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HYDRIDE BED ISOTOPIC EXCHANGE

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Summary

A hydride bed, was isotopically exchanged to recover tritium. Development and implementation of a new heating method led to recovery of significant additional tritium.

- Heating of the bed promotes increased exchange between deuterium and tritium.
- Cycling gas between loss-of-cooling tanks and the bed by varying temperature results in the equivalent of multiple traditional exchange cycles
- Maintaining the temperature above 80°C during cycling promotes deuterium-tritium exchange.
- **Additional 20% - 25% tritium heel recovered using technique described**

Introduction

LaNi_{4.25}Al_{0.75} can be used as a hydride storage material. Over time, as tritium stored on the beds decays, its byproduct, helium-3, accumulates. In addition, helium-3 in-growth traps a portion of the hydrogen (the heel) [1,2,3].

Prior to replacement, beds can be isotopically exchanged to recover tritium. Isotopic exchange involves repeatedly absorbing protium or deuterium into the LaNi_{4.25}Al_{0.75} and desorbing hydrogen isotopes from the bed [4].

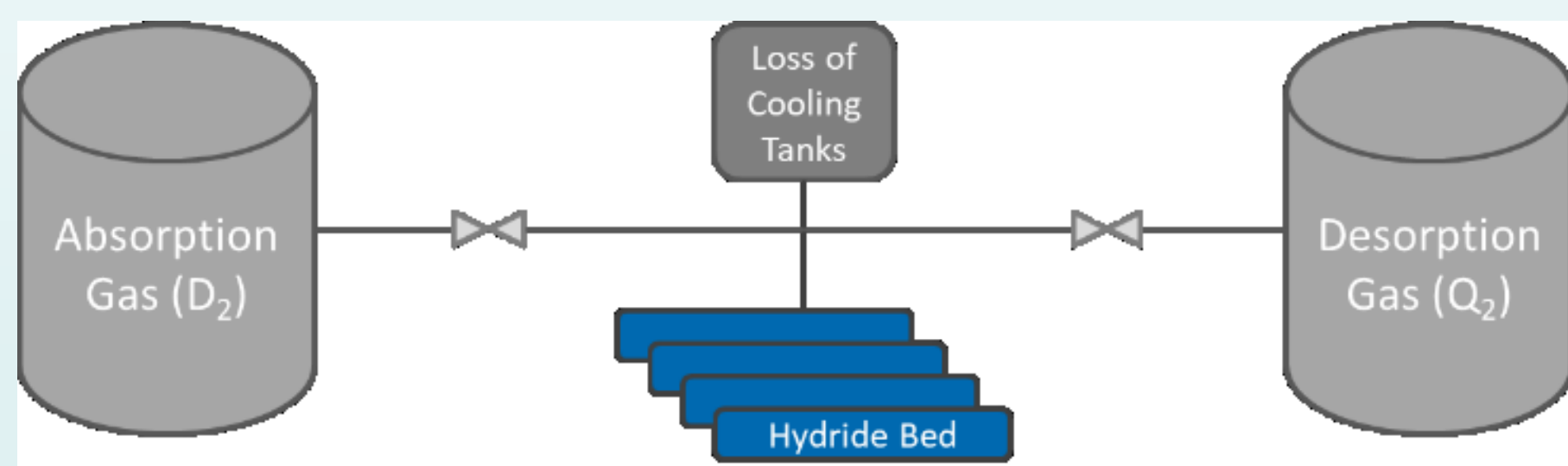


Figure 1. Depiction of isotopic exchange major components

Methodology

Twelve exchange cycles were performed.

Cycle	Type	Status
1-6	Traditional	Significant tritium heel remains, additional traditional exchanges unlikely to remove.
7-8	Heated	Tritium removal increases relative to previous traditional cycles.
9-12	Cycled	Tritium removal efficiency further increases.

Heated Exchanges - bed heated within a confined volume following absorption to encourage exchange between deuterium and tritium.

Cycled Exchanges - bed cycled between 80°C and 120°C 10 times to drive gas between bed and integral loss-of-cooling tanks. Takes advantage of the preferential absorption of lighter isotopes at temperatures above 80°C [5].

Results

Following traditional, heated, and cycling exchanges at elevated temperatures, at Cycle 12, heel reduced to approximately 1% of original. Tritium remaining on the bed during the final 4 cycles appears to follow a theoretical reduction assuming perfect mixing of the absorption gas with the heel gas.

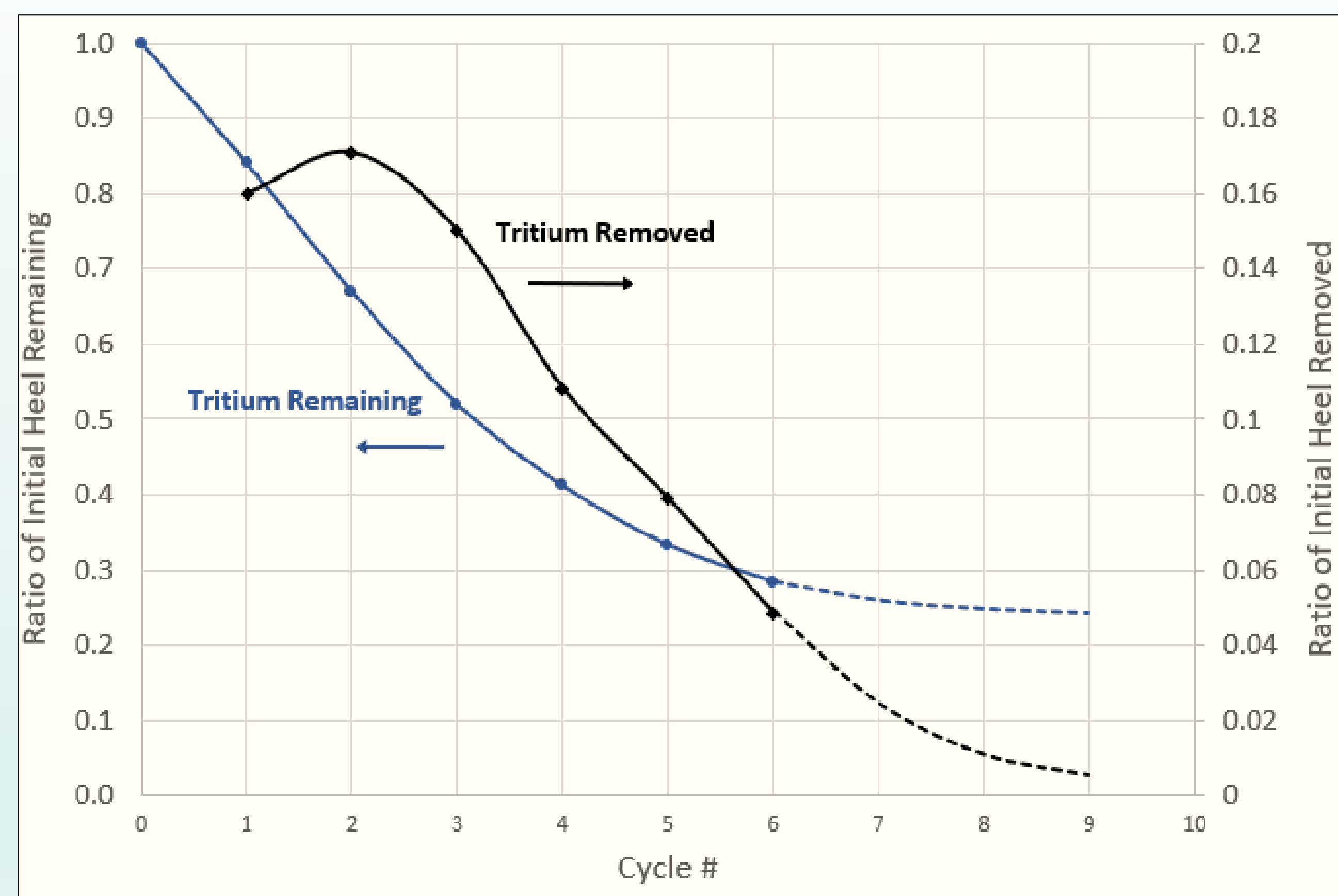


Figure 2. Ratio of Tritium remaining, tritium removed and projected path as a function of traditional cycles (1-6)

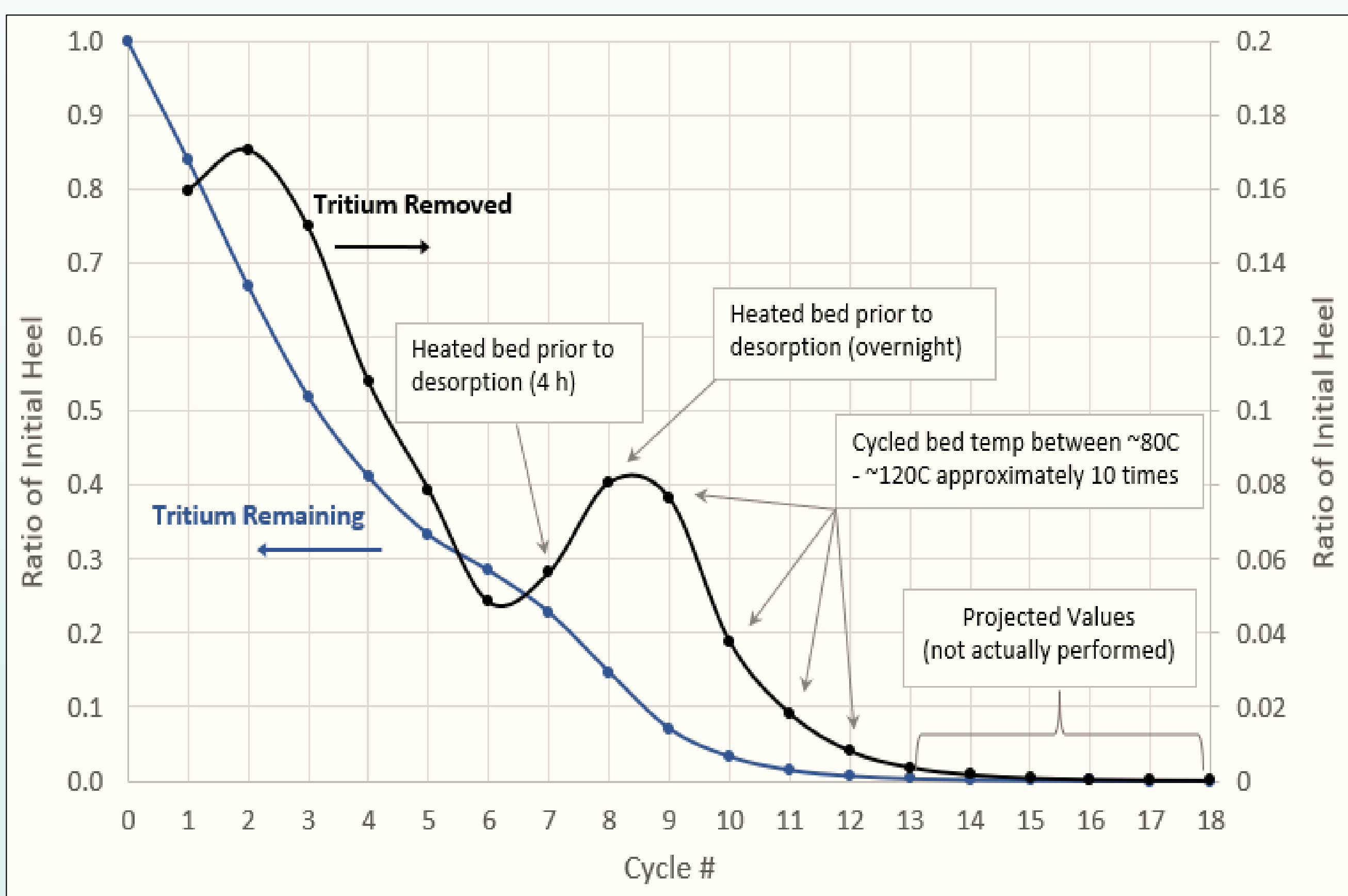


Figure 3. Ratio of Tritium remaining and tritium removed as a function of traditional cycling (1-6) – see Fig 1. – heated cycles (7-8), and cycles performed between 80°C and 120°C (9-12)

Conclusions

- Heating hydride bed promotes increased exchange between deuterium and tritium.
- Cycling gas between loss-of-cooling tank and the bed by varying temperature results in the equivalent of multiple traditional exchange cycles
- Maintaining temperature above 80°C during cycling promotes isotopic exchange.
- **Additional 20% - 25% of a tritium heel recovered that otherwise would have been wasted**

References

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