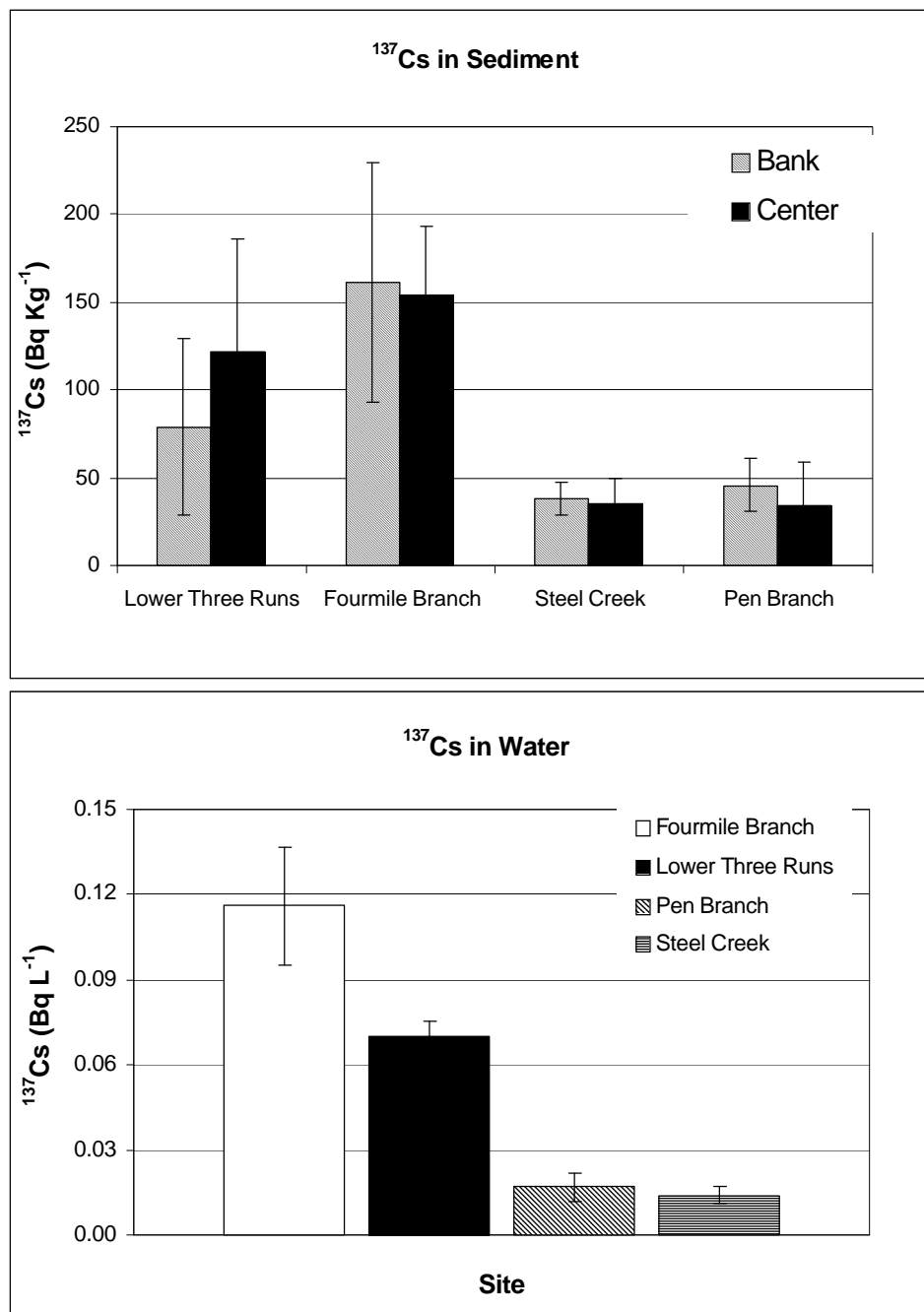


Figure 15. Concentration of ^{137}Cs (mean \pm Stdev.) in sediments and water in streams. A ranking of the ^{137}Cs concentrations in sediments and water in streams as chemical characteristics would be Fourmile Branch > Lower Three Runs > Pen Branch • Steel Creek.

6.0. CONCLUSIONS AND IMPLICATIONS

During this funding period (FY03) we re-sampled study sites in Pond A and R-Canal that had been treated with illite clays in 2001. The data revealed that ^{137}Cs concentrations in water are still lower than untreated control sites, some 112 weeks later. This encouraging result indicates that the *in situ* remediation technique has a longer-term effectiveness than what was previously reported. This positive finding occurred despite the study site experiencing a severe drought, as well as flooded conditions during the two-year period.

Analyses of particle settling and flocculation rates among three commercially available illites (trade name Todd Light) that differed in particle size and moisture content revealed the flexibility of the clay application method and the versatility of the illites to different remediation needs. Air-floated Todd Light Illites were found to disperse quickly throughout the water column and would be ideal for remediating lakes, or possibly long reaches of a contaminated stream. If, however, small localized contaminated hot spots within streams are the target of remediation, then the crude Todd Light Illite's more rapid settling characteristics would allow one to place the mineral specifically where needed, rather than broadly dispersing it within the stream. More work on field deployment methods for various contaminant dispersal scenarios would be useful.

Applying the illite clays caused water turbidity to increase by two to three orders of magnitude. Although water clarity improved at an exponential rate, with a half-time of 1.2 days, some 100 days were required for the high clay treatments to reach the turbidity levels of control sites. The environmental impact of the turbid water was assessed by examining other water quality measurements, phytoplankton productivity, chlorophyll-a production, and the abundance of macro-invertebrates. No significant differences in pH, electrical conductivity, dissolved oxygen or oxidation-reduction potential were found among treatments. Electrical conductivity values would be expected to decrease if the illites were absorbing ionic solutes from the water column, such as K or NH_4 . Thus, the clay application is not likely to decrease the availability of essential plant nutrients in treated waters. No significant differences were observed in phytoplankton productivity, chlorophyll-a, or respiration levels.

No differences in total numbers of macroinvertebrate insects were observed between controls and illite-treated limnocorrals. However, differences in species assemblages of macroinvertebrate insects were apparent between controls and high clay application treatments, two years after the initial clay application. No difference was found between the low clay applications and controls. The divergence in species assemblages among treatments significantly increased after the second clay application. These results are very encouraging in that they suggest that although impacts

did occur to the macroinvertebrate community, recovery was occurring and that the impacts are not permanent.

When the illite was applied to flowing streams (raceways at the Aquatic Ecology Laboratory), water clarity returned within a matter of hours and fish mortality was not affected. This is an encouraging first approximation at gross effects that might occur in a stream application of the technique. The experiments were acute exposures to turbidity and conducted in artificial raceways, but we know little about the long-term effects to a stream ecosystem.

It is obvious that turbidity from the illite clay applications does much less damage to sensitive wetlands than other ^{137}Cs remediation strategies in which the wetland is totally drained and the contaminated sediments are dredged up and trucked to another locality. In situations where it is deemed absolutely necessary to remediate a ^{137}Cs -contaminated wetland, the use of an *in situ* illite application would preserve more of the ecosystem function than more destructive remediation methods. Additional measurements are needed to determine the long-term effectiveness of the technique and environmental impacts on parameters not measured in this study. When coupled to the earlier work of Hinton et al. (1999, 2001, 2002), the results reported herein support the use of illite applications to ^{137}Cs -contaminated wetlands on the SRS as an alternative to more environmentally destructive remediation methods.

7.0. ACKNOWLEDGEMENTS

Authors are appreciative of the technical assistance provided by Jamie Marsh (SREL) and Yi Yi (SREL). This research was supported by WSRS-SGCP and by the Environmental Remediation Sciences Division of the Office of Biological and Environmental Research, U.S. Department of Energy, through Financial Assistance Award No. DE-FC09-96-SR18546 to the University of Georgia Research Foundation. Westinghouse Savannah River Company is operated for the U.S. Department of Energy under contract DE-AC09-96SR18500.

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9.0. APPENDIX A: TIMELINE OF ACTIVITY

Nov.25, 2002	The following final report was published on work done in FY02: Hinton, T.G., A. Knox, and D.I. Kaplan. 2002. Field Deployment of Illite Clay as an <i>In Situ</i> Method for Remediation ¹³⁷ Cs-contaminated Wetlands (U). Report No. WSRC-TR-2002-00516 and SREL-69 (UC-66E). 56 p.
Jan. 2003	Water samples were taken from each of the 16 limnocorrals in R-Canal and Pond A and assayed for ¹³⁷ Cs and a suite of stable elements. These data are valuable for longer-term effectiveness of the remediation method.
Feb. 3, 2003	Bonjun Koo, a postdoctoral researcher started to help on the project. He completed his radiation worker training (RW-II)
Jan.-Feb, 2003	We spent considerable time looking at WSRC databases for IOUs such as Fourmile Branch, Steel Creek, Upper Three Runs, Pen Branch, and Lower Three Runs.
Feb. 12-29, 2003	Sediment and water samples were taken from each IOUs for baseline data. They were counted for ¹³⁷ Cs. Took water quality measurements on each IOUs (table below). Physical parameters such as depth, width, and velocity were recorded for selecting the best field site for clay application.

Parameter	Fourmile Branch	Lower Three Runs	Pen Branch	Steel Creek
DO (mg L ⁻¹)	9.25	9.01	9.67	9.80
EC (• S cm ⁻¹)	56	74	79	66
ORP (mV)	316	256	279	304
pH	6.62	7.47	8.30	7.05
Temp. (°C)	14.8	13.1	7.85	13.7
TDS (g L ⁻¹)	0.00	0.10	0.11	0.00

Feb. 2003	90.7 kg of each three different Illite clays (air-floated, semi-dry, and crude) from Kentucky-Tennessee Clay Company were ordered for laboratory and AEL experiments.
March 5, 2003	Hester-Dendy samplers were deployed in the 16 limnocorrals in R-Canal and Pond A for pre-testing. Two H-D samplers in each limnocorrals (4 control; no clay application, 5 LCA, and 6 HCA) were introduced except for limnocorral #13 (LCA) that had a huge hole.
March-April, 2003	Lab experiments to determine which type of illite clays will be the best for field application. Settling velocities, removal of ¹³⁷ Cs in static water

and dynamic water systems of Todd Light Illites were examined.

Physicochemical properties of the illites were also determined.

March 10,2003 StowAway TidbiT Temperature Loggers were deployed in limnocorrals #1 (control), 12 (LCA), 14 (HCA), and R-canal.

March 25, 2003 Pre-turbidity was checked for 24 hours in 3 raceways at AEL (data in table below). The pipes of raceway #2 and 3 were plugged (indicated as N/A) and the data was not available.

Raceway	Date	Time	Turbidity (NTU)		
			Rep. 1	Rep. 2	Rep. 3
#1	3/24/03	8:30am	26.6	25.1	26.7
#2	3/24/03	8:30am	34.6	31.3	33.8
#3	3/24/03	8:30am	12.2	12.1	12.2
Blank (DI water)	3/24/03	8:30am	0.00	0.00	0.00
#1	3/25/03	8:30am	7.70	7.14	7.30
#2	3/25/03	8:30am	N/A	N/A	N/A
#3	3/25/03	8:30am	N/A	N/A	N/A
Blank (DI water)	3/25/03	8:30am	0.00	0.00	0.00

April 2, 2003 Hester-Dendy samplers were collected for pre-testing. Insects were isolated, sorted, and identified for counting. Results indicated that there was no significance for number of insects at $p > 0.05$ among treatments (table below).

Treatment	Sample No.	Number of Individual Invertebrates
Control	LC1	12
	LC6	27
	LC9	37
	LC10	23
	Treatment Avg \pm Stdev	24.8 \pm 10.3
LCA	LC4	7
	LC8	0
	LC12	22
	LC13	56
	LC15	14
	LC16	17
	Treatment Avg \pm Stdev	19.3 \pm 19.6
HCA	LC2	3
	LC3	20
	LC5	14
	LC7	14
	LC11	17
	LC14	12
	Treatment Avg \pm Stdev	13.3 \pm 5.79

April-May, 2003 Prepared AEL experiments. We cleaned the raceways and modified the filtering system. Continued analyses for sediment and water samples produced by laboratory experiments.

April 3, 2003 Water samples were also taken from 16 limnocorrals in R-canal for chlorophyll-a analysis. No significant differences existed at $p > 0.05$ (table below).

Treatment	Sample No.	Chlorophyll-a ($\mu\text{g L}^{-1}$)
Control	LC1	4.81
	LC9	8.02
	LC10	1.60
	Treatment Avg \pm Stdev	4.81 ± 3.21
LCA	LC4	0.89
	LC12	2.67
	LC13	3.56
	LC15	0.00
	LC16	1.60
	Treatment Avg \pm Stdev	1.75 ± 1.41
HCA	LC2	0.00
	LC3	2.14
	LC5	5.35
	LC7	2.14
	LC11	2.14
	LC14	2.67
	Treatment Avg \pm Stdev	2.41 ± 1.72

May 29-June 5, 2003 Raceways were examined for water velocity, depth and clay movement at AEL.

June 6, 2003 We fixed limnocorrals in R-canal for clay application that examined the environmental impacts.

June 9, 2003 Ordered 1.95 Mg (tonnes; 4,300 lbs) of air-floated illites for applying clay into the limnocorrals. 30 H-D multiplate samplers were placed in twelve of the 16 limnocorrals located in R-Canal and Pond A.

June 10-15, 2003 Collected 5 fish species such as mosquito fish (*Gambusia affinis*), dusky shiner (*Notropis cummingsae*), yellowfin shiner (*Notropis lutipinnis*), tessellated darter (*Etheostoma olmstedi*), and blackbanded darter (*Percina nigrofasciata*) in the Fourmile Branch and introduced them into a raceway for selecting two species for clay experiments. Based on the result of survival (table below) and literature review, we selected tessellated darters for a bottom-feeder and yellowfin shiners for a top-feeder as study organisms.

Fish	No. of Fish Introduced	No. of Fish Survived
Mosquito	12	9
Dusky shiner	12	11
Yellowfin shiner	12	12
Tessellated darter	12	12
Blackbanded darter	7	7

June 16, 2003	Collected 50 tessellated darters and 50 yellowfin shiner in Fourmile Branch. Started AEL experiments to evaluate the effect of clay application on fish mortality and sedimentation in continuous flowing raceways and static fish tanks. Twelve yellowfin shiners and 9 tessellated darters were introduced into a raceway. Those were collected on 16 July, 2003 (1 month after) for fish mortality. We also put 12 yellowfin shiner and 2 tessellated darters into 125.4 L static fish tanks (56.5 cm X 48 cm) and fish were collected on 16 September, 2003 (3 months after) for fish mortality. Results indicated that there are no significant differences between air-floated clay treatment and control ($p > 0.05$). During the experiment, turbidity was also checked.
June 20, 2003	0.76 kg of air-floated illite was applied into static fish tanks and measured turbidity and fish mortality. The applied amount of illite was equivalent amount, which we applied into limnocorrals can cover the sediment with 0.25 cm illite.
June 27, 2003	1.95 Mg (4,300 lbs) of air-floated air-floated were arrived.
June 30, 2003	14 pre-treated H-D samplers were collected.
July-Sep., 2003	Performed AEL experiments and limnocorral experiments in R-Canal and Pond A for evaluating environmental impacts of clay application.
July 1, 2003	Turbidity and water quality measurement were taken from limnocorrals for BCA baseline data. 3-liter water samples were also taken.
July 3-4, 2003	Turbidity samples were taken.
July 3, 2003	Clay amendments were added to the limnocorrals in R-Canal and Pond A. The limnocorrals were randomly chosen for low clay application, high clay application, and controls (no clay application). 27.22 kg of clay was added to the low application limnocorrals. 272.2 kg of clay was added to the high application limnocorrals.
July 10, 2003	15 H-D samplers were collected one week after clay application.

July 7, 10, 15, 17, and 21, 2003	Water quality was done on each limnocorral with the hydrolab. Also, turbidity samples were collected.
July 7, 15, 21, 2003	Phytoplankton productivity was measured using the dissolved oxygen method.
July 15, 21, 2003	Water samples were taken from 14 limnocorrals for chlorophyll-a analyses. Samples were taken about 5 cm below the water surface.
July 22-23, 2003	1 st replicate for fish mortality and sedimentation experiment: Using backpack electrofishers and a 3-mm mesh seine, fish were collected.
July 24-Aug. 4, 2003	50 yellowfin shiners and 20 tessellated darters introduced into raceways for clay experiment.
July 28, 2003	9.78 kg of air-floated and semi-dry illites was applied into raceways and measured turbidity on 28, 29, 30 July, and 1, 4 August, 2003. The applied amount of illite was equivalent amount, can cover the sediment with 0.25 cm illite.
Aug. 4, 2003	1 st fish collection from raceways for fish mortality and sedimentation was also measured for clay movement.
Aug. 11, 2003	38 days after the clay application, 15 H-D samplers were deployed to evaluate longer-term effect of clay application on macroinvertebrates. Final turbidity samples were taken from all 16 limnocorrals.
Aug. 14-15, 2003	2 nd replicate for fish mortality and sedimentation experiment.
Aug. 19, 2003	Air-floated and semi-dry illites was applied into raceways and measured turbidity on 19, 20, 21, 23, and 26 August, 2003.
Aug. 26, 2003	2 nd fish collection from raceways.
Sep. 3-5, 2003	3 rd replicate for fish mortality and sedimentation experiment.
Sep. 9, 2003	Illite were introduced and measured turbidity on 9, 10, 11, 13, and 16 September, 2003.
Sep. 16, 2003	3 rd fish collection from raceways and
Sep. 22, 2003	15 H-D samplers were collected from the limnocorrals.