

**Contract No:**

This document was prepared in conjunction with work accomplished under Contract No. 89303321CEM000080 with the U.S. Department of Energy (DOE) Office of Environmental Management (EM).

**Disclaimer:**

This work was prepared under an agreement with and funded by the U.S. Government. Neither the U.S. Government or its employees, nor any of its contractors, subcontractors or their employees, makes any express or implied:

- 1 ) warranty or assumes any legal liability for the accuracy, completeness, or for the use or results of such use of any information, product, or process disclosed; or
- 2 ) representation that such use or results of such use would not infringe privately owned rights; or
- 3) endorsement or recommendation of any specifically identified commercial product, process, or service.

Any views and opinions of authors expressed in this work do not necessarily state or reflect those of the United States Government, or its contractors, or subcontractors.



**Savannah River  
National Laboratory®**

A U.S. DEPARTMENT OF ENERGY NATIONAL LAB • SAVANNAH RIVER SITE • AIKEN, SC • USA

# **Analysis of Tank 38H (HTF-38-22-31, -32) and Tank 43H (HTF-43-22-33, -34) Samples for Support of the Enrichment Control and Corrosion Control Programs**

**M. S. Hay  
C. J. Coleman  
D. P Diprete**

May 2022  
SRNL-STI-2022-00214, Rev. 0

**SRNL.DOE.GOV**

## **DISCLAIMER**

This work was prepared under an agreement with and funded by the U.S. Government. Neither the U.S. Government or its employees, nor any of its contractors, subcontractors or their employees, makes any express or implied:

1. warranty or assumes any legal liability for the accuracy, completeness, or for the use or results of such use of any information, product, or process disclosed; or
2. representation that such use or results of such use would not infringe privately owned rights; or
3. endorsement or recommendation of any specifically identified commercial product, process, or service.

Any views and opinions of authors expressed in this work do not necessarily state or reflect those of the United States Government, or its contractors, or subcontractors.

**Printed in the United States of America**

**Prepared for  
U.S. Department of Energy**

**Keywords:** *2H Evaporator System*  
*Supernate Analysis, Radionuclides*

**Retention:** *Permanent*

## **Analysis of Tank 38H (HTF-38-22-31, -32) and Tank 43H (HTF-43-22-33, -34) Samples for Support of the Enrichment Control and Corrosion Control Programs**

M. S. Hay  
C. J. Coleman  
D. P. Diprete

May 2022

---

Savannah River National Laboratory is operated by  
Battelle Savannah River Alliance for the U.S. Department  
of Energy under Contract No. 89303321CEM000080.



## REVIEWS AND APPROVALS

### AUTHORS:

---

M. S. Hay, Chemical Flowsheet Development	Date
---	------

---

C. J. Coleman, Analytical Characterization & Sample Management	Date
--	------

---

D. P. Diprete, Environmental and Legacy Management	Date
--	------

### TECHNICAL REVIEW:

---

W. D. King, Chemical Flowsheet Development	Date
--	------

### APPROVAL:

---

G. A. Morgan, Manager Chemical Flowsheet Development	Date
---	------

---

F. M. Pennebaker, Director Chemical Processing & Characterization	Date
--	------

---

A. T. Hooker, Safety Programs & Regulatory Manager SRR, Tank Farm Facility Engineering	Date
---	------

## EXECUTIVE SUMMARY

SRNL analyzed samples from Tank 38H and Tank 43H to support ECP and CCP. The results indicate the concentrations of most soluble species in the Tank 38H surface sample increased from the previous surface sample. The Tank 38H sub-surface sample shows changes in concentration for soluble species in the solution with some increasing and some decreasing. The current Tank 38H sub-surface sample contains less sludge solids than the previous sample based on visual appearance. The small differences in the concentrations of major components between the Tank 38H surface and sub-surface samples indicate only minimal stratification of solution species within the tank.

The Tank 43H surface and sub-surface samples are similar in composition to the previous samples. The similar solution compositions measured in the Tank 43H surface and sub-surface samples indicate a minimal stratification within the tank.

The total uranium and plutonium in the Tank 38H surface sample increased in the current sample compared to the previous analysis. The Tank 38H sub-surface sample shows a decrease in uranium and plutonium concentrations compared to the previous sample likely because of a decrease in sludge solids in the current sample. The total uranium concentration in the two Tank 43H samples both increased slightly from previous sample results. The plutonium concentration in both the Tank 43H surface and sub-surface samples decreased from previous sample results.

The sum of the major cations versus the sum of the major anions shows a difference of <5% for both samples from Tank 38H and for both samples from Tank 43H providing an indication of good data quality for the non-radioactive analytes in the samples.

The silicon concentrations measured in the Tank 38H sub-surface sample decreased compared with the previous sample results likely due to the presence of less sludge solids in the current sample. The Tank 38H surface and Tank 43H surface and sub-surface samples silicon concentrations are similar to the previous sample results. The samples analyzed from Tanks 38H and 43H show silicon concentrations ranging from 58.4 to 71.2 mg/L.

## TABLE OF CONTENTS

LIST OF TABLES .....	vii
LIST OF FIGURES .....	vii
LIST OF ABBREVIATIONS .....	viii
1.0 Introduction .....	1
2.0 Experimental Procedure .....	1
3.0 Results and Discussion .....	3
4.0 Conclusions .....	6
5.0 Acknowledgements .....	6
6.0 References .....	7

## LIST OF TABLES

Table 2-1. Sampling Height and Sample Mass of the Tank 38H and 43H Samples.....	2
Table 3-1. ECP, CCP, and other Analytical Data for Tank 38H and Tank 43H Samples. (Averages and %RSD values are of triplicate measurements).....	5

## LIST OF FIGURES

Figure 2-1. Samples from Tank 38H and 43H .....	2
---	---



## LIST OF ABBREVIATIONS

DI	de-ionized
CCP	Corrosion Control Program
ECP	Enrichment Control Program
IC	ion chromatography
ICP-ES	inductively coupled plasma emission spectroscopy
%RSD	percent relative standard deviation
SaM	Sensing & Metrology
S/ICPMS	separation/ inductively coupled plasma mass spectrometry
SRNL	Savannah River National Laboratory
SRMC	Savannah River Mission Completion
TIC	total inorganic carbon
TTQAP	Task Technical and Quality Assurance Plan
TTR	Technical Task Request

## 1.0 Introduction

Feed limits have been established for the 2H-Evaporator system to ensure nuclear criticality is not possible and corrosion is minimized.<sup>1</sup> These limits are protected by the Enrichment Control Program (ECP) and the Corrosion Control Program (CCP) that require periodic sampling and analysis to confirm that the waste supernate composition stays within the limits.<sup>2,3</sup>

Savannah River Mission Completion (SRMC) obtained samples from two different heights within each of the two waste tanks supporting the 2H-Evaporator operations on March 15, 2022. The Tank 38H (evaporator drop tank) and Tank 43H (evaporator feed tank) samples were received by the Savannah River National Laboratory (SRNL) Shielded Cells on March 15, 2022. Analysis of these samples provides information necessary for determining compliance with the ECP and CCP. The sample characterization was requested via a Technical Task Request<sup>4</sup> (TTR) and conducted based on a Task Technical and Quality Assurance Plan (TTQAP).<sup>5</sup>

## 2.0 Experimental Procedure

The samples from Tank 38H and 43H were opened in the SRNL Shielded Cells and poured into clear plastic beakers. The beakers were photographed and the masses of the samples determined. Table 2-1 provides the sampling height and mass of each sample. Figure 2-1 shows a photograph of the samples in the clear beakers. Both the surface samples from Tank 38H and Tank 43H were mostly clear and showed no visible undissolved solids when poured into the plastic beakers. The sub-surface samples for Tank 38H and Tank 43H were slightly cloudy. Based on experience with past samples, the solids in the Tank 38H and Tank 43H subsurface samples represent less than 1 wt% insoluble solids.

All four samples received the analyses required by the ECP that includes determination of uranium isotopes by uranium separation/inductively coupled plasma-mass spectrometry (S/ICPMS) and determination of plutonium isotopes by radiochemical separation and counting methods. Additionally, the samples were also submitted for gamma spectroscopy to determine the Cs-137 concentration and inductively coupled plasma-emission spectroscopy (ICP-ES) to determine Na, Al, Si, and other metals. All four samples from both Tank 38H and Tank 43H also received the analyses required by the CCP. The CCP analysis suite includes determination of free hydroxide, and ion chromatography (IC). The total inorganic carbon (TIC) was also determined on the samples to provide a concentration for the carbonate anion present in the samples.

Density measurements were made on well-mixed (unfiltered) aliquots of the samples using calibrated volumetric tubes at ambient cell temperature (27 °C).

For the samples receiving the CCP analysis suite, de-ionized (DI) water dilutions were made in triplicate from a well-mixed (unfiltered) sample and submitted to SRNL Sensing & Metrology (SaM) for analysis. A blank of the DI water was also prepared along with the samples. The water dilutions were analyzed by ion chromatography, total inorganic carbon, and free hydroxide methods.

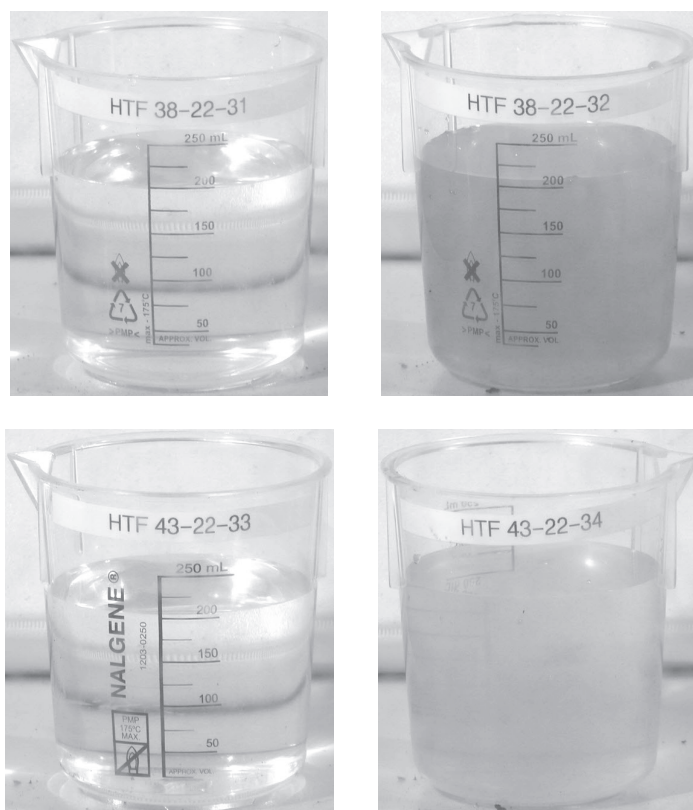
Triplicate aliquots of the well-mixed (unfiltered) sample from each sample receiving the ECP analysis suite were prepared for analysis using the warm acid strike method.<sup>6</sup> A reagent blank and three silicon standard solutions were submitted for analysis with the samples. The samples prepared by warm acid strike were submitted to SaM for analysis by ICP-ES, uranium isotopics, plutonium isotopics, and gamma spectroscopy.

### Quality Assurance

Requirements for performing reviews of technical reports and the extent of review are established in Manual E7, Procedure 2.60.<sup>7</sup> SRNL documents the extent of the Design Check using the SRNL Technical Report Design Checklist.<sup>8</sup> This review meets the acceptable criteria to comply with the TTR<sup>4</sup> requesting this work with a functional classification of Safety Class and per guidance in the TTQAP.<sup>5</sup> Data are recorded in the electronic laboratory notebook system as notebook/experiment number Y7081-00081-49.<sup>9</sup>

**Table 2-1. Sampling Height and Sample Mass of the Tank 38H and 43H Samples**

Sample ID	Sample Type	Sampling Height (inches from bottom)	Sample Mass (g)
HTF-38-22-31	Surface	surface	250.3
HTF-38-22-32	Sub-surface	224"	271.1
HTF-43-22-33	Surface	surface	254.4
HTF-43-22-34	Sub-surface	137"	262.5



**Figure 2-1. Samples from Tank 38H and 43H**

### 3.0 Results and Discussion

Table 3-1 contains the results from the analysis of the 2H-Evaporator samples. The table shows the average concentrations and the percent relative standard deviation (%RSD) for the triplicate sample preparations unless otherwise noted. Results preceded by “<” indicate the analyte was below the limits of quantification for all three replicate aliquots of the sample. In this case, the less than value in the table results from averaging the three less than values for the replicates. Results preceded by “≤” indicate that at least one of the replicates for the sample was above the limits of quantification while one or more of the replicates analyzed were below detection. The less than or equal to value in the table results from averaging all values for all three replicates. The %RSD presented in the table only includes the uncertainty associated with sub-sampling/sample preparation in the Shielded Cells and the analytical method. The estimated one sigma percent uncertainty provides an indication of the uncertainty associated with the analytical method as reported by SaM. Neither of these measures of uncertainty includes the uncertainty associated with sampling a large waste tank. Previous investigations indicate the uncertainty from taking a small sample from a large waste tank can be significant.<sup>10,11,12</sup>

The results in Table 3-1 indicate the concentrations of most soluble species in the Tank 38H surface sample increased from the previous surface sample with the sodium concentration rising from 4.8 M in the previous sample to 6.28 M in the current sample.<sup>13</sup> The Tank 38H sub-surface sample shows changes in concentration for soluble species in the solution with some increasing and some decreasing. The current Tank 38H sub-surface sample contains less sludge solids than the previous sample based on visual appearance. The sodium concentration in the Tank 38H sub-surface sample of 6.92 M is slightly lower the value of 7.18 M observed in the previous sample.<sup>13</sup> The total uranium and plutonium in the Tank 38H surface sample increased in the current sample compared to the previous analysis. The Tank 38H sub-surface sample shows a decrease in uranium and plutonium concentrations compared to the previous sample likely because of a decrease in sludge solids in the current sample. The weight fraction of U-235 to U-total in the Tank 38H surface sample at 0.86 is similar to the previous sample at 0.84. The U-235 to U-total weight fraction increased from 0.58 in the previous sub-surface sample to 0.81 in the current sub-surface sample.<sup>13</sup> The small differences in the sodium and other major component concentrations between the Tank 38H surface and sub-surface samples indicate only minimal stratification of solution species remains within the tank.

The Tank 43H surface sample soluble species composition is essentially unchanged from the previous surface sample as indicated by minimal change in the sodium concentration from 5.83 M to 5.96 M in the current sample.<sup>13</sup> The Tank 43H sub-surface sample is fairly similar to the previous sub-surface sample for most highly soluble species analyzed also. The sodium concentration in the Tank 43H sub-surface sample decreased with a concentration of 5.92 M in the current samples versus 6.28 M in the previous sample. The total uranium concentration in the two Tank 43H samples both increased from previous sample results. The plutonium concentration in both the Tank 43H surface and sub-surface samples decreased from previous sample results. The U-235 weight fraction is similar to the previous analysis in both the Tank 43H surface and sub-surface samples.<sup>13</sup> The composition of the Tank 43H surface and sub-surface samples are similar indicating minimal stratification within the tank.

The results for most species in the table, along with the density measurement results, generally show small %RSD values for the replicate analyses. The sum of the major cations versus the sum of the major anions shows a difference of <5% for the four Tank 38H and Tank 43H samples providing an indication of good data quality for the non-radioactive analytes in the samples. The Tank 43H sub-surface sample shows slightly higher %RSD's for the plutonium isotopes due to one replicate sample preparation being somewhat lower than the other two replicates.

The Cs-137 results for the both Tank 38H samples increased from the previous analysis while both the Tank 43H surface and sub-surface samples are similar to the previous samples.<sup>13</sup>

The silicon concentrations measured in the Tank 38H sub-surface sample decreased compared with the previous sample results likely due to the presence of less sludge solids in the current sample.<sup>13</sup> The Tank 38H surface and Tank 43H surface and sub-surface samples silicon concentrations are similar to the previous sample results.<sup>13</sup> The standards used for the silicon analysis (50 mg/L silicon in the solution prepared by warm acid strike diluted to final concentrations of 0.5, 1.0, and 2.0 mg/L) were close to the target concentrations with differences from the targeted concentrations of <5%. The silicon concentration was below detectable levels in the process blank. The samples analyzed show silicon concentrations ranging from 58.4 to 71.2 mg/L.

**Table 3-1. ECP, CCP, and other Analytical Data for Tank 38H and Tank 43H Samples.**  
(Averages and %RSD values are of triplicate measurements)

analyte	method	units	est. 1σ	HTF-38-22-31 average %RSD	HTF-38-22-32 average %RSD	HTF-43-22-33 average %RSD	HTF-43-22-34 average %RSD
Sample Type	--	--	--	Surface	Sub-surface	Surface	Sub-surface
density @ 27°C	grav.	g/mL	5%	1.26 0.4%	1.30 0.1%	1.26 0.7%	1.25 0.4%
U-233	S/ICPMS	mg/L	20%	4.75E-03 1.5%	3.96E-03 0.8%	4.34E-03 2.4%	4.29E-03 1.3%
U-234	S/ICPMS	mg/L	20%	9.69E-03 1.9%	1.09E-02 3.8%	8.86E-03 4.3%	8.62E-03 1.9%
U-235	S/ICPMS	mg/L	20%	3.13E-01 7.0%	2.46E-01 0.8%	2.71E-01 0.4%	2.70E-01 0.4%
U-236	S/ICPMS	mg/L	20%	2.13E-02 13%	1.70E-02 3.5%	1.76E-02 0.8%	1.70E-02 0.7%
U-238	S/ICPMS	mg/L	20%	3.60E+01 0.7%	3.00E+01 0.8%	3.34E+01 1.4%	3.34E+01 0.7%
Total U	calc.	mg/L	--	3.64E+01 0.8%	3.03E+01 0.8%	3.37E+01 1.4%	3.37E+01 0.7%
U-235 / U	calc.	%	--	0.86 6.4%	0.81 0.0%	0.80 1.1%	0.80 0.3%
Pu-238	PuTTA	mg/L	10%	4.08E-04	7.25E-04	4.09E-04	2.67E-04
		dpm/mL		1.55E+04 7.8%	2.76E+04 2.6%	1.55E+04 6.0%	1.02E+04 65%
Pu-239 <sup>a</sup>	PuTTA	mg/L	30%	5.28E-03	5.37E-03	6.23E-03	2.96E-03
Pu-239/240	PuTTA	dpm/mL		7.28E+02 7.1%	7.41E+02 16%	8.59E+02 29%	4.09E+02 102%
Pu-241	Pu238/41	mg/L	30%	<1.27E-05	<1.35E-05	<2.11E-05	<1.55E-05
		dpm/mL		<2.89E+03 --	<3.08E+03 --	<4.83E+03 --	<3.55E+03 --
Cs-137	gamma scan	dpm/mL	5%	3.24E+08	3.13E+08	2.84E+08	2.82E+08
Ba-137m				3.07E+08 1.3%	2.96E+08 3.3%	2.69E+08 2.3%	2.67E+08 1.8%
OH <sup>-</sup> <sub>free</sub>	titration	M	10%	1.94E+00 1.3%	2.21E+00 3.7%	1.86E+00 2.2%	1.83E+00 2.4%
F <sup>-</sup>	IC	M	10%	<1.06E-02 --	<9.98E-03 --	<1.06E-02 --	<1.06E-02 --
CHO <sub>2</sub> <sup>-</sup>	IC	M	10%	1.97E-02 1.4%	2.08E-02 0.4%	1.69E-02 0.8%	1.66E-02 2.2%
Cl <sup>-</sup>	IC	M	10%	<5.69E-03 --	6.53E-03 1.3%	5.88E-03 1.6%	5.79E-03 0.3%
NO <sub>2</sub> <sup>-</sup>	IC	M	10%	1.86E+00 0.7%	2.01E+00 1.1%	1.90E+00 0.8%	1.90E+00 1.0%
NO <sub>3</sub> <sup>-</sup>	IC	M	10%	8.78E-01 0.9%	1.09E+00 0.4%	9.41E-01 0.3%	9.24E-01 0.3%
PO <sub>4</sub> <sup>3-</sup>	IC	M	10%	3.33E-03 2.9%	1.22E-02 3.7%	3.96E-03 1.9%	3.94E-03 2.8%
SO <sub>4</sub> <sup>2-</sup>	IC	M	10%	5.05E-02 1.0%	6.18E-02 1.3%	4.69E-02 1.1%	4.53E-02 0.3%
C <sub>2</sub> O <sub>4</sub> <sup>2-</sup>	IC	M	10%	4.28E-03 2.4%	6.45E-03 1.9%	5.80E-03 2.1%	5.51E-03 0.5%
Br <sup>-</sup>	IC	M	10%	<1.26E-02 --	<1.19E-02 --	<1.26E-02 --	<1.26E-02 --
CO <sub>3</sub> <sup>2-</sup>	TIC	M	10%	5.84E-01 2.7%	6.38E-01 1.5%	5.48E-01 0.8%	5.39E-01 0.7%
Al	ICP-ES	mg/L	10%	1.65E+03 2.1%	2.22E+03 3.2%	1.57E+03 1.0%	1.63E+03 2.3%
B	ICP-ES	mg/L	10%	1.51E+02 2.1%	1.56E+02 2.9%	1.40E+02 1.1%	1.39E+02 2.6%
Ca	ICP-ES	mg/L	10%	3.22E+00 11%	2.06E+00 5.1%	2.48E+00 11%	2.88E+00 11%
Cr	ICP-ES	mg/L	10%	6.76E+01 2.2%	8.00E+01 3.1%	6.58E+01 0.7%	6.41E+01 2.5%
Fe	ICP-ES	mg/L	10%	2.70E+00 11%	3.20E+00 2.7%	2.51E+00 13%	2.83E+00 6.7%
K	ICP-ES	mg/L	10%	3.21E+02 0.8%	3.97E+02 3.9%	3.04E+02 3.3%	3.11E+02 4.1%
Li	ICP-ES	mg/L	10%	<9.92E+01 --	<9.99E+01 --	<1.02E+02 --	<1.02E+02 --
Na	ICP-ES	mg/L	10%	1.44E+05	1.59E+05	1.37E+05	1.36E+05
		M		6.28E+00 2.8%	6.92E+00 0.6%	5.96E+00 0.7%	5.92E+00 0.7%
P	ICP-ES	mg/L	10%	1.28E+02 2.5%	1.91E+02 3.4%	1.21E+02 1.0%	1.21E+02 2.5%
Si	ICP-ES	mg/L	10%	7.12E+01 3.3%	5.84E+01 4.9%	6.44E+01 1.6%	6.40E+01 3.6%
Zn	ICP-ES	mg/L	10%	2.91E+00 3.3%	3.48E+00 3.0%	2.66E+00 0.6%	2.78E+00 4.3%
TOC	TOC	mg C/L	10%	3.54E+02 2.8%	3.68E+02 2.4%	3.46E+02 1.1%	3.38E+02 0.9%
Anions	sum	M	--	6.05E+00 --	6.87E+00 --	5.99E+00 --	5.93E+00 --
Cations	sum	M	--	6.29E+00 --	6.93E+00 --	5.97E+00 --	5.92E+00 --

calc. = calculation; est. 1σ = estimated one sigma percent uncertainty as reported by AD.

<sup>a</sup> Pu-239 mass assumes entire Pu-239/240 activity is Pu-239



## 4.0 Conclusions

The results indicate the concentrations of most soluble species in the Tank 38H surface sample increased from the previous surface sample. The Tank 38H sub-surface sample shows changes in concentration for soluble species in the solution with some increasing and some decreasing. The current Tank 38H sub-surface sample contains less sludge solids than the previous sample based on visual appearance. The small differences in the concentrations of major components between the Tank 38H surface and sub-surface samples indicate only minimal stratification of solution species within the tank.

The Tank 43H surface and sub-surface samples are similar in composition to the previous samples. The similar solution compositions measured in the Tank 43H surface and sub-surface samples indicate a minimal stratification within the tank.

The total uranium and plutonium in the Tank 38H surface sample increased in the current sample compared to the previous analysis. The Tank 38H sub-surface sample shows a decrease in uranium and plutonium concentrations compared to the previous sample likely because of a decrease in sludge solids in the current sample. The total uranium concentration in the two Tank 43H samples both increased slightly from previous sample results. The plutonium concentration in both the Tank 43H surface and sub-surface samples decreased from previous sample results.

The sum of the major cations versus the sum of the major anions shows a difference of <5% for both samples from Tank 38H and for both samples from Tank 43H providing an indication of good data quality for the non-radioactive analytes in the samples.

The silicon concentrations measured in the Tank 38H sub-surface sample decreased compared with the previous sample results likely due to the presence of less sludge solids in the current sample. The Tank 38H surface and Tank 43H surface and sub-surface samples silicon concentrations are similar to the previous sample results. The samples analyzed from Tanks 38H and 43H show silicon concentrations ranging from 58.4 to 71.2 mg/L.

## 5.0 Acknowledgements

The contributions of Taylor Rush and Kevin Hauptfear, in preparing the samples, and those of Sonia Dyer, Matthew Nelson, Scott Brown, and Tom White, for providing analytical services, are appreciated and acknowledged.

## 6.0 References

1. H. C. Benhardt, *Nuclear Criticality Safety Evaluation: Operation of the 2H Evaporator System*, N-NCS-H-00180, Rev. 1, May 2018.
2. H. Bui, *CSTF Evaporator Feed Qualification Program*, WSRC-TR-2003-00055, Rev. 14, June 2021.
3. K. B. Martin., *CSTF Corrosion Control Program*, WSRC-TR-2002-00327, Rev. 9, December 2015.
4. J. R. Jacobs, *Enrichment Control Program Sample Analysis of Tanks 38 and 43*, X-TTR-H-00054, Rev. 0, November 2014.
5. C. J. Martino, *Task Technical and Quality Assurance Plan for Analysis of Tank 38H and Tank 43H Enrichment Control Program and Corrosion Control Samples*, SRNL-RP-2013-00522, Rev. 0, August 2013.
6. F. M. Pennebaker, C. J. Coleman, M. A. Jones, W. R. Wilmarth, C. M. Jantzen, D. R. Click, *Evaluation of Warm Acid Strike Treatment for Silicon Analysis in High Level Waste*, WSRC-TR-2003-00036, Rev. 0, January 2003.
7. Manual E7, Procedure 2.60, Technical Reviews, Rev. 18, December 2, 2019.
8. Savannah River National Laboratory, *Technical Report Design Check Guidelines*, WSRC-IM-2002-00011, Rev. 2, August, 2004.
9. M. S. Hay, *Tank 38/43 ECP-CCP Semi-Annual March 2022*, Y7081-00081-49, SRNL E-Notebook (Production), Savannah River National Laboratory, March 2022.
10. C. J. Coleman, T. B. Edwards, C. A. Nash, *Statistical Analysis of Sample Data from Tank 48H*, WSRC-TR-95-0325, Rev. 0, September 29, 1995.
11. D. D. Walker, W. T. Boyce, C. J. Coleman, D. P. Diprete, T. B. Edwards, A. A. Ekechukwu, C. W. Hsu, S. F. Peterson, L. L. Tovo, M. J. Whitaker, *Tank 48H Waste Composition and Results of Investigations of Analytical Methods*, WSRC-TR-97-00063, Rev. 0, April 2, 1997.
12. M. S. Hay, T. B. Edwards, *Statistical Analysis of ESP Verification Test Samples*, WSRC-RP-94-1224, Rev. 0, November 4, 1994.
13. M. S. Hay, C. J. Coleman, D. P. Diprete, *Analysis of Tank 38H (HTF-38-21-95, -96) and Tank 43H (HTF-43-21-97, -98) Samples for Support of the Enrichment Control and Corrosion Control Programs*, SRNL-STI-2021-00620, Rev. 0, November 2021.



**Distribution:**

[cj.bannochie@srnl.doe.gov](mailto:cj.bannochie@srnl.doe.gov)  
[William.bates@srnl.doe.gov](mailto:William.bates@srnl.doe.gov)  
[marion.cofer@srnl.doe.gov](mailto:marion.cofer@srnl.doe.gov)  
[alex.cozzi@srnl.doe.gov](mailto:alex.cozzi@srnl.doe.gov)  
[Marissa.reigel@srnl.doe.gov](mailto:Marissa.reigel@srnl.doe.gov)  
[Brady.lee@srnl.doe.gov](mailto:Brady.lee@srnl.doe.gov)  
[connie.herman@srnl.doe.gov](mailto:connie.herman@srnl.doe.gov)  
[Joseph.Manna@srnl.doe.gov](mailto:Joseph.Manna@srnl.doe.gov)  
[Gregg.Morgan@srnl.doe.gov](mailto:Gregg.Morgan@srnl.doe.gov)  
[frank.pennebaker@srnl.doe.gov](mailto:frank.pennebaker@srnl.doe.gov)  
[William.Ramsey@srnl.doe.gov](mailto:William.Ramsey@srnl.doe.gov)  
[William.swift@srnl.doe.gov](mailto:William.swift@srnl.doe.gov)  
[eric.skidmore@srnl.doe.gov](mailto:eric.skidmore@srnl.doe.gov)  
[michael.stone@srnl.doe.gov](mailto:michael.stone@srnl.doe.gov)  
[Boyd.Wiedenman@srnl.doe.gov](mailto:Boyd.Wiedenman@srnl.doe.gov)  
[chris.martino@srnl.doe.gov](mailto:chris.martino@srnl.doe.gov)  
[kim.crapse@srnl.doe.gov](mailto:kim.crapse@srnl.doe.gov)  
[david.diprete@srnl.doe.gov](mailto:david.diprete@srnl.doe.gov)  
[charles02.coleman@srnl.doe.gov](mailto:charles02.coleman@srnl.doe.gov)  
[lawrence.oji@srnl.doe.gov](mailto:lawrence.oji@srnl.doe.gov)  
[Azadeh.Samadi-Dezfouli@srs.gov](mailto:Azadeh.Samadi-Dezfouli@srs.gov)  
[keisha.martin@srs.gov](mailto:keisha.martin@srs.gov)  
[hilary.bui@srs.gov](mailto:hilary.bui@srs.gov)  
[Bailey.Scott@srs.gov](mailto:Bailey.Scott@srs.gov)  
[Andrea.Bridges@srs.gov](mailto:Andrea.Bridges@srs.gov)  
[celia.aponte@srs.gov](mailto:celia.aponte@srs.gov)  
[vijay.jain@srs.gov](mailto:vijay.jain@srs.gov)  
[david02.martin@srs.gov](mailto:david02.martin@srs.gov)  
[timothy.baughman@srs.gov](mailto:timothy.baughman@srs.gov)  
[john.jacobs@srs.gov](mailto:john.jacobs@srs.gov)  
[phillip.norris@srs.gov](mailto:phillip.norris@srs.gov)  
[john.occhipinti@srs.gov](mailto:john.occhipinti@srs.gov)  
[Richard.Edwards@srs.gov](mailto:Richard.Edwards@srs.gov)  
[Azikiwe.hooker@srs.gov](mailto:Azikiwe.hooker@srs.gov)  
[Ryan.McNew@srs.gov](mailto:Ryan.McNew@srs.gov)  
[Thomas.Huff@srs.gov](mailto:Thomas.Huff@srs.gov)  
[arthur.wiggins@srs.gov](mailto:arthur.wiggins@srs.gov)  
[bill.clark@srs.gov](mailto:bill.clark@srs.gov)  
[jeffrey.crenshaw@srs.gov](mailto:jeffrey.crenshaw@srs.gov)  
[james.folk@srs.gov](mailto:james.folk@srs.gov)  
[Curtis.Gardner@srs.gov](mailto:Curtis.Gardner@srs.gov)  
[Pauline.hang@srs.gov](mailto:Pauline.hang@srs.gov)  
[Anna.Murphy@srs.gov](mailto:Anna.Murphy@srs.gov)  
[tony.polk@srs.gov](mailto:tony.polk@srs.gov)

[Anthony.Robinson@srs.gov](mailto:Anthony.Robinson@srs.gov)  
[mark-a.smith@srs.gov](mailto:mark-a.smith@srs.gov)  
[patricia.suggs@srs.gov](mailto:patricia.suggs@srs.gov)  
[thomas.temple@srs.gov](mailto:thomas.temple@srs.gov)  
[kenneth.wells@srs.gov](mailto:kenneth.wells@srs.gov)  
[Heather.Capogreco@srnl.doe.gov](mailto:Heather.Capogreco@srnl.doe.gov)  
Records Administration (EDWS)