

Contract No:

This document was prepared in conjunction with work accomplished under Contract No. 89303321CEM000080 with the U.S. Department of Energy (DOE) Office of Environmental Management (EM).

Disclaimer:

This work was prepared under an agreement with and funded by the U.S. Government. Neither the U.S. Government or its employees, nor any of its contractors, subcontractors or their employees, makes any express or implied:

- 1) warranty or assumes any legal liability for the accuracy, completeness, or for the use or results of such use of any information, product, or process disclosed; or
- 2) representation that such use or results of such use would not infringe privately owned rights; or
- 3) endorsement or recommendation of any specifically identified commercial product, process, or service.

Any views and opinions of authors expressed in this work do not necessarily state or reflect those of the United States Government, or its contractors, or subcontractors.



Savannah River
National Laboratory®

A U.S. DEPARTMENT OF ENERGY NATIONAL LAB • SAVANNAH RIVER SITE • AIKEN, SC • USA

Characterization of the Sulfur-Saturated Melt Versions of the EMHQ LAW Glasses

M. C. Hsieh

February 2022

SRNL-STI-2022-00003, Revision 0

SRNL.DOE.GOV

DISCLAIMER

This work was prepared under an agreement with and funded by the U.S. Government. Neither the U.S. Government or its employees, nor any of its contractors, subcontractors or their employees, makes any express or implied:

1. warranty or assumes any legal liability for the accuracy, completeness, or for the use or results of such use of any information, product, or process disclosed; or
2. representation that such use or results of such use would not infringe privately owned rights; or
3. endorsement or recommendation of any specifically identified commercial product, process, or service.

Any views and opinions of authors expressed in this work do not necessarily state or reflect those of the United States Government, or its contractors, or subcontractors.

Printed in the United States of America

**Prepared for
U.S. Department of Energy**

Keywords: *Hanford, WTP, low-activity waste, waste glass*

Retention: *Permanent*

Characterization of the Sulfur-Saturated Melt Versions of the EMHQ LAW Glasses

M. C. Hsieh

February 2022

Savannah River National Laboratory is operated by Battelle Savannah River Alliance for the U.S. Department of Energy under Contract No. 89303321CEM000080.



REVIEWS AND APPROVALS

AUTHORS:

M. C. Hsieh, Applied Materials Research

TECHNICAL REVIEW:

A. N. Stanfield, Applied Materials Research, Reviewed per E7 2.60

APPROVAL:

J. Manna, Director
Environmental and Legacy Management

ACKNOWLEDGEMENTS

The author would like to thank Daniel Jones, Kandice Miles, Whitney Riley, and Kim Wyszynski at Savannah River National Laboratory for their skilled assistance with the sample and data analyses described in this report. The author thanks Renee Russell at Pacific Northwest National Laboratory for helpful discussions and review of these data and the report. Funding from the U.S. Department of Energy through Inter-Entity Work Order HAN-M0SRV00101 as managed by Albert A. Kruger is gratefully acknowledged.

EXECUTIVE SUMMARY

This report provides the results from the chemical analyses of a series of sulfur-saturated melt versions of the EMHQ Low-Activity Waste study glasses, a series of simulated nuclear waste glasses designed and fabricated at Pacific Northwest National Laboratory. These data will be used in the development, validation, and implementation of enhanced property/composition models for waste glass vitrification at Hanford.

Chemical analyses were performed on a representative sample of each of the sulfur-saturated melt versions of the glasses to allow for comparisons with targeted compositions, as well as the measured compositions of the quenched glasses. The relative differences between the targeted and measured concentrations of B_2O_3 , Na_2O , and SO_3 for several of the glasses and V_2O_5 and ZrO_2 in two individual glasses were greater than $\pm 10\%$.

The wash solutions contained mainly sodium and sulfur.

TABLE OF CONTENTS

LIST OF TABLES.....	viii
LIST OF ABBREVIATIONS.....	ix
1.0 Introduction.....	1
2.0 Experimental Procedure.....	1
2.1 Quality Assurance	1
2.2 Glasses Selected for Study	1
2.3 Glass Composition Analysis	2
2.4 Wash Solution Analysis	3
3.0 Results and Discussion	4
3.1 Review and Evaluation of the SSM Glass Composition Measurements.....	4
3.1.1 Treatment of Detection Limits.....	4
3.1.2 Composition Measurements by Glass Identifier.....	4
3.1.3 Results for the LRM Standard Glass	5
3.1.4 Measured versus Target Compositions.....	5
3.2 Comparison of Measured Compositions of Baseline and SSM Glasses	5
3.3 Review and Evaluation of the Wash Solution Measurements.....	5
3.3.1 Treatment of Detection Limits.....	6
3.3.2 Composition Measurements by Wash Solution Identifier.....	6
3.3.3 Results for the Standard Solutions.....	6
3.3.4 Measured Compositions of the Wash Solutions	6
4.0 Summary	6
5.0 References.....	7
Appendix A . Tables and Exhibits Supporting the EMHQ LAW Glass Composition Measurements	A-1
Appendix B . Tables and Exhibits Supporting the Wash Solution Composition Measurements	B-1

LIST OF TABLES

Table 2-1. Identifiers for the EMHQ LAW SSM Study Glasses	2
Table 2-2. Identifiers for the EMHQ LAW SSM Wash Solutions	2
Table 2-3. Preparation and Measurement Methods Used in Reporting the Analyte Concentrations of the Study Glasses.....	3
Table 2-4. Measurement Methods Used in Reporting the Analyte Concentrations of the Wash Solutions .	4

LIST OF ABBREVIATIONS

BDL	below detection limit
DOE	Department of Energy
hp std	High Purity Standards ICP multi-element custom solution SM-744-063
IC	ion chromatography
ICP-OES	inductively coupled plasma – optical emission spectroscopy
ID	identifier
KH	potassium hydroxide fusion
LAW	low-activity waste
LM	lithium metaborate fusion
LRM	low-activity test reference material
ORP	Office of River Protection
PF	sodium peroxide fusion
PNNL	Pacific Northwest National Laboratory
seq	sequence
SRNL	Savannah River National Laboratory
SSM	sulfur-saturated melt
std	High Purity Standards ICP multi-element custom solution SM-744-013
TTQAP	Task Technical and Quality Assurance Plan
wt.%	weight percent
WTP	Hanford Waste Treatment and Immobilization Plant

1.0 Introduction

The U.S. Department of Energy (DOE) is responsible for building the Hanford Tank Waste Treatment and Immobilization Plant (WTP) at the Hanford site in Washington to remediate 55 million gallons of radioactive waste that is temporarily stored in 177 underground tanks. The Office of River Protection (ORP) has requested that the Savannah River National Laboratory (SRNL) contribute in areas of recognized capabilities and expertise for glass waste form development to support successful startup of the WTP.

Successful efforts have allowed for demonstration of greatly enhanced treatment efficiencies of those projected from the minimum requirements set forth in the WTP Contract^a. Additional flexibility and expansion of the qualified glass forming region are the current focus.¹ SRNL support of this work is defined in the Task Technical and Quality Assurance Plan (TTQAP).²

This report provides results from the chemical analyses of the sulfur-saturated melt (SSM) versions of the Low-Activity Waste (LAW) composition boundary expansion study glasses, a series of simulated nuclear waste glasses designed and fabricated at Pacific Northwest National Laboratory (PNNL). The glasses were selected as part of a broader study of the influence of glass composition on chemical durability, sulfur retention, and other properties.³ The resulting data will be used in the development, validation, and implementation of enhanced property/composition models for nuclear waste glasses.¹

2.0 Experimental Procedure

2.1 Quality Assurance

Requirements for performing reviews of technical reports and the extent of review are established in Manual E7, Procedure 2.60.⁴ SRNL documents the extent and type of review using the SRNL Technical Report Design Checklist contained in WSRC-IM-2002-00011.⁵ Laboratory data for this study were recorded in the SRNL Electronic Laboratory Notebook system, experiment L6390-00441-04. The glasses provided by PNNL were designed and fabricated following a Task Plan.¹

2.2 Glasses Selected for Study

The baseline (quenched) glass compositions in this study were designed and fabricated by PNNL. Characterization of the baseline glasses were reported earlier.⁶ Samples of each of the SSM versions of the study glasses, along with samples of the wash solutions resulting from the preparation of each of the SSM glasses, were received at SRNL for chemical composition analysis. PNNL identifiers (IDs) for the glass samples and associated SRNL sample identifiers are listed in Table 2-1. The identifiers for the wash solutions are likewise listed in Table 2-2.

^aContract DE-AC27-01RV14136, as amended, U.S. Department of Energy, Richland, WA (2000).

Table 2-1. Identifiers for the EMHQ LAW SSM Study Glasses

PNNL Glass ID	Lab ID
EMHQ-LBE-01-SSM-S	S-13419
EMHQ-LBE-02-SSM-S	S-13420
EMHQ-LBE-03-SSM-S	S-13421
EMHQ-LBE-04-SSM-S	S-13422
EMHQ-LBE-05-SSM-S	S-13423
EMHQ-LBE-06-SSM-S	S-13424
EMHQ-LBE-07-SSM-S	S-13425
EMHQ-LBE-08-SSM-S	S-13426
EMHQ-LBE-09-SSM-S	S-13427
EMHQ-LBE-10-SSM-S	S-13428
EMHQ-LBE-SSM-S	S-13429

Table 2-2. Identifiers for the EMHQ LAW SSM Wash Solutions

PNNL Wash Solution ID	Lab ID
EMHQ-LBE-01-SSM-W	S-13430
EMHQ-LBE-02-SSM-W	S-13431
EMHQ-LBE-03-SSM-W	S-13432
EMHQ-LBE-04-SSM-W	S-13433
EMHQ-LBE-05-SSM-W	S-13434
EMHQ-LBE-06-SSM-W	S-13435
EMHQ-LBE-07-SSM-W	S-13436
EMHQ-LBE-08-SSM-W	S-13437
EMHQ-LBE-09-SSM-W	S-13438
EMHQ-LBE-10-SSM-W	S-13439
EMHQ-LBE-SSM-W	S-13440

2.3 Glass Composition Analysis

Chemical analyses were performed under the auspices of an analytical plan⁷ on a representative sample of each of the glasses listed in Table 2-1 to allow for comparisons with the targeted compositions. Three dissolution techniques were used for preparing each of the glass samples, in duplicate, for analysis (potassium hydroxide fusion (KH), lithium metaborate fusion (LM), and sodium peroxide fusion (PF)).⁸⁻¹⁰ Note that for some analytes, the analytical plan specified more than one preparation method for analysis. The results were reviewed and, in general, the method that provided better recovery of the analyte was selected for reporting.

Each of the duplicate samples was analyzed twice for each element of interest by inductively coupled plasma – optical emission spectroscopy (ICP-OES)¹¹ or ion chromatography (IC),¹² for a total of four measurements per element per glass. Glass standards were also intermittently measured to assess the performance of the ICP-OES and IC instruments over the course of these analyses. Specifically, several samples of the low-activity test reference material (LRM) were included as part of the analytical plans. The LRM composition reported as the “Consensus Average” is used as the reference composition of this glass.¹³ The preparation and measurement methods used for each of the reported glass components are listed in Table 2-3.

Table 2-3. Preparation and Measurement Methods Used in Reporting the Analyte Concentrations of the Study Glasses

Analyte	Measurement Method	Preparation Method
Al	ICP-OES	PF
B	ICP-OES	PF
Ca	ICP-OES	PF
Cl	IC	KH
Cr	ICP-OES	LM
F	IC	KH
Fe	ICP-OES	LM
K	ICP-OES	LM
Li	ICP-OES	PF
Mg	ICP-OES	LM
Mn	ICP-OES	LM
Na	ICP-OES	LM
P	ICP-OES	LM
S	ICP-OES	LM
Si	ICP-OES	PF
Sn	ICP-OES	PF
Ti	ICP-OES	LM
V	ICP-OES	LM
Zr	ICP-OES	PF

2.4 Wash Solution Analysis

Chemical analyses were performed under the auspices of an analytical plan⁷ on a representative sample of each of the wash solutions resulting from the preparation of the SSM versions of the glasses, as listed in Table 2-2. The samples were diluted at SRNL based on the expected concentrations of the species in solution in preparation for the analysis.

Each of the samples was analyzed in triplicate for each element of interest by ICP-OES¹¹ and IC¹². Solution standards,^b and blanks were also intermittently measured to assess the performance of the ICP-OES and IC instruments over the course of these analyses. The measurement methods used for each of the reported wash solution components are listed in Table 2-4.

^b ICP multi-element custom solutions, product numbers SM-744-013 and SM-744-063, High Purity Standards, North Charleston, SC.

Table 2-4. Measurement Methods Used in Reporting the Analyte Concentrations of the Wash Solutions

Analyte	Measurement Method
Al	ICP-OES
B	ICP-OES
Ca	ICP-OES
Cl ⁻	IC
Cr	ICP-OES
F ⁻	IC
Fe	ICP-OES
K	ICP-OES
Li	ICP-OES
Mg	ICP-OES
Mn	ICP-OES
Na	ICP-OES
P	ICP-OES
PO ₄ ³⁻	IC
S	ICP-OES
SO ₄ ²⁻	IC
Si	ICP-OES
Sn	ICP-OES
Ti	ICP-OES
V	ICP-OES
Zr	ICP-OES

3.0 Results and Discussion

JMP® Version 16.0.0 (SAS Institute, Inc.)¹⁴ was used to support these analyses.

3.1 Review and Evaluation of the SSM Glass Composition Measurements

Table A-1, Table A-2, and Table A-3 in Appendix A provide the elemental concentration measurements in weight percent (wt.%) from glasses prepared using KH, LM, and PF methods, respectively. Elemental measurements for samples of the LRM glass are also included in these tables of Appendix A.

3.1.1 *Treatment of Detection Limits*

The elemental concentrations in Table A-1, Table A-2, and Table A-3 in Appendix A were converted to oxide concentrations by multiplying the values of each element by the gravimetric factor for the corresponding oxide. A concentration measurement that was reported to be below the detection limit was set to the detection limit for the purposes of data review and calculating a sum of oxides for each glass in Table A-4. Concentration measurements that were below the detection limit (BDL) are denoted with a less than symbol (<).

3.1.2 *Composition Measurements by Glass Identifier*

Exhibit A-1 in Appendix A provides plots of the oxide concentration measurements by the PNNL Glass ID (including the LRM glasses) by Lab ID grouped by targeted concentration. Different symbols and colors are used to represent the different glasses. These plots show the individual measurements across the duplicates of each preparation method and the two instrument calibrations for each glass. Plotting the data

in this format provides an opportunity to review the values for each individual glass as a function of the duplicate preparations and duplicate measurements. A review of the plots presented in these exhibits reveals the repeatability of the four individual values for each oxide for each glass. There were no indications of errors in preparation or measurement that had to be addressed in treatment of the data.

3.1.3 Results for the LRM Standard Glass

Exhibit A-2 in Appendix A provides a comparison of the LRM results to their acceptability limits utilized by SRNL.¹¹ The review is in the form of plots of the measurements arranged by element, framed by upper and lower acceptability limits for the concentration of each element of interest. The results show that all measurements of the LRM elements of interest were within the acceptability limits during the execution of these analyses.

3.1.4 Measured versus Target Compositions

All measurements for each oxide for each glass (Table A-1, Table A-2, and Table A-3 in Appendix A) were used in calculating oxide values, which were then averaged to determine a representative chemical composition for each glass. A sum of oxides was also computed for each glass based upon the averaged oxide values. Exhibit A-3 in Appendix A provides plots showing the result for each glass for each oxide to allow PNNL to draw comparisons between the measured and targeted values.

Table A-4 in Appendix A provides a summary of the average compositions, targeted compositions and some associated differences and relative differences. The measured sums of oxides for all glasses fall within the interval of 96.1 wt.% to 98.7 wt. %, indicating acceptable recovery of the glass components.¹⁵ Entries in Table A-4 show the relative differences between the measured and targeted values for the analytes with measured and targeted values above 1 wt.%. The relative differences were shaded if they are 10% or more and are summarized below.

- B_2O_3 relative differences were 10% or greater for EMHQ-LBE-04-SSM-S, EMHQ-LBE-05-SSM-S, EMHQ-LBE-06-SSM-S, EMHQ-LBE-08-SSM-S, and EMHQ-LBE-10-SSM-S.
- Na_2O relative differences were 10% or greater for EMHQ-LBE-04-SSM-S, EMHQ-LBE-05-SSM-S, and EMHQ-LBE-06-SSM-S.
- The V_2O_5 relative difference was greater than 10% for EMHQ-LBE-05-SSM-S.
- The ZrO_2 relative difference was 10% for EMHQ-LBE-SSM-S.
- As expected, the measured SO_3 concentrations in the glasses were higher than targeted due to the use of the sulfur saturation method in fabricating these glasses.

3.2 Comparison of Measured Compositions of Baseline and SSM Glasses

Exhibit A-4 in Appendix A provides a comparison of the measured oxide concentrations among the baseline (quenched) and SSM versions of the study glasses. A review of Exhibit A-4 led to the following observation:

- The measured SO_3 concentrations were higher for SSM versions of the study glasses, as expected, due to the use of the sulfur saturation method in fabricating these glasses.

The discussion of the analyses of the wash solutions, provided in Section 3.3, may provide further insight into the measured compositions of the SSM glasses.

3.3 Review and Evaluation of the Wash Solution Measurements

Table B-1 in Appendix B provides the elemental concentration measurements in mg/L for the wash solutions as measured by ICP-OES. Table B-2 in Appendix B provides the anion concentration measurements in mg/L for the wash solutions as measured by IC. Elemental measurements of the blanks and standard solutions are included in these tables of Appendix B. These unprocessed data are provided so that the values are readily available should they be of interest for future reviews.

3.3.1 Treatment of Detection Limits

The elemental and anion concentrations in Table B-1 and Table B-2 of Appendix B include measurements that were reported to be below the detection limit. These values were set to the detection limit for the purposes of data review and of calculating an average composition for each wash solution.

3.3.2 Composition Measurements by Wash Solution Identifier

Exhibit B-1 in Appendix B provides plots of the elemental and anion concentration measurements grouped by the wash solution identifier (including the blanks and standard solutions). Different symbols and colors are used to represent the different solutions. Plotting the data in this format provides an opportunity to review the values for each individual solution as a function of the triplicate measurements. A review of the plots presented in these exhibits reveals the repeatability of the three individual values for each analyte for each solution. These observations were not considered to indicate an error in preparation or measurement that had to be addressed in treatment of the data. Therefore, the entire set of measurement data was used in determining representative, measured compositions for the wash solutions.

3.3.3 Results for the Standard Solutions

Table B-3 in Appendix B provides comparisons of the standard solution results to their reference values. The results in this table indicate no issues with the performance of the analyses.

3.3.4 Measured Compositions of the Wash Solutions

From the discussion of Section 3.3.2, all the measurements for each analyte for each wash solution (Table B-1 and Table B-2 of Appendix B) were averaged to determine a representative chemical composition for each solution. Table B-4 in Appendix B provides a summary of the average measured compositions of the wash solutions. The concentrations of PO_4^{3-} and SO_4^{2-} reported in these tables include the measured values from both ICP-OES and IC for comparison. The measured S and P concentrations from the ICP-OES analyses were converted to PO_4^{3-} and SO_4^{2-} concentrations by multiplying by the appropriate gravimetric factors to support these comparisons. A graphical representation of the average chemical composition data for each wash solution is provided in Exhibit B-2. The following observations are offered from the review of Table B-4 and Exhibit B-2:

- The wash solutions contained mainly Na (194-958 mg/L), S (124-567 mg/L), and SO_4^{2-} (386-1730 mg/L), which could be attributed to the excess sodium sulfate added as part of the SSM preparation process.
- The measured concentrations of B, Ca, Cr, K, Li, P, Si, and V in the wash solutions were generally below 100 mg/L.
- The measured concentrations of Al, Cl⁻, F⁻, Fe, Mg, Mn, PO_4^{3-} , Sn, Ti, and Zr in the wash solutions were near or below the detection limits.
- The ICP-OES measured concentrations of P were converted to PO_4^{3-} for comparison and were in the range of <3.07 to 16.4 mg/L. PO_4^{3-} values as measured by IC and were all <10.0.
- The ICP-OES measured concentrations of S were converted to SO_4^{2-} for comparison with IC values and were in the range of 370 to 1700 mg/L. SO_4^{2-} values as measured by IC were in the range of 386 to 1730 mg/L.

4.0 Summary

Chemical analyses were performed on a series of SSM versions of simulated nuclear waste glasses and resulting wash solutions with ICP-OES and IC. The glasses were designed and fabricated by PNNL as part of a broader study of the influence of glass composition on chemical durability, sulfur retention, and other properties.

The relative differences between the targeted and measured concentrations of B_2O_3 and Na_2O for several of the glasses and V_2O_5 and ZrO_2 for two individual glasses were greater than $\pm 10\%$. As expected, the measured concentrations of SO_3 in most of the glasses were higher than targeted due to the use of the sulfur saturation method in fabricating these glasses.

The wash solutions contained mainly sodium and sulfur.

5.0 References

1. R.L. Russell, "EMHQ LAW Glass Composition Boundary Expansion," Pacific Northwest National Laboratory, Richland, WA, EWG-TP-169, Revision 0.0, 2021.
2. K.M. Fox, "Task Technical and Quality Assurance Plan for Hanford Waste Glass Development and Characterization," Savannah River National Laboratory, Aiken, SC, SRNL-RP-2013-00692, Revision 1, 2016.
3. D.K. Peeler, D.S. Kim, J.D. Vienna, M.J. Schweiger, and G.F. Piepel, "Office of River Protection Advanced Low-Activity Waste Glass Research and Development Plan," Pacific Northwest National Laboratory, Richland, WA, PNNL-24883, EWG-RPT-008, Revision 0, 2015.
4. "Technical Reviews," Savannah River Site, Aiken, SC, Manual E7, Procedure 2.60, Rev. 20, 2021.
5. "Savannah River National Laboratory Technical Report Design Check Guidelines," Westinghouse Savannah River Company, Aiken, SC, WSRC-IM-2002-00011, Rev. 2, 2004.
6. M.C. Hsieh, "Composition Measurements of the EMHQ LAW Glasses," Savannah River National Laboratory, Aiken, SC, SRNL-STI-2021-00678, Revision 0, 2022.
7. M.C. Hsieh, "An Analytical Plan for Measuring the Chemical Compositions of the Sulfur-Saturated Melt Versions of the EMHQ LAW Study Glasses and Wash Solutions," Savannah River National Laboratory, Aiken, SC, SRNL-L3310-2021-00016, Revision 0, 2021.
8. "Sample Dissolution Using Potassium Hydroxide Fusion," Savannah River National Laboratory, Aiken, SC, Manual L29, Procedure ITS-0035, Rev. 3, 2015.
9. "Lithium Metaborate Fusion Preparation," Savannah River National Laboratory, Aiken, SC, Manual L33, Procedure 0071, Rev. 0, 2021.
10. "Dissolution of Glass, Sludge, and Slurry Samples Using $Na_2O_2/NaOH/HCl$," Savannah River National Laboratory, Aiken, SC, Manual L29, Procedure ITS-0040, Rev. 2, 2013.
11. "Calibration, Verification, and Operation of the Agilent 5110 ICP-OES Inductively Coupled Plasma-Optical Emission Spectrometer," Savannah River National Laboratory, Aiken, SC, Manual L33, Procedure 0242, Rev. 1, 2021.
12. "Anion Analysis Using the Dionex ICS 6000 Ion Chromatograph," Savannah River National Laboratory, Aiken, SC, Manual L33, Procedure 0244, Revision 1, 2020.

13. W.L. Ebert and S.F. Wolf, "Round-Robin Testing of a Reference Glass for Low-Activity Waste Forms," Argonne National Laboratory, Argonne, IL, ANL-99/22, Revision 0, 1999.
14. JMP(R) Version 16.0.0, SAS Institute Inc., Cary, NC, 2021.
15. C.M. Jantzen, "Verification of Glass Composition and Strategy for SGM and DWPF Glass Composition Determination," E.I du Pont de Nemours & Co., Savannah River Laboratory, Aiken, SC, DPST-86-708, 1987.

Appendix A. Tables and Exhibits Supporting the EMHQ LAW Glass Composition Measurements

Table A-1. KH Measurements (wt.%) of the SSM Study Glasses

PNNL ID	Block	Sub – Block	Seq	Lab ID	Cl ⁻	F ⁻
LRM	1	1	1	LRMKH111	<0.0250	0.820
EMHQ-LBE-08-SSM-S	1	1	2	S-13426KH11	0.0286	0.0543
EMHQ-LBE-01-SSM-S	1	1	3	S-13419KH11	0.0463	0.0692
EMHQ-LBE-02-SSM-S	1	1	4	S-13420KH21	0.0484	0.0722
EMHQ-LBE-03-SSM-S	1	1	5	S-13421KH11	0.0455	0.0702
EMHQ-LBE-01-SSM-S	1	1	6	S-13419KH21	0.0497	0.0758
EMHQ-LBE-05-SSM-S	1	1	7	S-13423KH21	0.0298	0.0482
EMHQ-LBE-06-SSM-S	1	1	8	S-13424KH11	0.0309	0.0499
EMHQ-LBE-SSM-S	1	1	9	S-13429KH11	<0.0250	0.875
EMHQ-LBE-04-SSM-S	1	1	10	S-13422KH21	0.0323	0.0514
EMHQ-LBE-03-SSM-S	1	1	11	S-13421KH21	0.0429	0.0644
EMHQ-LBE-02-SSM-S	1	1	12	S-13420KH11	0.0465	0.0716
LRM	1	1	13	LRMKH112	<0.0250	0.853
EMHQ-LBE-08-SSM-S	1	1	14	S-13426KH21	0.0286	0.0557
EMHQ-LBE-07-SSM-S	1	1	15	S-13425KH11	0.0280	0.0543
EMHQ-LBE-10-SSM-S	1	1	16	S-13428KH11	0.0271	0.0573
EMHQ-LBE-09-SSM-S	1	1	17	S-13427KH11	0.0304	0.0606
EMHQ-LBE-05-SSM-S	1	1	18	S-13423KH11	0.0296	0.0488
EMHQ-LBE-04-SSM-S	1	1	19	S-13422KH11	0.0334	0.0514
EMHQ-LBE-SSM-S	1	1	20	S-13429KH21	<0.0250	0.856
EMHQ-LBE-09-SSM-S	1	1	21	S-13427KH21	0.0267	0.0579
EMHQ-LBE-07-SSM-S	1	1	22	S-13425KH21	0.0291	0.0541
EMHQ-LBE-06-SSM-S	1	1	23	S-13424KH21	0.0302	0.0496
EMHQ-LBE-10-SSM-S	1	1	24	S-13428KH21	0.0276	0.0564
LRM	1	1	25	LRMKH113	<0.0250	0.888
LRM	1	2	1	LRMKH121	<0.0250	0.807
EMHQ-LBE-02-SSM-S	1	2	2	S-13420KH12	0.0425	0.0700
EMHQ-LBE-09-SSM-S	1	2	3	S-13427KH22	<0.0250	0.0556
EMHQ-LBE-05-SSM-S	1	2	4	S-13423KH22	0.0265	0.0463
EMHQ-LBE-06-SSM-S	1	2	5	S-13424KH22	0.0285	0.0482
EMHQ-LBE-03-SSM-S	1	2	6	S-13421KH22	0.0392	0.0628
EMHQ-LBE-09-SSM-S	1	2	7	S-13427KH12	0.0274	0.0590
EMHQ-LBE-08-SSM-S	1	2	8	S-13426KH12	0.0253	0.0538
EMHQ-LBE-10-SSM-S	1	2	9	S-13428KH12	<0.0250	0.0560
EMHQ-LBE-07-SSM-S	1	2	10	S-13425KH12	0.0260	0.0521
EMHQ-LBE-SSM-S	1	2	11	S-13429KH12	<0.0250	0.858
EMHQ-LBE-01-SSM-S	1	2	12	S-13419KH12	0.0424	0.0677
LRM	1	2	13	LRMKH122	<0.0250	0.836
EMHQ-LBE-05-SSM-S	1	2	14	S-13423KH12	0.0276	0.0471
EMHQ-LBE-SSM-S	1	2	15	S-13429KH22	<0.0250	0.836
EMHQ-LBE-02-SSM-S	1	2	16	S-13420KH22	0.0443	0.0705
EMHQ-LBE-03-SSM-S	1	2	17	S-13421KH12	0.0422	0.0692
EMHQ-LBE-07-SSM-S	1	2	18	S-13425KH22	0.0256	0.0524
EMHQ-LBE-04-SSM-S	1	2	19	S-13422KH22	0.0293	0.0496
EMHQ-LBE-08-SSM-S	1	2	20	S-13426KH22	0.0253	0.0528
EMHQ-LBE-01-SSM-S	1	2	21	S-13419KH22	0.0452	0.0730

Table A-1. KH Measurements (wt.%) of the SSM Study Glasses (continued)

PNNL ID	Block	Sub – Block	Seq	Lab ID	Cl ⁻	F ⁻
EMHQ-LBE-06-SSM-S	1	2	22	S-13424KH12	0.0290	0.0470
EMHQ-LBE-10-SSM-S	1	2	23	S-13428KH22	<0.0250	0.0546
EMHQ-LBE-04-SSM-S	1	2	24	S-13422KH12	0.0303	0.0503
LRM	1	2	25	LRMKH123	<0.0250	0.882

Table A-2. LM Measurements (wt.%) of the SSM Study Glasses (continued)

PNNL ID	Block	Sub – Block	Seq	Lab ID	Cr	Fe	K	Mg	Mn	Na	P	S	Ti	V
EMHQ-LBE-02-SSM-S	1	2	22	S-13420LM22	0.0254	0.0852	0.0772	0.0931	0.393	12.8	0.113	0.833	0.0678	2.00
EMHQ-LBE-06-SSM-S	1	2	23	S-13424LM12	0.0363	0.0875	0.0422	0.114	0.395	8.08	0.0517	0.885	0.0455	1.97
EMHQ-LBE-05-SSM-S	1	2	24	S-13423LM22	0.0177	0.0912	0.0471	0.116	0.396	7.85	0.0607	0.981	0.0459	1.62
LRM	1	2	25	LRMLM123	0.129	1.01	1.15	0.0649	0.0581	14.7	0.205	0.0864	0.0611	<0.100

Table A-3. PF Measurements (wt.%) of the SSM Study Glasses

PNNL ID	Block	Sub – Block	Seq	Lab ID	Al	B	Ca	Li	Si	Sn	Zr
LRM	1	1	1	LRMPF111	5.36	2.34	0.480	<0.100	25.6	<0.100	0.735
EMHQ-LBE-06-SSM-S	1	1	2	S-13424PF11	1.80	3.75	8.38	0.829	22.1	<0.100	1.39
EMHQ-LBE-10-SSM-S	1	1	3	S-13428PF21	4.50	2.44	5.61	<0.100	18.2	3.37	4.20
EMHQ-LBE-02-SSM-S	1	1	4	S-13420PF21	2.84	3.68	7.01	<0.100	19.0	<0.100	2.77
EMHQ-LBE-SSM-S	1	1	5	S-13429PF11	5.25	2.31	0.611	<0.100	25.1	<0.100	0.750
EMHQ-LBE-01-SSM-S	1	1	6	S-13419PF21	2.89	3.74	7.13	<0.100	19.4	<0.100	2.78
EMHQ-LBE-09-SSM-S	1	1	7	S-13427PF11	4.68	2.49	5.79	<0.100	18.6	3.49	4.36
EMHQ-LBE-04-SSM-S	1	1	8	S-13422PF21	1.83	3.85	8.55	0.865	22.4	<0.100	1.42
EMHQ-LBE-07-SSM-S	1	1	9	S-13425PF11	4.58	2.51	5.72	<0.100	18.7	3.43	4.28
EMHQ-LBE-06-SSM-S	1	1	10	S-13424PF21	1.78	3.79	8.39	0.823	22.0	<0.100	1.40
EMHQ-LBE-08-SSM-S	1	1	11	S-13426PF11	4.56	2.49	5.73	<0.100	18.4	3.37	4.27
EMHQ-LBE-05-SSM-S	1	1	12	S-13423PF11	1.84	3.85	8.65	0.859	22.4	<0.100	1.43
LRM	1	1	13	LRMPF112	5.45	2.38	0.496	<0.100	26.0	<0.100	0.781
EMHQ-LBE-10-SSM-S	1	1	14	S-13428PF11	4.61	2.49	5.70	<0.100	18.5	3.39	4.26
EMHQ-LBE-08-SSM-S	1	1	15	S-13426PF21	4.66	2.49	5.80	<0.100	18.7	3.48	4.32
EMHQ-LBE-03-SSM-S	1	1	16	S-13421PF11	2.91	3.71	7.12	<0.100	19.2	<0.100	2.78
EMHQ-LBE-SSM-S	1	1	17	S-13429PF21	5.52	2.41	0.652	<0.100	26.3	<0.100	0.786
EMHQ-LBE-07-SSM-S	1	1	18	S-13425PF21	4.69	2.52	5.84	<0.100	18.9	3.46	4.38
EMHQ-LBE-03-SSM-S	1	1	19	S-13421PF21	2.87	3.72	7.12	<0.100	19.3	<0.100	2.76
EMHQ-LBE-09-SSM-S	1	1	20	S-13427PF21	4.66	2.51	5.75	<0.100	18.7	3.41	4.27
EMHQ-LBE-05-SSM-S	1	1	21	S-13423PF21	1.86	3.85	8.70	0.867	22.4	<0.100	1.46
EMHQ-LBE-02-SSM-S	1	1	22	S-13420PF11	2.92	3.76	7.09	<0.100	19.6	<0.100	2.81
EMHQ-LBE-01-SSM-S	1	1	23	S-13419PF11	2.99	3.79	7.29	<0.100	19.8	<0.100	2.89
EMHQ-LBE-04-SSM-S	1	1	24	S-13422PF11	1.88	3.91	8.71	0.845	22.7	<0.100	1.48
LRM	1	1	25	LRMPF113	5.50	2.40	0.507	<0.100	26.2	<0.100	0.784
LRM	1	2	1	LRMPF121	5.04	2.27	0.455	<0.100	24.3	<0.100	0.715
EMHQ-LBE-03-SSM-S	1	2	2	S-13421PF22	2.72	3.56	6.92	<0.100	18.2	<0.100	2.68
EMHQ-LBE-02-SSM-S	1	2	3	S-13420PF22	2.75	3.56	6.95	<0.100	18.3	<0.100	2.74
EMHQ-LBE-05-SSM-S	1	2	4	S-13423PF22	1.78	3.72	8.54	0.808	21.3	<0.100	1.43
EMHQ-LBE-10-SSM-S	1	2	5	S-13428PF22	4.35	2.39	5.61	<0.100	17.6	3.28	4.19
EMHQ-LBE-09-SSM-S	1	2	6	S-13427PF12	4.44	2.40	5.68	<0.100	17.8	3.31	4.26
EMHQ-LBE-05-SSM-S	1	2	7	S-13423PF12	1.76	3.68	8.49	0.803	21.3	<0.100	1.39
EMHQ-LBE-SSM-S	1	2	8	S-13429PF22	5.06	2.27	0.607	<0.100	24.5	<0.100	0.754
EMHQ-LBE-08-SSM-S	1	2	9	S-13426PF22	4.40	2.39	5.61	<0.100	17.7	3.32	4.18
EMHQ-LBE-07-SSM-S	1	2	10	S-13425PF22	4.38	2.40	5.65	<0.100	17.8	3.33	4.27
EMHQ-LBE-01-SSM-S	1	2	11	S-13419PF12	2.73	3.54	6.98	<0.100	18.4	<0.100	2.74
EMHQ-LBE-03-SSM-S	1	2	12	S-13421PF12	2.75	3.55	6.83	<0.100	18.2	<0.100	2.701
LRM	1	2	13	LRMPF122	5.03	2.24	0.459	<0.100	24.4	<0.100	0.737
EMHQ-LBE-06-SSM-S	1	2	14	S-13424PF12	1.70	3.57	8.15	0.755	20.8	<0.100	1.34
EMHQ-LBE-10-SSM-S	1	2	15	S-13428PF12	4.28	2.35	5.45	<0.100	17.3	3.18	4.09
EMHQ-LBE-07-SSM-S	1	2	16	S-13425PF12	4.23	2.36	5.46	<0.100	17.4	3.22	4.12
EMHQ-LBE-04-SSM-S	1	2	17	S-13422PF12	1.72	3.65	8.26	0.754	21.1	<0.100	1.40
EMHQ-LBE-09-SSM-S	1	2	18	S-13427PF22	4.34	2.36	5.52	<0.100	17.6	3.15	4.11
EMHQ-LBE-02-SSM-S	1	2	19	S-13420PF12	2.67	3.51	6.80	<0.100	18.2	<0.100	2.68
EMHQ-LBE-01-SSM-S	1	2	20	S-13419PF22	2.62	3.46	6.68	<0.100	17.9	<0.100	2.63
EMHQ-LBE-SSM-S	1	2	21	S-13429PF12	4.91	2.20	0.577	<0.100	23.7	<0.100	0.738

Table A-3. PF Measurements (wt.%) of the SSM Study Glasses (continued)

PNNL ID	Block	Sub – Block	Seq	Lab ID	Al	B	Ca	Li	Si	Sn	Zr
EMHQ-LBE-04-SSM-S	1	2	22	S-13422PF22	1.68	3.56	8.17	0.757	20.7	<0.100	1.34
EMHQ-LBE-08-SSM-S	1	2	23	S-13426PF12	4.22	2.33	5.49	<0.100	17.3	3.16	4.10
EMHQ-LBE-06-SSM-S	1	2	24	S-13424PF22	1.72	3.63	8.20	0.760	21.0	<0.100	1.37
LRM	1	2	25	LRMPF123	4.94	2.20	0.445	<0.100	24.1	<0.100	0.738

Table A-4. Comparison of Measured versus Target Compositions

PNNL ID	Oxide	Mean Measurement (wt.%)	Target (wt.%)	Difference of Measured versus Target	% Difference Measured versus Target
LRM	Al ₂ O ₃	9.86	9.51	0.353	4%
LRM	B ₂ O ₃	7.42	7.85	-0.428	-5%
LRM	CaO	0.663	0.540	0.123	
LRM	Cl ⁻	<0.0250	0	0.025	
LRM	Cr ₂ O ₃	0.186	0.190	-0.004	
LRM	F ⁻	0.848	0.860	-0.012	
LRM	Fe ₂ O ₃	1.46	1.38	0.083	6%
LRM	K ₂ O	1.51	1.48	0.026	2%
LRM	Li ₂ O	<0.215	0.110	0.105	
LRM	MgO	0.107	0.100	0.007	
LRM	MnO	0.0746	0.0800	-0.005	
LRM	Na ₂ O	20.2	20.0	0.145	1%
LRM	P ₂ O ₅	0.463	0.540	-0.077	
LRM	SiO ₂	53.7	54.2	-0.504	-1%
LRM	SnO ₂	<0.127	0	0.127	
LRM	SO ₃	0.214	0.300	-0.086	
LRM	TiO ₂	0.105	0.100	0.005	
LRM	V ₂ O ₅	<0.179	0	0.179	
LRM	ZrO ₂	1.01	0.930	0.081	9%
LRM	Sum of Oxides	98.3	98.2	0.142	0%
EMHQ-LBE-01-SSM-S	Al ₂ O ₃	5.30	5.35	-0.046	-1%
EMHQ-LBE-01-SSM-S	B ₂ O ₃	11.7	12.9	-1.180	-9%
EMHQ-LBE-01-SSM-S	CaO	9.82	9.75	0.069	1%
EMHQ-LBE-01-SSM-S	Cl ⁻	0.0459	0.118	-0.072	
EMHQ-LBE-01-SSM-S	Cr ₂ O ₃	0.0436	0.0540	-0.010	
EMHQ-LBE-01-SSM-S	F ⁻	0.0714	0.114	-0.043	
EMHQ-LBE-01-SSM-S	Fe ₂ O ₃	0.121	0.132	-0.011	
EMHQ-LBE-01-SSM-S	K ₂ O	0.0977	0.126	-0.028	
EMHQ-LBE-01-SSM-S	Li ₂ O	<0.215	0	0.215	
EMHQ-LBE-01-SSM-S	MgO	0.157	0.144	0.013	
EMHQ-LBE-01-SSM-S	MnO	<0.0129	0.0100	0.003	
EMHQ-LBE-01-SSM-S	Na ₂ O	19.8	19.4	0.411	2%
EMHQ-LBE-01-SSM-S	P ₂ O ₅	0.259	0.305	-0.046	
EMHQ-LBE-01-SSM-S	SiO ₂	40.4	41.3	-0.913	-2%
EMHQ-LBE-01-SSM-S	SnO ₂	<0.127	0	0.127	
EMHQ-LBE-01-SSM-S	SO ₃	2.10	1.54	0.559	36%
EMHQ-LBE-01-SSM-S	TiO ₂	0.114	0.126	-0.012	
EMHQ-LBE-01-SSM-S	V ₂ O ₅	3.75	4.01	-0.258	-6%
EMHQ-LBE-01-SSM-S	ZrO ₂	3.73	3.75	-0.017	0%
EMHQ-LBE-01-SSM-S	Sum of Oxides	97.9	99.1	-1.240	-1%
EMHQ-LBE-02-SSM-S	Al ₂ O ₃	5.28	5.35	-0.070	-1%
EMHQ-LBE-02-SSM-S	B ₂ O ₃	11.7	12.9	-1.200	-9%
EMHQ-LBE-02-SSM-S	CaO	9.74	9.75	-0.011	0%
EMHQ-LBE-02-SSM-S	Cl ⁻	0.0454	0.118	-0.073	
EMHQ-LBE-02-SSM-S	Cr ₂ O ₃	0.0383	0.0540	-0.016	
EMHQ-LBE-02-SSM-S	F ⁻	0.0711	0.114	-0.043	

Table A-4. Comparison of Measured versus Target Compositions (continued)

PNNL ID	Oxide	Mean Measurement (wt.%)	Target (wt.%)	Difference of Measured versus Target	% Difference Measured versus Target
EMHQ-LBE-02-SSM-S	Fe ₂ O ₃	0.124	0.132	-0.008	
EMHQ-LBE-02-SSM-S	K ₂ O	0.0968	0.126	-0.029	
EMHQ-LBE-02-SSM-S	Li ₂ O	<0.215	0	0.215	
EMHQ-LBE-02-SSM-S	MgO	0.157	0.144	0.013	
EMHQ-LBE-02-SSM-S	MnO	0.516	0.555	-0.039	
EMHQ-LBE-02-SSM-S	Na ₂ O	19.3	19.4	-0.095	0%
EMHQ-LBE-02-SSM-S	P ₂ O ₅	0.268	0.305	-0.038	
EMHQ-LBE-02-SSM-S	SiO ₂	40.2	41.3	-1.130	-3%
EMHQ-LBE-02-SSM-S	SnO ₂	<0.127	0	0.127	
EMHQ-LBE-02-SSM-S	SO ₃	2.14	1.54	0.600	39%
EMHQ-LBE-02-SSM-S	TiO ₂	0.114	0.126	-0.012	
EMHQ-LBE-02-SSM-S	V ₂ O ₅	3.70	4.01	-0.303	-8%
EMHQ-LBE-02-SSM-S	ZrO ₂	3.71	3.75	-0.030	-1%
EMHQ-LBE-02-SSM-S	Sum of Oxides	97.5	99.6	-2.130	-2%
EMHQ-LBE-03-SSM-S	Al ₂ O ₃	5.31	5.30	0.017	0%
EMHQ-LBE-03-SSM-S	B ₂ O ₃	11.7	12.7	-1.040	-8%
EMHQ-LBE-03-SSM-S	CaO	9.79	9.56	0.235	2%
EMHQ-LBE-03-SSM-S	Cl ⁻	0.0425	0.117	-0.075	
EMHQ-LBE-03-SSM-S	Cr ₂ O ₃	0.0614	0.054	0.007	
EMHQ-LBE-03-SSM-S	F ⁻	0.0667	0.113	-0.046	
EMHQ-LBE-03-SSM-S	Fe ₂ O ₃	0.128	0.131	-0.003	
EMHQ-LBE-03-SSM-S	K ₂ O	0.0965	0.125	-0.029	
EMHQ-LBE-03-SSM-S	Li ₂ O	<0.215	0	0.215	
EMHQ-LBE-03-SSM-S	MgO	0.161	0.142	0.019	
EMHQ-LBE-03-SSM-S	MnO	0.534	0.555	-0.021	
EMHQ-LBE-03-SSM-S	Na ₂ O	18.9	19.2	-0.303	-2%
EMHQ-LBE-03-SSM-S	P ₂ O ₅	0.266	0.301	-0.035	
EMHQ-LBE-03-SSM-S	SiO ₂	40.1	40.9	-0.816	-2%
EMHQ-LBE-03-SSM-S	SnO ₂	<0.127	0	0.127	
EMHQ-LBE-03-SSM-S	SO ₃	2.12	1.53	0.598	39%
EMHQ-LBE-03-SSM-S	TiO ₂	0.116	0.125	-0.009	
EMHQ-LBE-03-SSM-S	V ₂ O ₅	3.68	3.97	-0.288	-7%
EMHQ-LBE-03-SSM-S	ZrO ₂	3.69	3.71	-0.019	-1%
EMHQ-LBE-03-SSM-S	Sum of Oxides	97.1	98.5	-1.470	-1%
EMHQ-LBE-04-SSM-S	Al ₂ O ₃	3.36	3.54	-0.179	-5%
EMHQ-LBE-04-SSM-S	B ₂ O ₃	12.1	13.5	-1.490	-11%
EMHQ-LBE-04-SSM-S	CaO	11.8	12.2	-0.455	-4%
EMHQ-LBE-04-SSM-S	Cl ⁻	0.0313	0.0760	-0.045	
EMHQ-LBE-04-SSM-S	Cr ₂ O ₃	0.0239	0.0250	-0.001	
EMHQ-LBE-04-SSM-S	F ⁻	0.0507	0.0950	-0.044	
EMHQ-LBE-04-SSM-S	Fe ₂ O ₃	0.122	0.131	-0.009	
EMHQ-LBE-04-SSM-S	K ₂ O	0.0628	0.0870	-0.024	
EMHQ-LBE-04-SSM-S	Li ₂ O	1.73	1.67	0.069	4%
EMHQ-LBE-04-SSM-S	MgO	0.19	0.180	0.010	
EMHQ-LBE-04-SSM-S	MnO	<0.0129	0.0130	0.000	
EMHQ-LBE-04-SSM-S	Na ₂ O	12.9	9.81	3.100	32%
EMHQ-LBE-04-SSM-S	P ₂ O ₅	0.139	0.158	-0.019	

Table A-4. Comparison of Measured versus Target Compositions (continued)

PNNL ID	Oxide	Mean Measurement (wt.%)	Target (wt.%)	Difference of Measured versus Target	% Difference Measured versus Target
EMHQ-LBE-04-SSM-S	SiO ₂	46.5	49.6	-3.170	-6%
EMHQ-LBE-04-SSM-S	SnO ₂	<0.127	0	0.127	
EMHQ-LBE-04-SSM-S	SO ₃	2.38	1.69	0.697	41%
EMHQ-LBE-04-SSM-S	TiO ₂	0.0756	0.0910	-0.015	
EMHQ-LBE-04-SSM-S	V ₂ O ₅	3.80	4.03	-0.233	-6%
EMHQ-LBE-04-SSM-S	ZrO ₂	1.90	2.00	-0.093	-5%
EMHQ-LBE-04-SSM-S	Sum of Oxides	97.2	99.0	-1.780	-2%
EMHQ-LBE-05-SSM-S	Al ₂ O ₃	3.42	3.54	-0.118	-3%
EMHQ-LBE-05-SSM-S	B ₂ O ₃	12.2	13.5	-1.380	-10%
EMHQ-LBE-05-SSM-S	CaO	12.0	12.2	-0.214	-2%
EMHQ-LBE-05-SSM-S	Cl ⁻	0.0284	0.0760	-0.048	
EMHQ-LBE-05-SSM-S	Cr ₂ O ₃	0.0262	0.0250	0.001	
EMHQ-LBE-05-SSM-S	F ⁻	0.0476	0.0950	-0.047	
EMHQ-LBE-05-SSM-S	Fe ₂ O ₃	0.131	0.131	0.000	
EMHQ-LBE-05-SSM-S	K ₂ O	0.0614	0.0870	-0.026	
EMHQ-LBE-05-SSM-S	Li ₂ O	1.80	1.67	0.131	8%
EMHQ-LBE-05-SSM-S	MgO	0.192	0.180	0.012	
EMHQ-LBE-05-SSM-S	MnO	0.513	0.558	-0.045	
EMHQ-LBE-05-SSM-S	Na ₂ O	12.5	9.81	2.720	28%
EMHQ-LBE-05-SSM-S	P ₂ O ₅	0.140	0.158	-0.018	
EMHQ-LBE-05-SSM-S	SiO ₂	46.7	49.6	-2.910	-6%
EMHQ-LBE-05-SSM-S	SnO ₂	<0.127	0	0.127	
EMHQ-LBE-05-SSM-S	SO ₃	2.47	1.69	0.787	47%
EMHQ-LBE-05-SSM-S	TiO ₂	0.0763	0.0910	-0.015	
EMHQ-LBE-05-SSM-S	V ₂ O ₅	3.56	4.03	-0.470	-12%
EMHQ-LBE-05-SSM-S	ZrO ₂	1.93	2.00	-0.070	-3%
EMHQ-LBE-05-SSM-S	Sum of Oxides	98.0	99.6	-1.580	-2%
EMHQ-LBE-06-SSM-S	Al ₂ O ₃	3.31	3.50	-0.196	-6%
EMHQ-LBE-06-SSM-S	B ₂ O ₃	11.9	13.4	-1.540	-11%
EMHQ-LBE-06-SSM-S	CaO	11.6	12.1	-0.531	-4%
EMHQ-LBE-06-SSM-S	Cl ⁻	0.0297	0.0750	-0.045	
EMHQ-LBE-06-SSM-S	Cr ₂ O ₃	0.0416	0.0250	0.017	
EMHQ-LBE-06-SSM-S	F ⁻	0.0487	0.0940	-0.045	
EMHQ-LBE-06-SSM-S	Fe ₂ O ₃	0.126	0.129	-0.003	
EMHQ-LBE-06-SSM-S	K ₂ O	0.0566	0.0870	-0.030	
EMHQ-LBE-06-SSM-S	Li ₂ O	1.70	1.64	0.067	4%
EMHQ-LBE-06-SSM-S	MgO	0.191	0.179	0.012	
EMHQ-LBE-06-SSM-S	MnO	0.516	0.558	-0.042	
EMHQ-LBE-06-SSM-S	Na ₂ O	12.5	9.71	2.810	29%
EMHQ-LBE-06-SSM-S	P ₂ O ₅	0.129	0.157	-0.028	
EMHQ-LBE-06-SSM-S	SiO ₂	45.9	49.1	-3.210	-7%
EMHQ-LBE-06-SSM-S	SnO ₂	<0.127	0	0.127	
EMHQ-LBE-06-SSM-S	SO ₃	2.31	1.67	0.638	38%
EMHQ-LBE-06-SSM-S	TiO ₂	0.0759	0.0900	-0.014	
EMHQ-LBE-06-SSM-S	V ₂ O ₅	3.72	3.99	-0.268	-7%
EMHQ-LBE-06-SSM-S	ZrO ₂	1.86	1.98	-0.121	-6%
EMHQ-LBE-06-SSM-S	Sum of Oxides	96.1	98.5	-2.400	-2%

Table A-4. Comparison of Measured versus Target Compositions (continued)

PNNL ID	Oxide	Mean Measurement (wt.%)	Target (wt.%)	Difference of Measured versus Target	% Difference Measured versus Target
EMHQ-LBE-07-SSM-S	Al ₂ O ₃	8.45	8.69	-0.247	-3%
EMHQ-LBE-07-SSM-S	B ₂ O ₃	7.88	8.64	-0.756	-9%
EMHQ-LBE-07-SSM-S	CaO	7.93	7.63	0.298	4%
EMHQ-LBE-07-SSM-S	Cl ⁻	0.0272	0.0770	-0.050	
EMHQ-LBE-07-SSM-S	Cr ₂ O ₃	0.128	0.167	-0.039	
EMHQ-LBE-07-SSM-S	F ⁻	0.0532	0.0670	-0.014	
EMHQ-LBE-07-SSM-S	Fe ₂ O ₃	0.117	0.124	-0.007	
EMHQ-LBE-07-SSM-S	K ₂ O	0.0519	0.0710	-0.019	
EMHQ-LBE-07-SSM-S	Li ₂ O	<0.215	0	0.215	
EMHQ-LBE-07-SSM-S	MgO	0.130	0.115	0.015	
EMHQ-LBE-07-SSM-S	MnO	<0.0129	0.00800	0.005	
EMHQ-LBE-07-SSM-S	Na ₂ O	22.8	23.2	-0.395	-2%
EMHQ-LBE-07-SSM-S	P ₂ O ₅	0.266	0.310	-0.044	
EMHQ-LBE-07-SSM-S	SiO ₂	38.9	39.4	-0.428	-1%
EMHQ-LBE-07-SSM-S	SnO ₂	4.27	4.34	-0.071	-2%
EMHQ-LBE-07-SSM-S	SO ₃	1.32	0.190	1.130	596%
EMHQ-LBE-07-SSM-S	TiO ₂	0.134	0.139	-0.005	
EMHQ-LBE-07-SSM-S	V ₂ O ₅	<0.179	0	0.179	
EMHQ-LBE-07-SSM-S	ZrO ₂	5.76	5.82	-0.062	-1%
EMHQ-LBE-07-SSM-S	Sum of Oxides	98.7	99.0	-0.294	0%
EMHQ-LBE-08-SSM-S	Al ₂ O ₃	8.43	8.69	-0.266	-3%
EMHQ-LBE-08-SSM-S	B ₂ O ₃	7.81	8.64	-0.829	-10%
EMHQ-LBE-08-SSM-S	CaO	7.92	7.63	0.284	4%
EMHQ-LBE-08-SSM-S	Cl ⁻	0.0270	0.0770	-0.050	
EMHQ-LBE-08-SSM-S	Cr ₂ O ₃	0.0728	0.167	-0.094	
EMHQ-LBE-08-SSM-S	F ⁻	0.0542	0.0670	-0.013	
EMHQ-LBE-08-SSM-S	Fe ₂ O ₃	0.133	0.124	0.009	
EMHQ-LBE-08-SSM-S	K ₂ O	0.0522	0.0710	-0.019	
EMHQ-LBE-08-SSM-S	Li ₂ O	<0.215	0	0.215	
EMHQ-LBE-08-SSM-S	MgO	0.130	0.115	0.015	
EMHQ-LBE-08-SSM-S	MnO	0.528	0.553	-0.025	
EMHQ-LBE-08-SSM-S	Na ₂ O	22.4	23.2	-0.833	-4%
EMHQ-LBE-08-SSM-S	P ₂ O ₅	0.258	0.310	-0.052	
EMHQ-LBE-08-SSM-S	SiO ₂	38.6	39.4	-0.802	-2%
EMHQ-LBE-08-SSM-S	SnO ₂	4.23	4.34	-0.106	-2%
EMHQ-LBE-08-SSM-S	SO ₃	1.30	0.190	1.110	586%
EMHQ-LBE-08-SSM-S	TiO ₂	0.135	0.139	-0.004	
EMHQ-LBE-08-SSM-S	V ₂ O ₅	<0.179	0	0.179	
EMHQ-LBE-08-SSM-S	ZrO ₂	5.70	5.82	-0.123	-2%
EMHQ-LBE-08-SSM-S	Sum of Oxides	98.1	99.5	-1.400	-1%
EMHQ-LBE-09-SSM-S	Al ₂ O ₃	8.56	8.61	-0.046	-1%
EMHQ-LBE-09-SSM-S	B ₂ O ₃	7.86	8.55	-0.693	-8%
EMHQ-LBE-09-SSM-S	CaO	7.95	7.56	0.399	5%
EMHQ-LBE-09-SSM-S	Cl ⁻	<0.0274	0.0760	-0.049	
EMHQ-LBE-09-SSM-S	Cr ₂ O ₃	0.0694	0.165	-0.096	
EMHQ-LBE-09-SSM-S	F ⁻	0.0583	0.0660	-0.008	
EMHQ-LBE-09-SSM-S	Fe ₂ O ₃	0.113	0.123	-0.010	

Table A-4. Comparison of Measured versus Target Compositions (continued)

PNNL ID	Oxide	Mean Measurement (wt.%)	Target (wt.%)	Difference of Measured versus Target	% Difference Measured versus Target
EMHQ-LBE-09-SSM-S	K ₂ O	0.0490	0.0710	-0.022	
EMHQ-LBE-09-SSM-S	Li ₂ O	<0.215	0	0.215	
EMHQ-LBE-09-SSM-S	MgO	0.127	0.114	0.013	
EMHQ-LBE-09-SSM-S	MnO	0.529	0.553	-0.024	
EMHQ-LBE-09-SSM-S	Na ₂ O	22.2	23.0	-0.835	-4%
EMHQ-LBE-09-SSM-S	P ₂ O ₅	0.261	0.307	-0.046	
EMHQ-LBE-09-SSM-S	SiO ₂	38.9	39.0	-0.083	0%
EMHQ-LBE-09-SSM-S	SnO ₂	4.24	4.29	-0.054	-1%
EMHQ-LBE-09-SSM-S	SO ₃	1.25	0.188	1.060	565%
EMHQ-LBE-09-SSM-S	TiO ₂	0.140	0.138	0.002	
EMHQ-LBE-09-SSM-S	V ₂ O ₅	<0.179	0	0.179	
EMHQ-LBE-09-SSM-S	ZrO ₂	5.74	5.76	-0.020	0%
EMHQ-LBE-09-SSM-S	Sum of Oxides	98.4	98.5	-0.113	0%
EMHQ-LBE-10-SSM-S	Al ₂ O ₃	8.38	8.69	-0.313	-4%
EMHQ-LBE-10-SSM-S	B ₂ O ₃	7.78	8.64	-0.853	-10%
EMHQ-LBE-10-SSM-S	CaO	7.83	7.63	0.193	3%
EMHQ-LBE-10-SSM-S	Cl ⁻	<0.0262	0.0770	-0.051	
EMHQ-LBE-10-SSM-S	Cr ₂ O ₃	0.0753	0.167	-0.092	
EMHQ-LBE-10-SSM-S	F ⁻	0.0561	0.0670	-0.011	
EMHQ-LBE-10-SSM-S	Fe ₂ O ₃	0.111	0.124	-0.013	
EMHQ-LBE-10-SSM-S	K ₂ O	0.0526	0.0710	-0.018	
EMHQ-LBE-10-SSM-S	Li ₂ O	<0.215	0	0.215	
EMHQ-LBE-10-SSM-S	MgO	0.128	0.115	0.013	
EMHQ-LBE-10-SSM-S	MnO	0.267	0.281	-0.014	
EMHQ-LBE-10-SSM-S	Na ₂ O	22.4	23.2	-0.867	-4%
EMHQ-LBE-10-SSM-S	P ₂ O ₅	0.258	0.310	-0.052	
EMHQ-LBE-10-SSM-S	SiO ₂	38.3	39.4	-1.070	-3%
EMHQ-LBE-10-SSM-S	SnO ₂	4.20	4.34	-0.141	-3%
EMHQ-LBE-10-SSM-S	SO ₃	1.30	0.190	1.110	584%
EMHQ-LBE-10-SSM-S	TiO ₂	0.140	0.139	0.001	
EMHQ-LBE-10-SSM-S	V ₂ O ₅	<0.179	0	0.179	
EMHQ-LBE-10-SSM-S	ZrO ₂	5.65	5.82	-0.167	-3%
EMHQ-LBE-10-SSM-S	Sum of Oxides	97.3	99.3	-1.950	-2%
EMHQ-LBE-SSM-S	Al ₂ O ₃	9.80	9.51	0.287	3%
EMHQ-LBE-SSM-S	B ₂ O ₃	7.40	7.85	-0.452	-6%
EMHQ-LBE-SSM-S	CaO	0.856	0.540	0.316	
EMHQ-LBE-SSM-S	Cl ⁻	<0.0250	0	0.025	
EMHQ-LBE-SSM-S	Cr ₂ O ₃	0.190	0.190	0.000	
EMHQ-LBE-SSM-S	F ⁻	0.856	0.860	-0.004	
EMHQ-LBE-SSM-S	Fe ₂ O ₃	1.41	1.38	0.031	2%
EMHQ-LBE-SSM-S	K ₂ O	1.40	1.48	-0.083	-6%
EMHQ-LBE-SSM-S	Li ₂ O	<0.215	0.110	0.105	
EMHQ-LBE-SSM-S	MgO	0.111	0.100	0.011	
EMHQ-LBE-SSM-S	MnO	0.0761	0.0800	-0.004	
EMHQ-LBE-SSM-S	Na ₂ O	20.2	20.0	0.156	1%
EMHQ-LBE-SSM-S	P ₂ O ₅	0.446	0.540	-0.094	
EMHQ-LBE-SSM-S	SiO ₂	53.3	54.2	-0.931	-2%

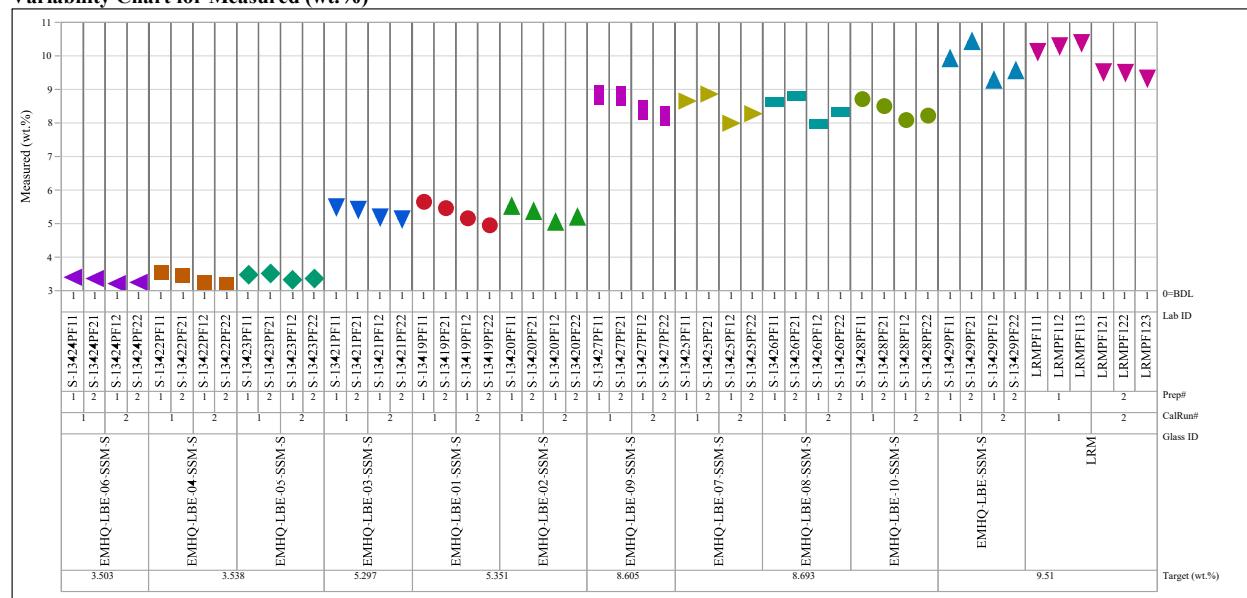
Table A-4. Comparison of Measured versus Target Compositions (continued)

PNNL ID	Oxide	Mean Measurement (wt.%)	Target (wt.%)	Difference of Measured versus Target	% Difference Measured versus Target
EMHQ-LBE-SSM-S	SnO ₂	<0.127	0	0.127	
EMHQ-LBE-SSM-S	SO ₃	0.207	0.300	-0.093	
EMHQ-LBE-SSM-S	TiO ₂	0.110	0.100	0.010	
EMHQ-LBE-SSM-S	V ₂ O ₅	<0.179	0	0.179	
EMHQ-LBE-SSM-S	ZrO ₂	1.02	0.930	0.093	10%
EMHQ-LBE-SSM-S	Sum of Oxides	97.9	98.2	-0.321	0%

Exhibit A-1. Plots of Oxide Measurements by Glass Identifier by Target Concentrations

Oxide=Al₂O₃ (wt%)

Variability Chart for Measured (wt.%)



Oxide=B₂O₃ (wt%)

Variability Chart for Measured (wt.%)

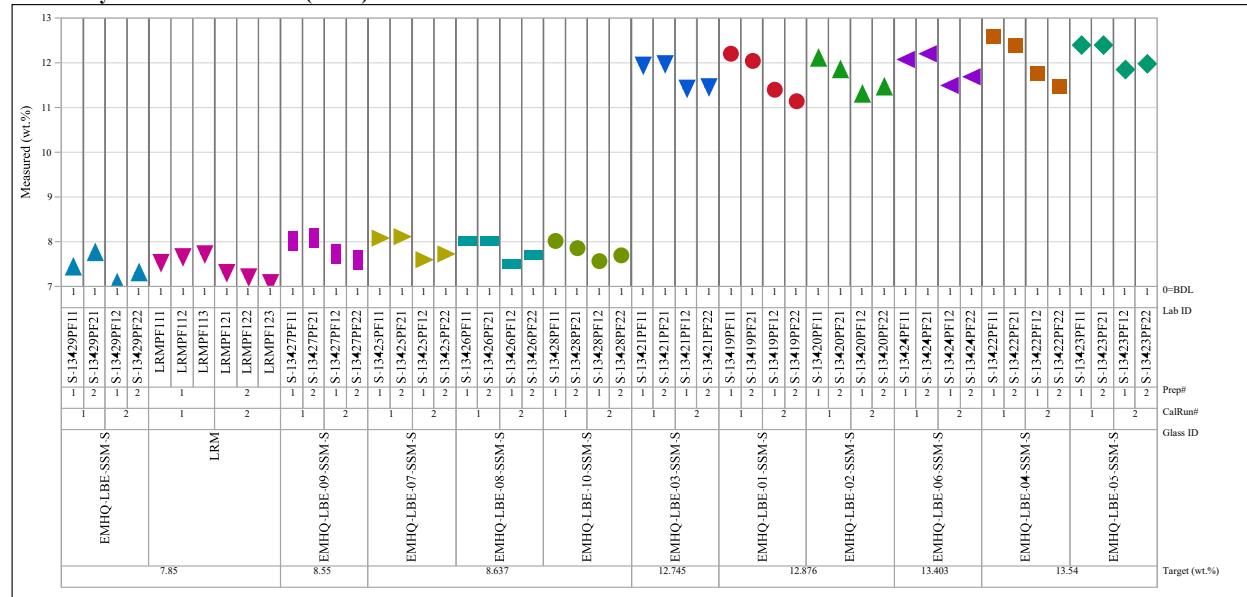
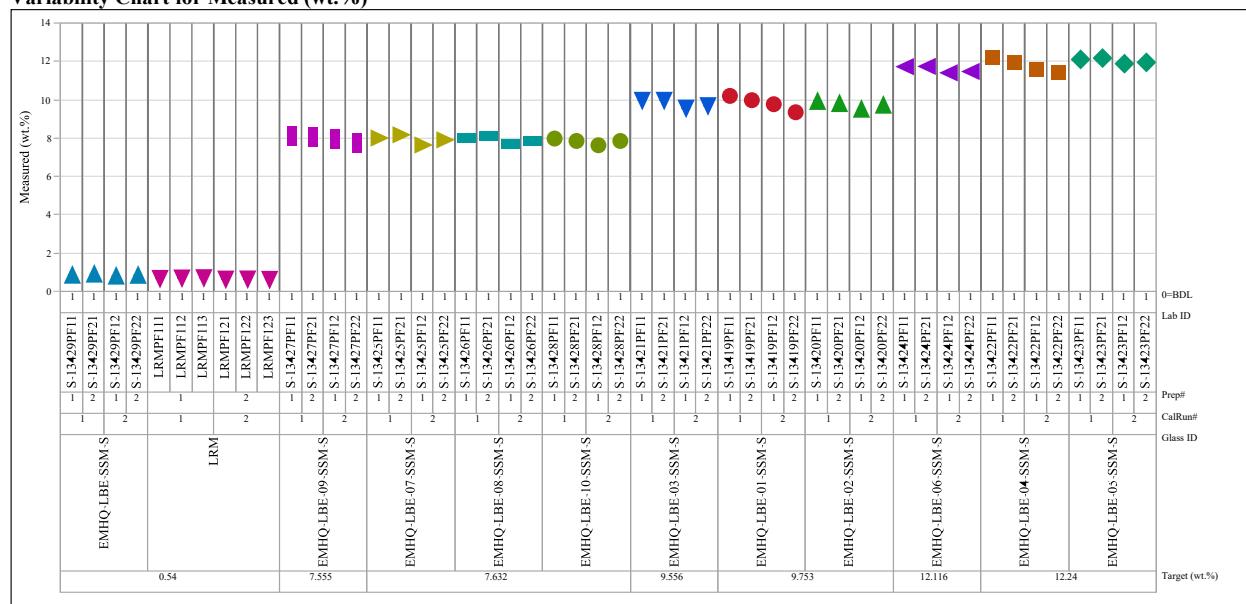


Exhibit A-1. Plots of Oxide Measurements by Glass Identifier by Target Concentrations (continued)

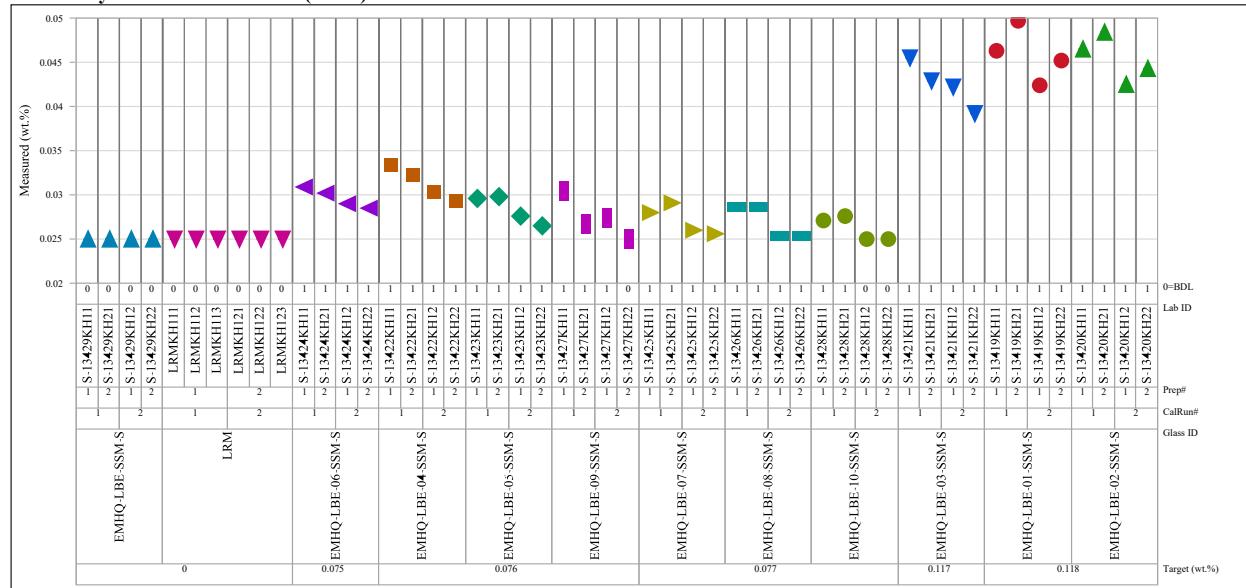
Oxide=CaO (wt%)

Variability Chart for Measured (wt.%)



Oxide=Cl (wt%)

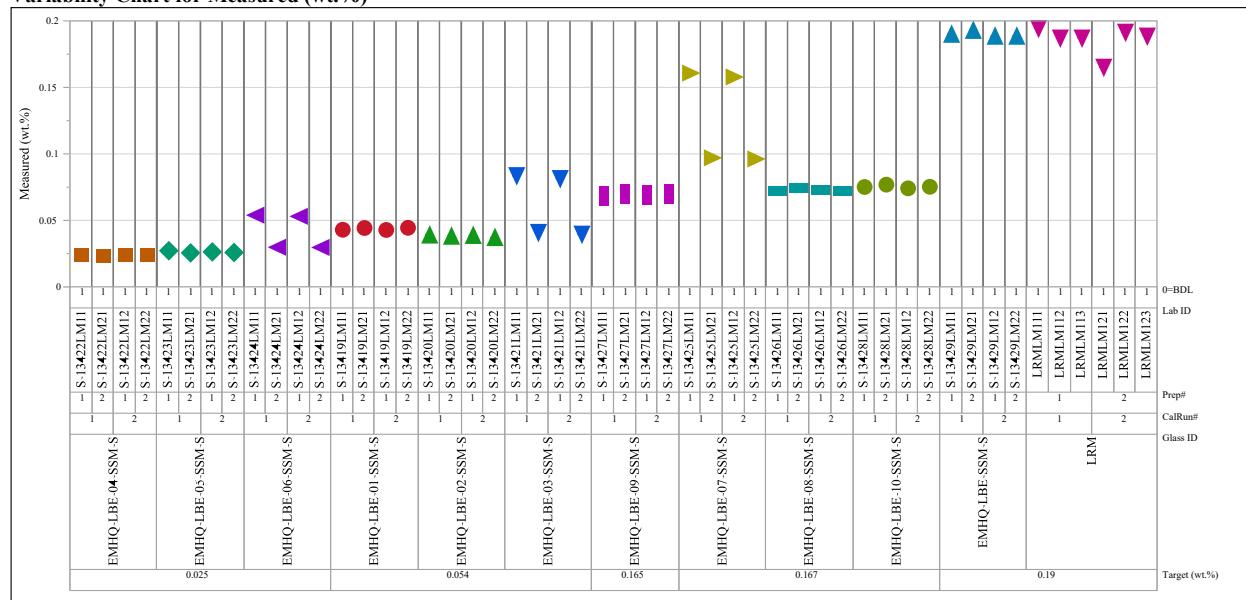
Variability Chart for Measured (wt.%)



**Exhibit A-1. Plots of Oxide Measurements by Glass Identifier by Target Concentrations
(continued)**

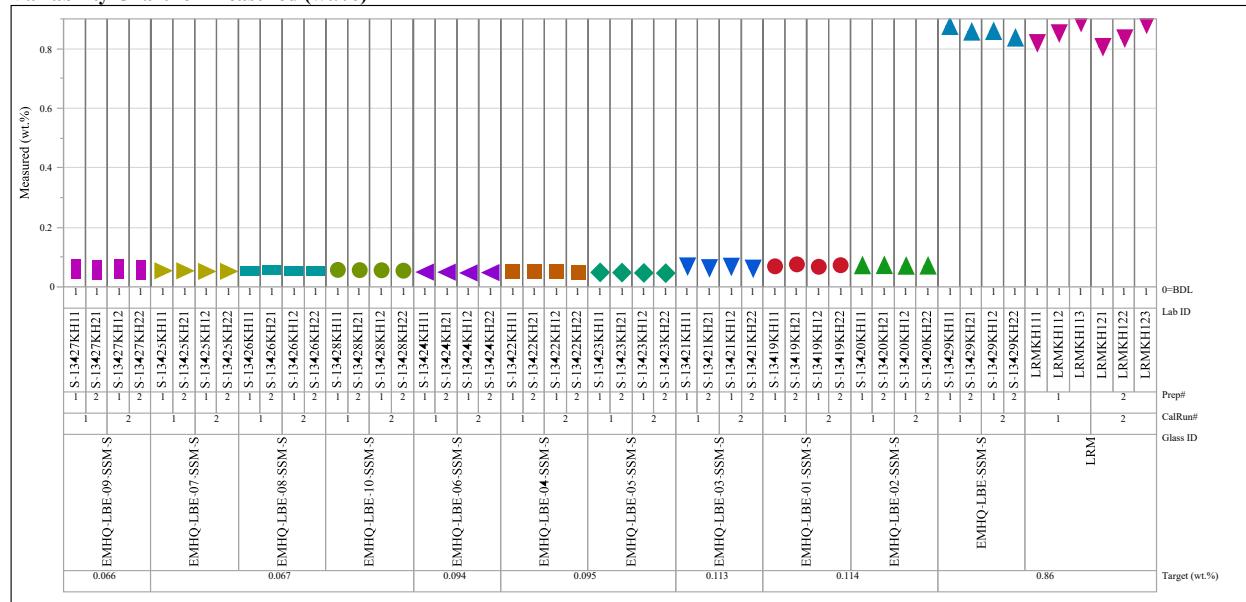
Oxide=Cr₂O₃ (wt%)

Variability Chart for Measured (wt.%)



Oxide=F⁻ (wt%)

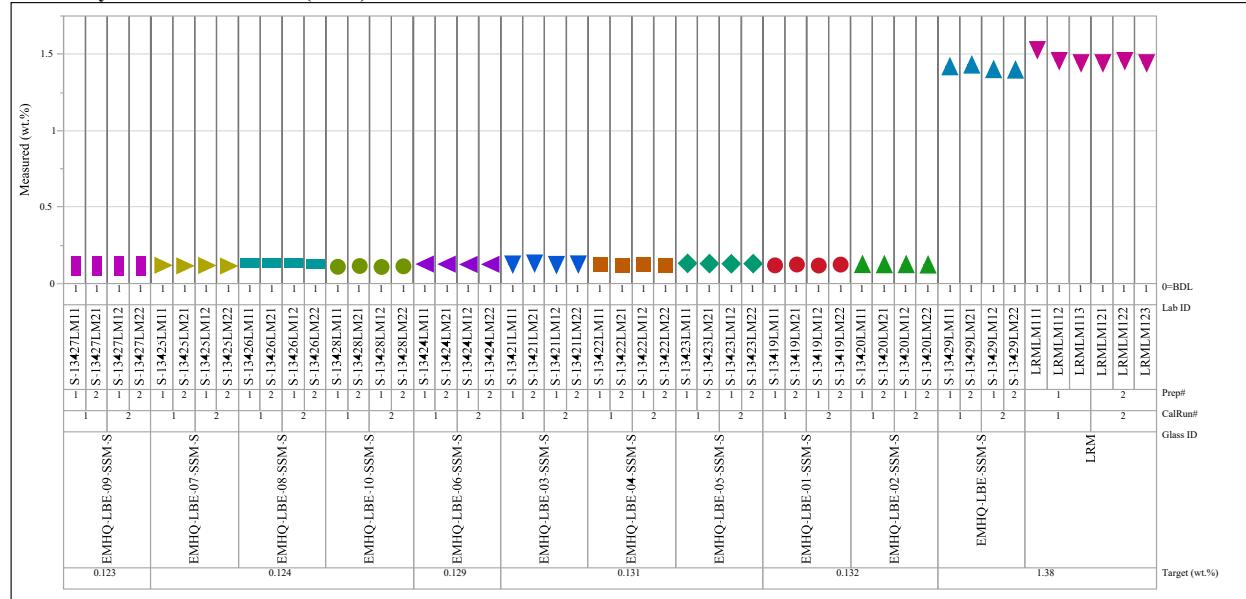
Variability Chart for Measured (wt.%)



**Exhibit A-1. Plots of Oxide Measurements by Glass Identifier by Target Concentrations
(continued)**

Oxide= Fe_2O_3 (wt%)

Variability Chart for Measured (wt.%)



Oxide= K_2O (wt%)

Variability Chart for Measured (wt.%)

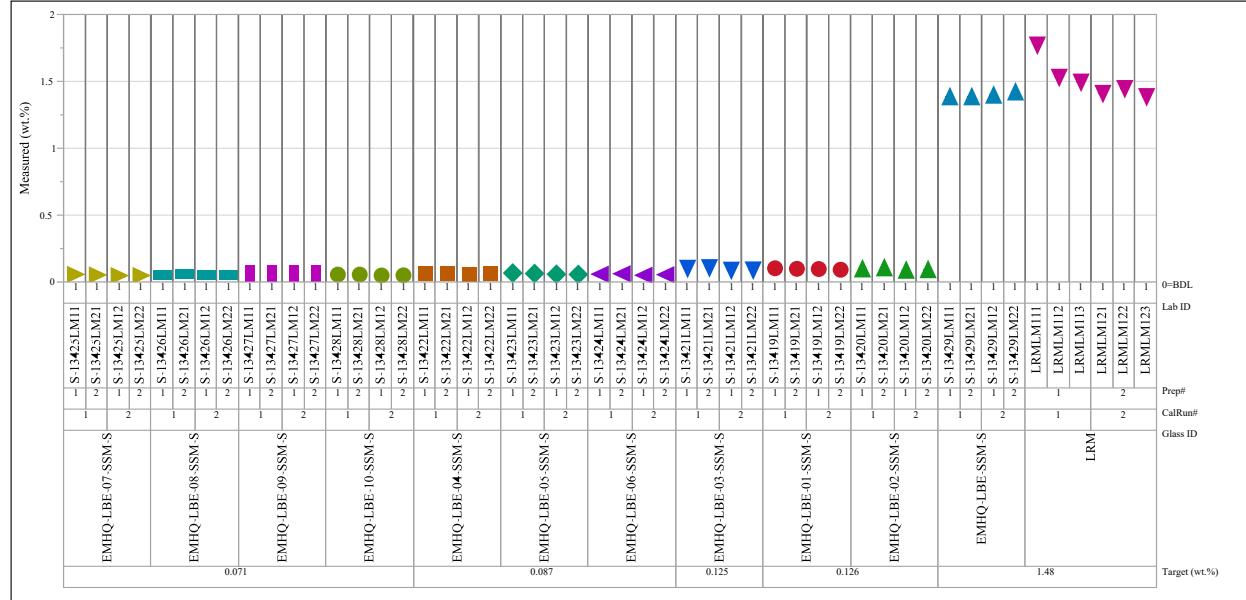
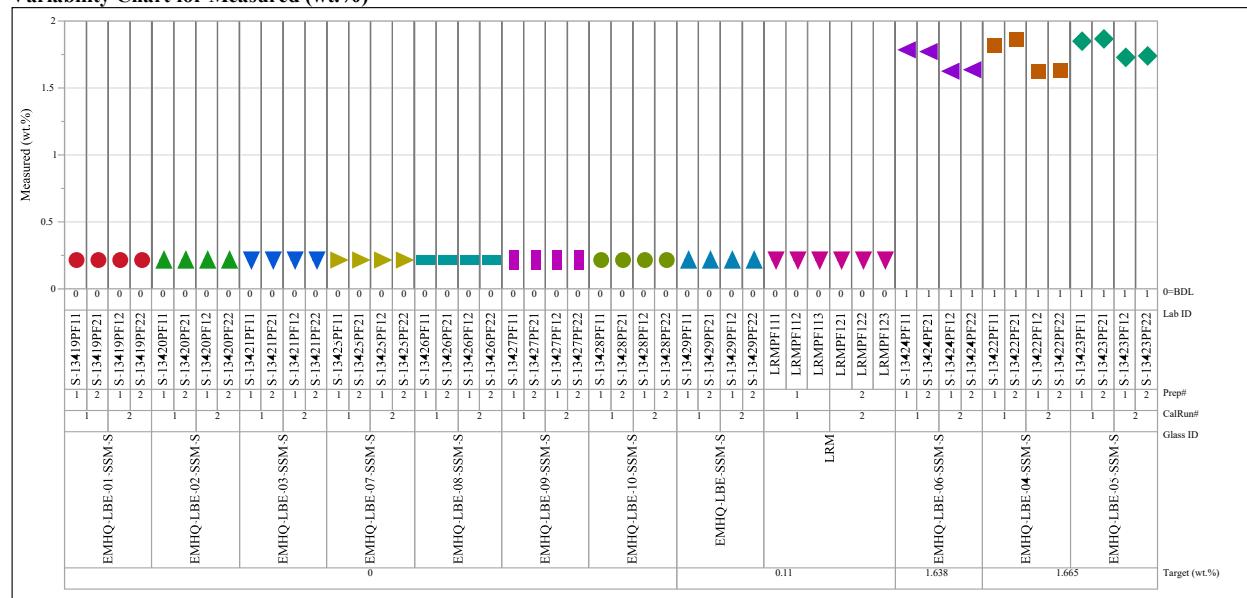


Exhibit A-1. Plots of Oxide Measurements by Glass Identifier by Target Concentrations (continued)

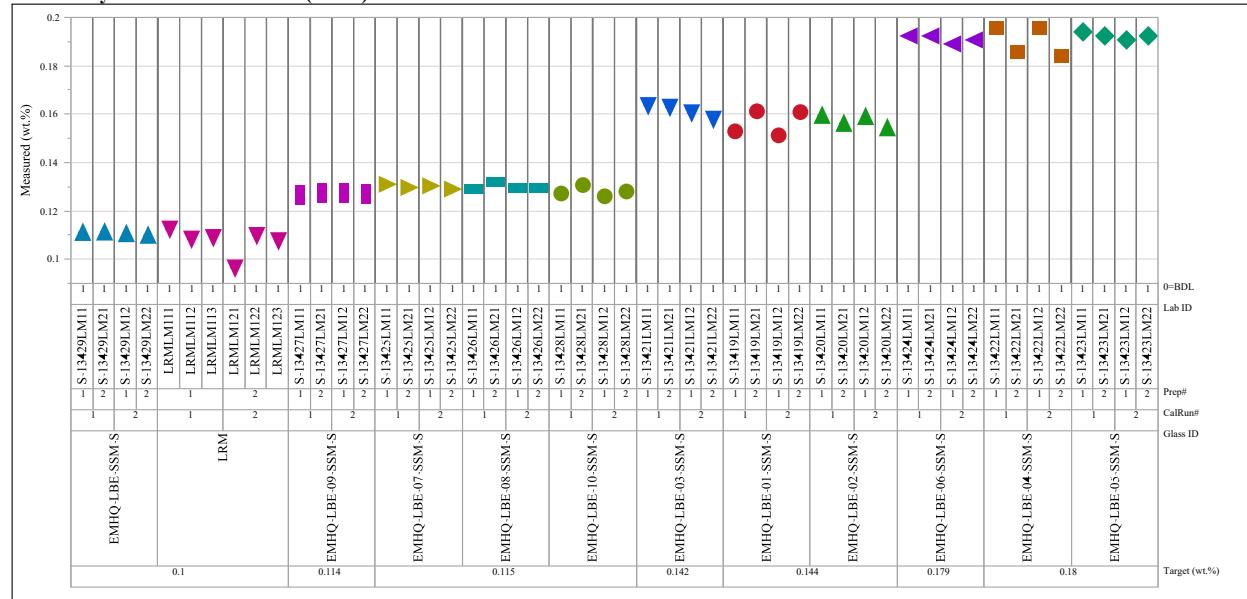
Oxide=Li₂O (wt%)

Variability Chart for Measured (wt.%)



Oxide=MgO (wt%)

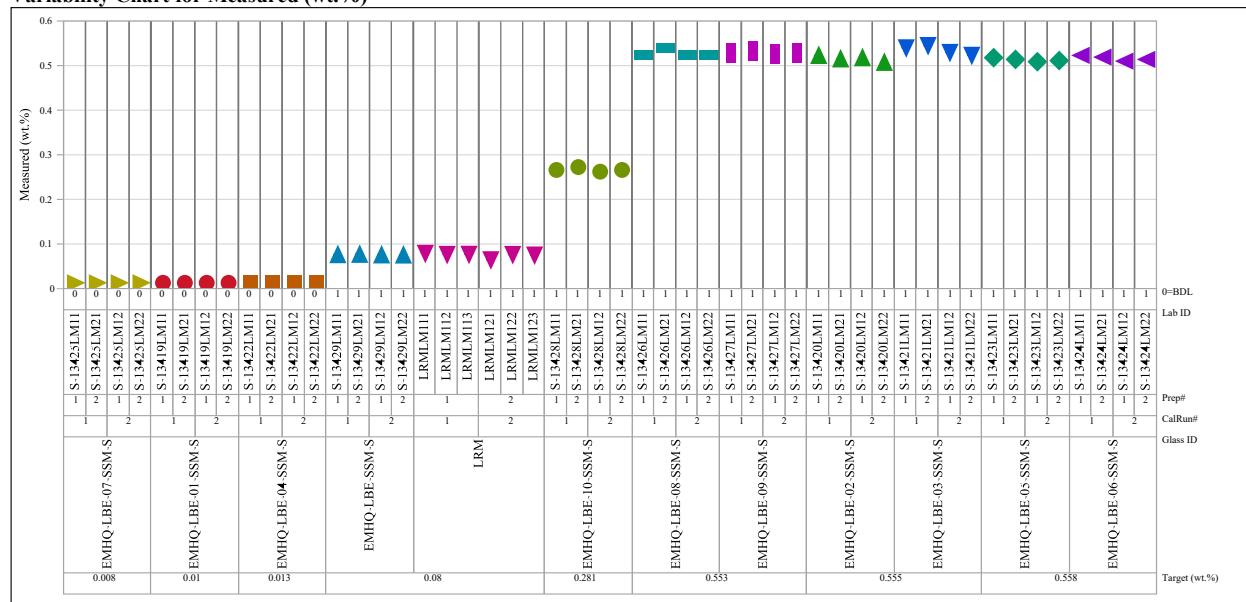
Variability Chart for Measured (wt.%)



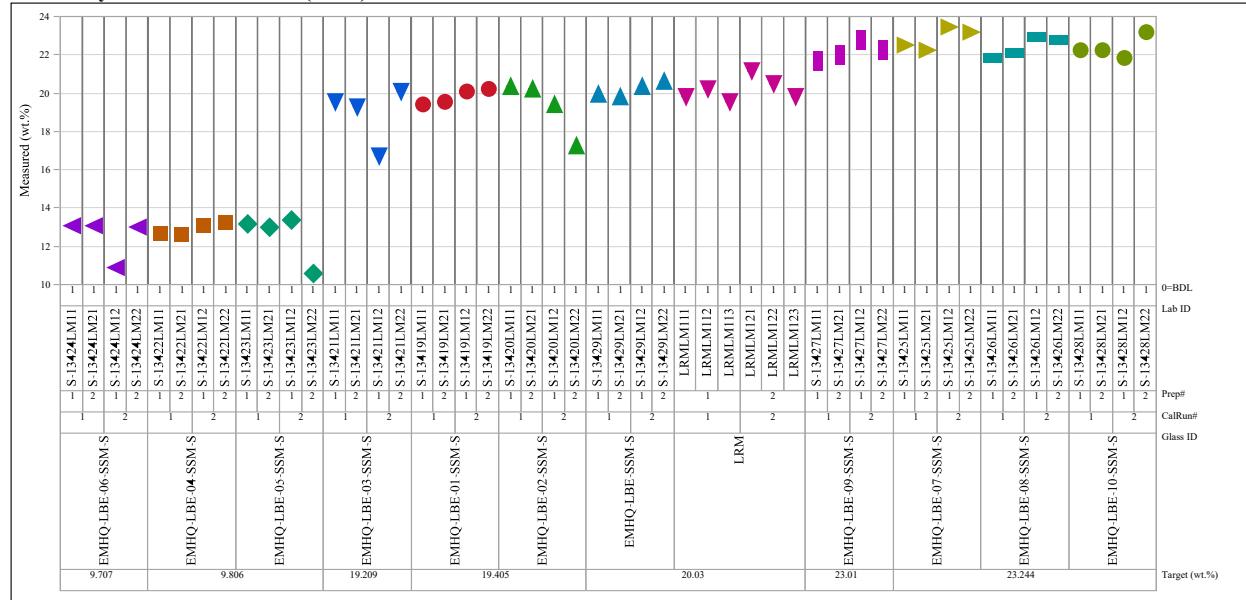
**Exhibit A-1. Plots of Oxide Measurements by Glass Identifier by Target Concentrations
(continued)**

Oxide=MnO (wt%)

Variability Chart for Measured (wt.%)

Oxide=Na₂O (wt%)

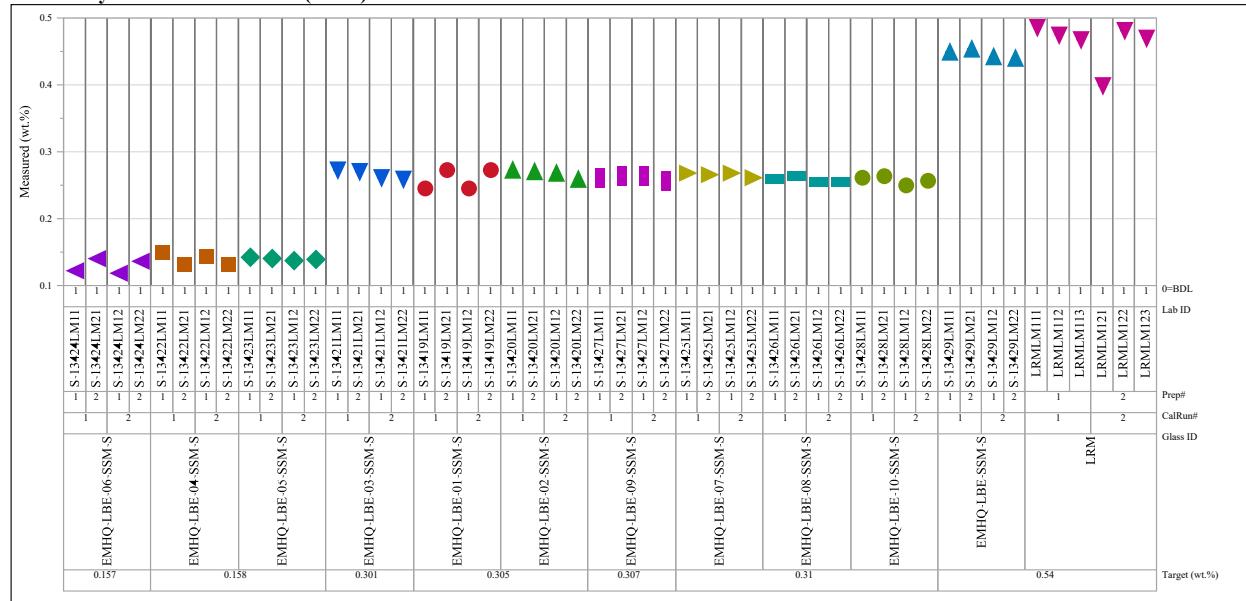
Variability Chart for Measured (wt.%)



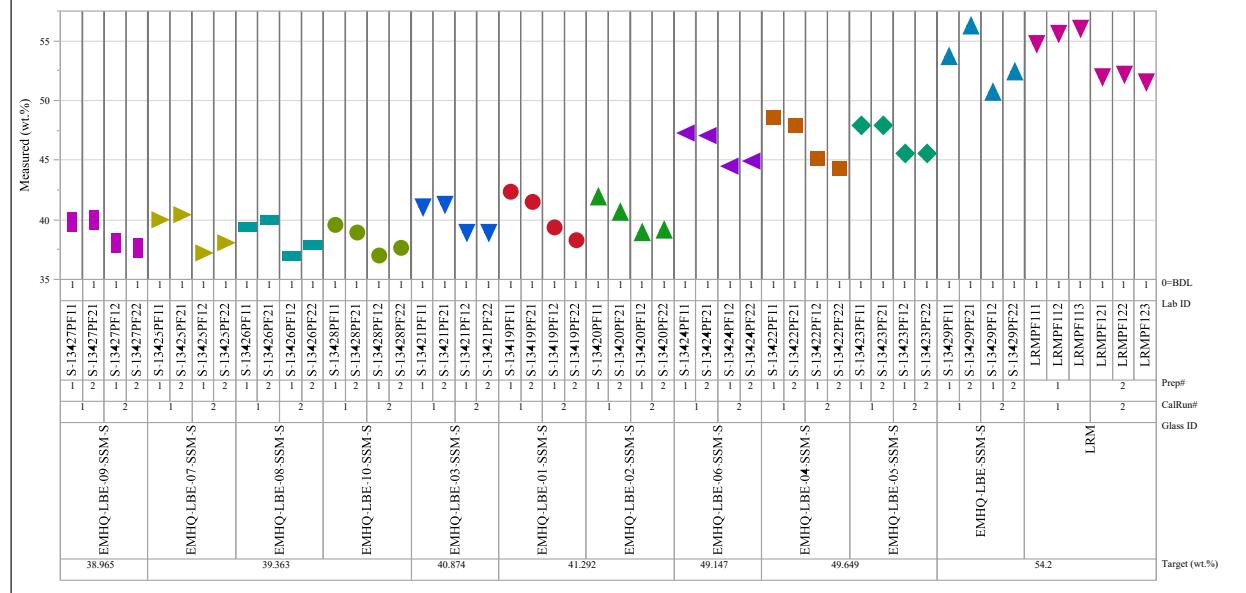
**Exhibit A-1. Plots of Oxide Measurements by Glass Identifier by Target Concentrations
(continued)**

Oxide= P_2O_5 (wt%)

Variability Chart for Measured (wt.%)

Oxide= SiO_2 (wt%)

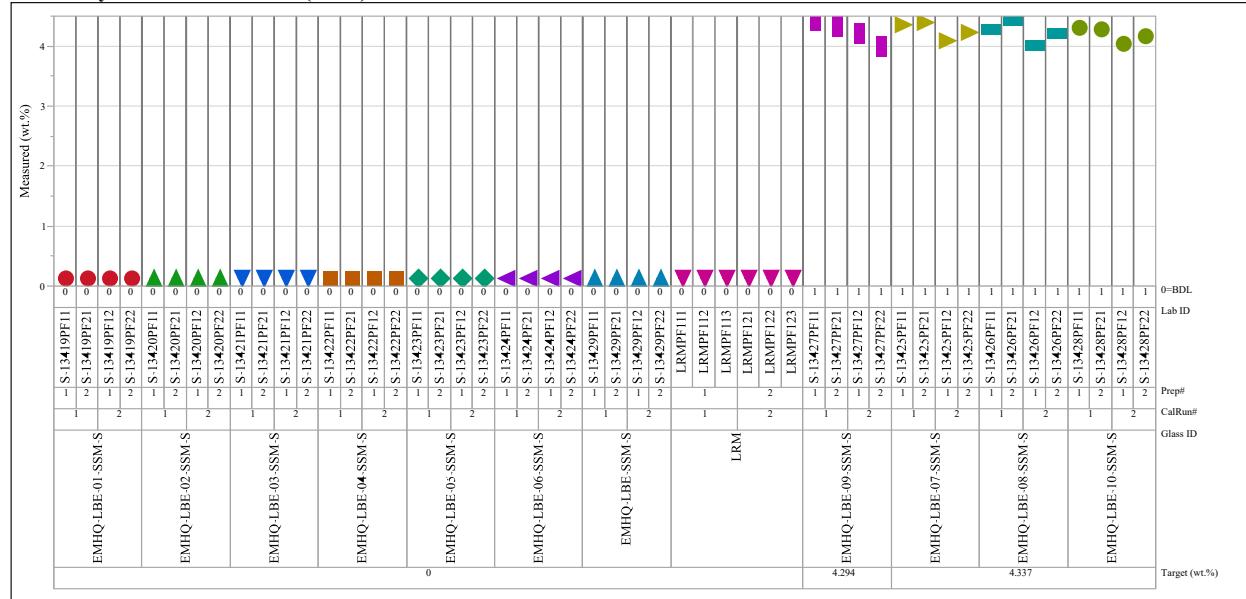
Variability Chart for Measured (wt.%)



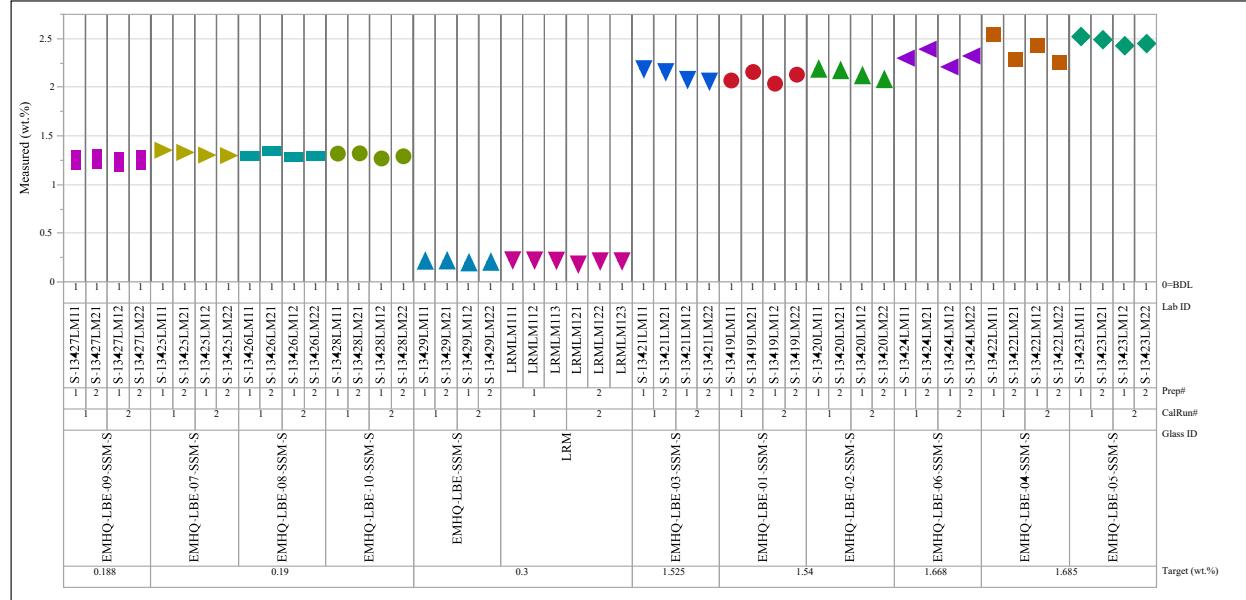
**Exhibit A-1. Plots of Oxide Measurements by Glass Identifier by Target Concentrations
(continued)**

Oxide=SnO₂ (wt%)

Variability Chart for Measured (wt.%)

Oxide=SO₃ (wt%)

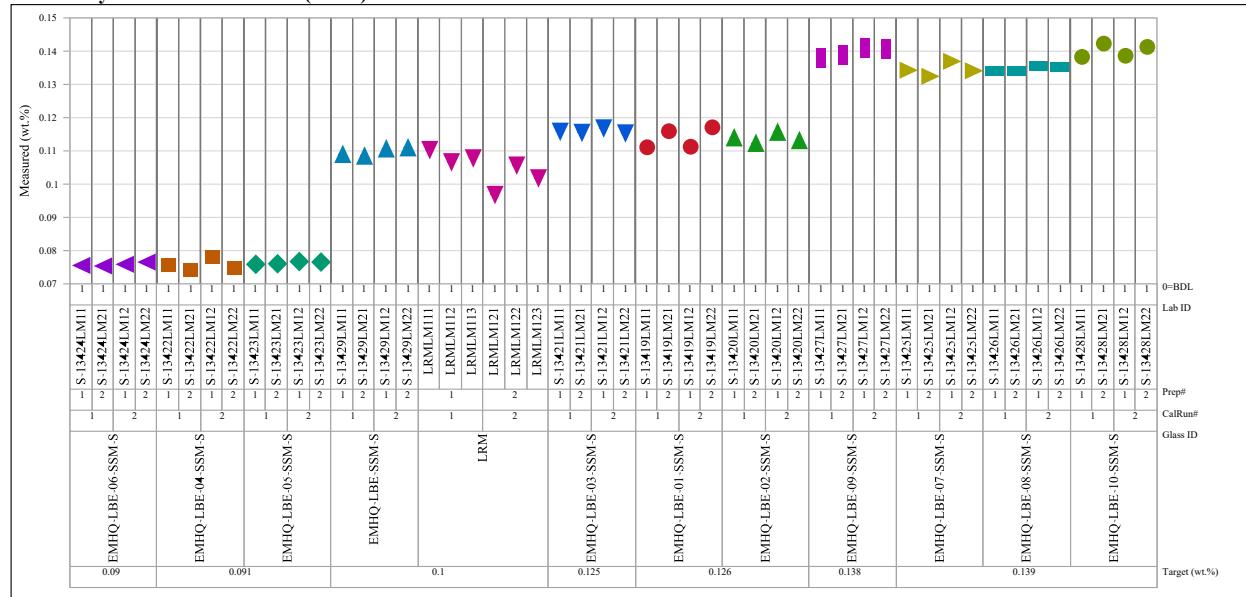
Variability Chart for Measured (wt.%)



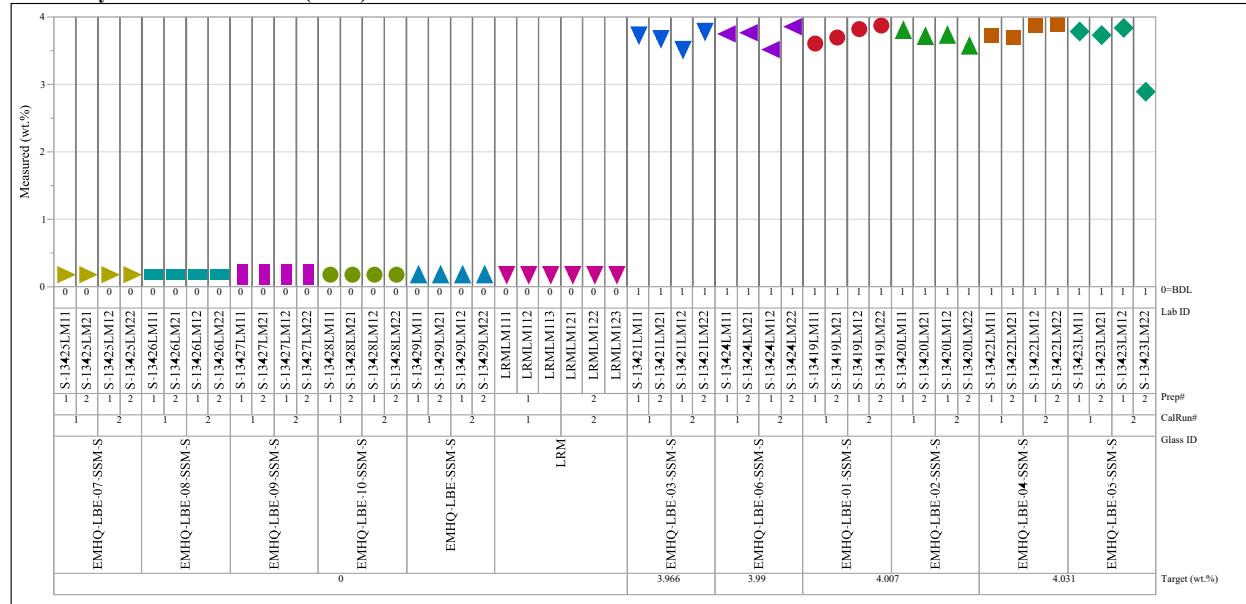
**Exhibit A-1. Plots of Oxide Measurements by Glass Identifier by Target Concentrations
(continued)**

Oxide= TiO_2 (wt%)

Variability Chart for Measured (wt.%)

Oxide= V_2O_5 (wt%)

Variability Chart for Measured (wt.%)



**Exhibit A-1. Plots of Oxide Measurements by Glass Identifier by Target Concentrations
(continued)**

Oxide=ZrO₂ (wt%)

Variability Chart for Measured (wt.%)

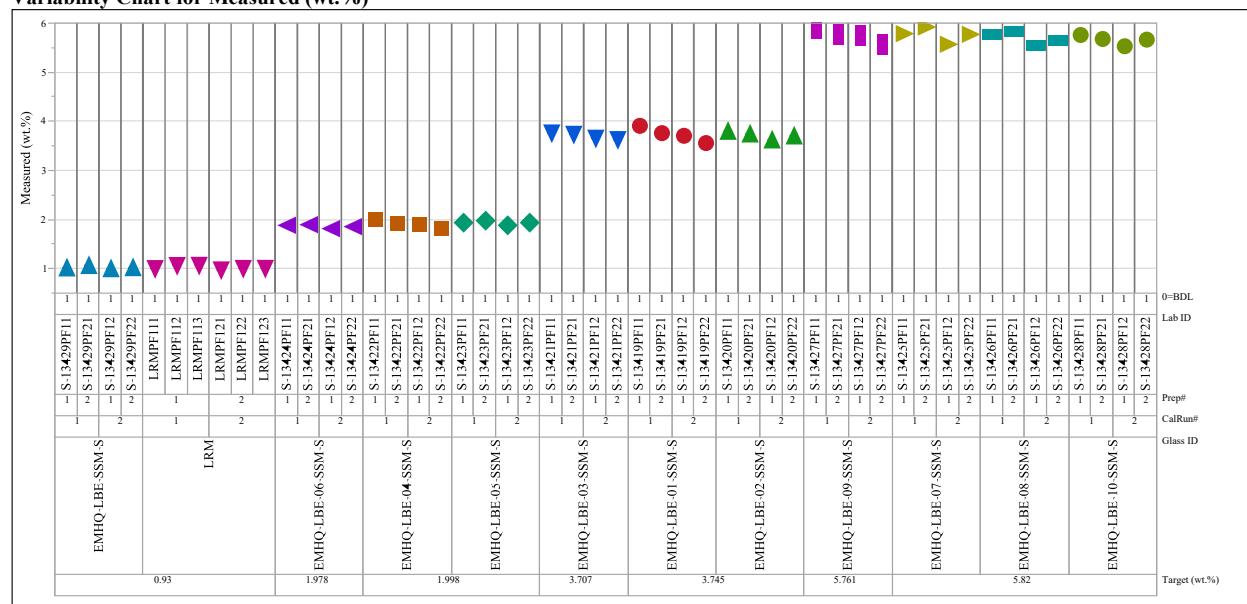
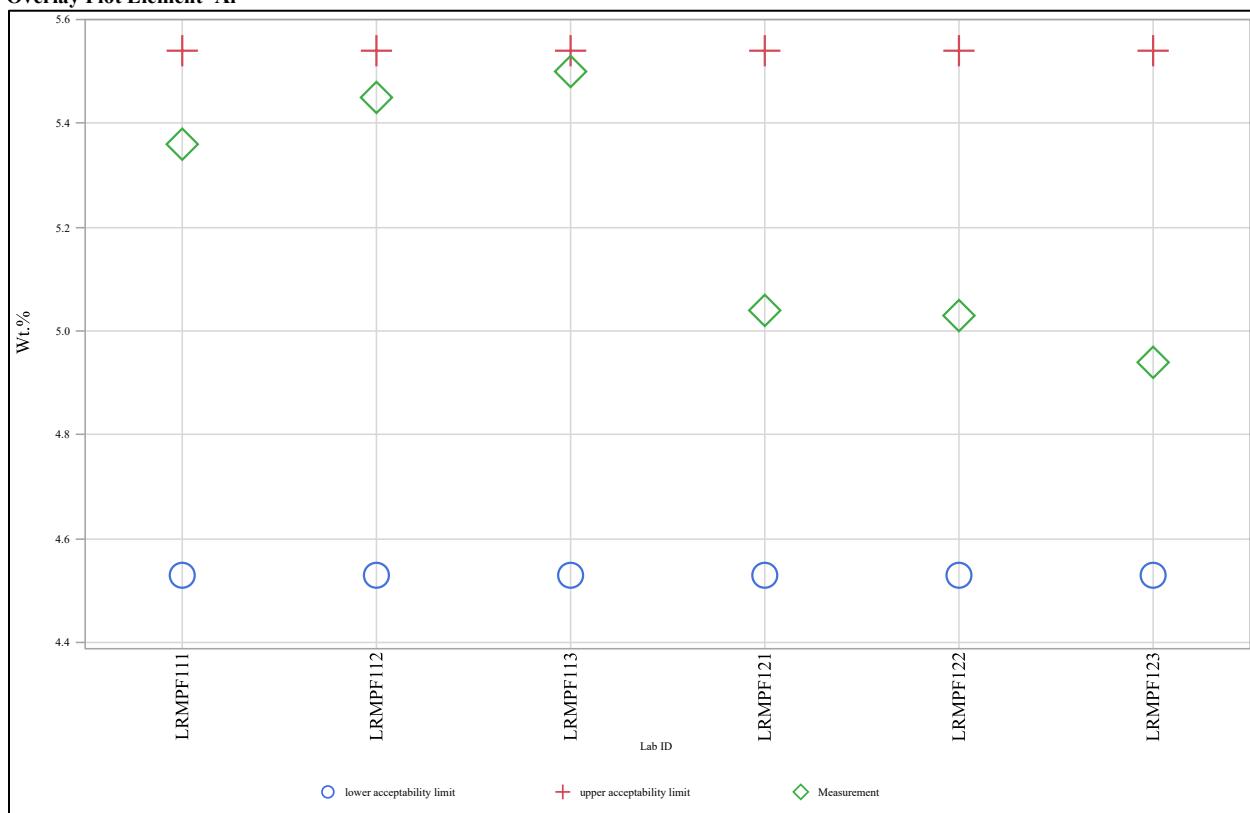


Exhibit A-2. Acceptability Evaluation for Measurements of the LRM Glass

Overlay Plot Element=A1



Overlay Plot Element=B

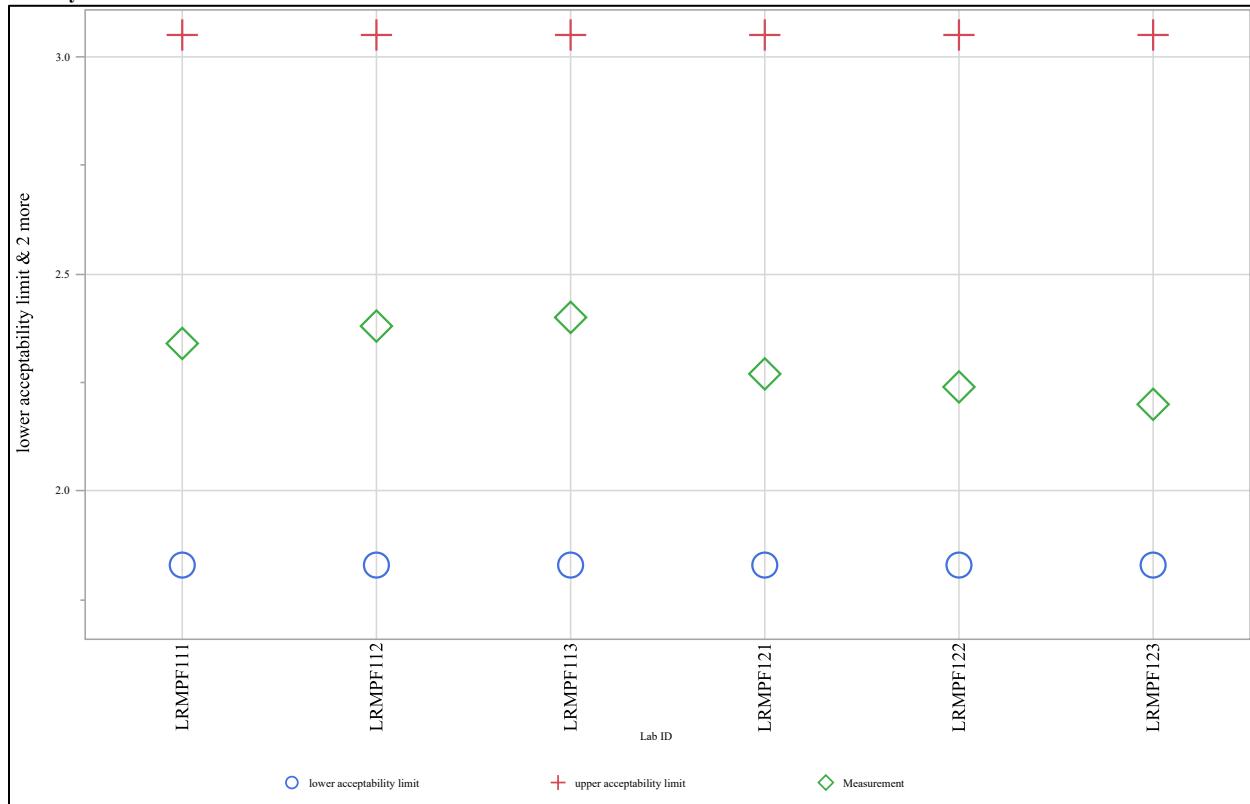
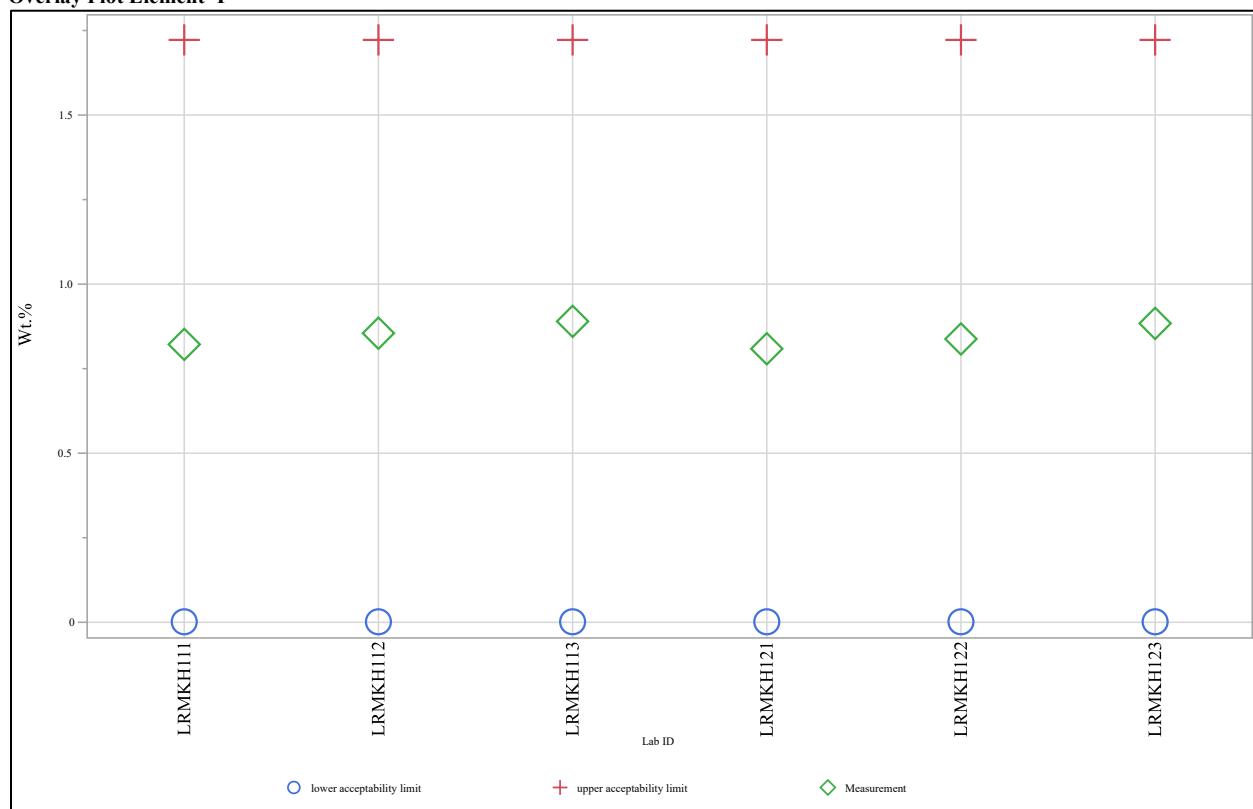


Exhibit A-2. Acceptability Evaluation for Measurements of the LRM Glass (continued)Overlay Plot Element=F⁻

Overlay Plot Element=Fe

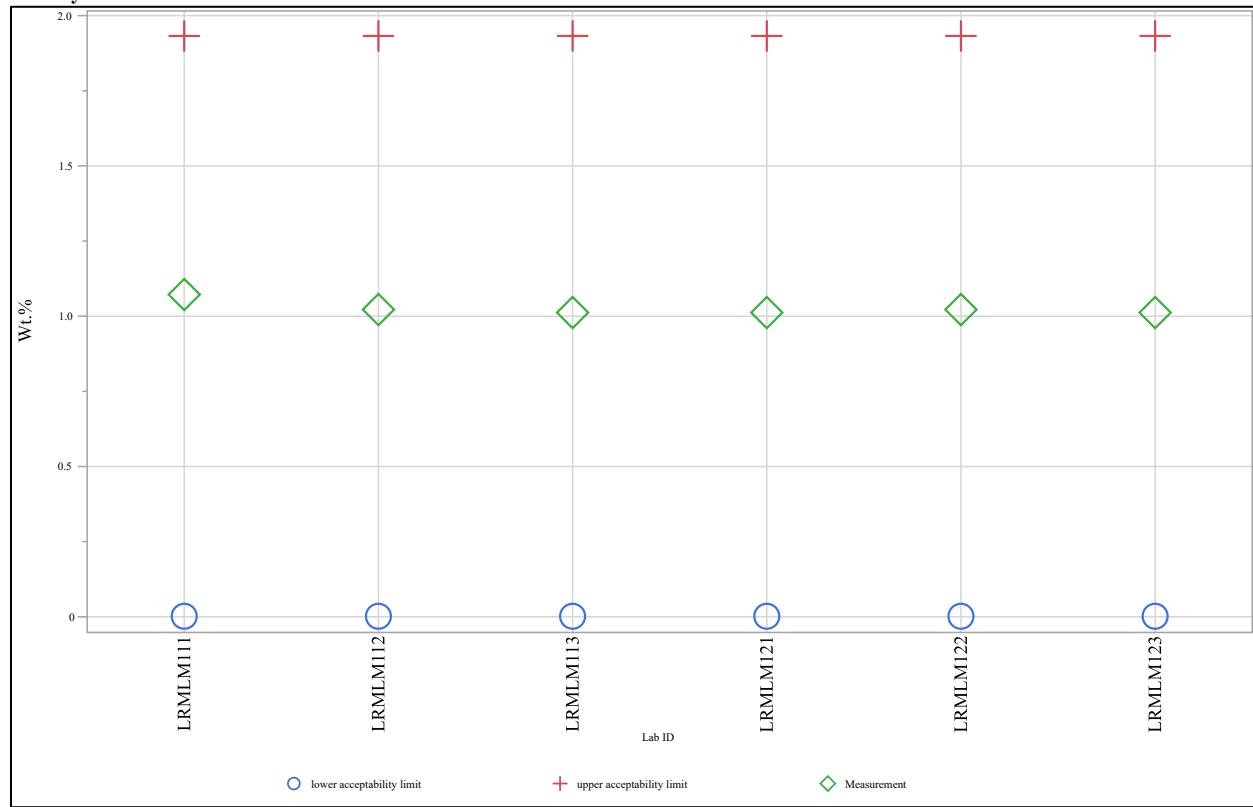
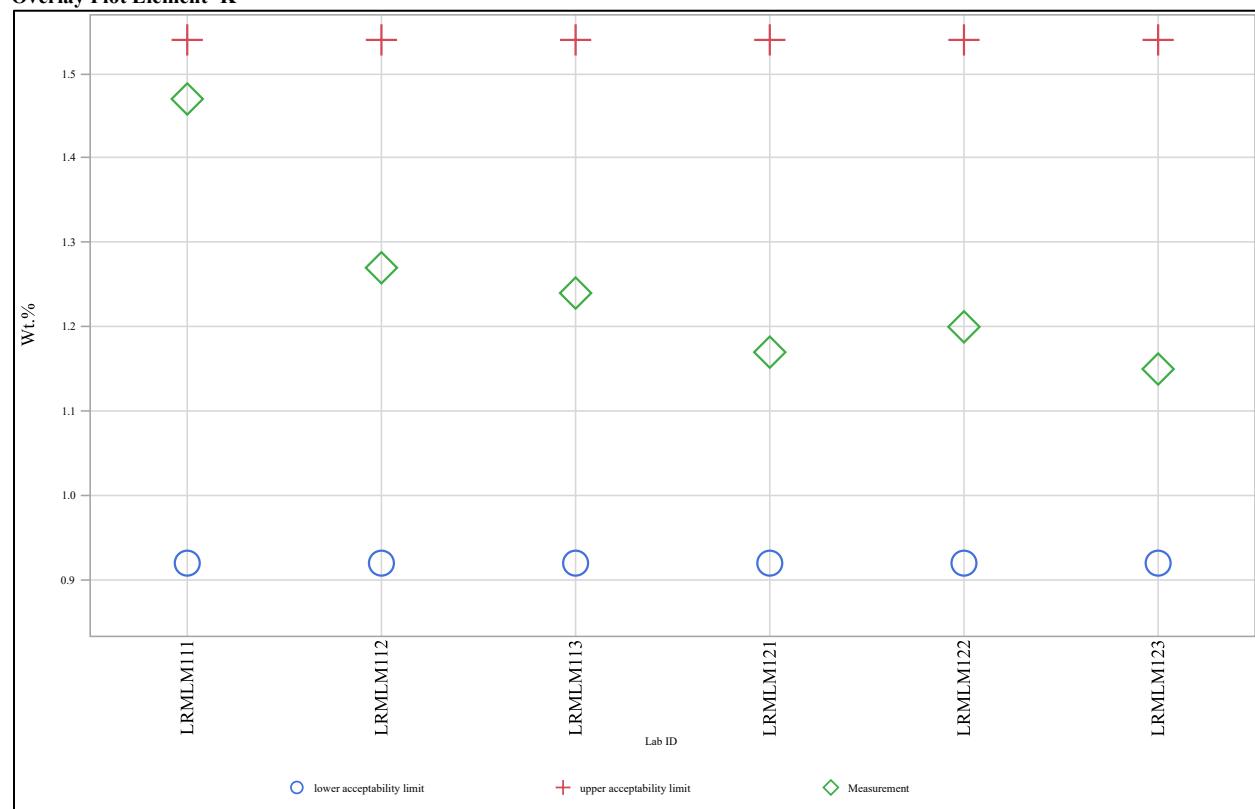


Exhibit A-2. Acceptability Evaluation for Measurements of the LRM Glass (continued)

Overlay Plot Element=K



Overlay Plot Element=Na

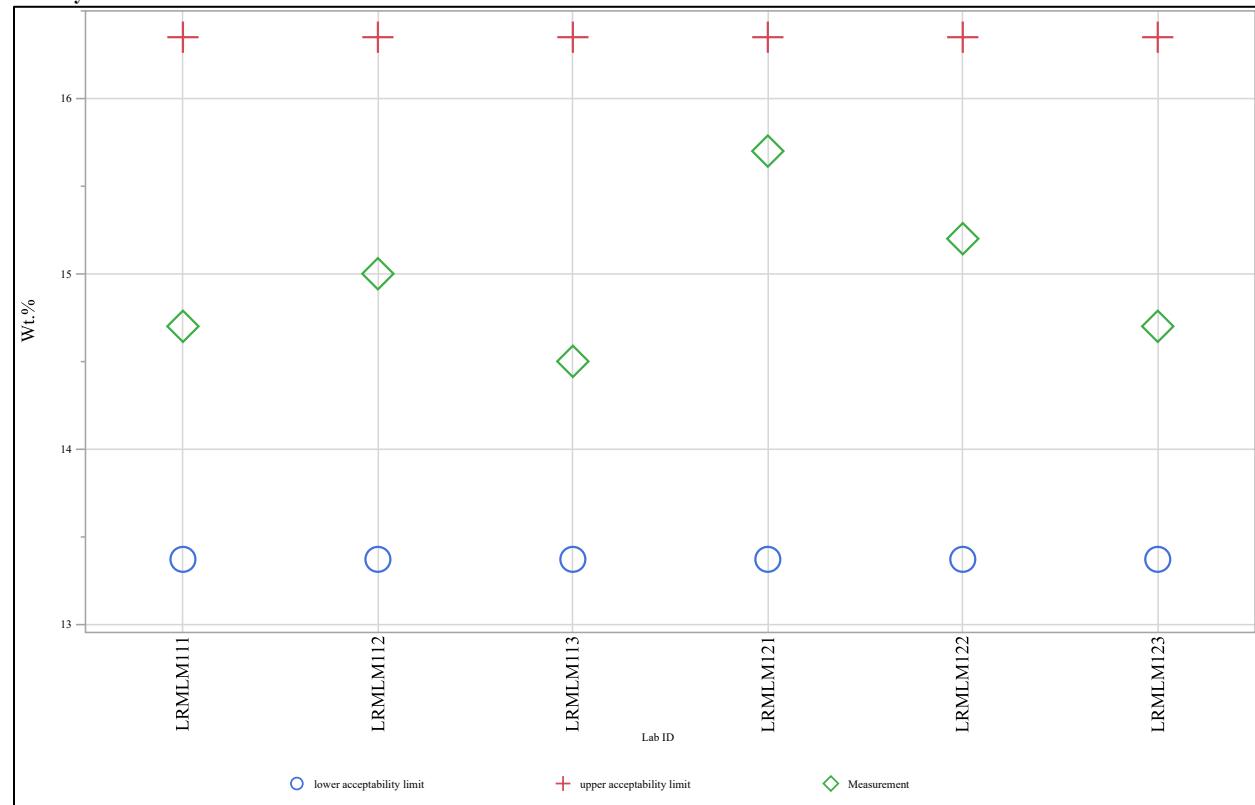
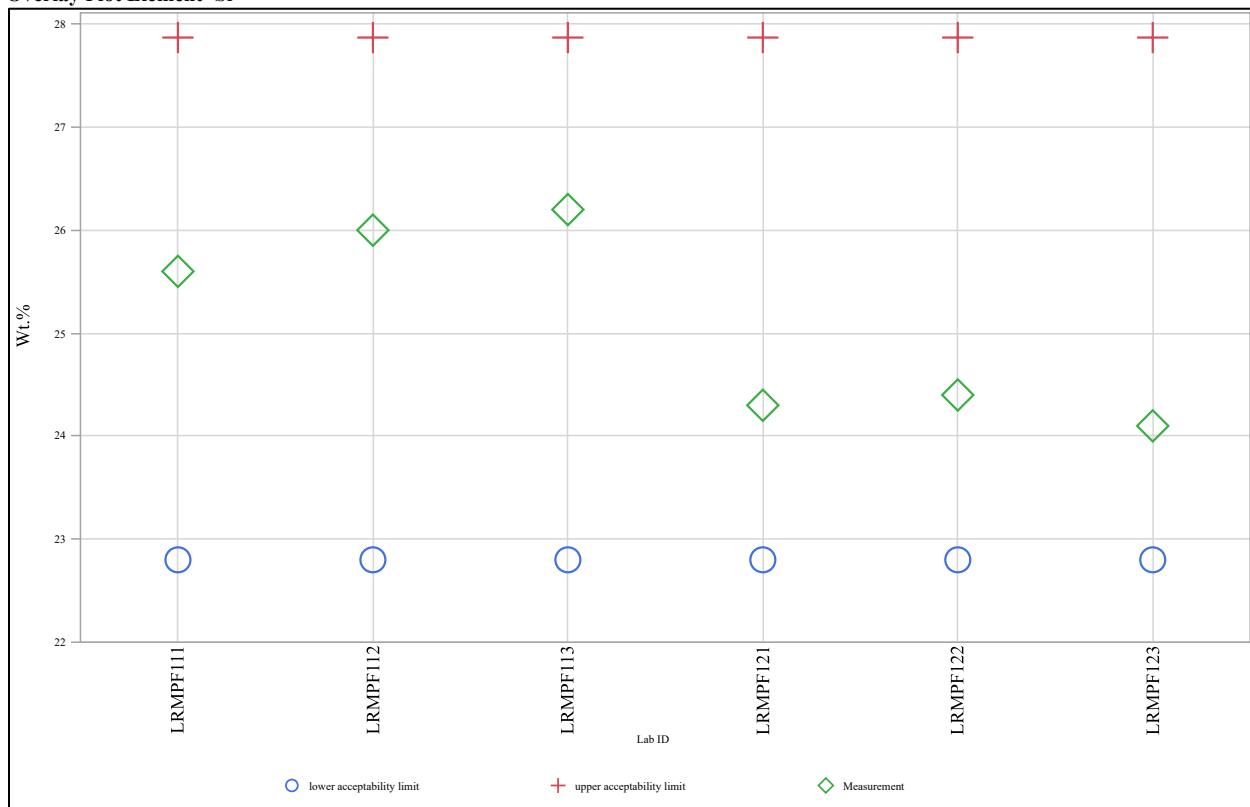


Exhibit A-2. Acceptability Evaluation for Measurements of the LRM Glass (continued)

Overlay Plot Element=Si



Overlay Plot Element=Zr

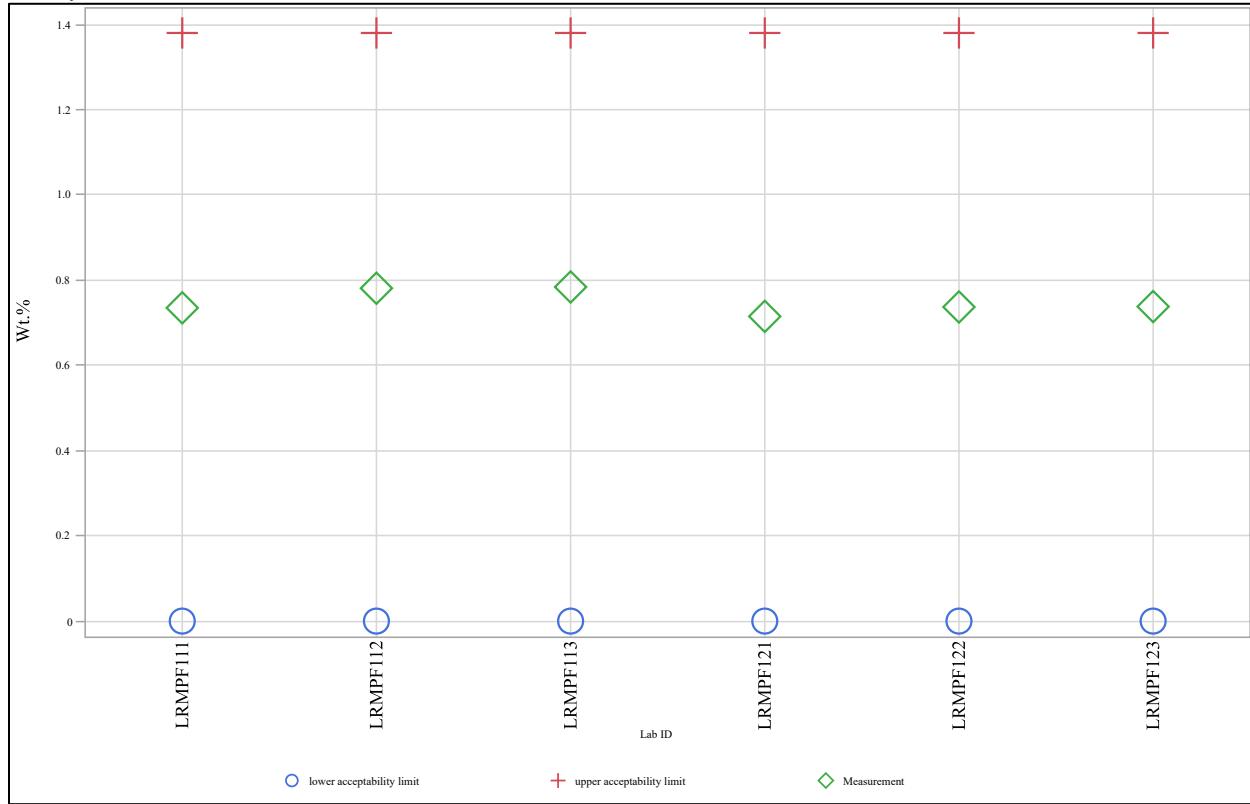


Exhibit A-3. Measured versus Target Concentrations by Glass ID by Oxide

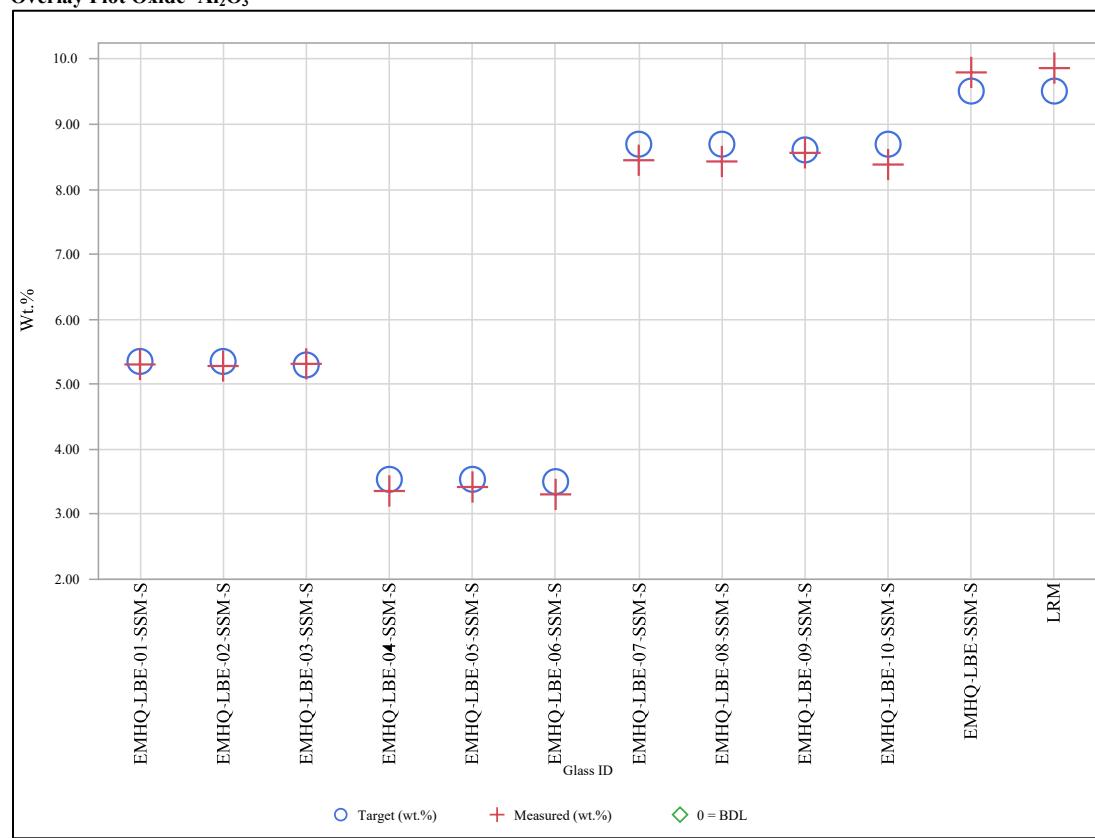
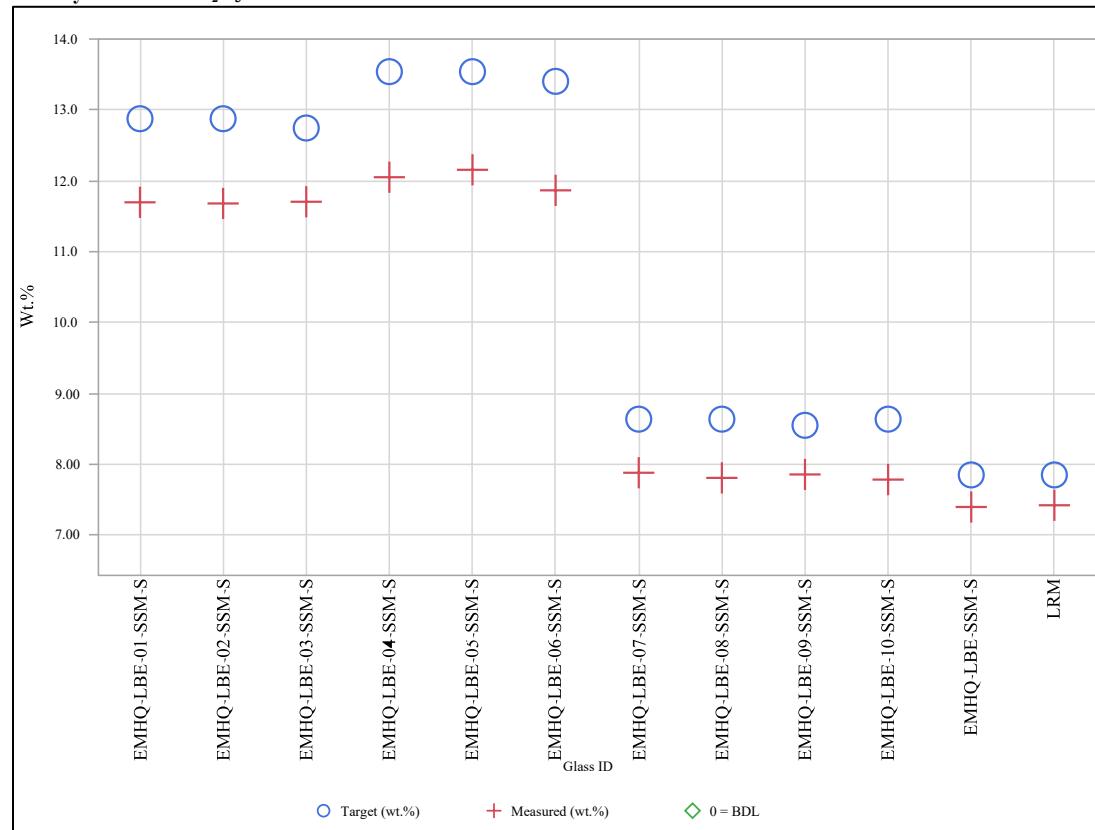
Overlay Plot Oxide= Al_2O_3 Overlay Plot Oxide= B_2O_3 

Exhibit A-3. Measured versus Target Concentrations by Glass ID by Oxide (continued)

Overlay Plot Oxide=CaO

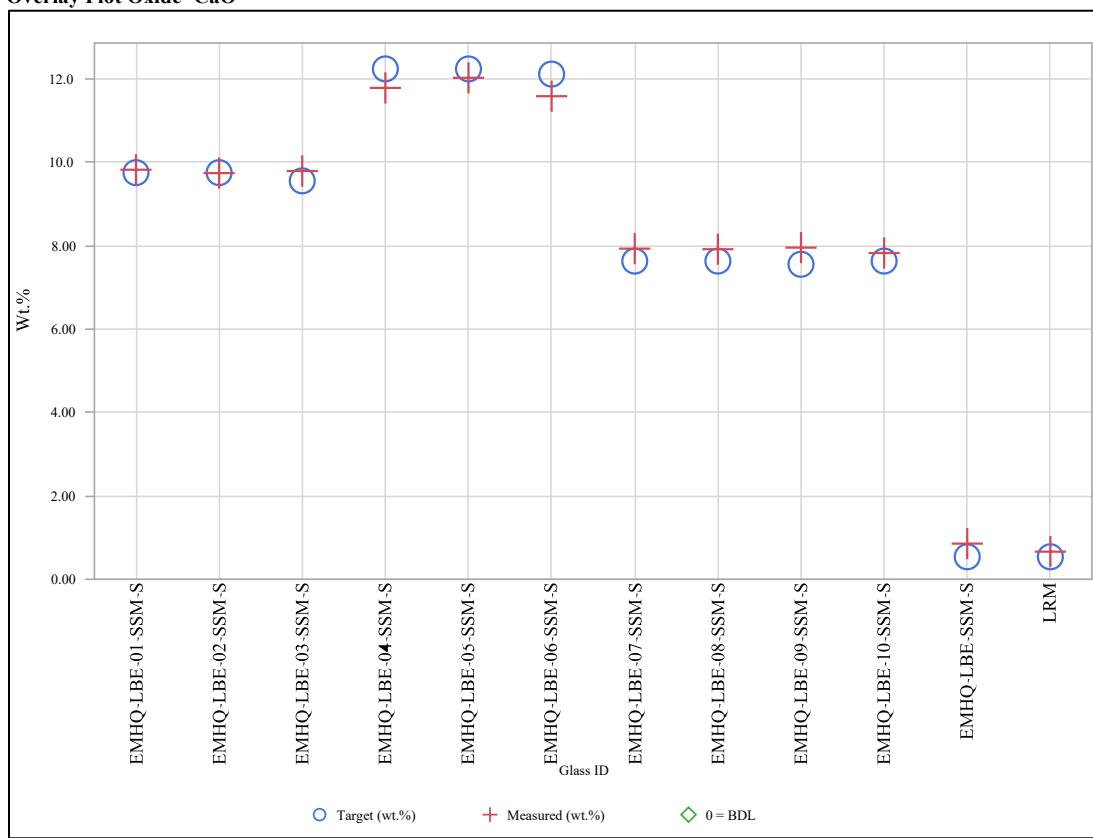
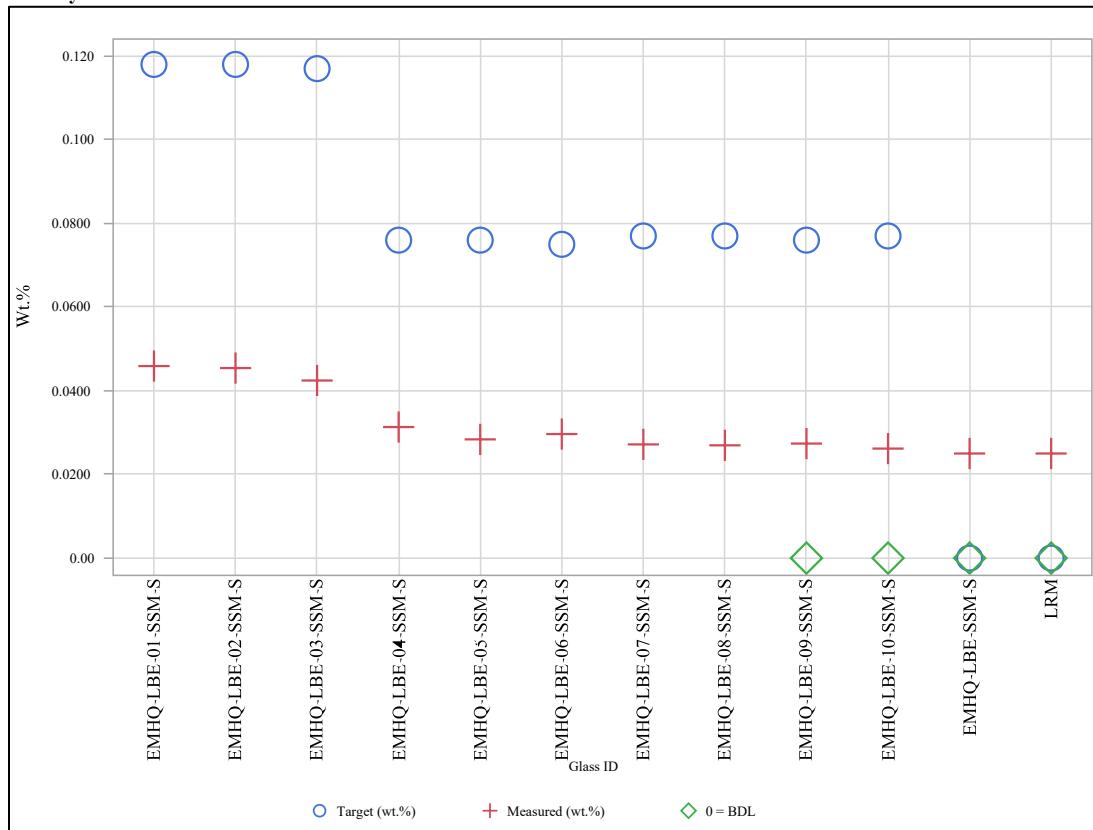
Overlay Plot Oxide=Cl⁻

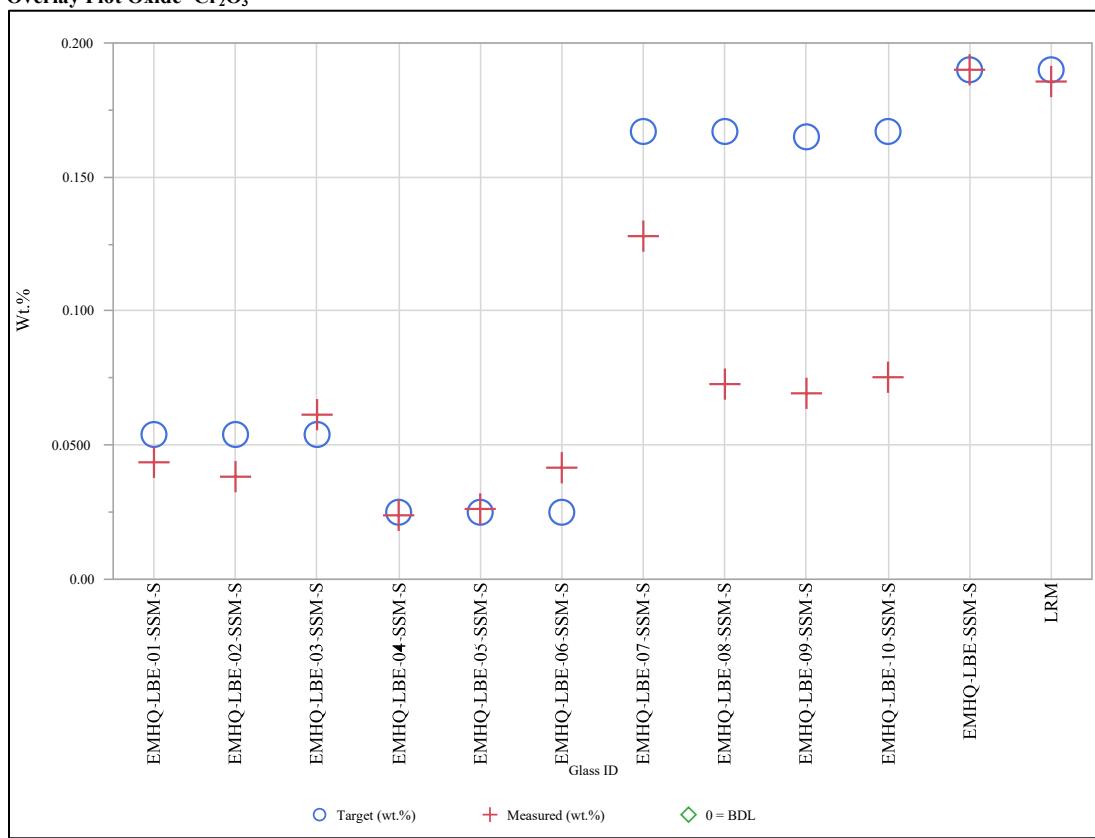
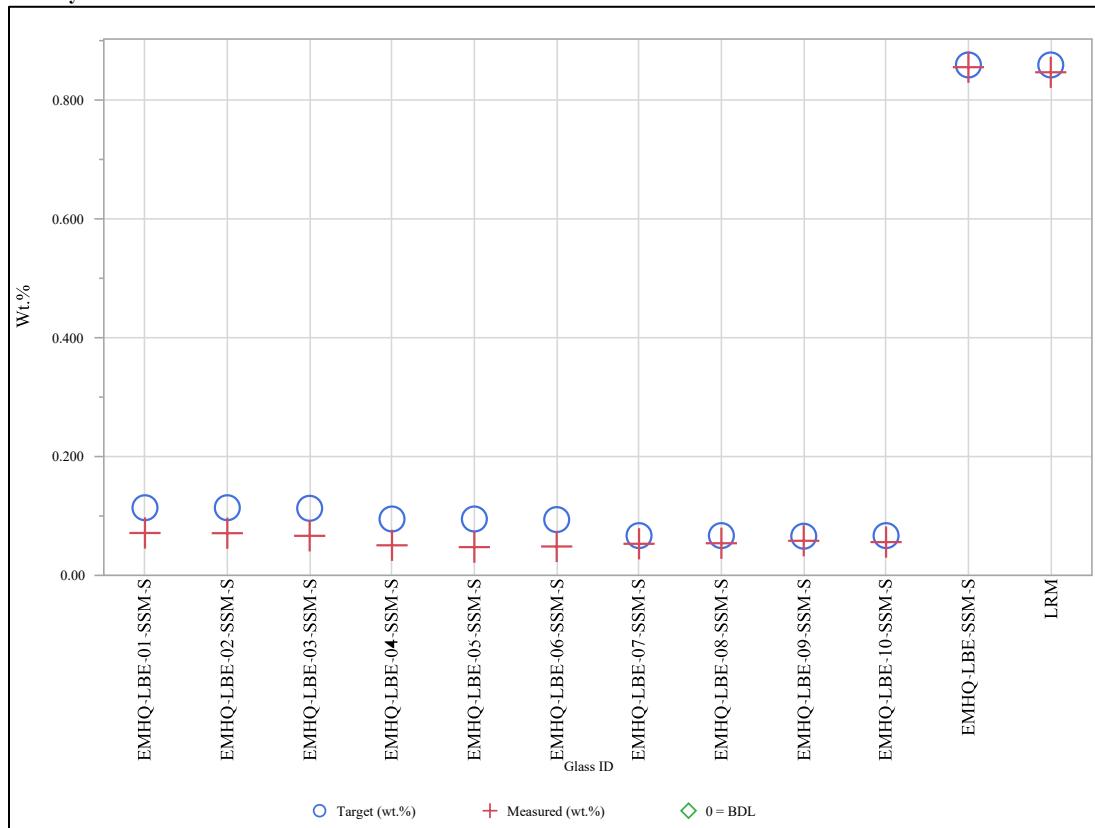
Exhibit A-3. Measured versus Target Concentrations by Glass ID by Oxide (continued)Overlay Plot Oxide= Cr_2O_3 Overlay Plot Oxide=F⁻

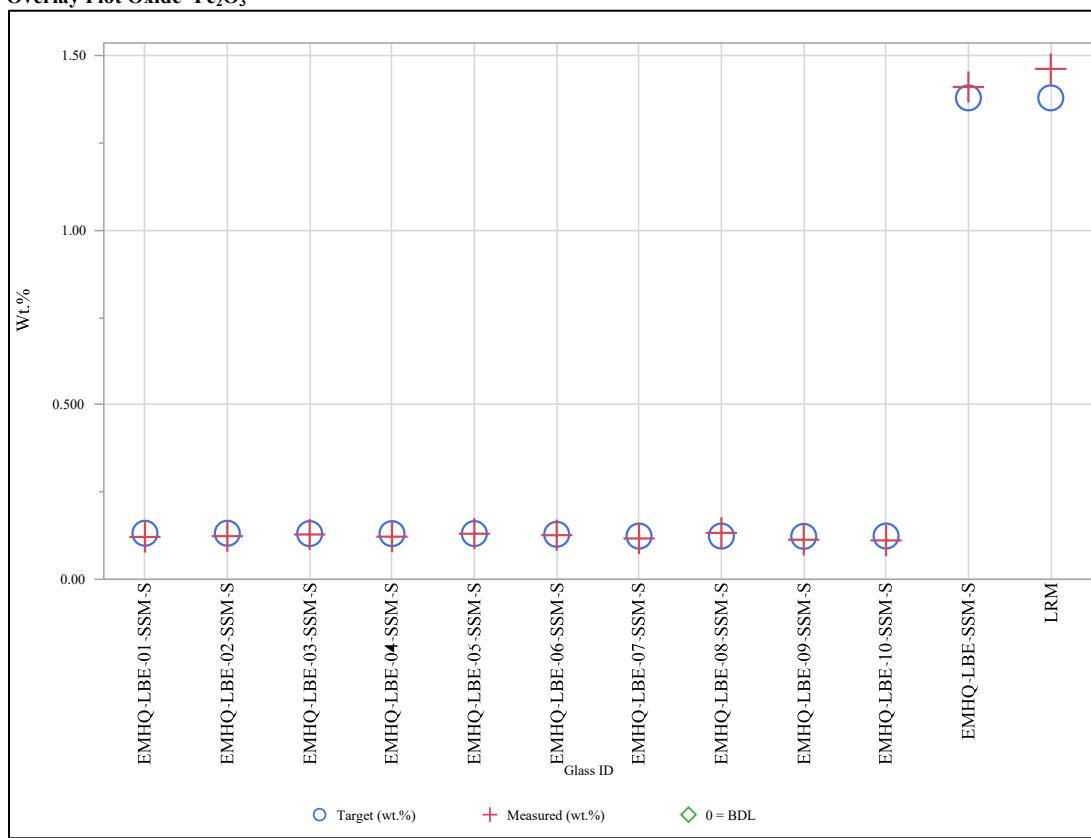
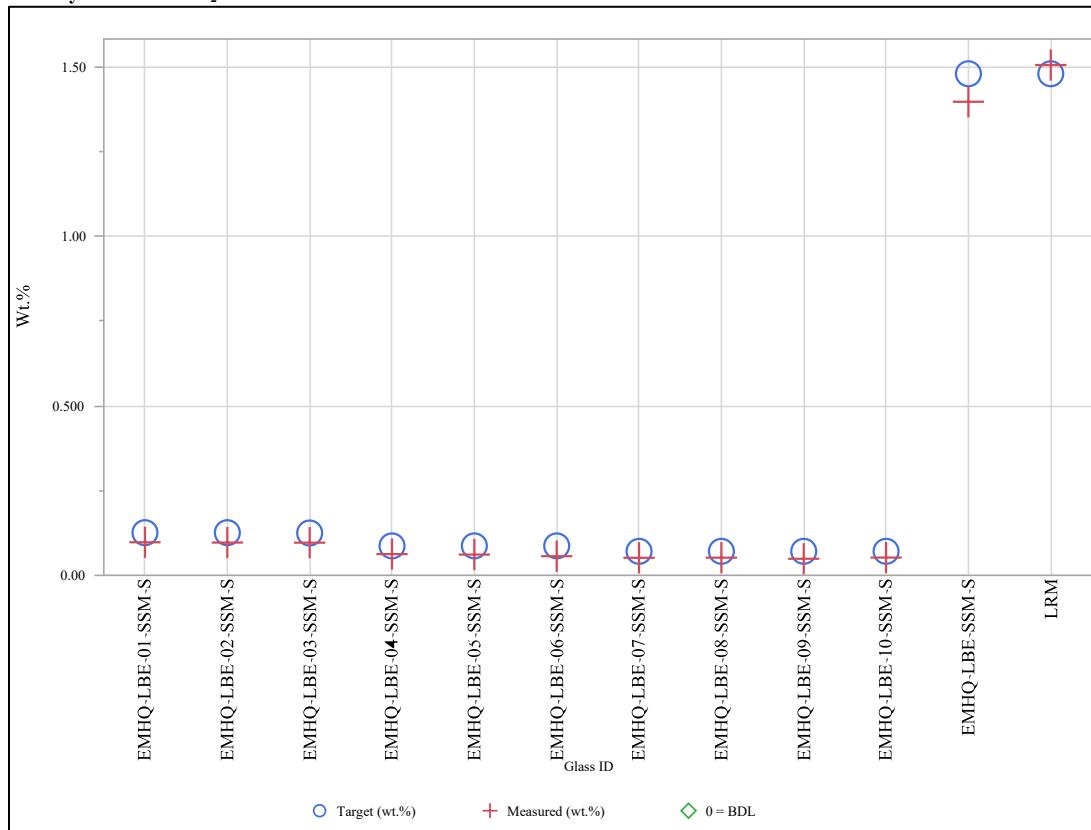
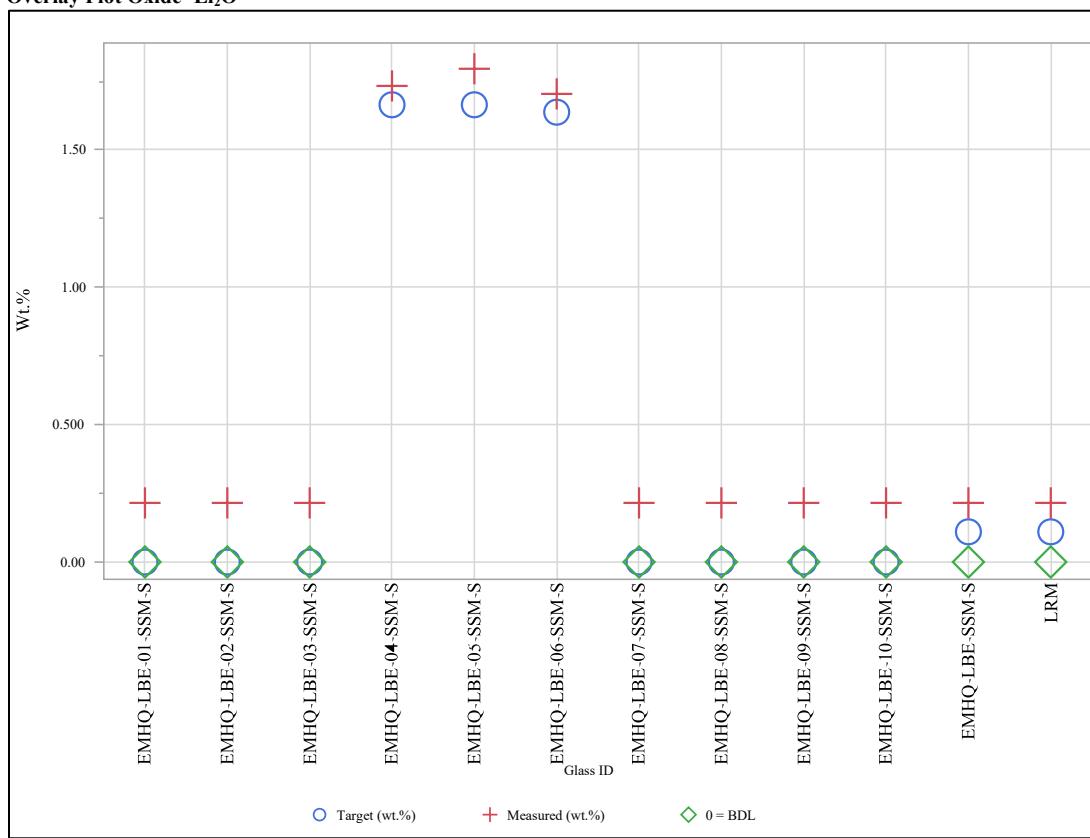
Exhibit A-3. Measured versus Target Concentrations by Glass ID by Oxide (continued)Overlay Plot Oxide= Fe_2O_3 Overlay Plot Oxide= K_2O 

Exhibit A-3. Measured versus Target Concentrations by Glass ID by Oxide (continued)Overlay Plot Oxide= Li_2O 

Overlay Plot Oxide=MgO

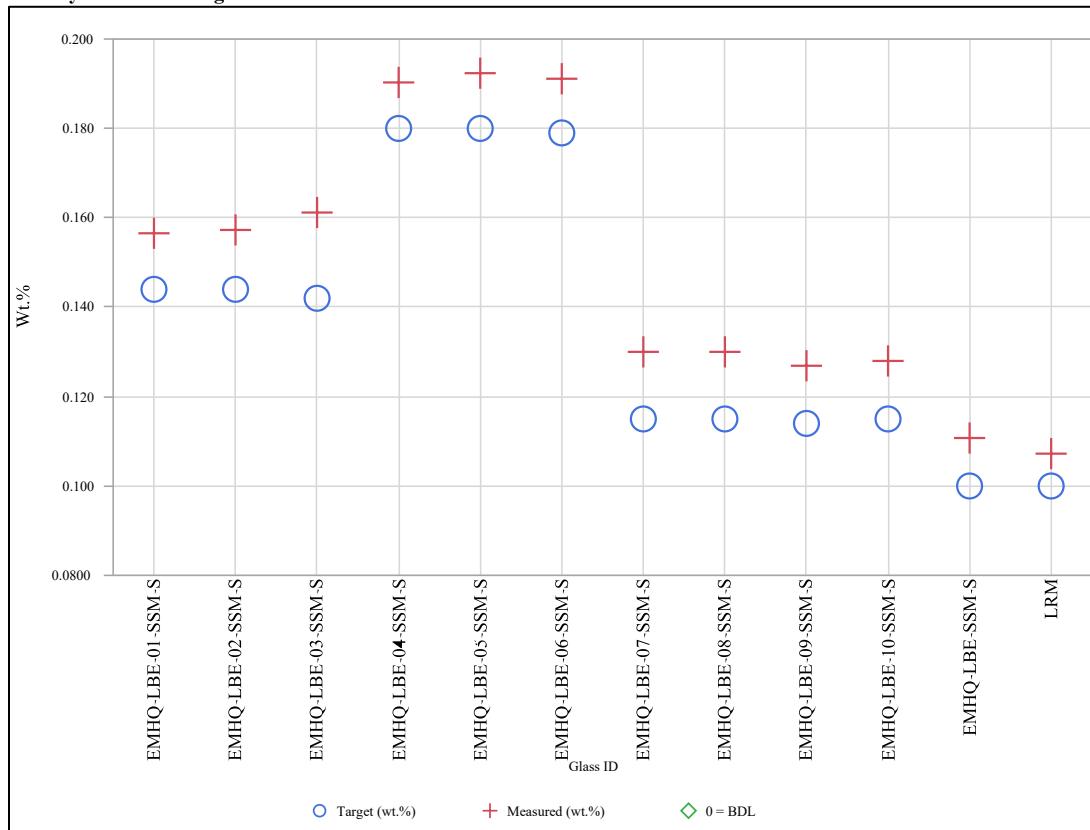


Exhibit A-3. Measured versus Target Concentrations by Glass ID by Oxide (continued)

Overlay Plot Oxide=MnO

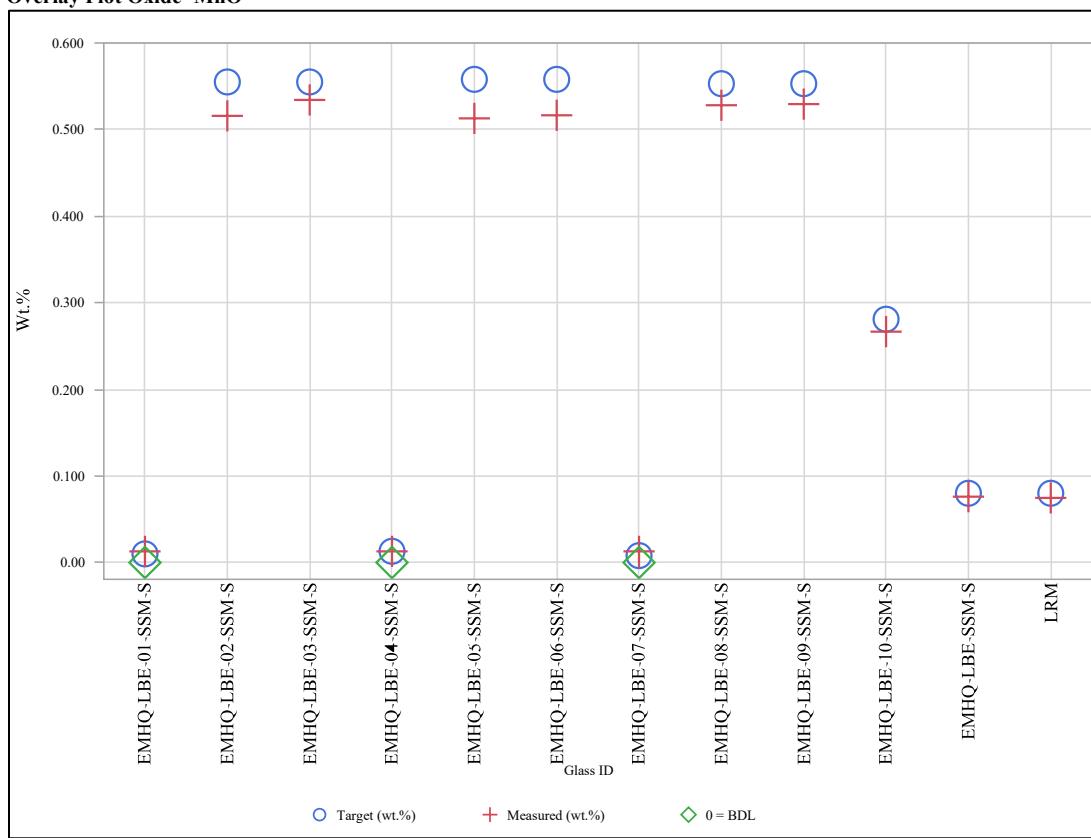
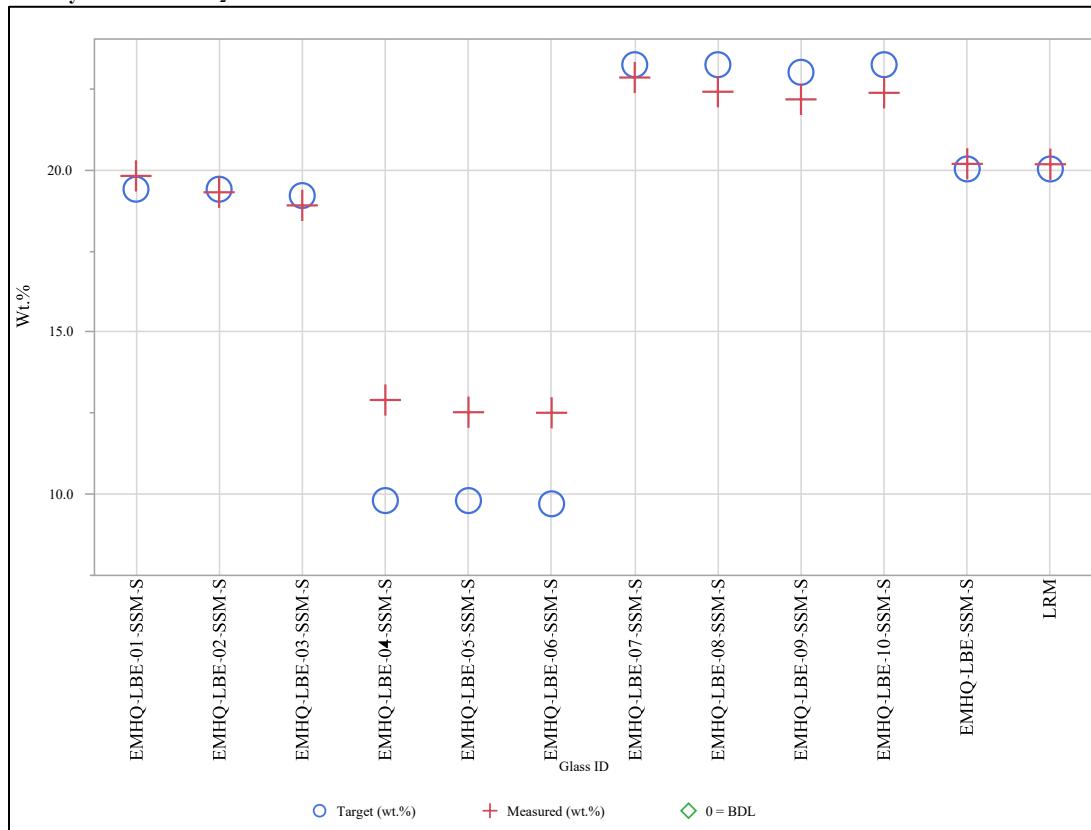
Overlay Plot Oxide=Na₂O

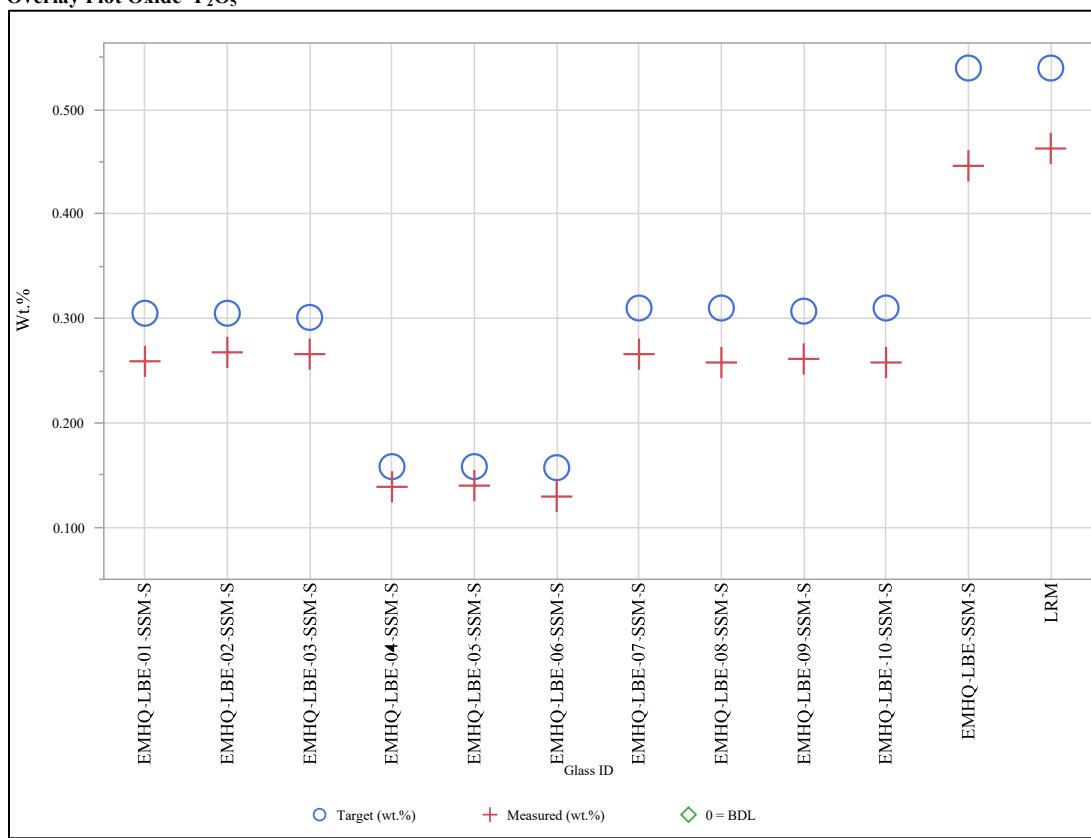
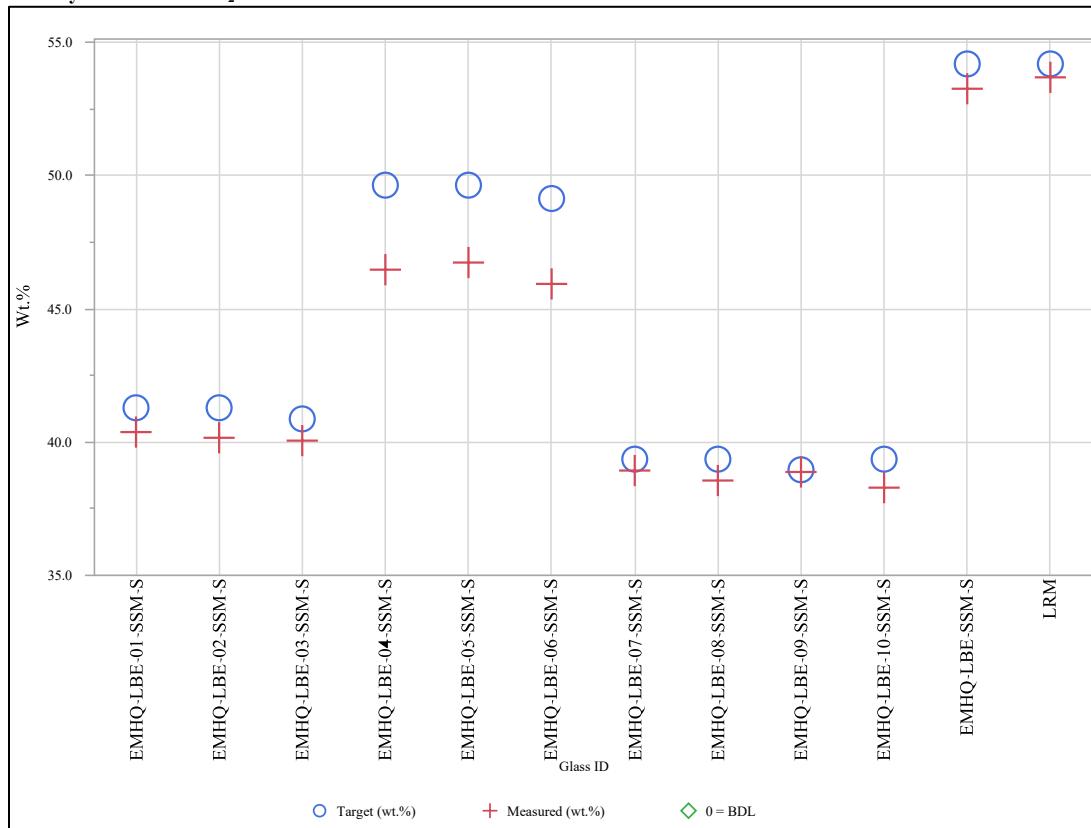
Exhibit A-3. Measured versus Target Concentrations by Glass ID by Oxide (continued)Overlay Plot Oxide= P_2O_5 Overlay Plot Oxide= SiO_2 

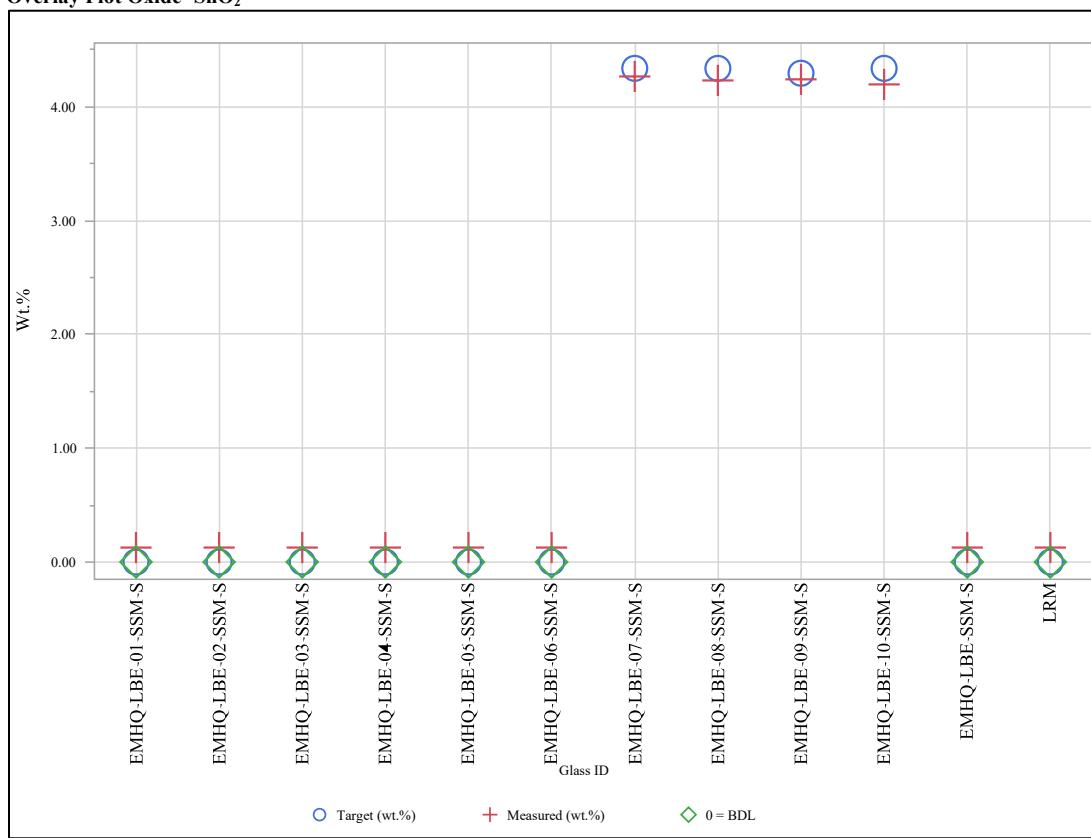
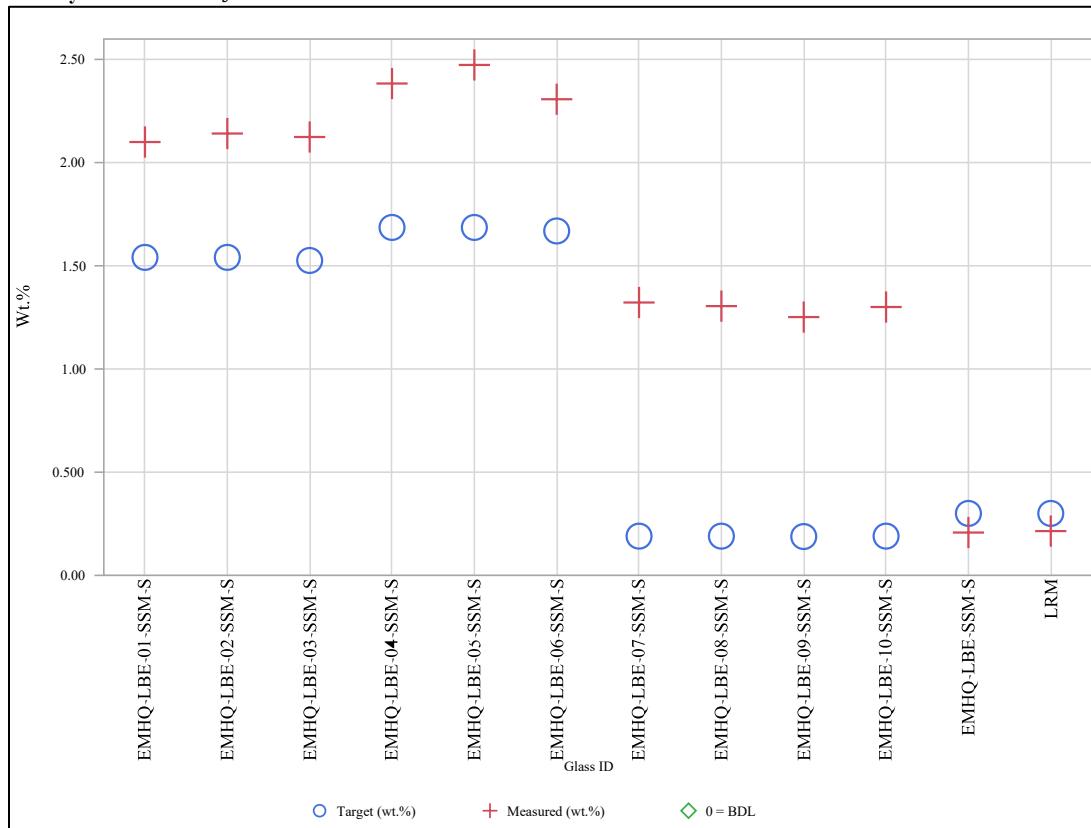
Exhibit A-3. Measured versus Target Concentrations by Glass ID by Oxide (continued)Overlay Plot Oxide=SnO₂Overlay Plot Oxide=SO₃

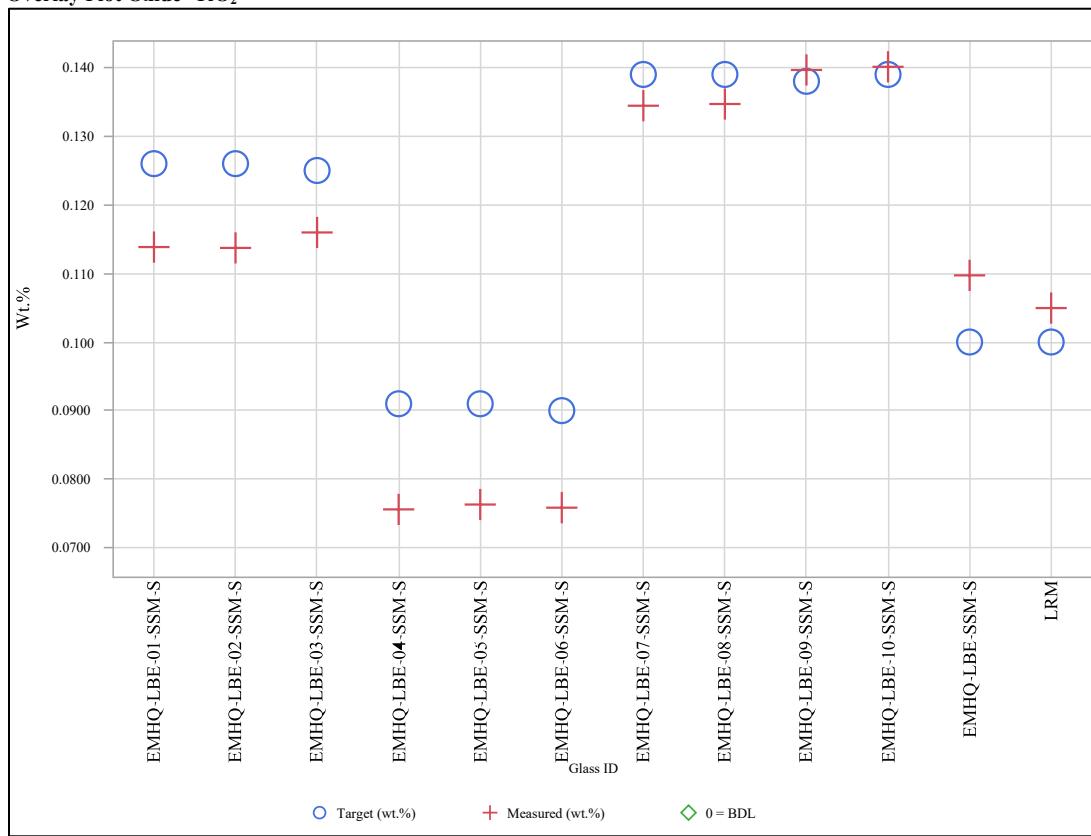
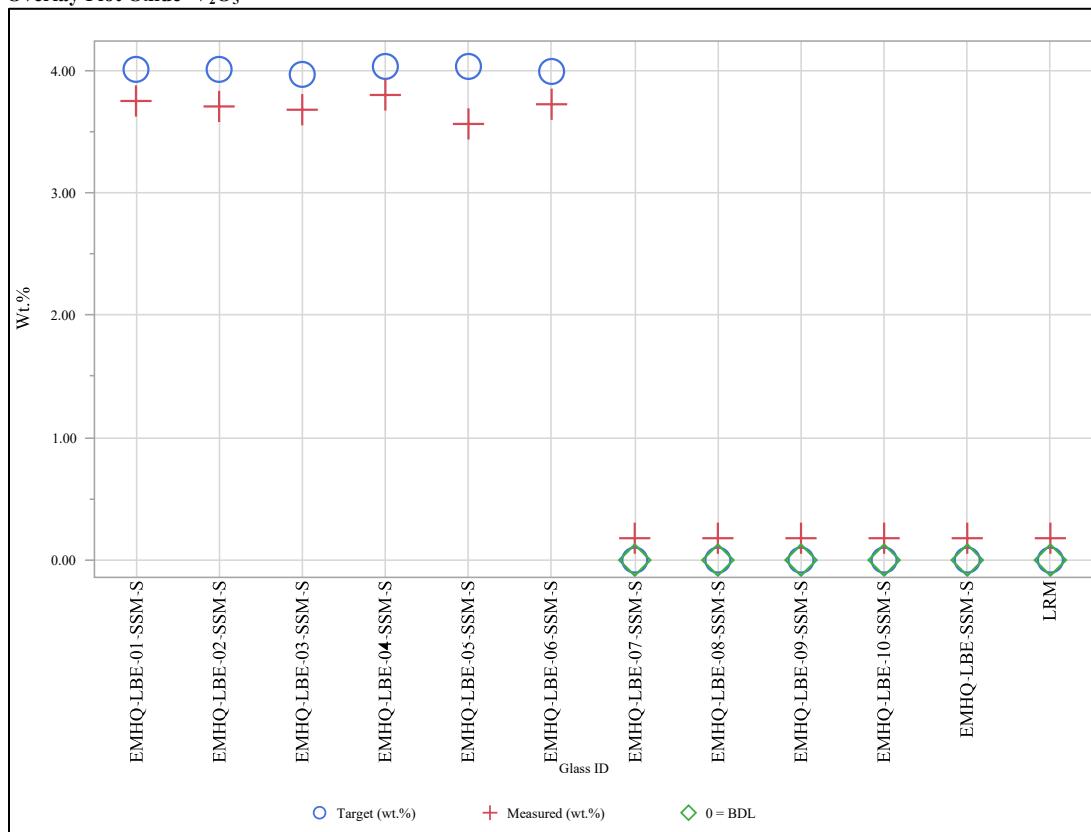
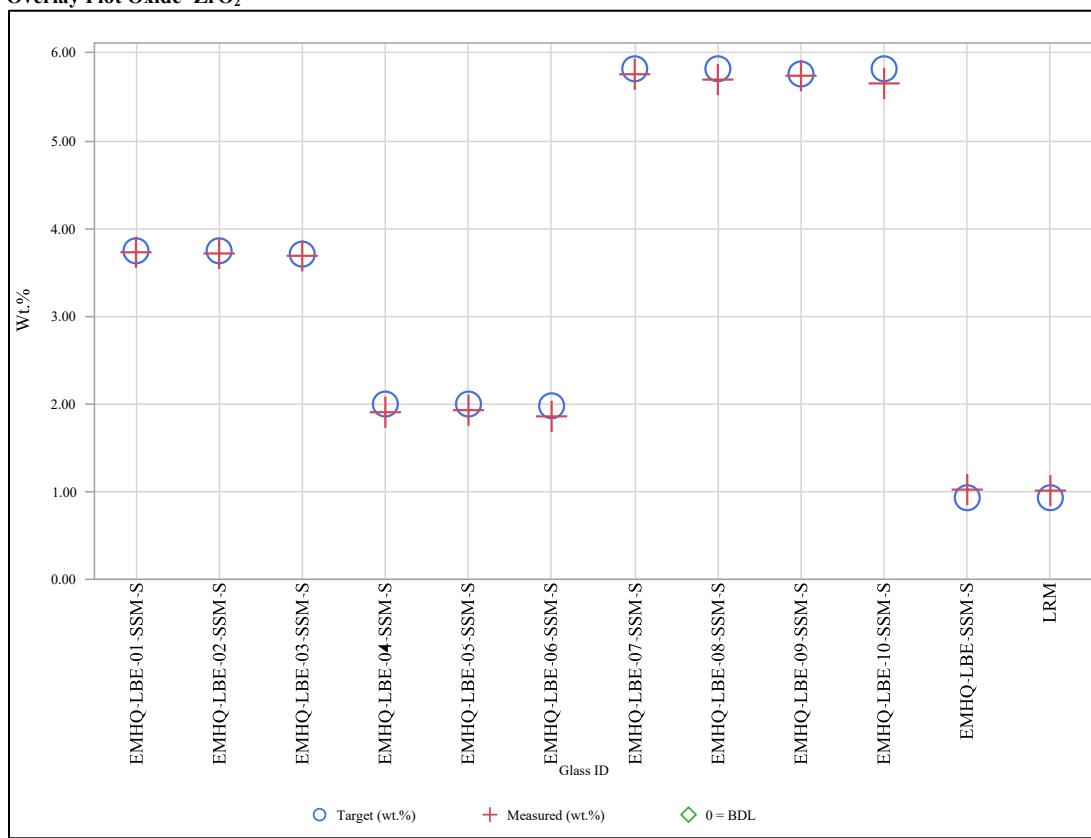
Exhibit A-3. Measured versus Target Concentrations by Glass ID by Oxide (continued)Overlay Plot Oxide= TiO_2 Overlay Plot Oxide= V_2O_5 

Exhibit A-3. Measured versus Target Concentrations by Glass ID by Oxide (continued)Overlay Plot Oxide=ZrO₂

Overlay Plot Oxide=Sum of Oxides

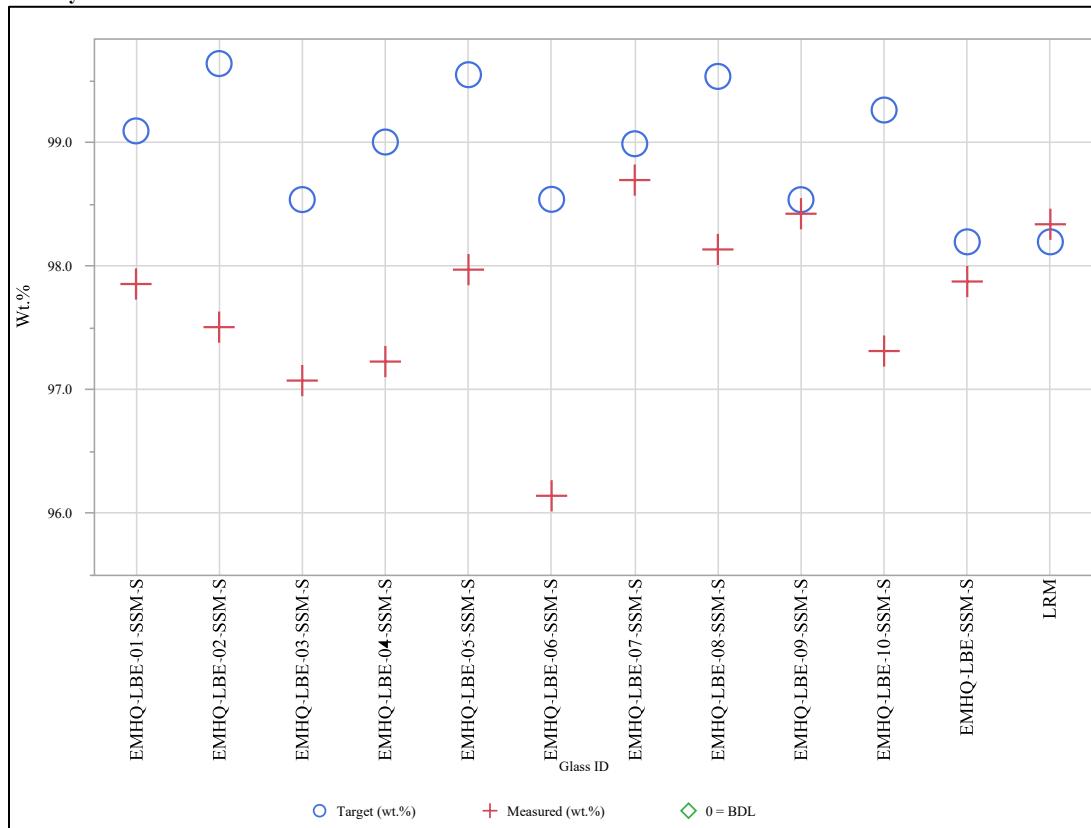
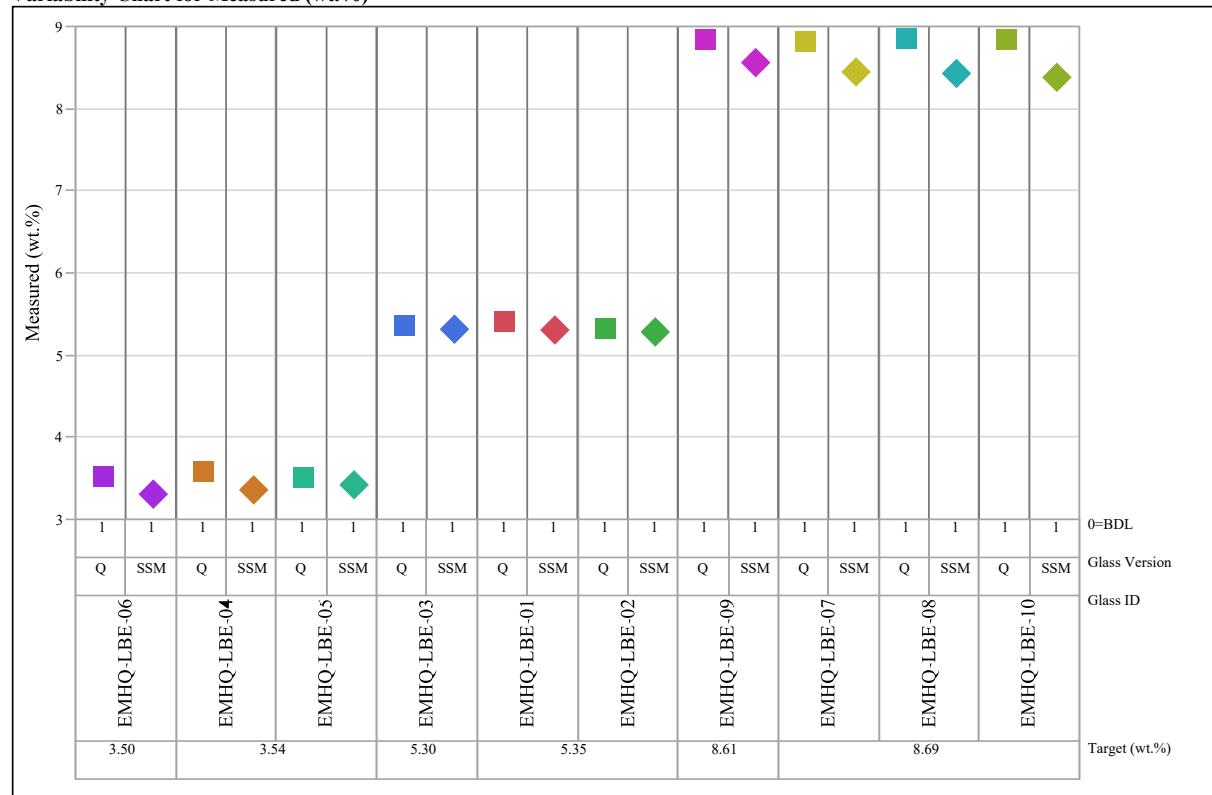


Exhibit A-4. Comparisons of the Measured Compositions of the Quenched and SSM Versions of the Study Glasses

Oxide= Al_2O_3

Variability Chart for Measured (wt.%)

Oxide= B_2O_3

Variability Chart for Measured (wt.%)

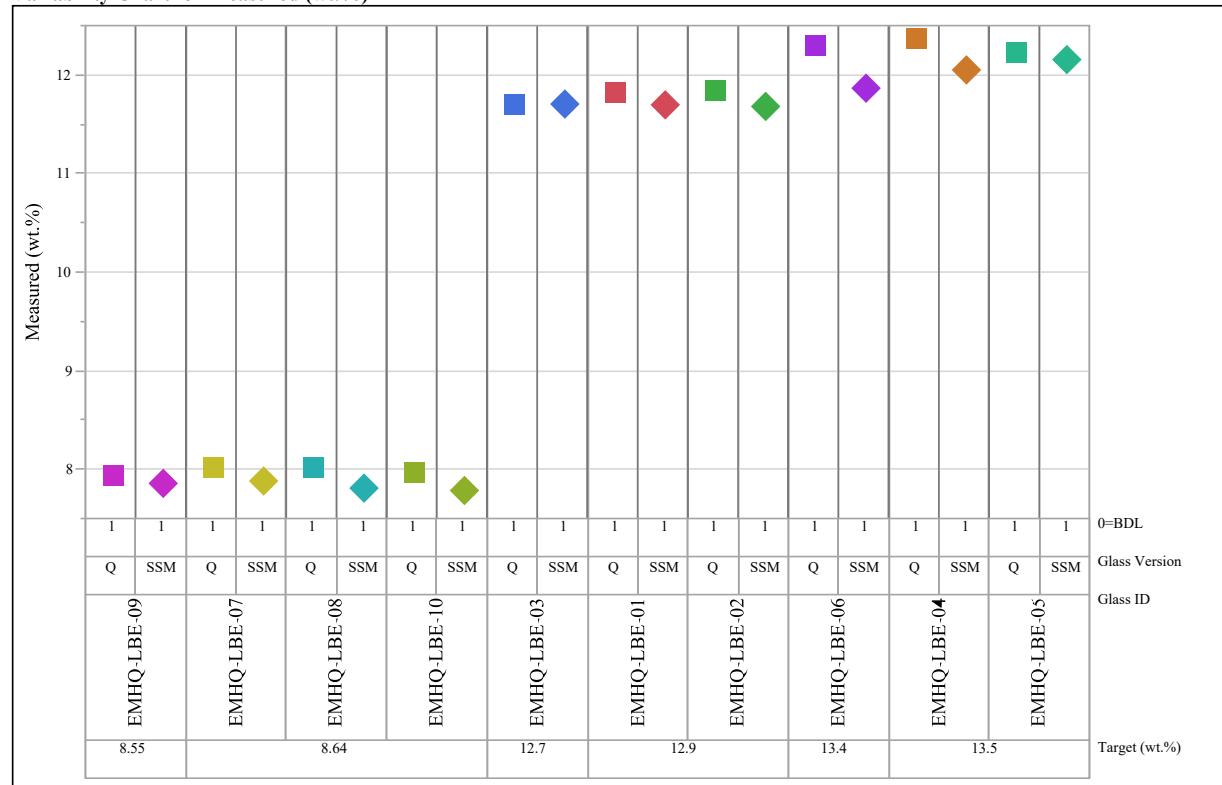
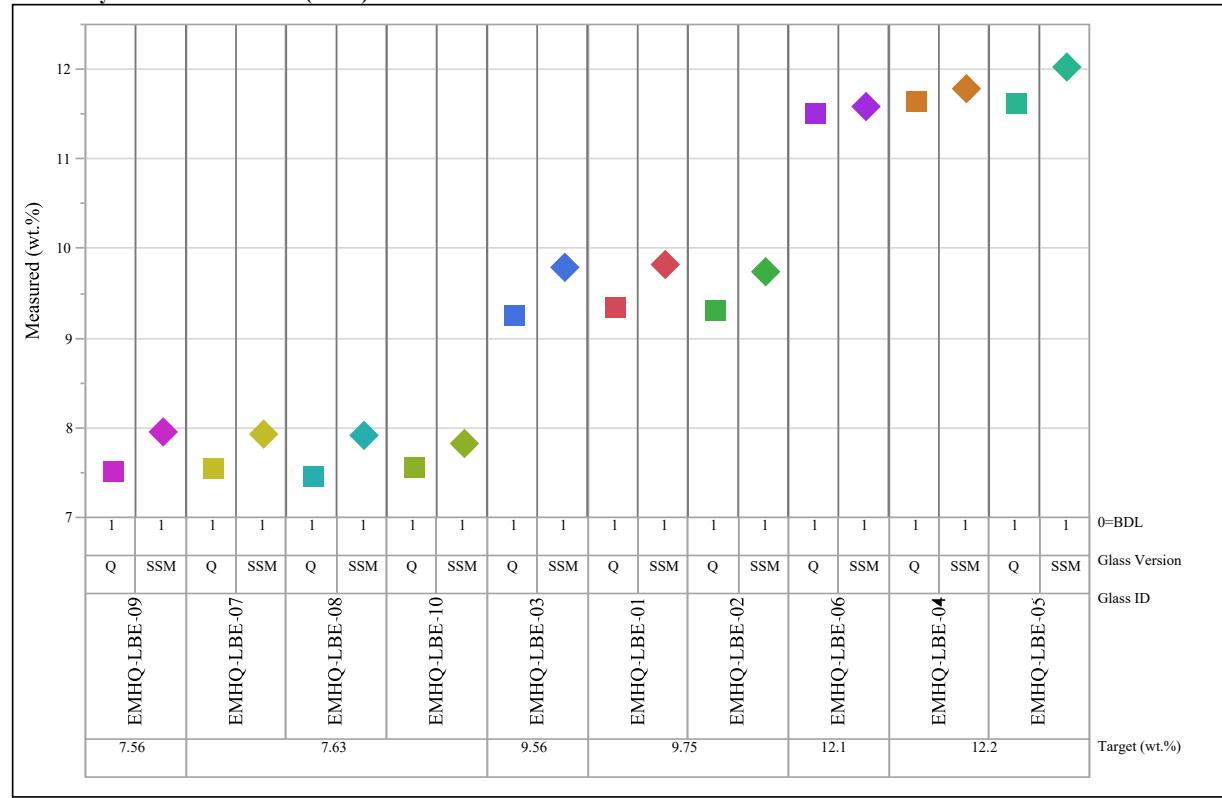


Exhibit A-4. Comparisons of the Measured Compositions of the Quenched and SSM Versions of the Study Glasses (continued)

Oxide=CaO

Variability Chart for Measured (wt.%)

Oxide=Cl⁻

Variability Chart for Measured (wt.%)

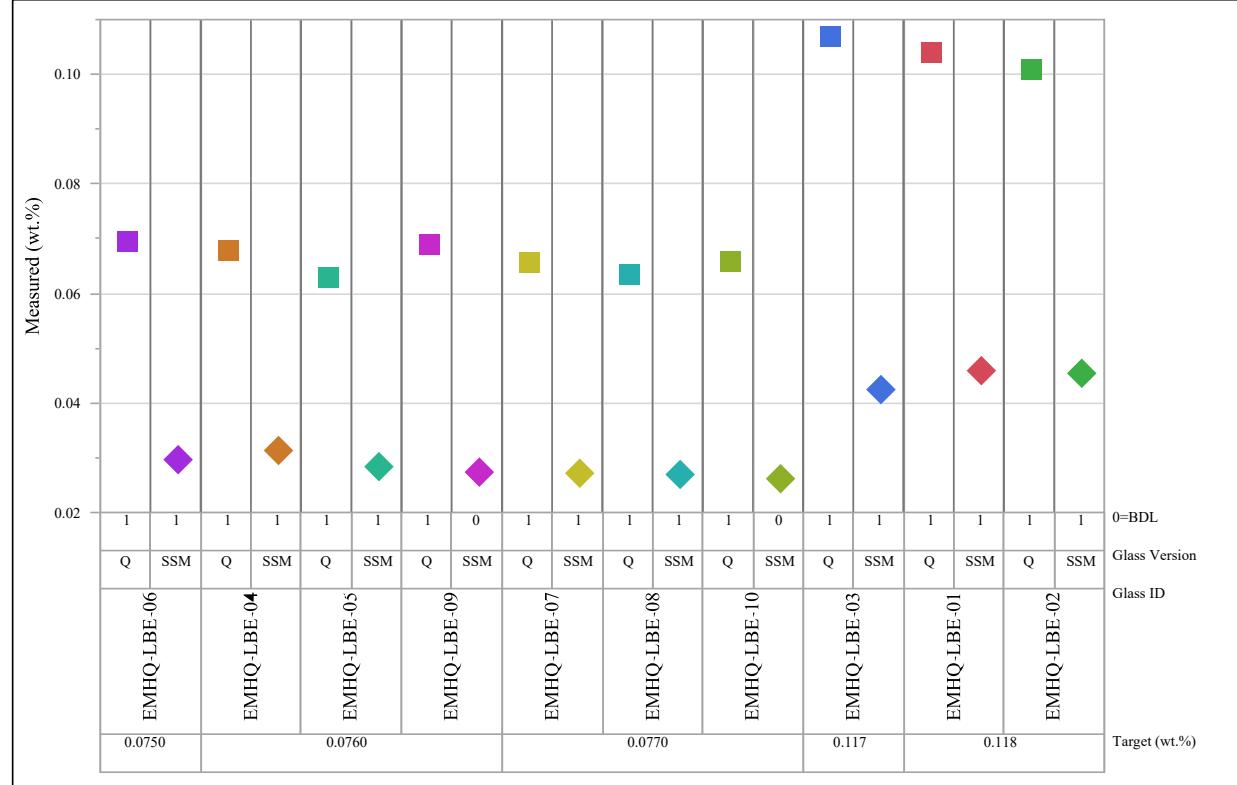
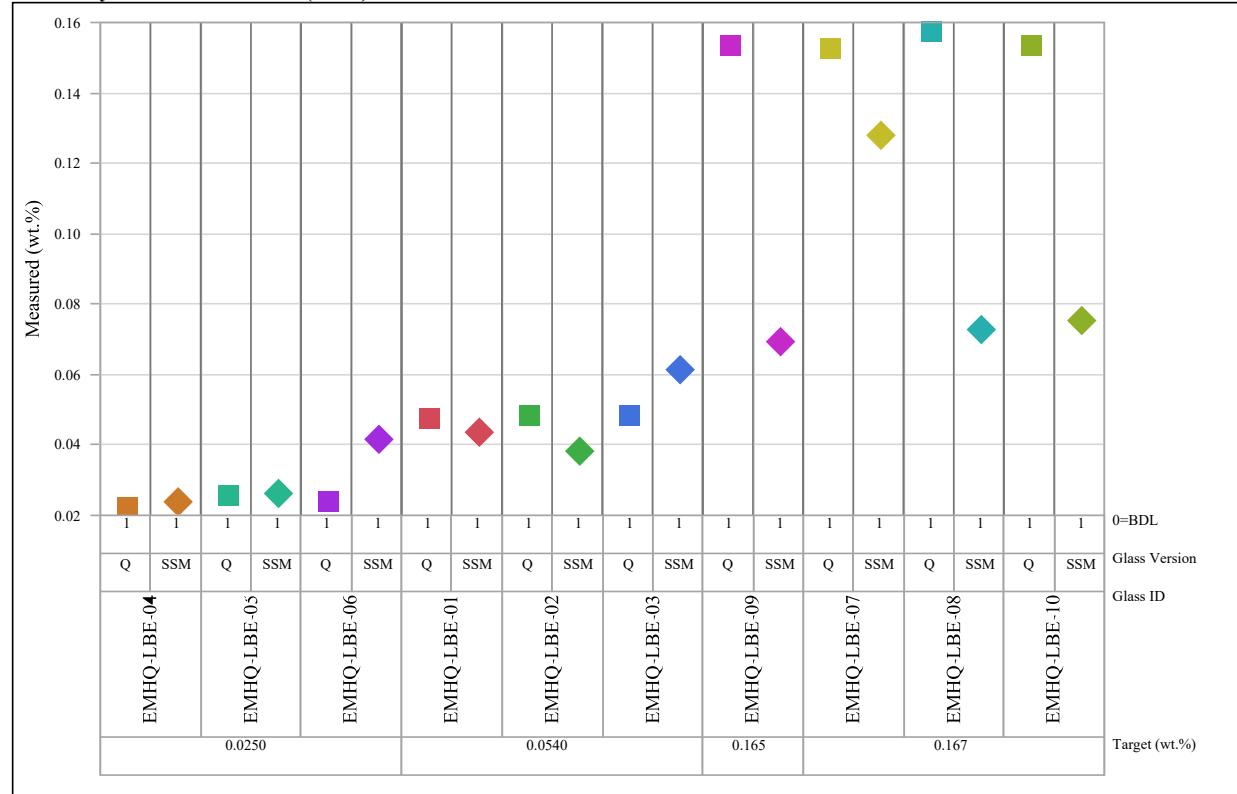


Exhibit A-4. Comparisons of the Measured Compositions of the Quenched and SSM Versions of the Study Glasses (continued)

Oxide=Cr₂O₃

Variability Chart for Measured (wt.%)



Oxide=F

Variability Chart for Measured (wt.%)

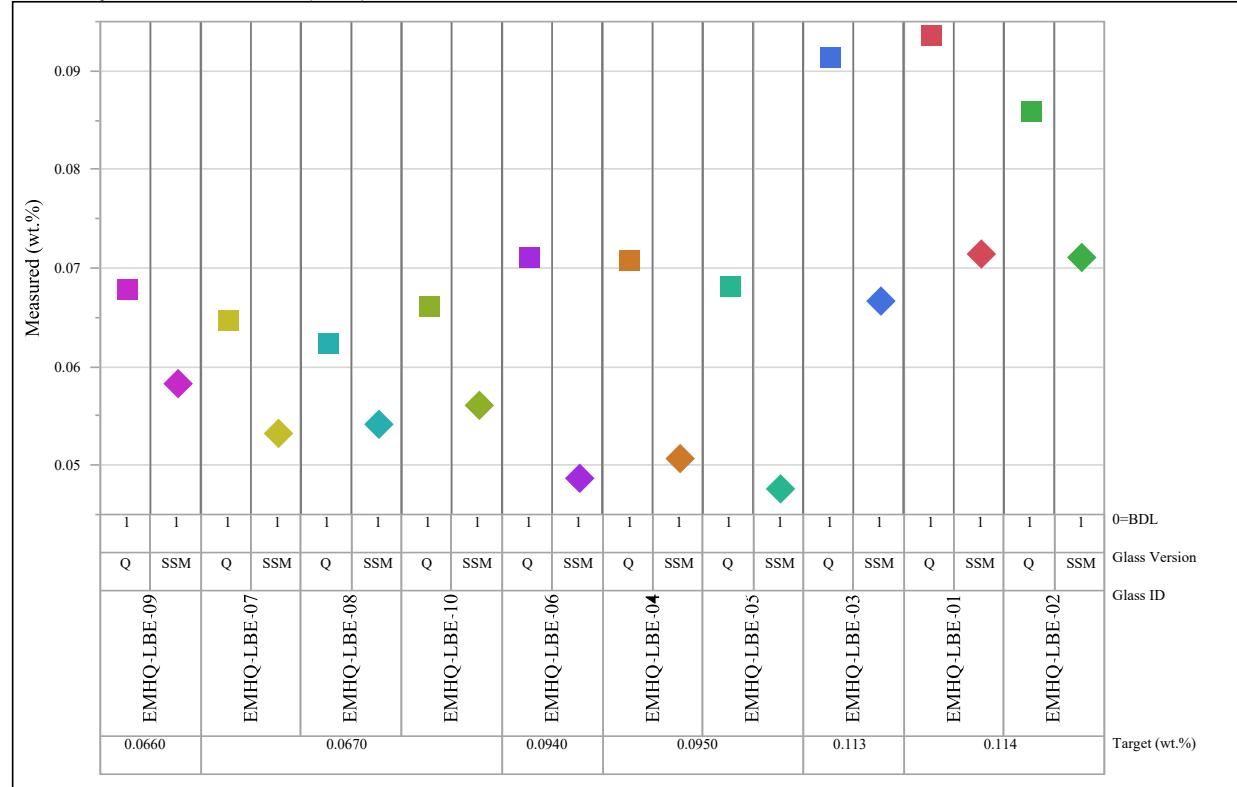
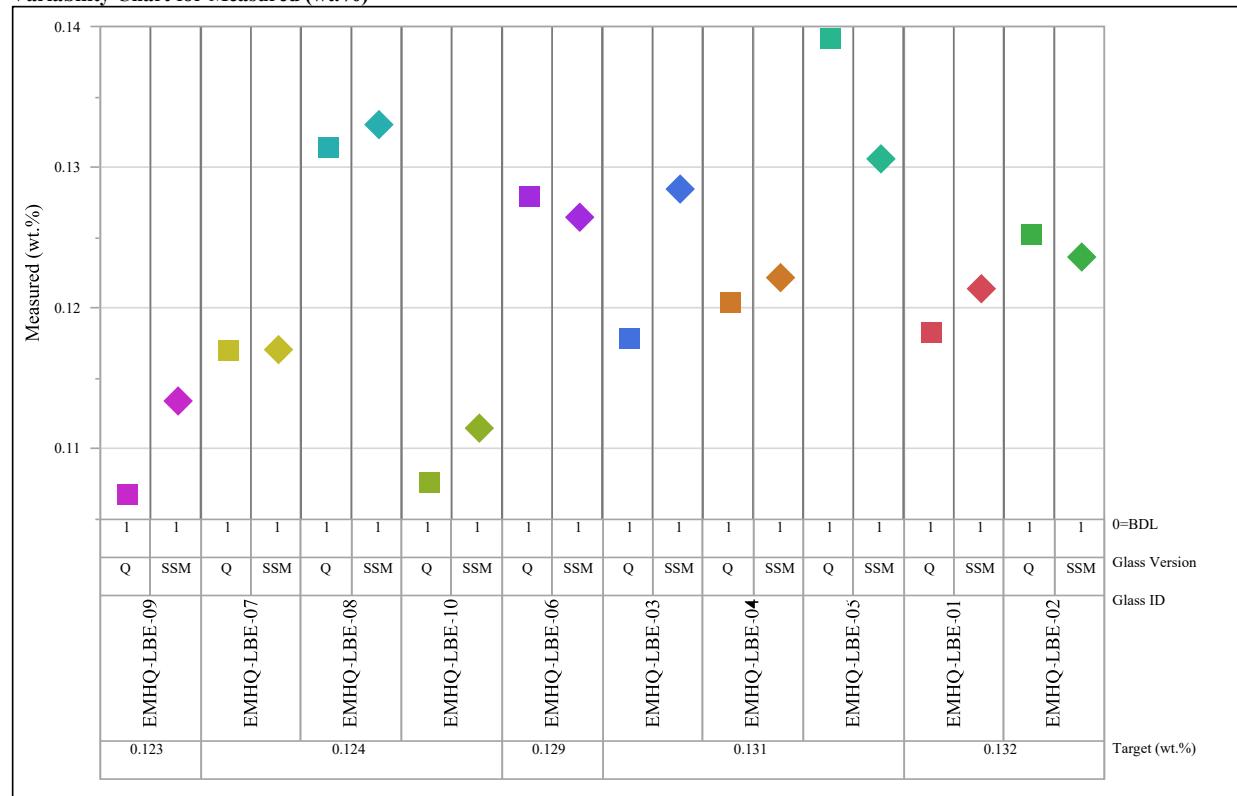


Exhibit A-4. Comparisons of the Measured Compositions of the Quenched and SSM Versions of the Study Glasses (continued)

Oxide=Fe₂O₃

Variability Chart for Measured (wt.%)

Oxide=K₂O

Variability Chart for Measured (wt.%)

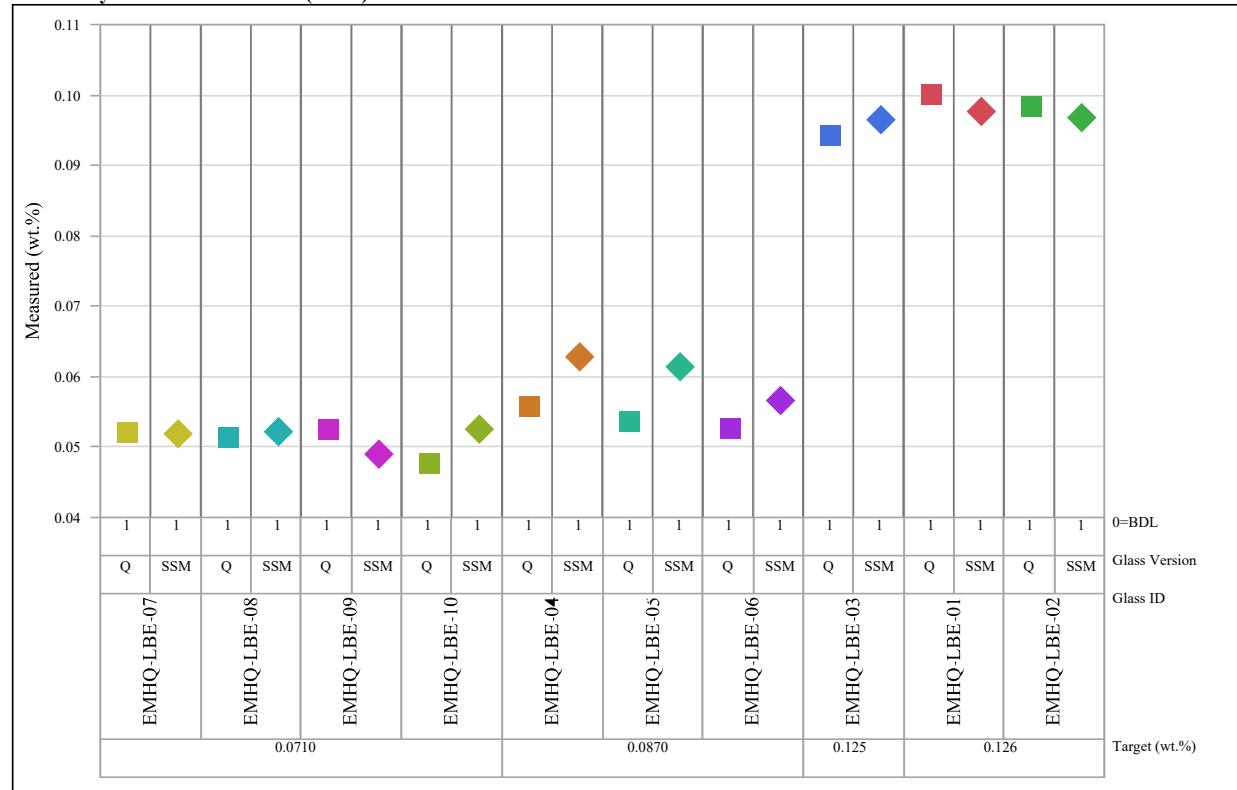
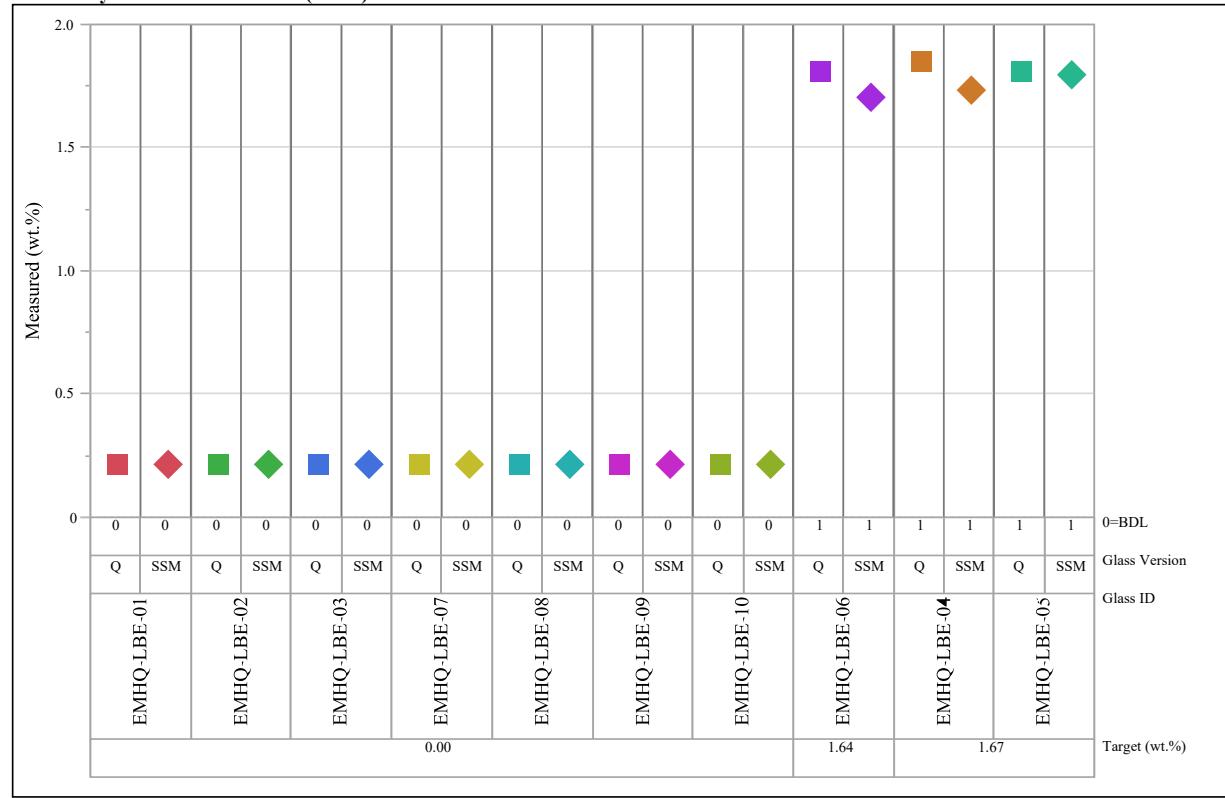


Exhibit A-4. Comparisons of the Measured Compositions of the Quenched and SSM Versions of the Study Glasses (continued)

Oxide=Li₂O

Variability Chart for Measured (wt.%)



Oxide=MgO

Variability Chart for Measured (wt.%)

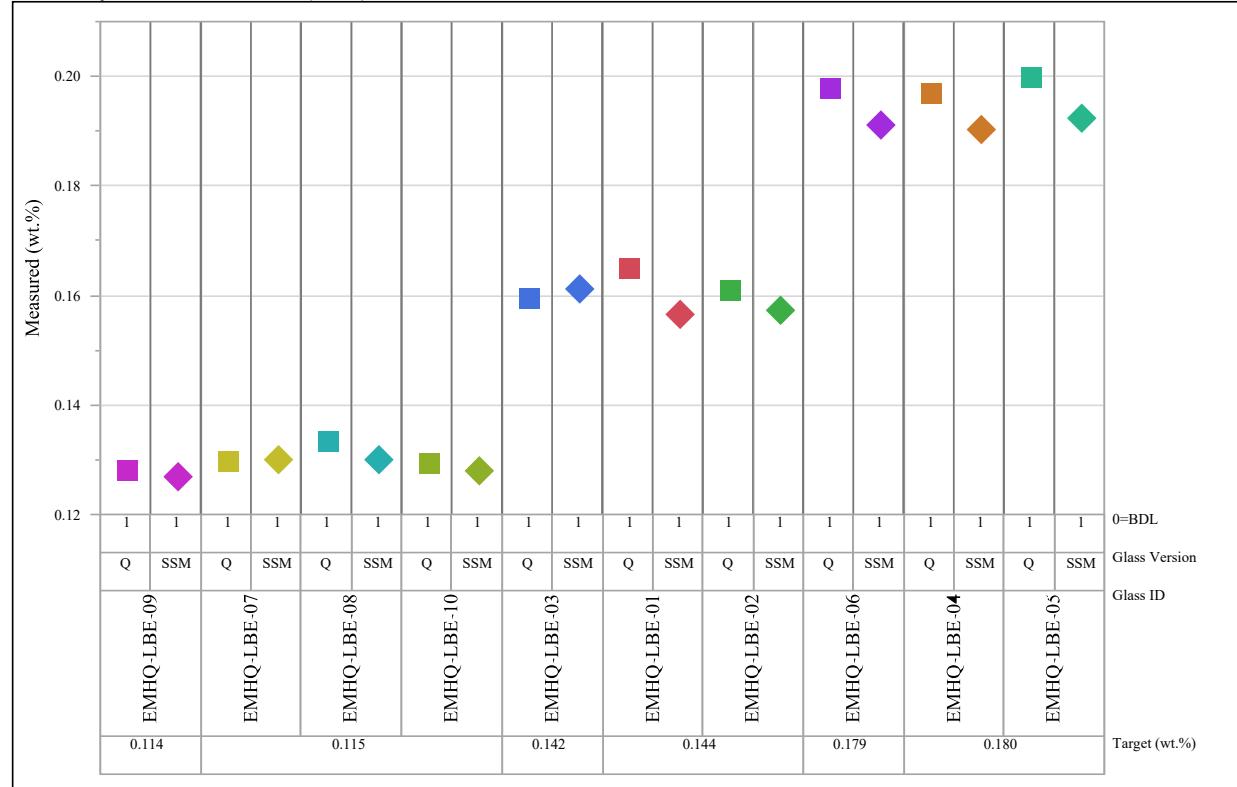
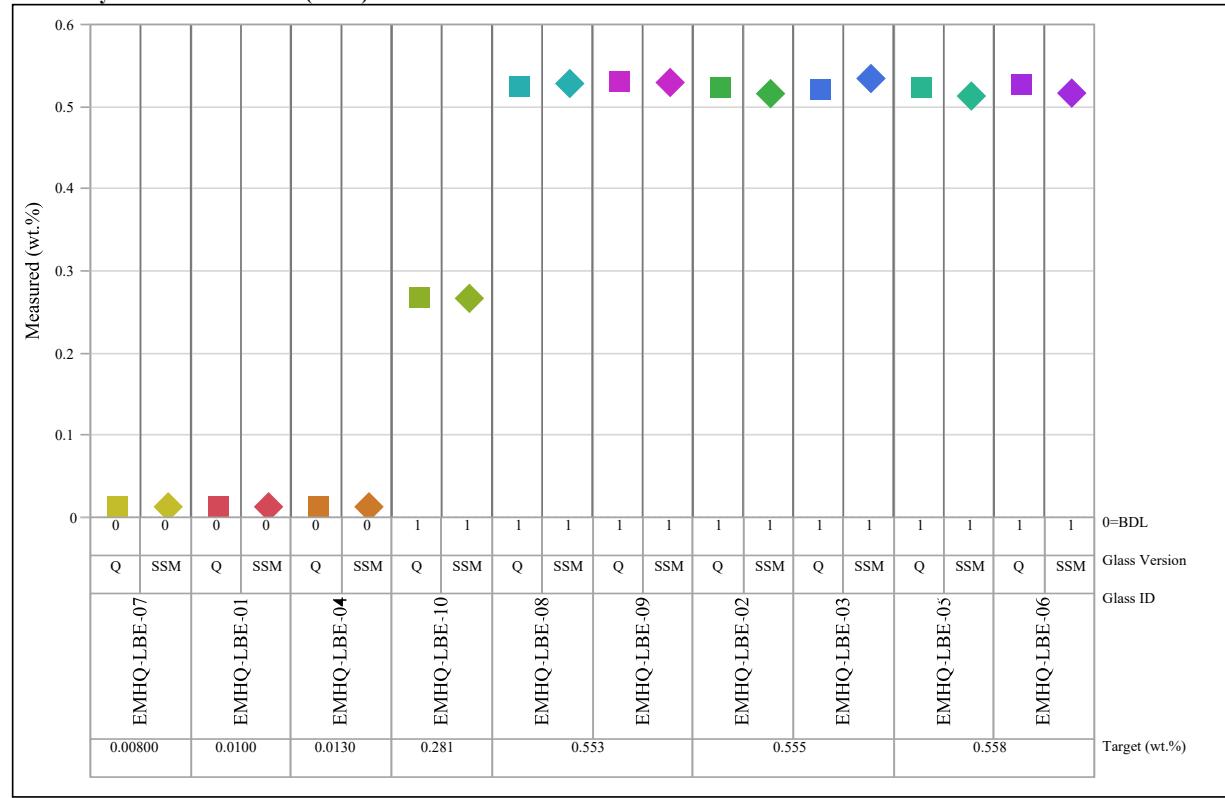


Exhibit A-4. Comparisons of the Measured Compositions of the Quenched and SSM Versions of the Study Glasses (continued)

Oxide=MnO

Variability Chart for Measured (wt.%)

Oxide=Na₂O

Variability Chart for Measured (wt.%)

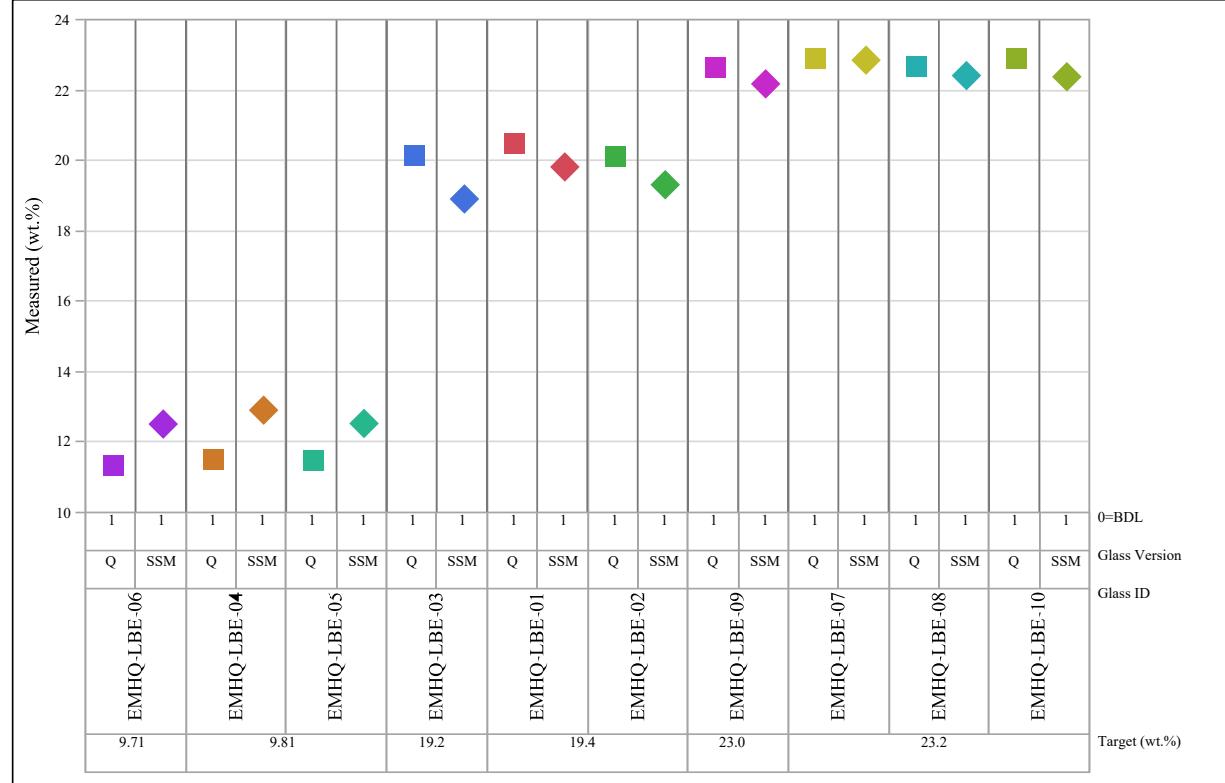
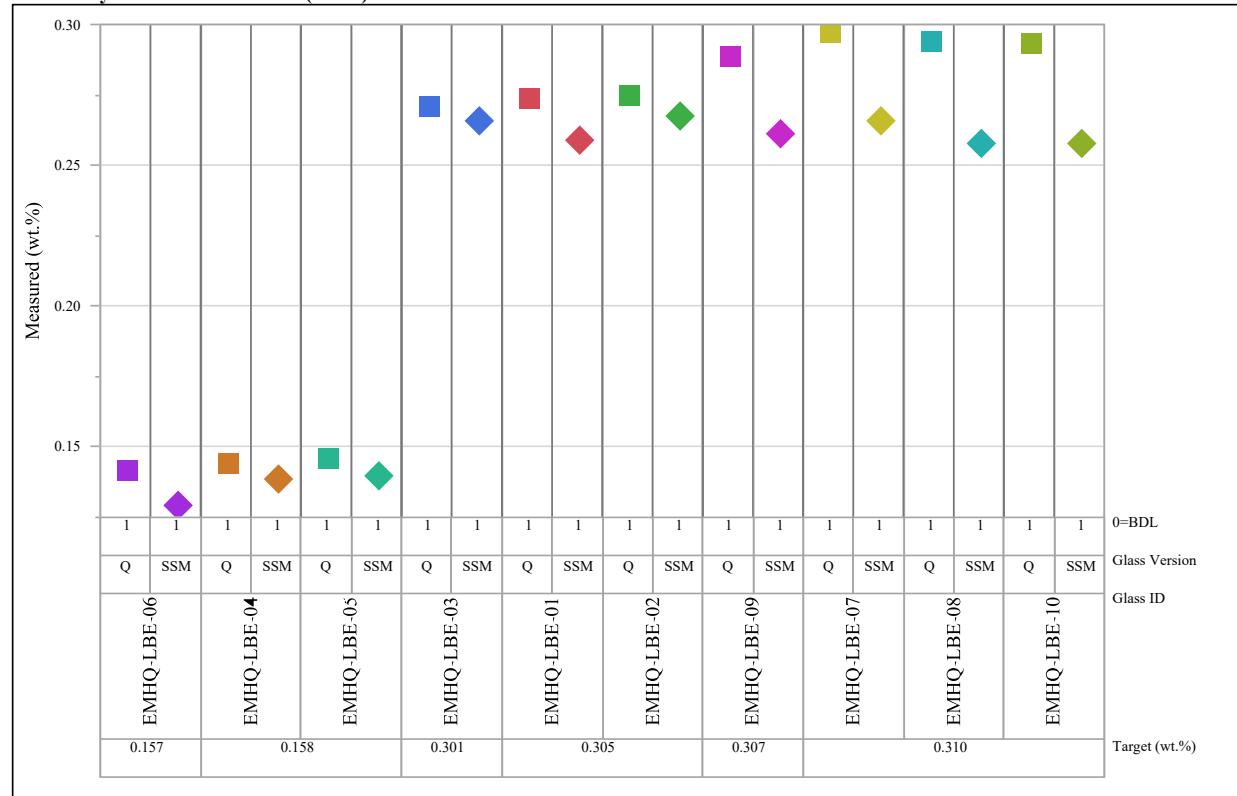


Exhibit A-4. Comparisons of the Measured Compositions of the Quenched and SSM Versions of the Study Glasses (continued)

Oxide= P_2O_5

Variability Chart for Measured (wt.%)

Oxide= SiO_2

Variability Chart for Measured (wt.%)

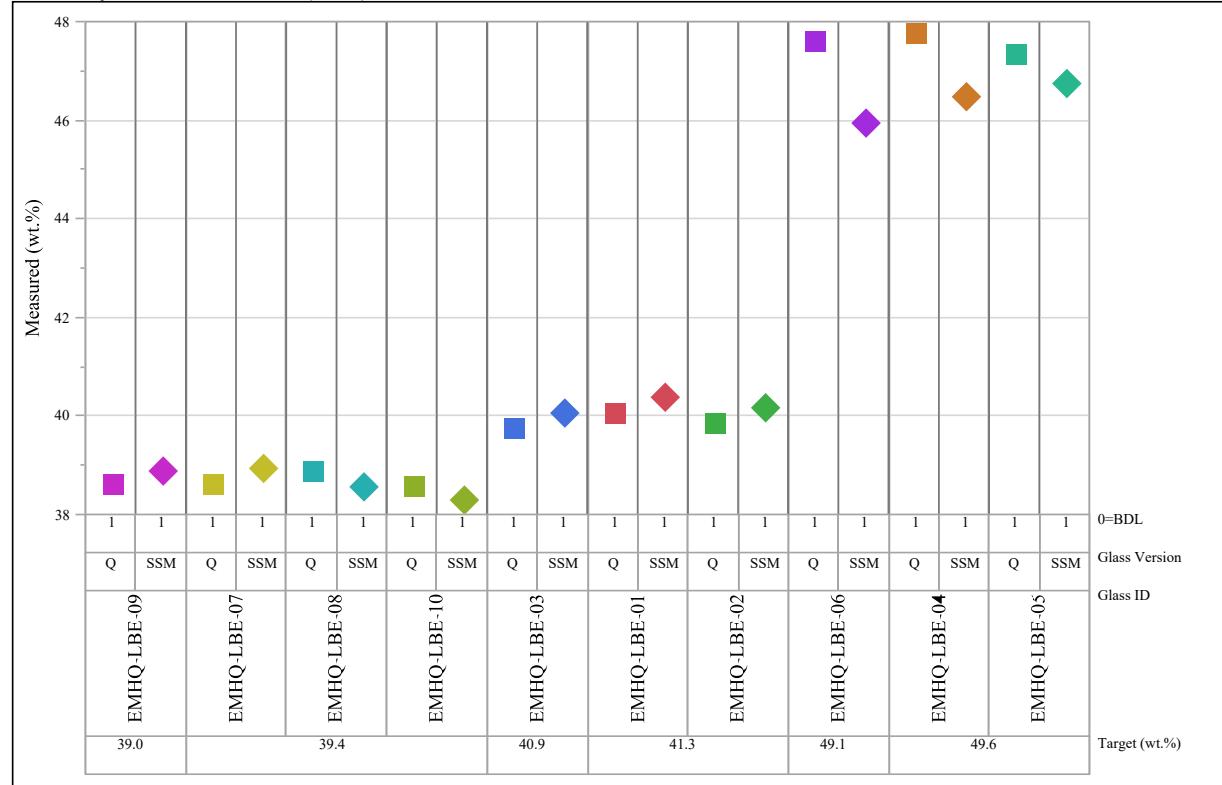
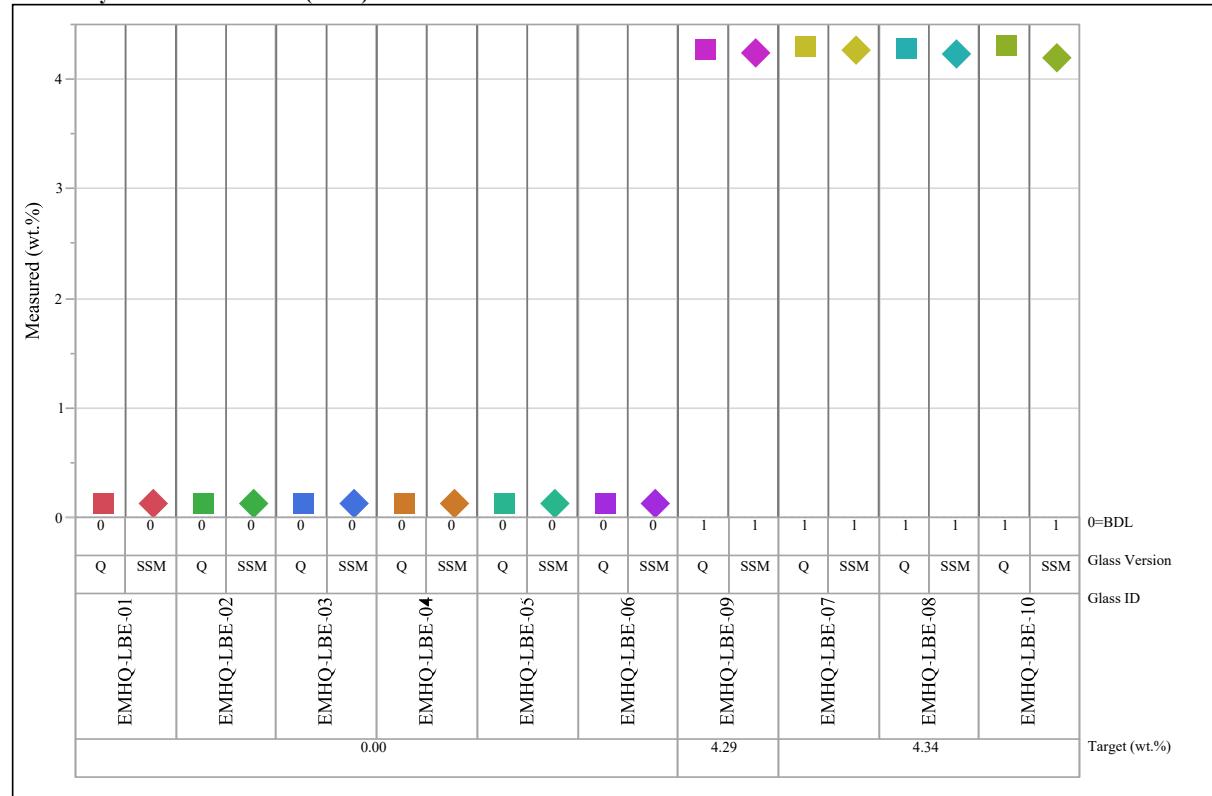


Exhibit A-4. Comparisons of the Measured Compositions of the Quenched and SSM Versions of the Study Glasses (continued)

Oxide= SnO_2

Variability Chart for Measured (wt.%)

Oxide= SO_3

Variability Chart for Measured (wt.%)

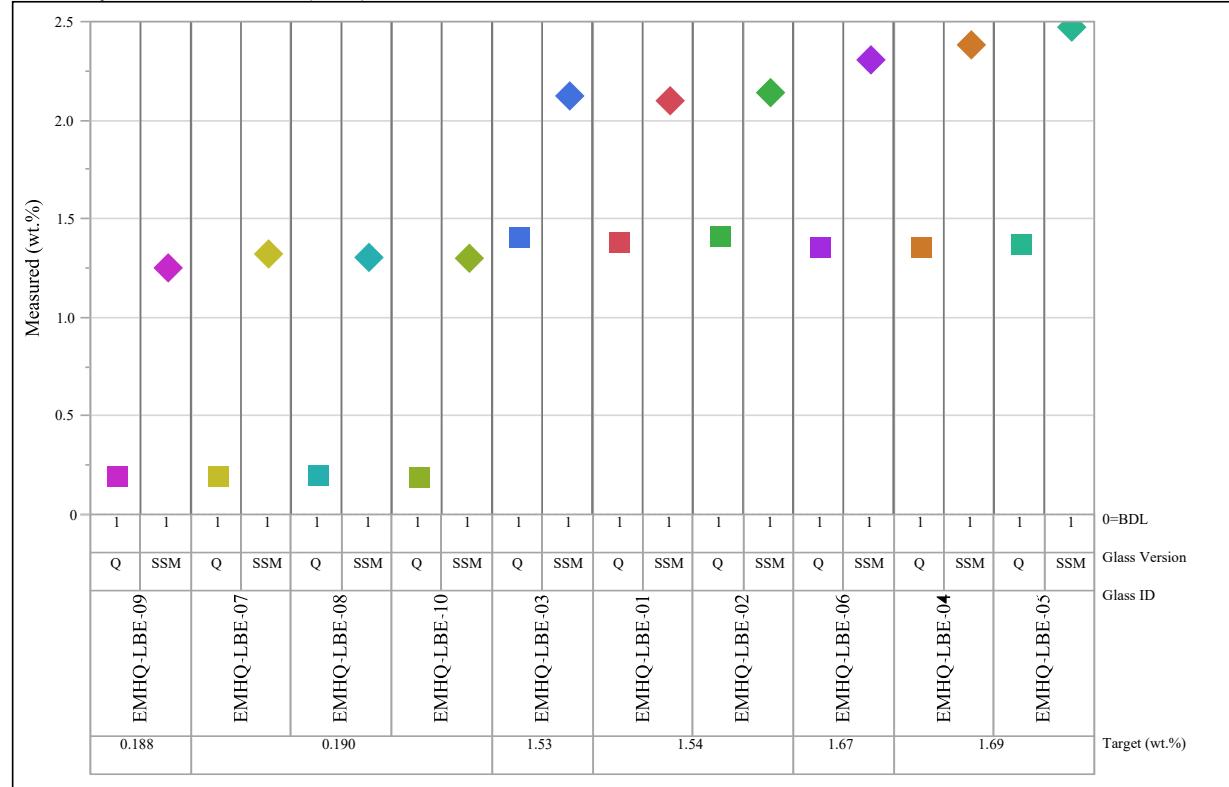
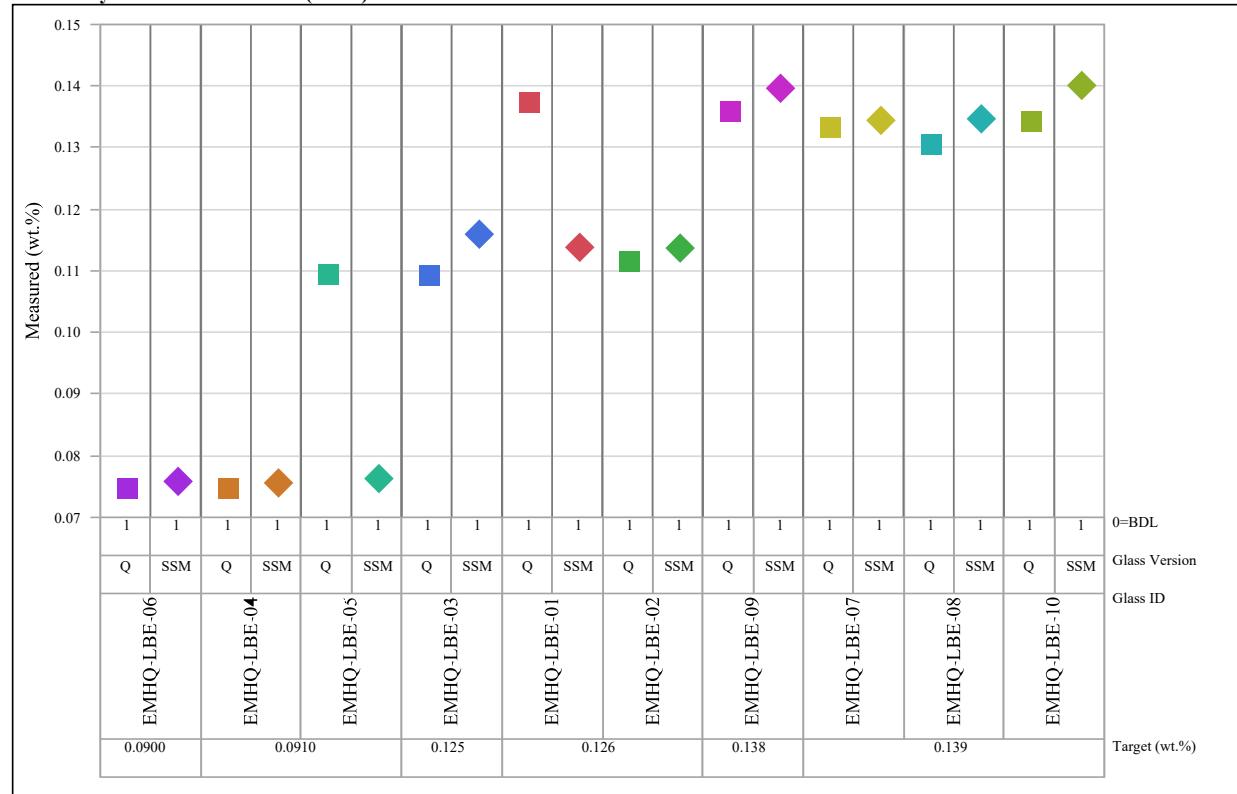


Exhibit A-4. Comparisons of the Measured Compositions of the Quenched and SSM Versions of the Study Glasses (continued)

Oxide= TiO_2

Variability Chart for Measured (wt.%)

Oxide= V_2O_5

Variability Chart for Measured (wt.%)

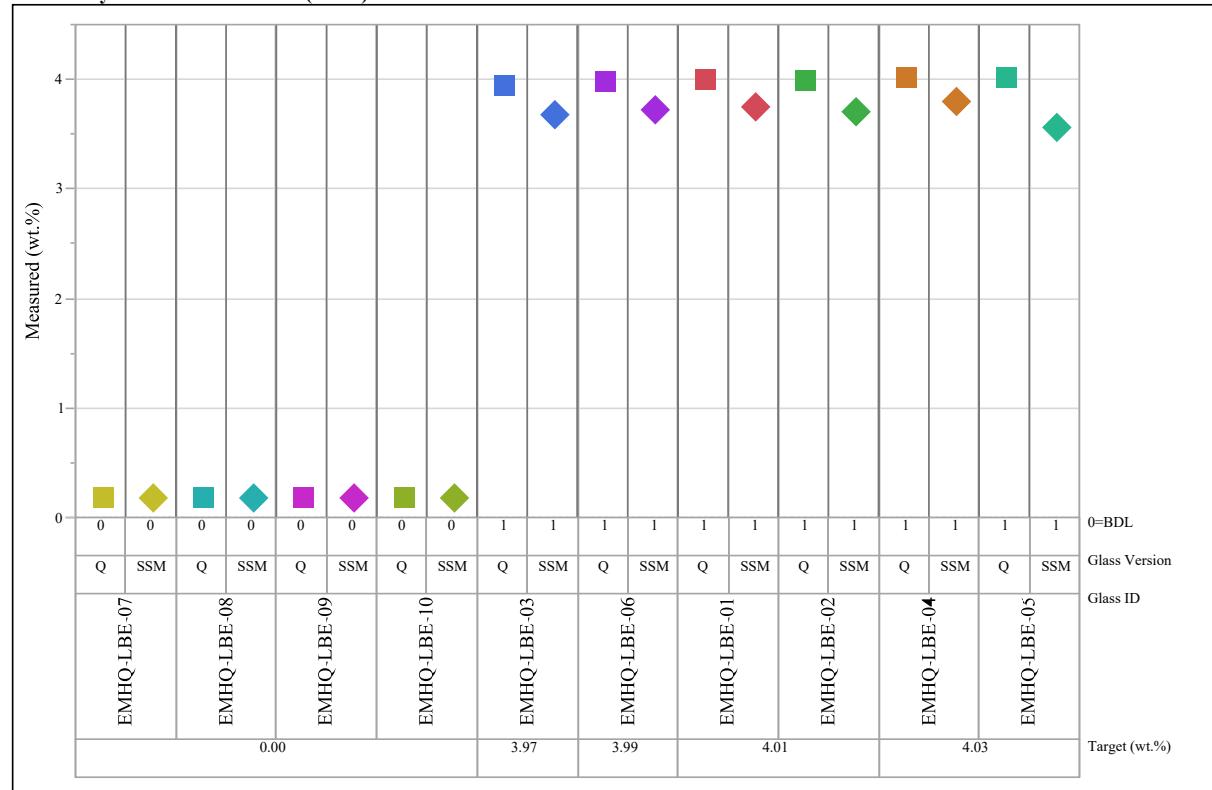
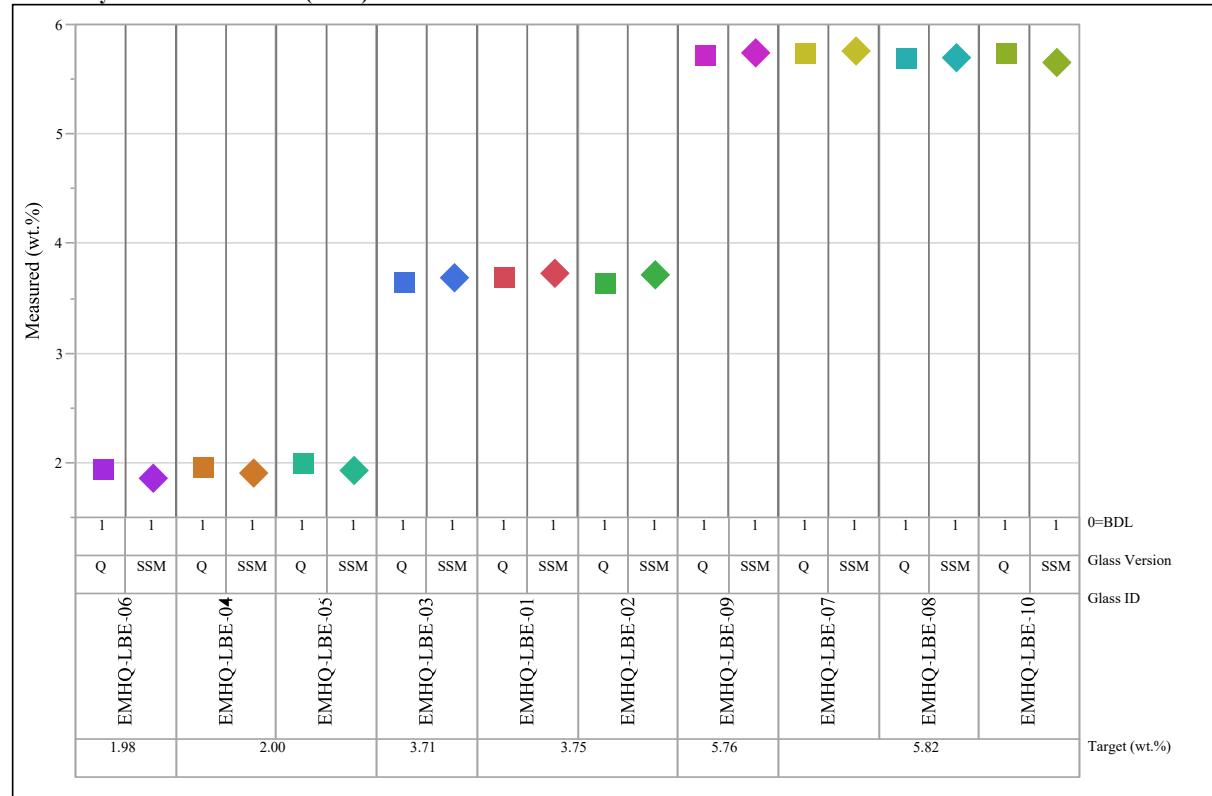
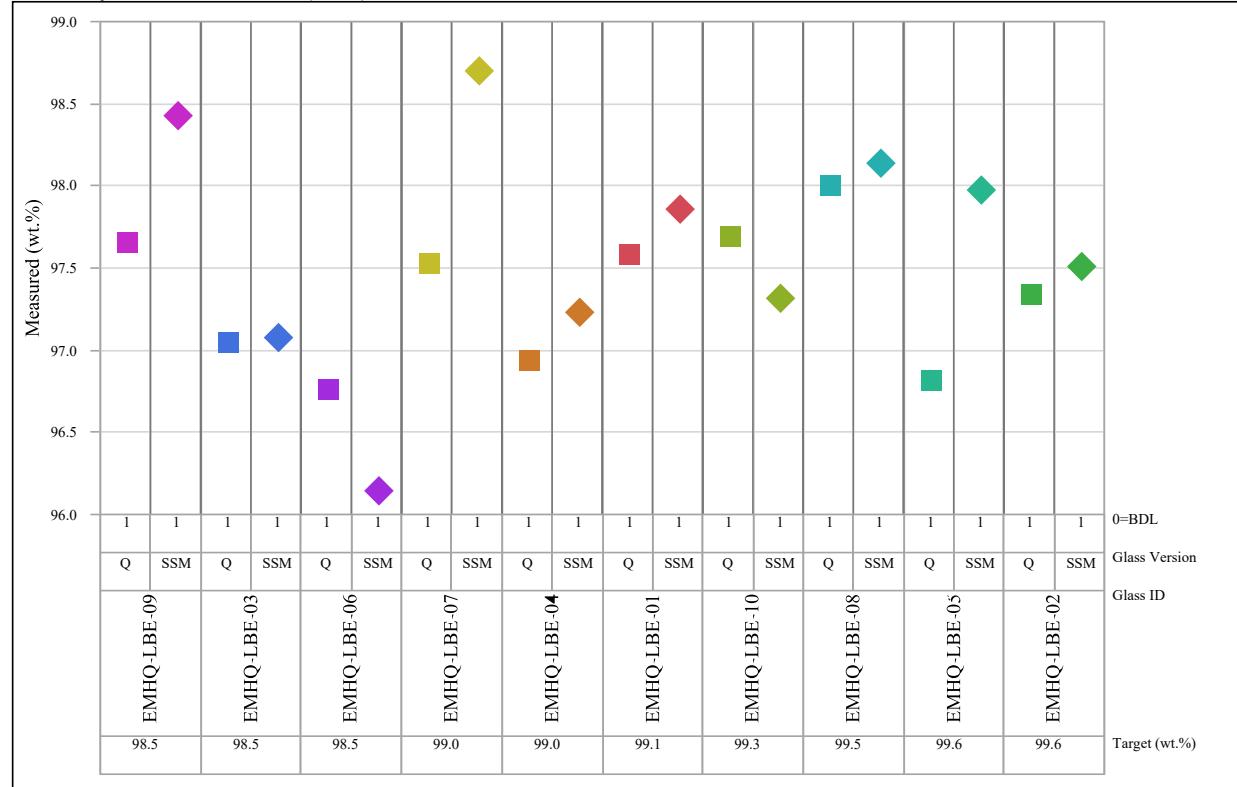


Exhibit A-4. Comparisons of the Measured Compositions of the Quenched and SSM Versions of the Study Glasses (continued)

Oxide=ZrO₂
Variability Chart for Measured (wt.%)



Oxide=Sum of Oxides
Variability Chart for Measured (wt.%)



Appendix B. Tables and Exhibits Supporting the Wash Solution Composition Measurements

Table B-2. Measurements (mg/L) of the SSM Wash Solutions Measured by IC

PNNL ID	Block	Seq.	Lab ID	Cl⁻	F⁻	PO₄³⁻	SO₄²⁻
1ppm ckstd	1	1	1ppm ckstd	0.947	1.00	0.986	1.08
EMHQ-LBE-02-SSM-W	1	2	S-13431-1	6.08	<5.00	<10.0	1710
EMHQ-LBE-01-SSM-W	1	3	S-13430-1	6.19	<5.00	<10.0	1640
EMHQ-LBE-07-SSM-W	1	4	S-13436-1	<5.00	<5.00	<10.0	1600
EMHQ-LBE-06-SSM-W	1	5	S-13435-1	<5.00	<5.00	<10.0	402
EMHQ-LBE-09-SSM-W	1	6	S-13438-1	<5.00	<5.00	<10.0	1570
EMHQ-LBE-08-SSM-W	1	7	S-13437-1	<5.00	<5.00	<10.0	1590
IC BLANK	1	8	IC BLANK 1-1	<5.00	<5.00	<10.0	<10.0
EMHQ-LBE-10-SSM-W	1	9	S-13439-1	<5.00	<5.00	<10.0	1560
EMHQ-LBE-SSM-W	1	10	S-13440-1	<5.00	<5.00	<10.0	1340
IC BLANK	1	11	IC BLANK 1-2	<5.00	<5.00	<10.0	<10.0
EMHQ-LBE-03-SSM-W	1	12	S-13432-1	5.88	<5.00	<10.0	1720
EMHQ-LBE-05-SSM-W	1	13	S-13434-1	<5.00	<5.00	<10.0	387
EMHQ-LBE-04-SSM-W	1	14	S-13433-1	<5.00	<5.00	<10.0	501
1ppm ckstd	1	15	1ppm ckstd	0.950	1.00	0.948	0.950
1ppm ckstd	2	1	1ppm ckstd	0.954	1.00	0.958	1.00
EMHQ-LBE-05-SSM-W	2	2	S-13434-2	<5.00	<5.00	<10.0	385
IC BLANK	2	3	IC BLANK 2-1	<5.00	<5.00	<10.0	<10.0
EMHQ-LBE-10-SSM-W	2	4	S-13439-2	<5.00	<5.00	<10.0	1570
IC BLANK	2	5	IC BLANK 2-2	<5.00	<5.00	<10.0	<10.0
EMHQ-LBE-SSM-W	2	6	S-13440-2	<5.00	<5.00	<10.0	1350
EMHQ-LBE-03-SSM-W	2	7	S-13432-2	5.91	<5.00	<10.0	1730
EMHQ-LBE-08-SSM-W	2	8	S-13437-2	<5.00	<5.00	<10.0	1580
EMHQ-LBE-07-SSM-W	2	9	S-13436-2	<5.00	<5.00	<10.0	1620
EMHQ-LBE-04-SSM-W	2	10	S-13433-2	<5.00	<5.00	<10.0	501
EMHQ-LBE-01-SSM-W	2	11	S-13430-2	6.26	<5.00	<10.0	1660
EMHQ-LBE-09-SSM-W	2	12	S-13438-2	<5.00	<5.00	<10.0	1580
EMHQ-LBE-06-SSM-W	2	13	S-13435-2	<5.00	<5.00	<10.0	406
EMHQ-LBE-02-SSM-W	2	14	S-13431-2	6.04	<5.00	<10.0	1740
1ppm ckstd	2	15	1ppm ckstd	0.945	1.02	1.07	0.967
1ppm ckstd	3	1	1ppm ckstd	0.949	0.999	0.984	1.10
EMHQ-LBE-07-SSM-W	3	2	S-13436-3	<5.00	<5.00	<10.0	1630
EMHQ-LBE-01-SSM-W	3	3	S-13430-3	6.27	<5.00	<10.0	1670
EMHQ-LBE-02-SSM-W	3	4	S-13431-3	5.89	<5.00	<10.0	1700
EMHQ-LBE-10-SSM-W	3	5	S-13439-3	<5.00	<5.00	<10.0	1660
IC BLANK	3	6	IC BLANK 3-1	<5.00	<5.00	<10.0	<10.0
EMHQ-LBE-06-SSM-W	3	7	S-13435-3	<5.00	<5.00	<10.0	407
EMHQ-LBE-SSM-W	3	8	S-13440-3	<5.00	<5.00	<10.0	1350
EMHQ-LBE-08-SSM-W	3	9	S-13437-3	<5.00	<5.00	<10.0	1600
IC BLANK	3	10	IC BLANK 3-2	<5.00	<5.00	<10.0	<10.0
EMHQ-LBE-05-SSM-W	3	11	S-13434-3	<5.00	<5.00	<10.0	387
EMHQ-LBE-09-SSM-W	3	12	S-13438-3	<5.00	<5.00	<10.0	1570
EMHQ-LBE-04-SSM-W	3	13	S-13433-3	<5.00	<5.00	<10.0	503
EMHQ-LBE-03-SSM-W	3	14	S-13432-3	5.88	<5.00	<10.0	1730
1ppm ckstd	3	15	1ppm ckstd	0.963	1.01	0.968	1.07

Table B-3. Results for Standards and Blanks Utilized During the Measurement of the Wash Solutions

Solution ID	Instrument	Analyte	Reference Value (mg/L)	Mean Measurement (mg/L)
1ppm std	IC	Cl ⁻	1	0.951
1ppm std	IC	F ⁻	1	1.00
1ppm std	IC	PO ₄ ³⁻	1	0.986
1ppm std	IC	SO ₄ ²⁻	1	1.03
hp std	ICP-OES	Al	50	52.8
hp std	ICP-OES	B	0	<1.00
hp std	ICP-OES	Ca	0	<1.00
hp std	ICP-OES	Cr	0	<1.00
hp std	ICP-OES	Fe	50	49.8
hp std	ICP-OES	K	0	<1.00
hp std	ICP-OES	Li	0	<1.00
hp std	ICP-OES	Mg	0	<1.00
hp std	ICP-OES	Mn	20	21.0
hp std	ICP-OES	Na	150	145
hp std	ICP-OES	P	0	<1.00
hp std	ICP-OES	S	10	9.88
hp std	ICP-OES	Si	0	<1.00
hp std	ICP-OES	Sn	0	<1.00
hp std	ICP-OES	Ti	0	<1.00
hp std	ICP-OES	V	0	<1.00
hp std	ICP-OES	Zr	0	<1.00
std	ICP-OES	Al	4	4.03
std	ICP-OES	B	20	18.8
std	ICP-OES	Ca	0	<1.00
std	ICP-OES	Cr	0	<1.00
std	ICP-OES	Fe	4	4.09
std	ICP-OES	K	10	9.62
std	ICP-OES	Li	10	10.3
std	ICP-OES	Mg	0	<1.00
std	ICP-OES	Mn	0	<1.00
std	ICP-OES	Na	81	79.7
std	ICP-OES	P	0	<1.00
std	ICP-OES	S	0	<1.00
std	ICP-OES	Si	50	50.7
std	ICP-OES	Sn	0	<1.00
std	ICP-OES	Ti	0	<1.00
std	ICP-OES	V	0	<1.00
std	ICP-OES	Zr	0	<1.00

Table B-4. Average Measurements (mg/L) of the SSM Wash Solutions

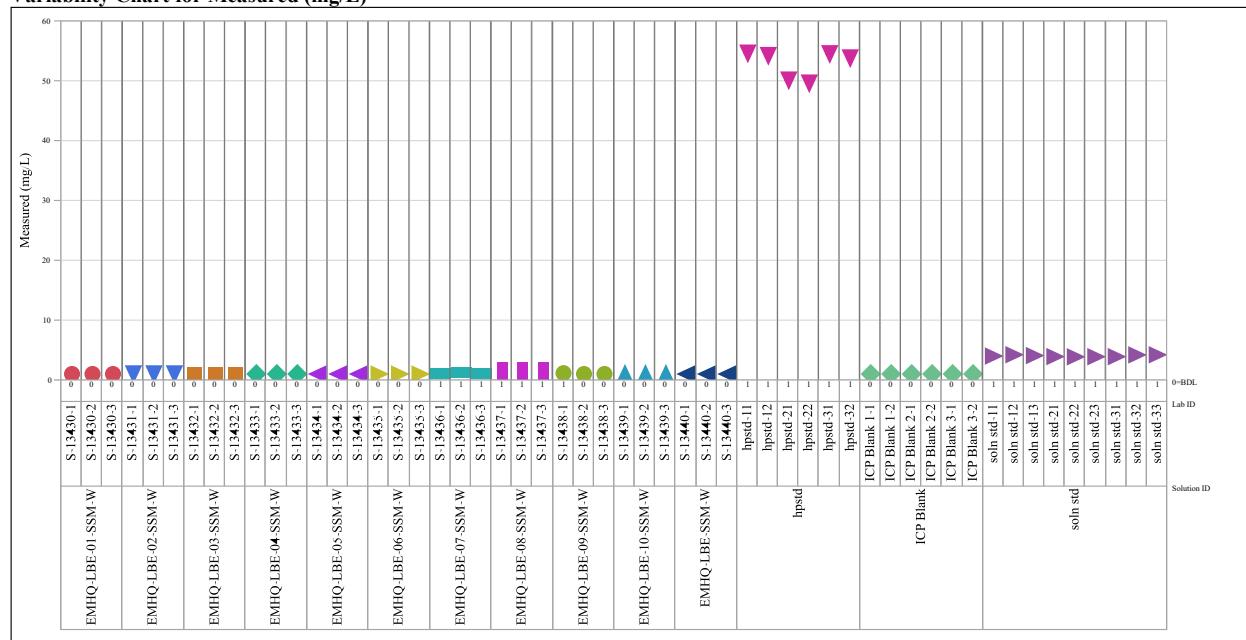
Lab ID	PNNL ID	IC				ICP-OES																		
		Cl ⁻	F ⁻	PO ₄ ³⁻	SO ₄ ²⁻	Al	B	Ca	Cr	Fe	K	Li	Mg	Mn	Na	P	PO ₄ ^{3-*}	S	Si	Sn	SO ₄ ^{2-*}	Ti	V	Zr
S-13430	EMHQ-LBE-01-SSM-W	6.24	<5.00	<10.0	1660	<1.00	32.6	29.3	2.55	<1.00	5.05	<1.00	<1.00	<1.00	908	<1.00	<3.07	541	29.4	<1.00	1620	<1.00	36.1	<1.00
S-13431	EMHQ-LBE-02-SSM-W	6.00	<5.00	<10.0	1720	<1.00	28.0	29.3	4.44	<1.00	5.17	<1.00	<1.00	<1.00	924	<1.00	<3.07	555	23.5	<1.00	1660	<1.00	34.9	<1.00
S-13432	EMHQ-LBE-03-SSM-W	5.89	<5.00	<10.0	1730	<1.00	31.1	28.8	4.77	<1.00	5.28	<1.00	<1.00	<1.00	958	<1.00	<3.07	567	27.2	<1.00	1700	<1.00	36.3	<1.00
S-13433	EMHQ-LBE-04-SSM-W	<5.00	<5.00	<10.0	502	<1.00	14.0	42.7	<1.00	<1.00	<1.00	7.88	<1.00	<1.00	238	<1.00	<3.07	163	33.3	<1.00	488	<1.00	11.8	<1.00
S-13434	EMHQ-LBE-05-SSM-W	<5.00	<5.00	<10.0	386	<1.00	16.0	36.9	<1.00	<1.00	<1.00	6.82	<1.00	<1.00	197	<1.00	<3.07	126	38.2	<1.00	376	<1.00	11.9	<1.00
S-13435	EMHQ-LBE-06-SSM-W	<5.00	<5.00	<10.0	405	<1.00	14.7	37.4	<1.00	<1.00	<1.00	7.04	<1.00	<1.00	194	<1.00	<3.07	124	35.7	<1.00	370	<1.00	11.6	<1.00
S-13436	EMHQ-LBE-07-SSM-W	<5.00	<5.00	<10.0	1620	1.08	19.4	3.15	19.7	<1.00	3.44	<1.00	<1.00	<1.00	869	4.96	15.2	515	10.9	<1.00	1540	<1.00	<1.00	<1.00
S-13437	EMHQ-LBE-08-SSM-W	<5.00	<5.00	<10.0	1590	1.27	19.3	3.55	29.0	<1.00	3.28	<1.00	<1.00	<1.00	868	5.11	15.7	515	10.4	<1.00	1540	<1.00	<1.00	<1.00
S-13438	EMHQ-LBE-09-SSM-W	<5.00	<5.00	<10.0	1570	<1.05	18.0	3.14	27.6	<1.00	3.43	<1.00	<1.00	<1.00	856	4.60	14.1	506	9.19	<1.00	1520	<1.00	<1.00	<1.00
S-13439	EMHQ-LBE-10-SSM-W	<5.00	<5.00	<10.0	1600	<1.00	17.1	4.47	25.3	<1.00	3.79	<1.00	<1.00	<1.00	869	5.34	16.4	515	7.64	<1.00	1540	<1.00	<1.00	<1.00
S-13440	EMHQ-LBE-SSM-W	<5.00	<5.00	<10.0	1350	<1.00	<1.00	<1.00	<1.00	<1.00	<1.00	<1.00	<1.00	<1.00	619	<1.00	<3.07	441	<1.00	<1.00	1320	<1.00	<1.00	<1.00

*ICP-OES PO₄³⁻ and SO₄²⁻ values were calculated from ICP-OES P and S values.

Exhibit B-1. Measurements of SSM Wash Solutions by Solution ID by Analyte

Analyte=A1, Analysis=ICP

Variability Chart for Measured (mg/L)



Analyte=B, Analysis=ICP

Variability Chart for Measured (mg/L)

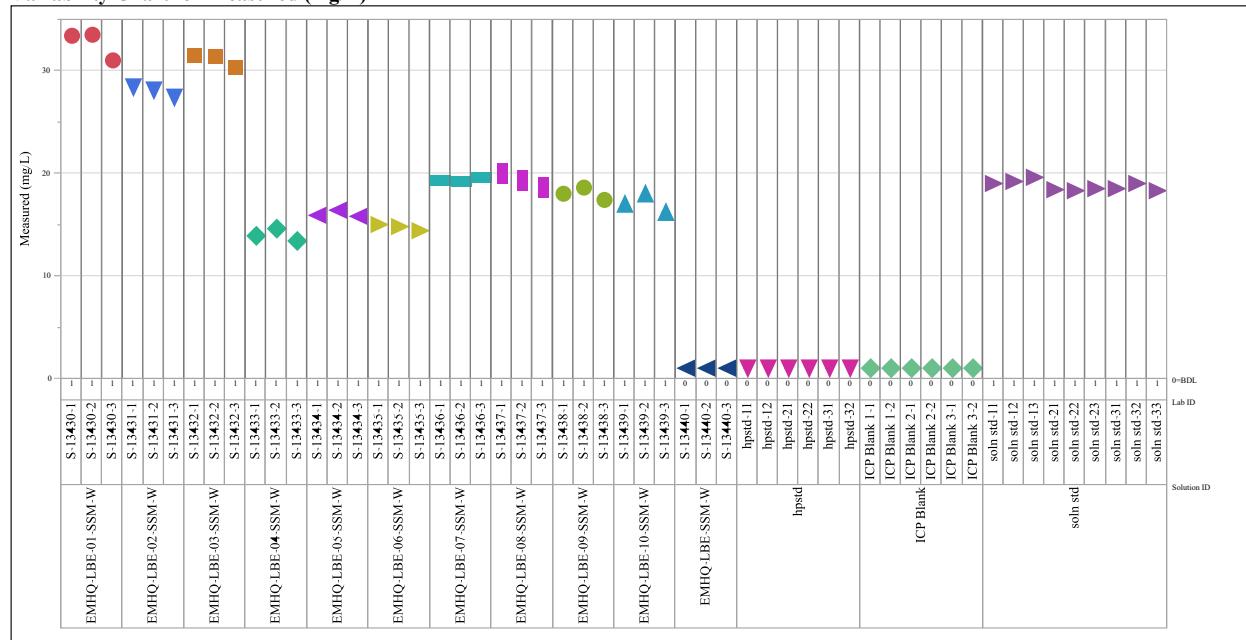
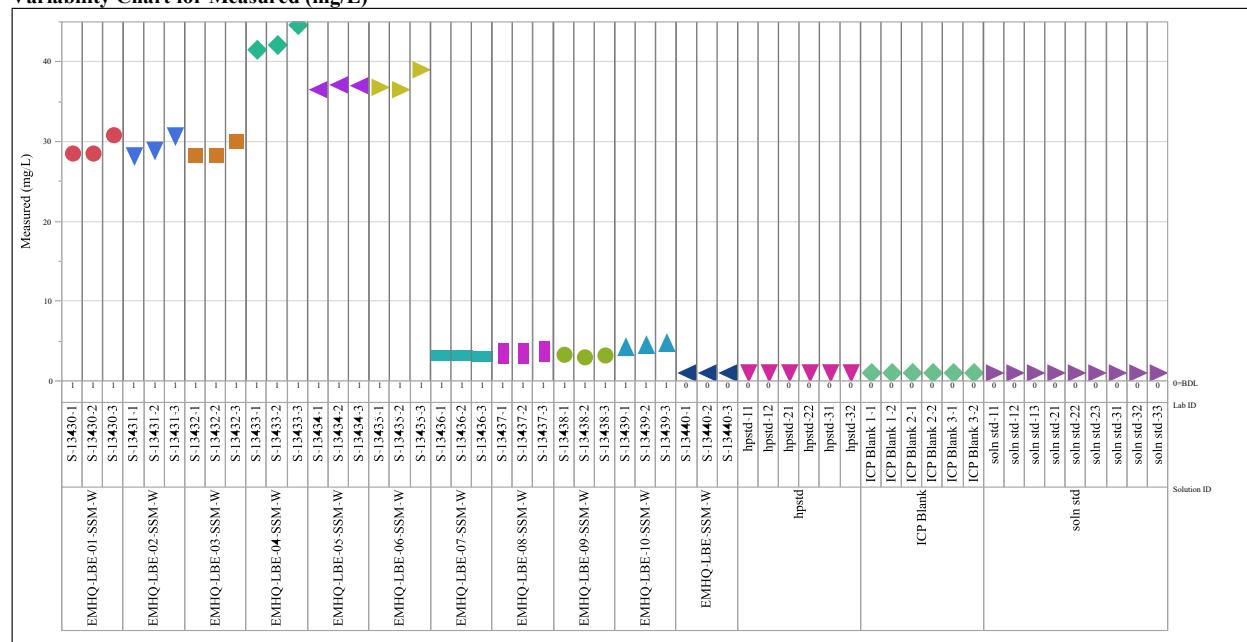


Exhibit B-1. Measurements of SSM Wash Solutions by Solution ID by Analyte (continued)

Analyte=Ca, Analysis=ICP

Variability Chart for Measured (mg/L)



Analyte=Cl, Analysis=IC

Variability Chart for Measured (mg/L)

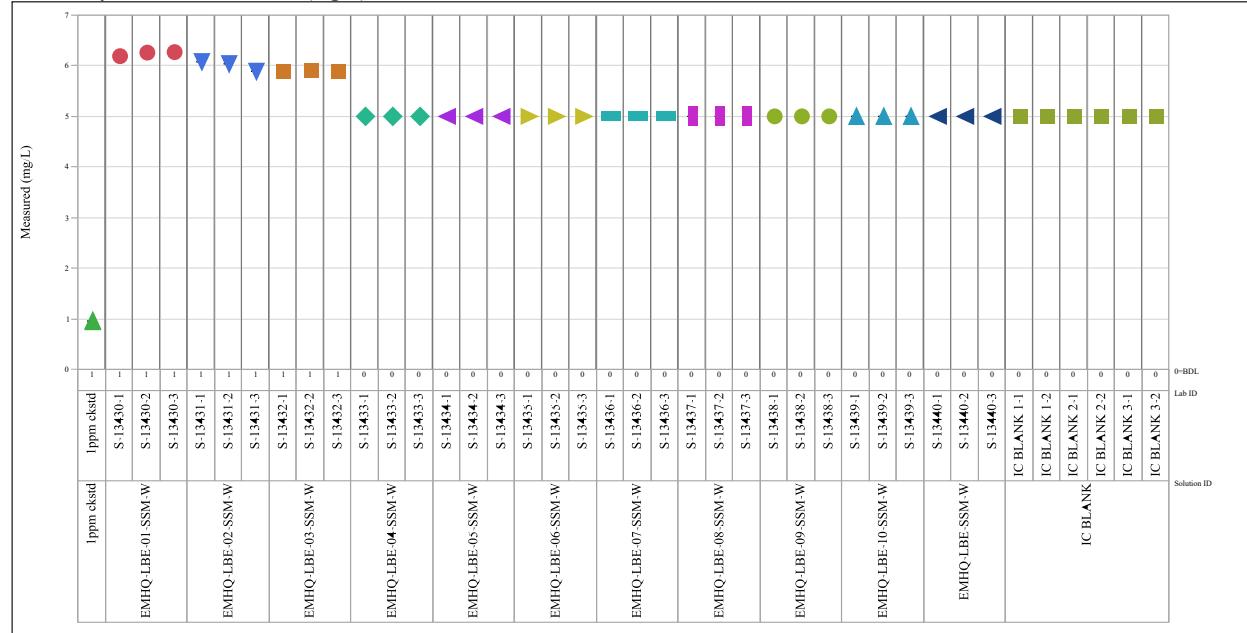
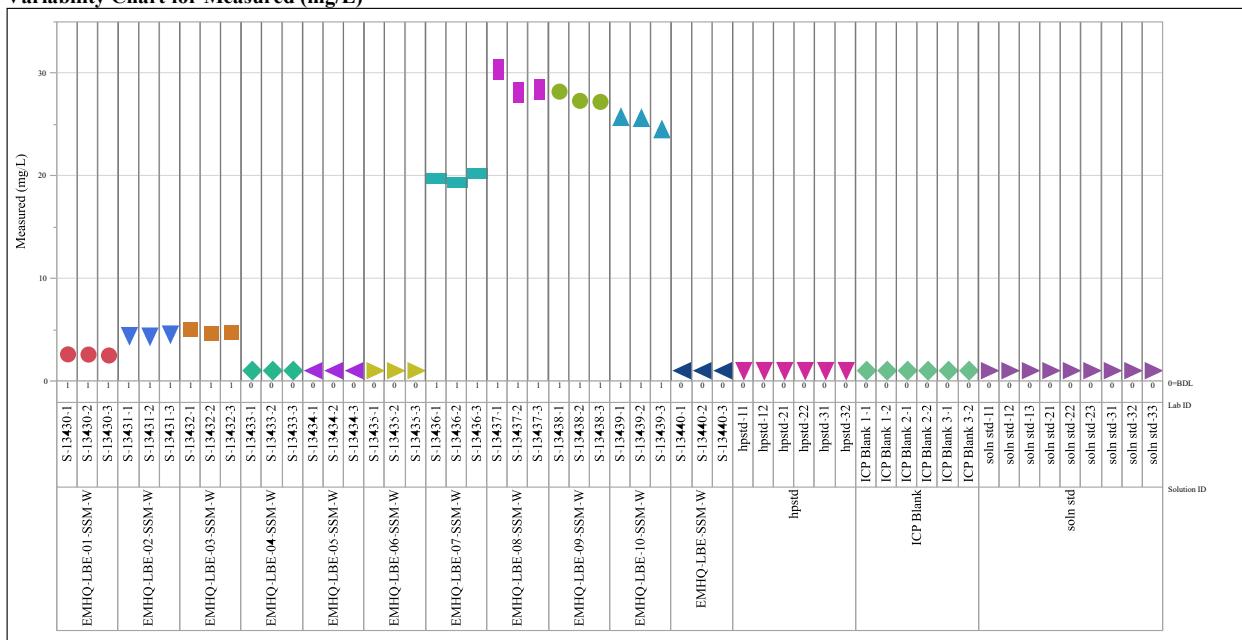


Exhibit B-1. Measurements of SSM Wash Solutions by Solution ID by Analyte (continued)

Analyte=Cr, Analysis=ICP

Variability Chart for Measured (mg/L)



Analyte=F, Analysis=IC

Variability Chart for Measured (mg/L)

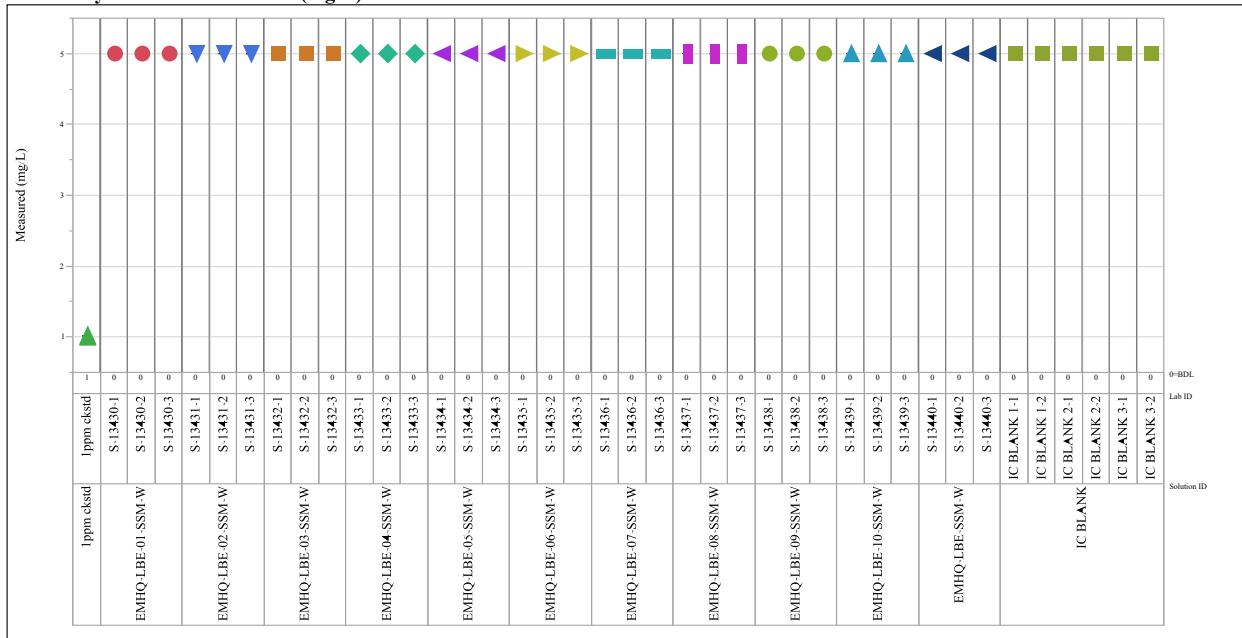
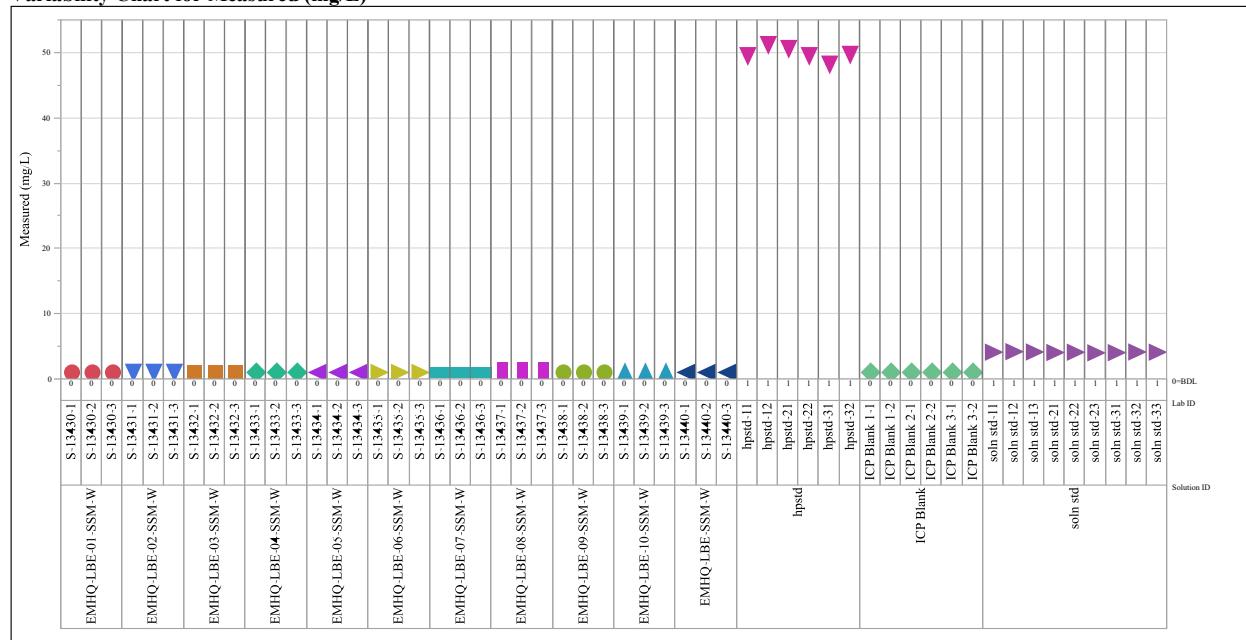


Exhibit B-1. Measurements of SSM Wash Solutions by Solution ID by Analyte (continued)

Analyte=Fe, Analysis=ICP

Variability Chart for Measured (mg/L)



Analyte=K, Analysis=ICP

Variability Chart for Measured (mg/L)

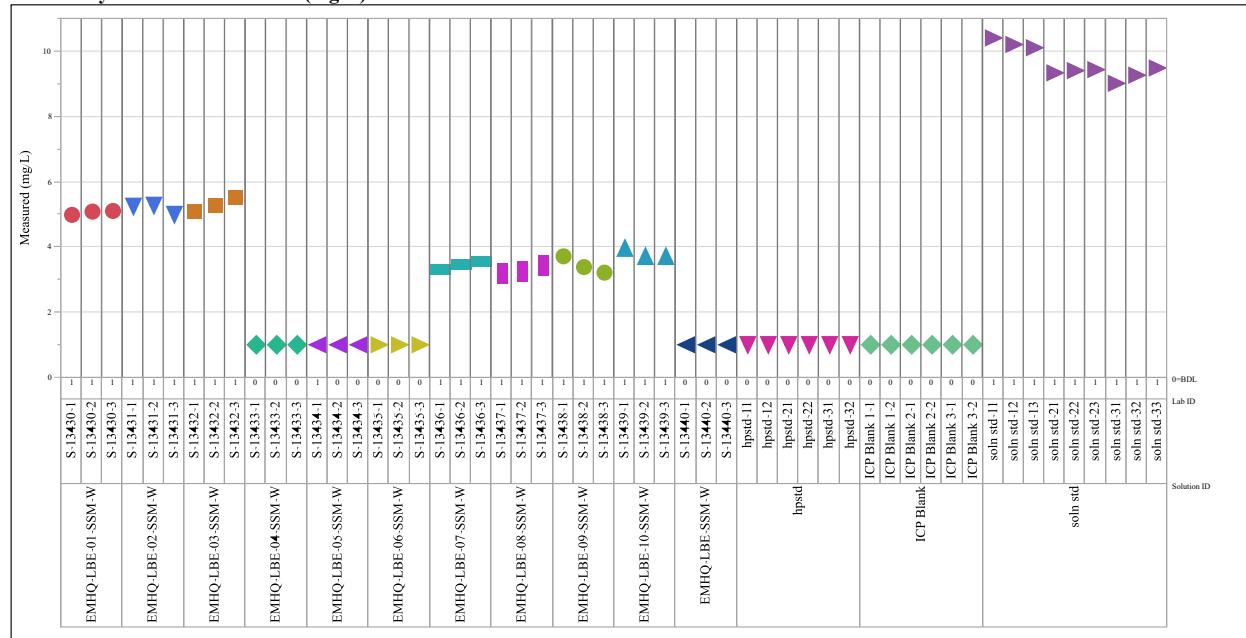
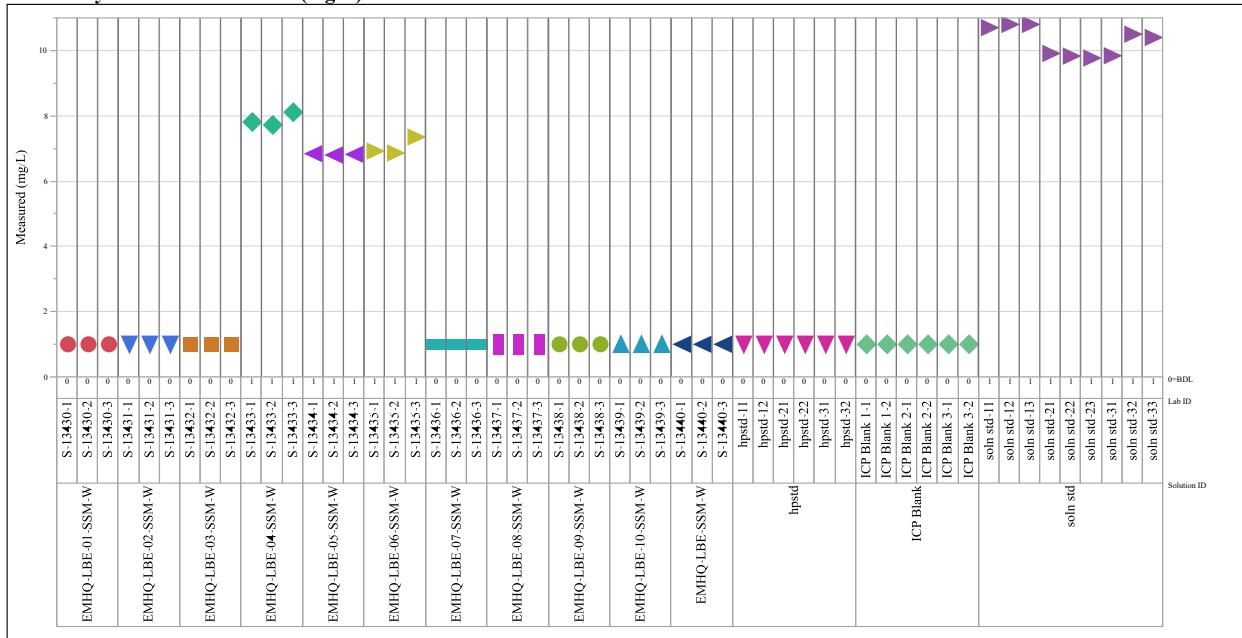


Exhibit B-1. Measurements of SSM Wash Solutions by Solution ID by Analyte (continued)

Analyte=Li, Analysis=ICP

Variability Chart for Measured (mg/L)



Analyte=Mg, Analysis=ICP

Variability Chart for Measured (mg/L)

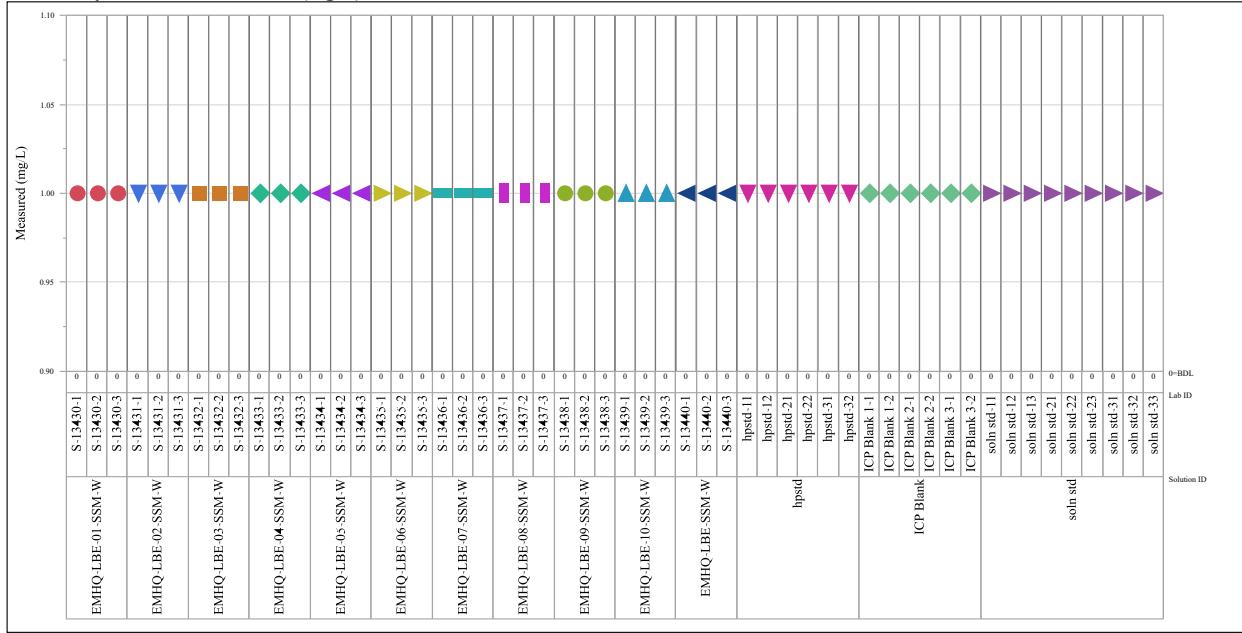
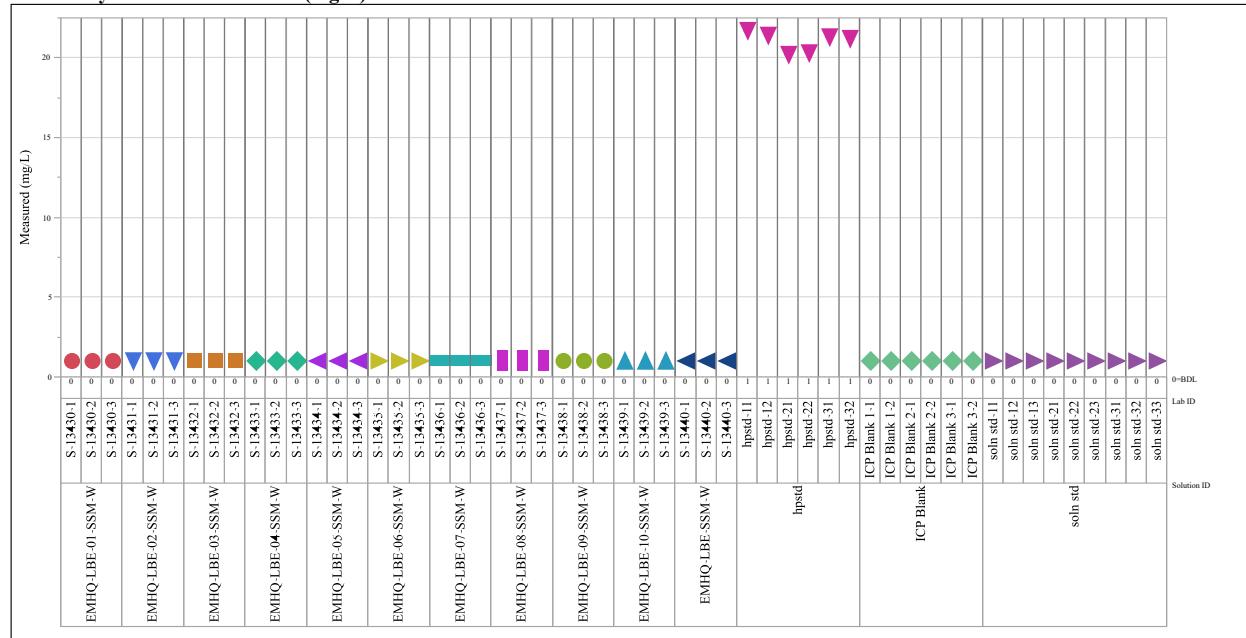


Exhibit B-1. Measurements of SSM Wash Solutions by Solution ID by Analyte (continued)

Analyte=Mn, Analysis=ICP

Variability Chart for Measured (mg/L)



Analyte=Na, Analysis=ICP

Variability Chart for Measured (mg/L)

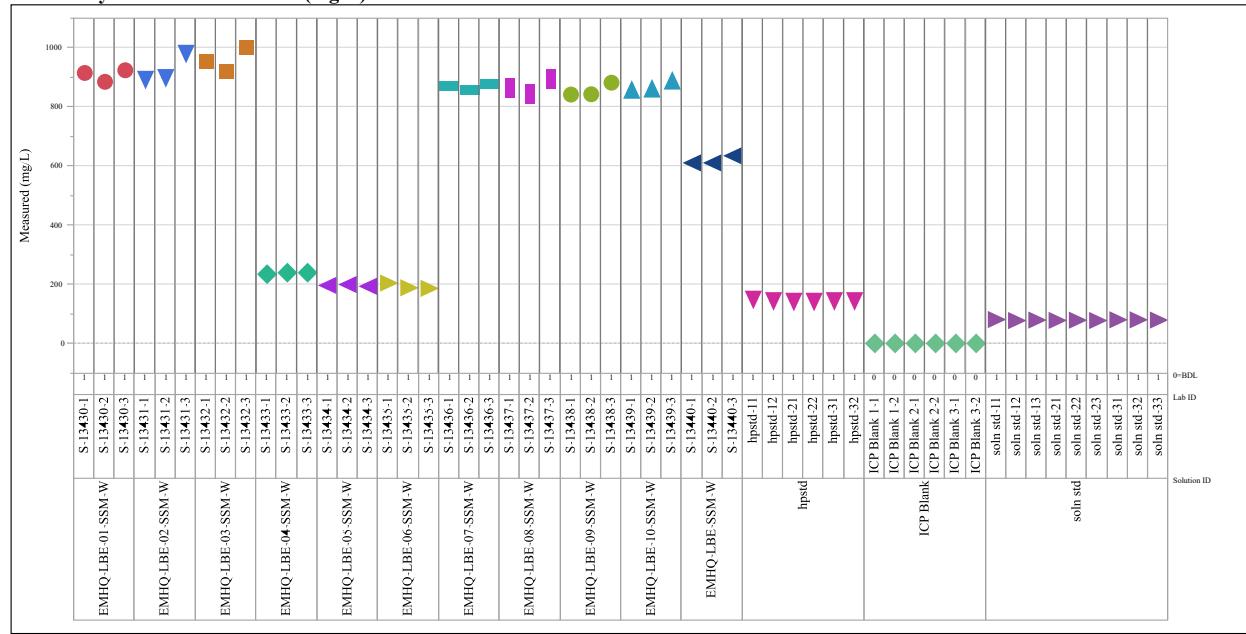
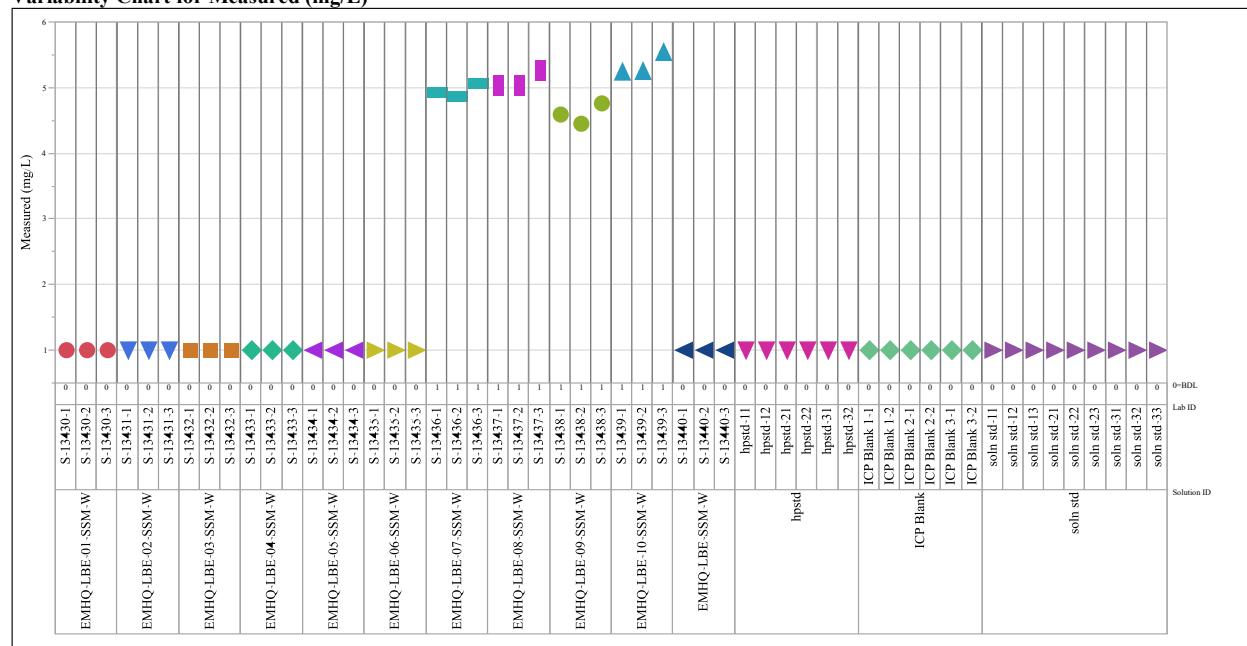


Exhibit B-1. Measurements of SSM Wash Solutions by Solution ID by Analyte (continued)

Analyte=P, Analysis=ICP

Variability Chart for Measured (mg/L)

Analyte=PO₄, Analysis=IC

Variability Chart for Measured (mg/L)

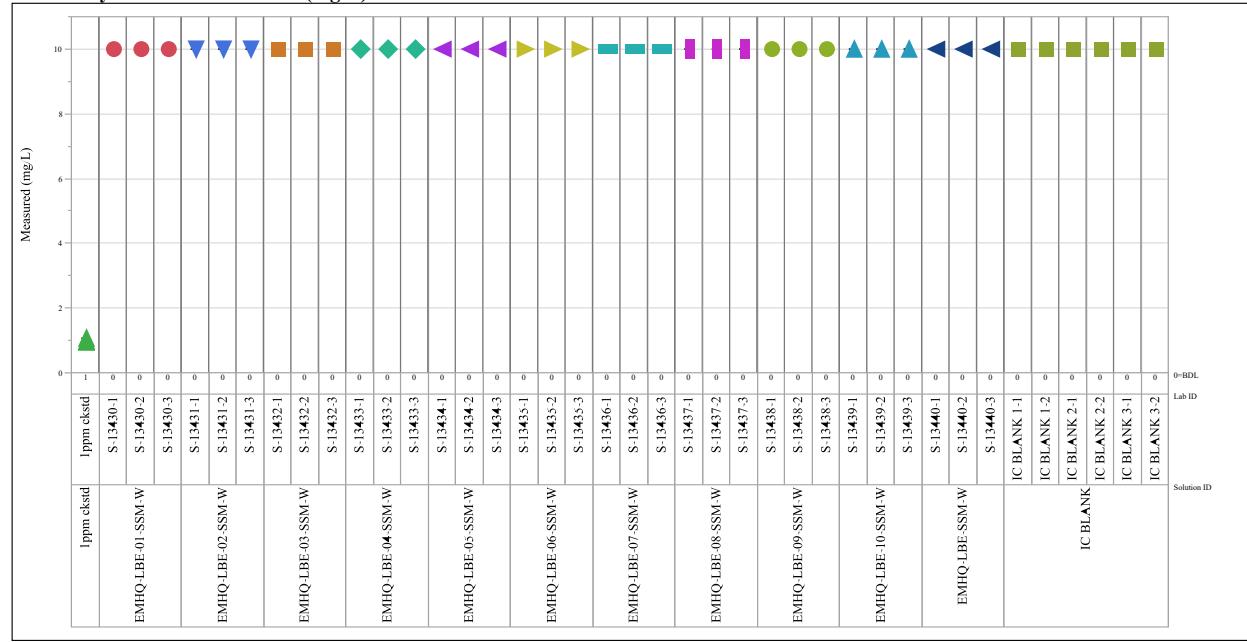
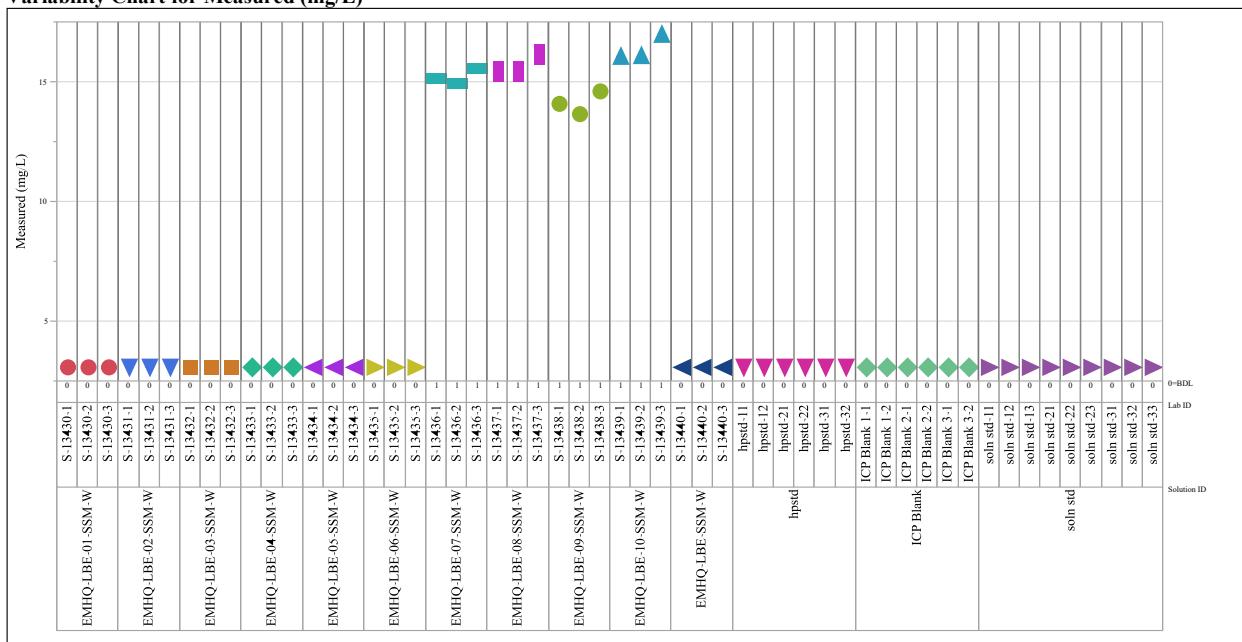


Exhibit B-1. Measurements of SSM Wash Solutions by Solution ID by Analyte (continued)

Analyte=PO₄, Analysis=ICP

Variability Chart for Measured (mg/L)



Analyte=S, Analysis=ICP

Variability Chart for Measured (mg/L)

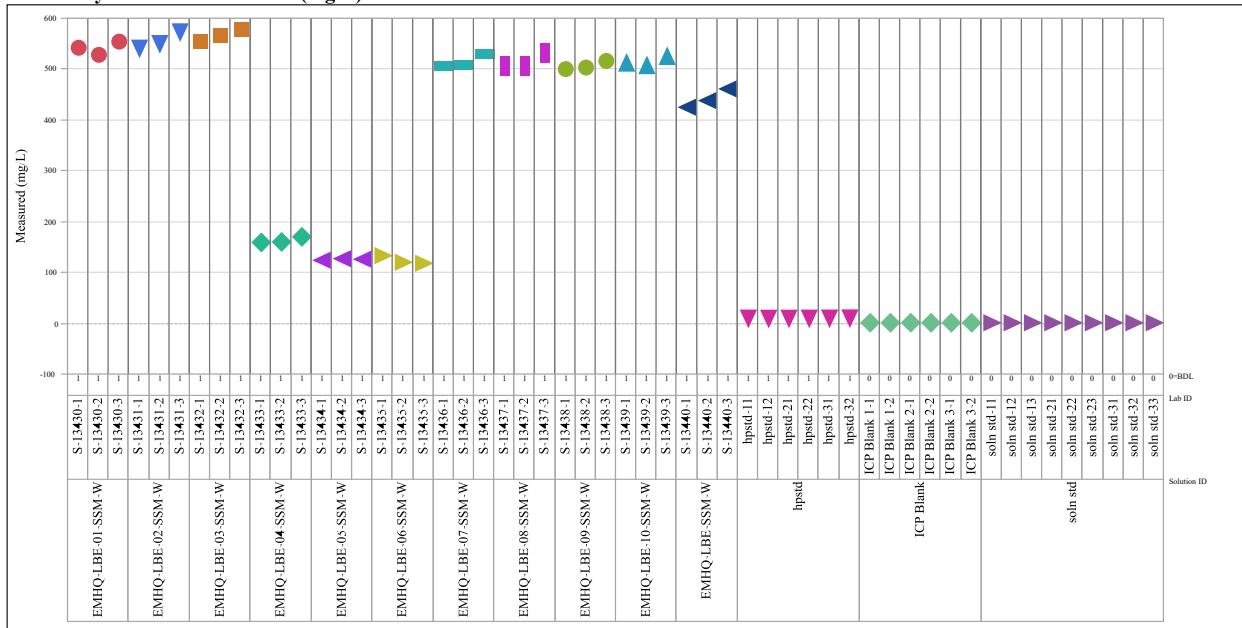
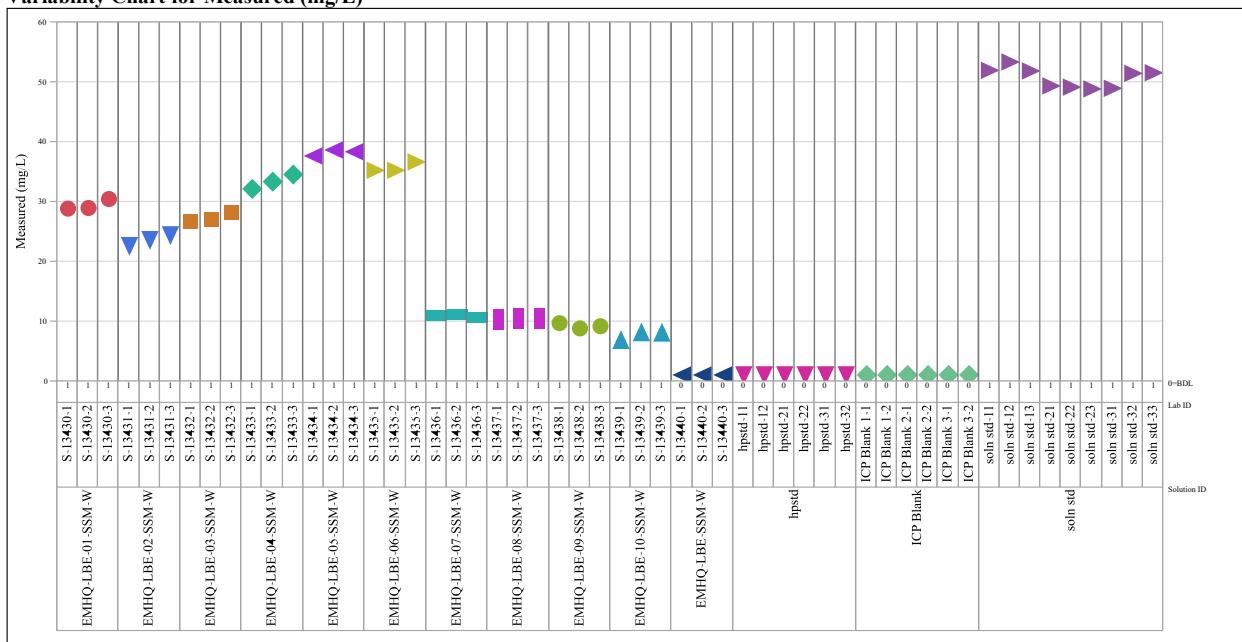


Exhibit B-1. Measurements of SSM Wash Solutions by Solution ID by Analyte (continued)

Analyte=Si, Analysis=ICP

Variability Chart for Measured (mg/L)



Analyte=Sn, Analysis=ICP

Variability Chart for Measured (mg/L)

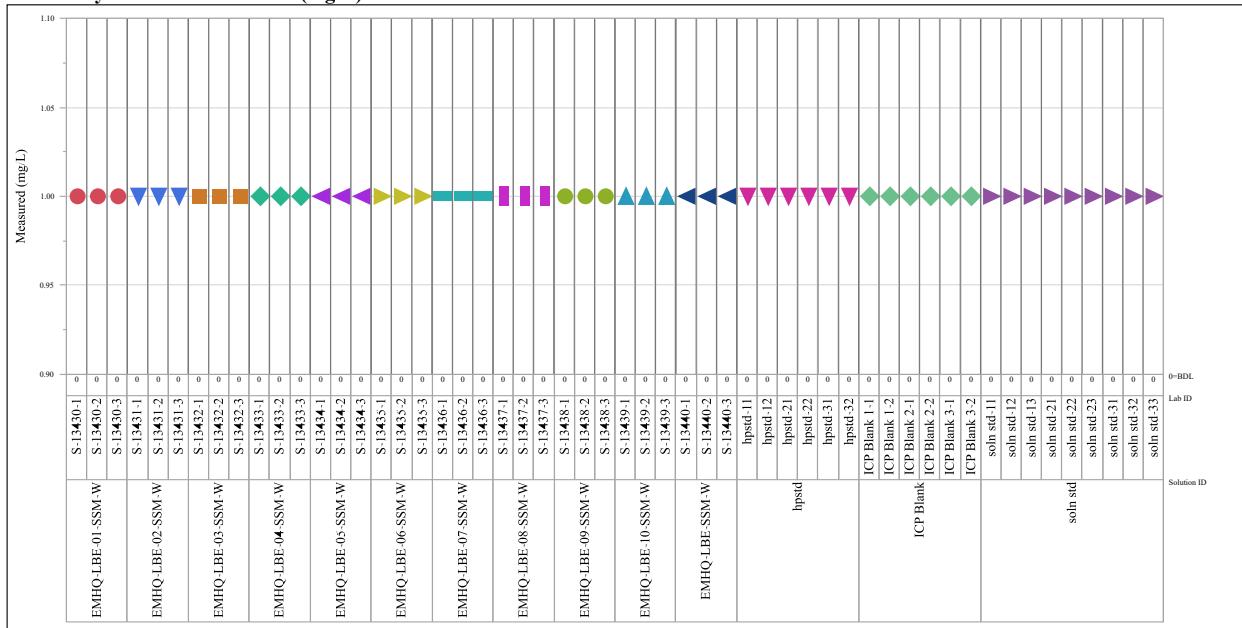
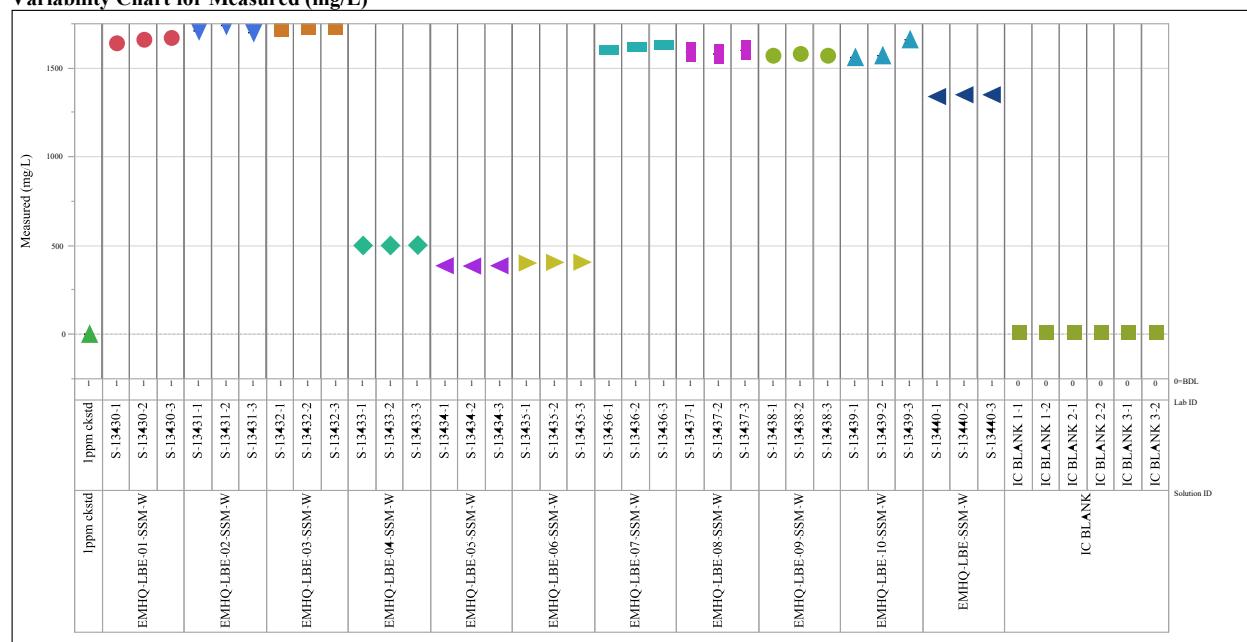


Exhibit B-1. Measurements of SSM Wash Solutions by Solution ID by Analyte (continued)

Analyte=SO₄, Analysis=IC

Variability Chart for Measured (mg/L)

Analyte=SO₄, Analysis=ICP

Variability Chart for Measured (mg/L)

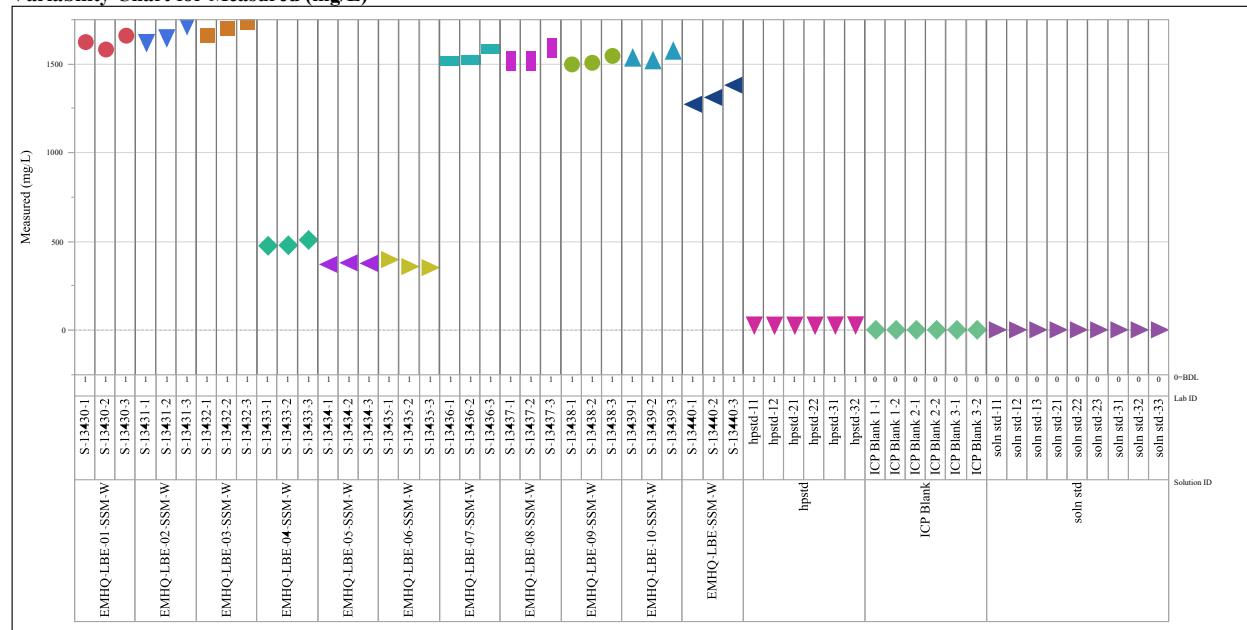
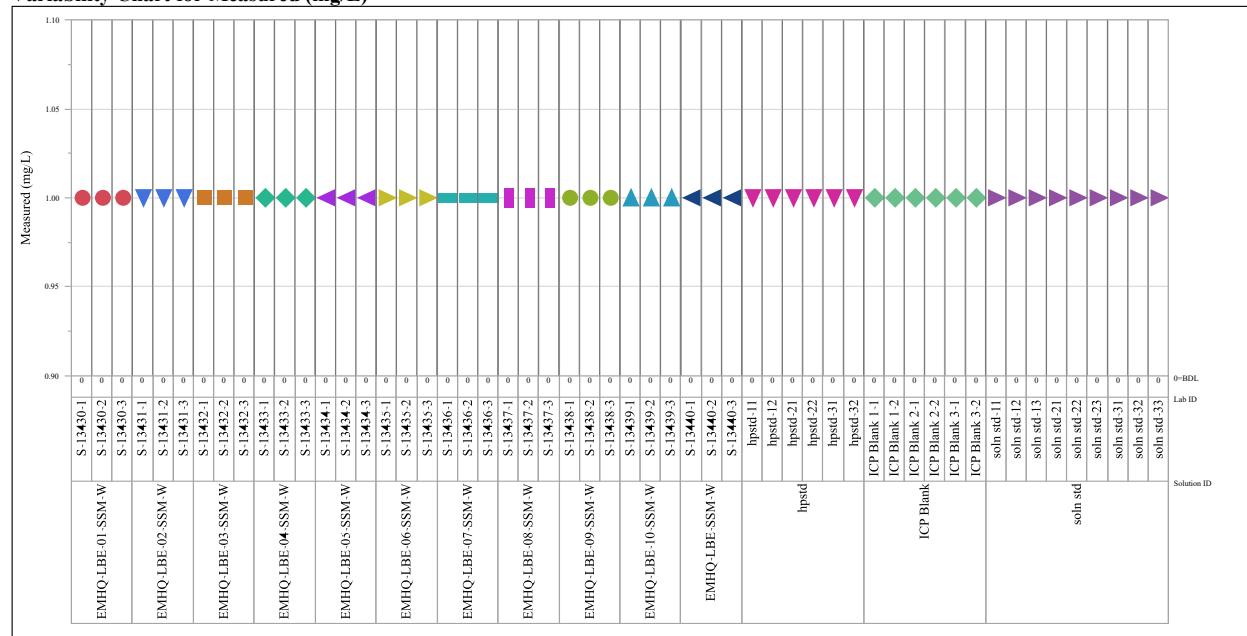


Exhibit B-1. Measurements of SSM Wash Solutions by Solution ID by Analyte (continued)

Analyte=Ti, Analysis=ICP

Variability Chart for Measured (mg/L)



Analyte=V, Analysis=ICP

Variability Chart for Measured (mg/L)

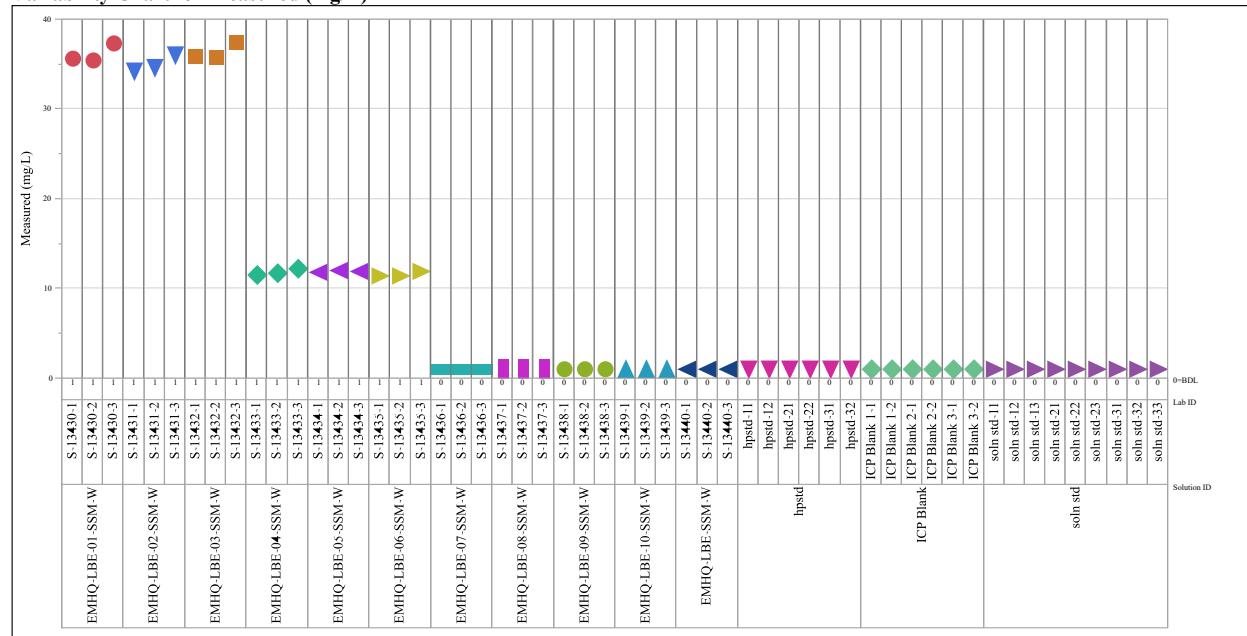


Exhibit B-1. Measurements of SSM Wash Solutions by Solution ID by Analyte (continued)

Analyte=Zr, Analysis=ICP

Variability Chart for Measured (mg/L)

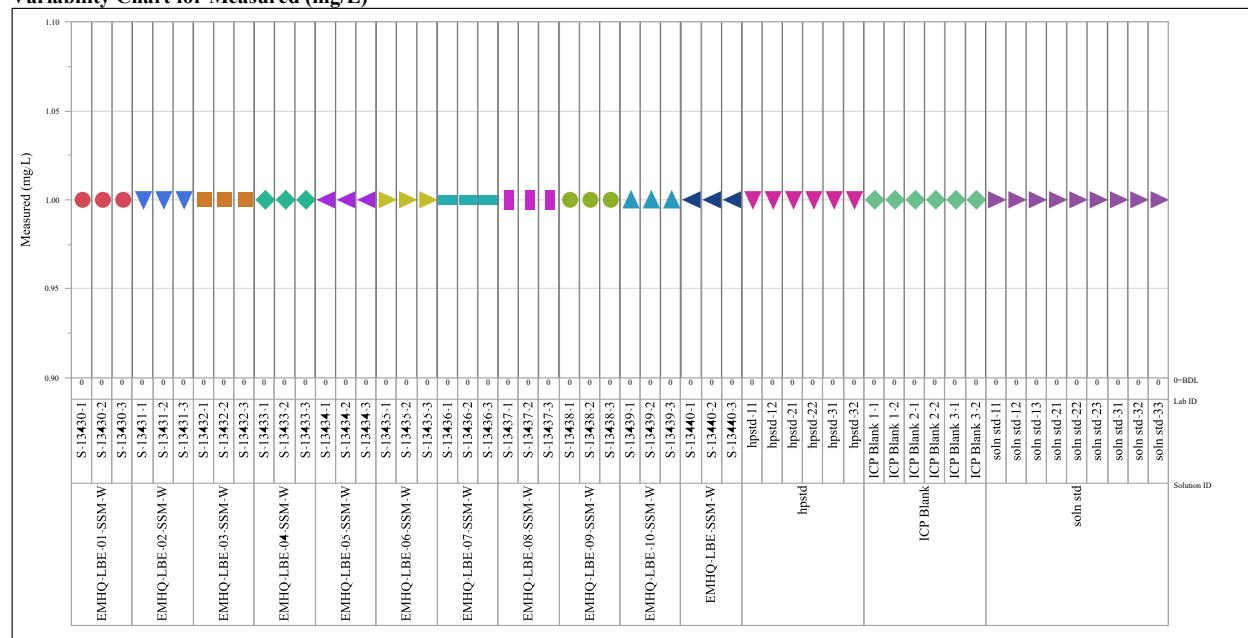
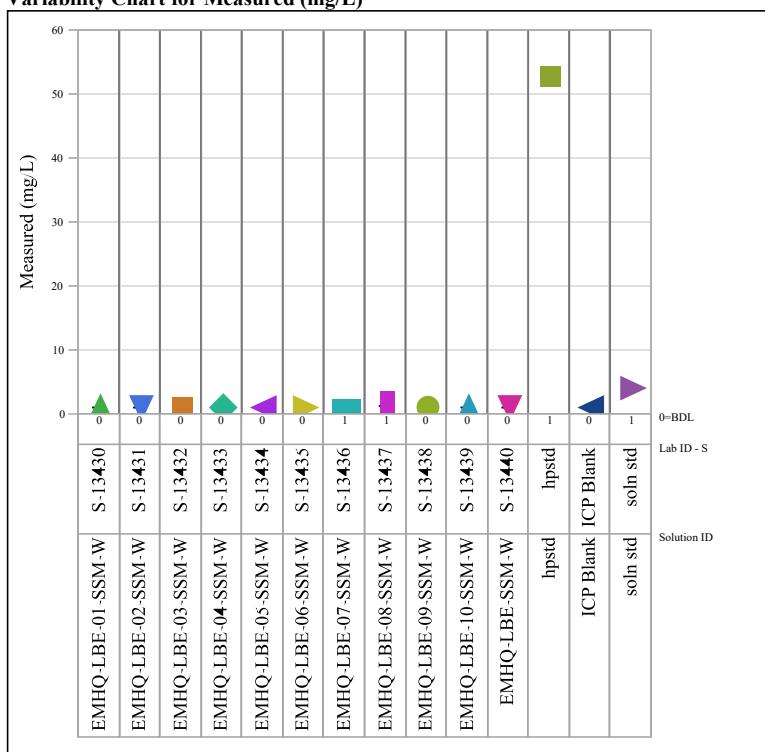


Exhibit B-2. Average Measurements of the SSM Wash Solutions

Analyte=Al, Analysis=ICP

Variability Chart for Measured (mg/L)



Analyte=B, Analysis=ICP

Variability Chart for Measured (mg/L)

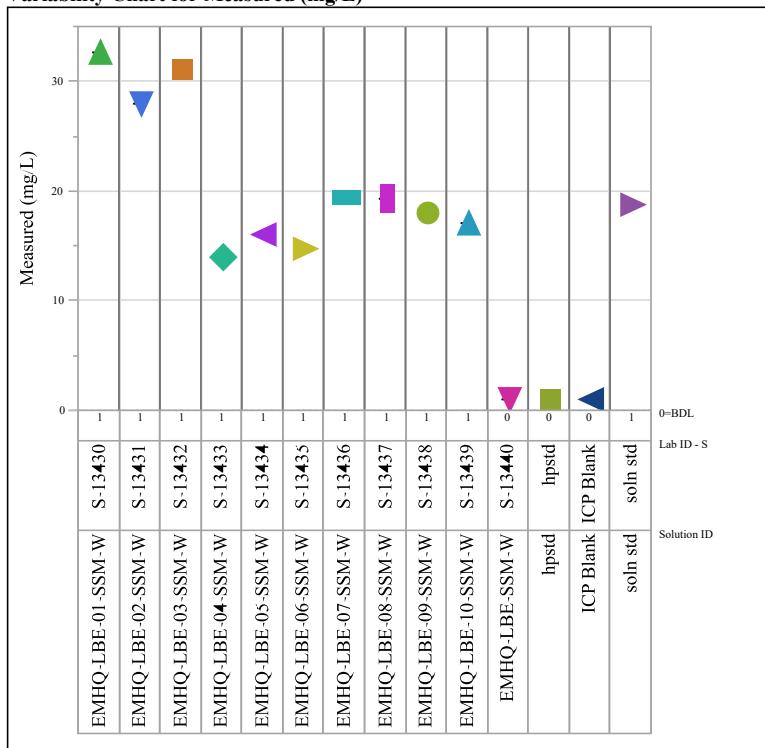
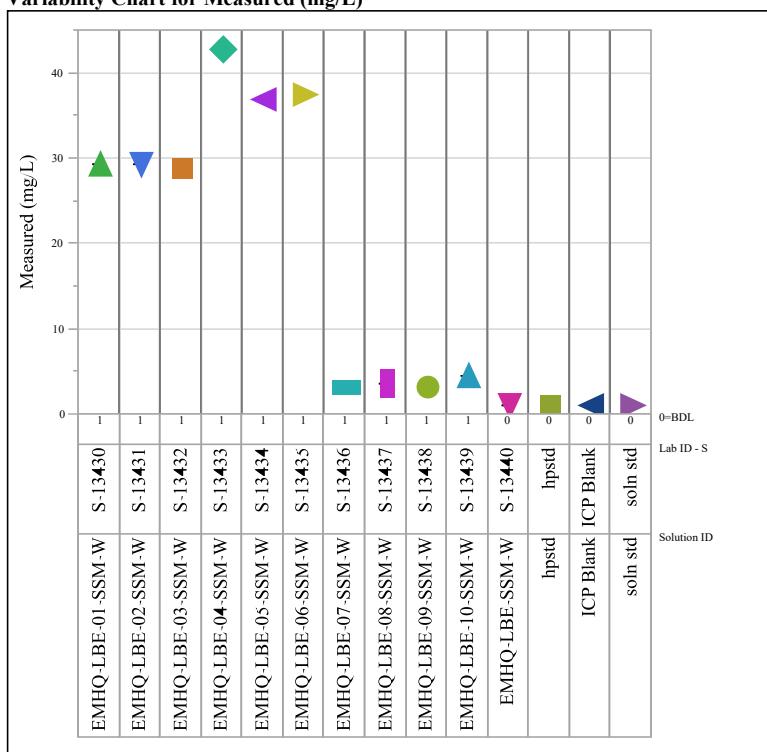


Exhibit B-2. Average Measurements of the SSM Wash Solutions (continued)

Analyte=Ca, Analysis=ICP

Variability Chart for Measured (mg/L)

Analyte=Cl⁻, Analysis=IC

Variability Chart for Measured (mg/L)

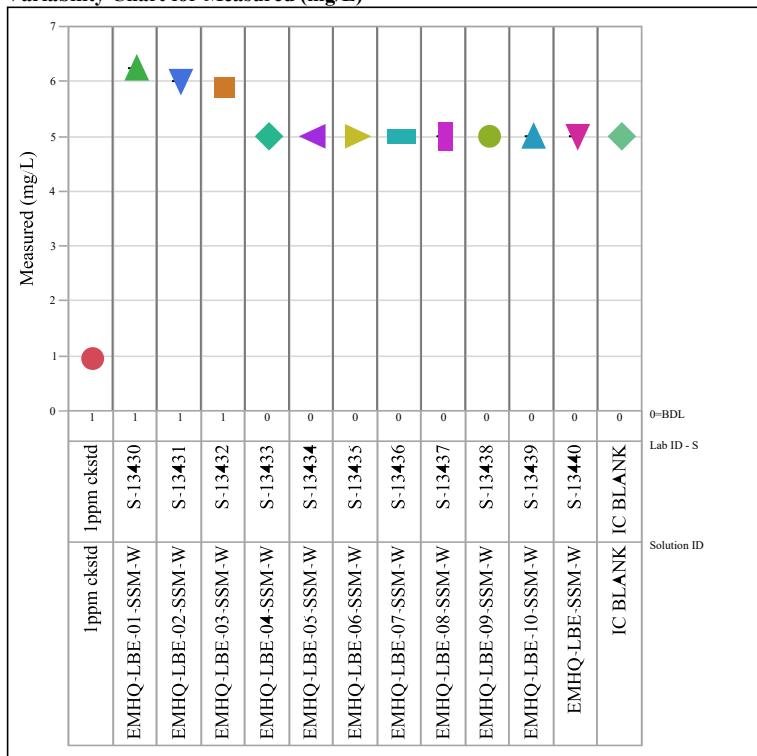
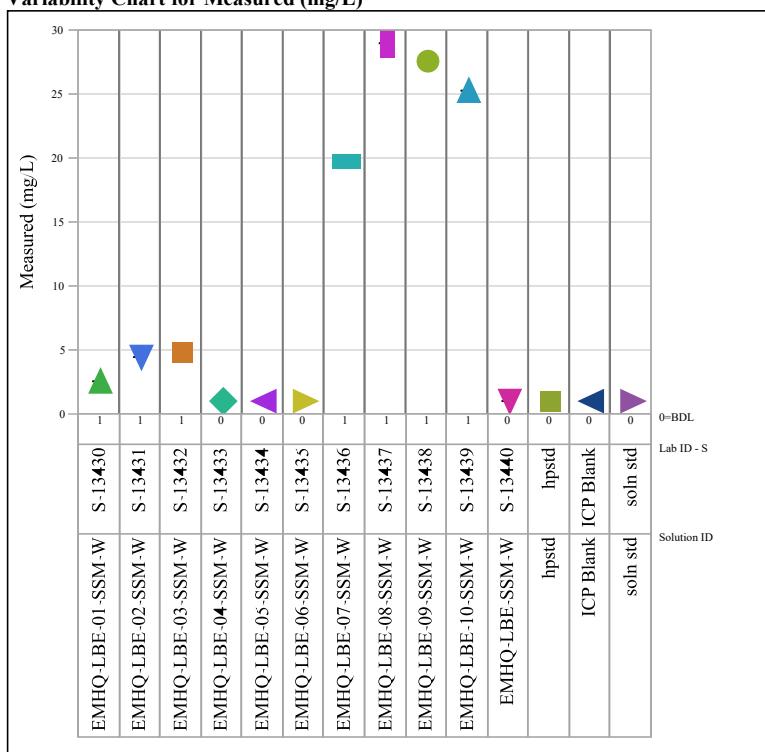


Exhibit B-2. Average Measurements of the SSM Wash Solutions (continued)

Analyte=Cr, Analysis=ICP

Variability Chart for Measured (mg/L)



Analyte=F, Analysis=IC

Variability Chart for Measured (mg/L)

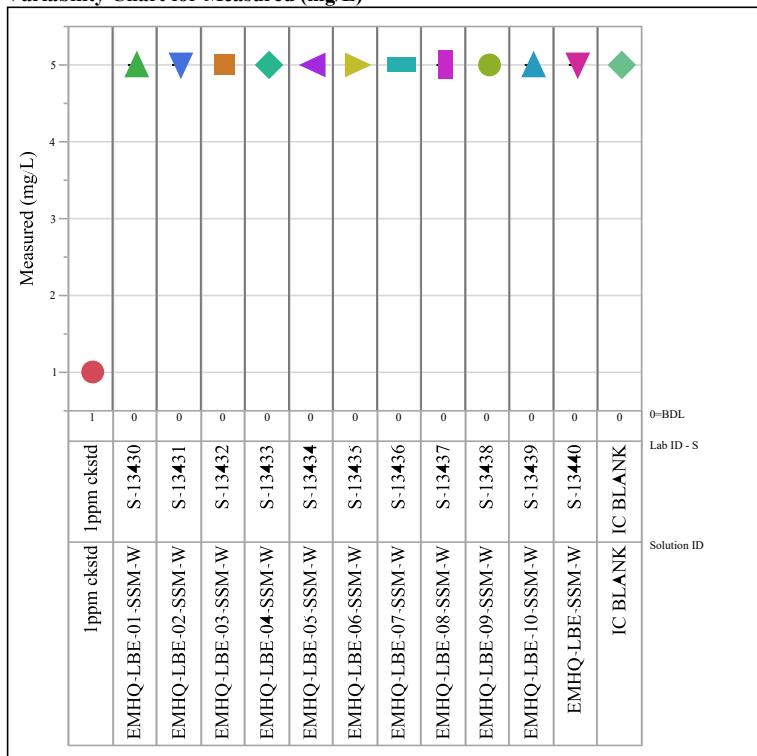
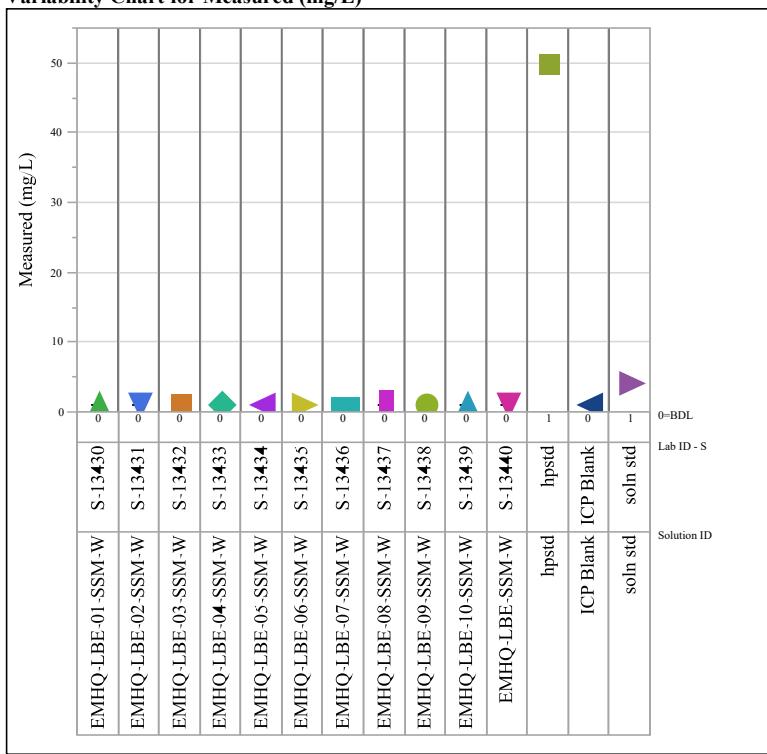


Exhibit B-2. Average Measurements of the SSM Wash Solutions (continued)

Analyte=Fe, Analysis=ICP

Variability Chart for Measured (mg/L)



Analyte=K, Analysis=ICP

Variability Chart for Measured (mg/L)

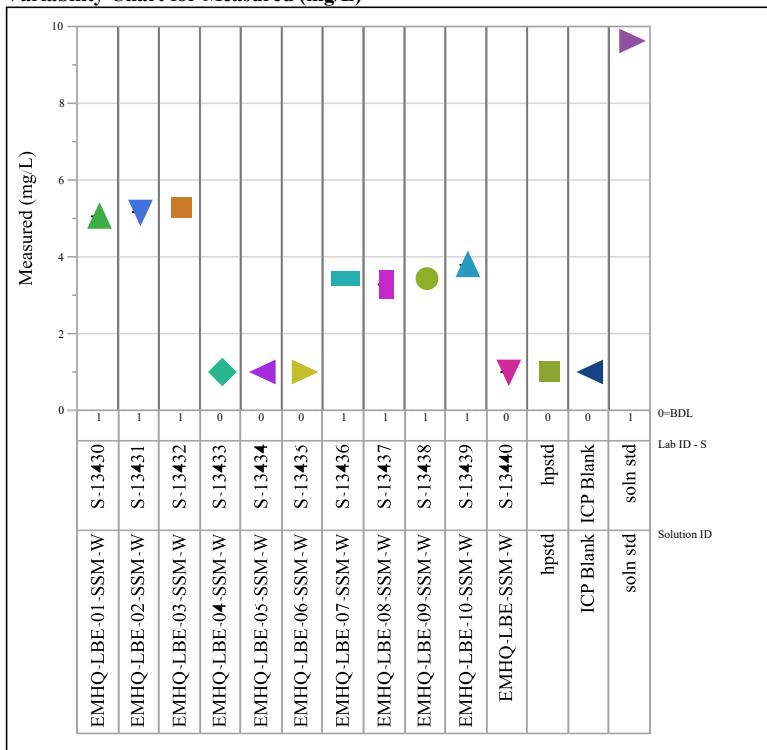
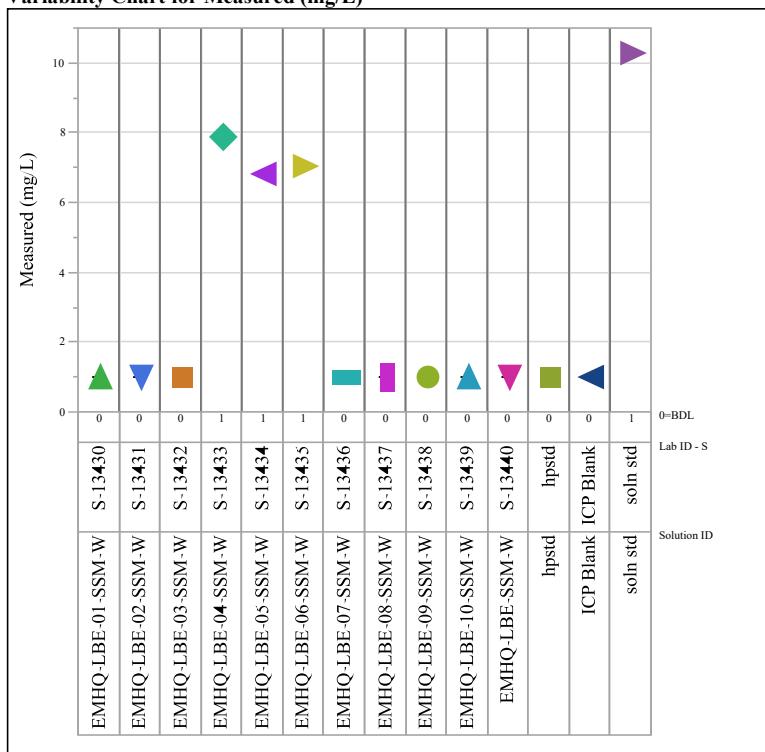


Exhibit B-2. Average Measurements of the SSM Wash Solutions (continued)

Analyte=Li, Analysis=ICP

Variability Chart for Measured (mg/L)



Analyte=Mg, Analysis=ICP

Variability Chart for Measured (mg/L)

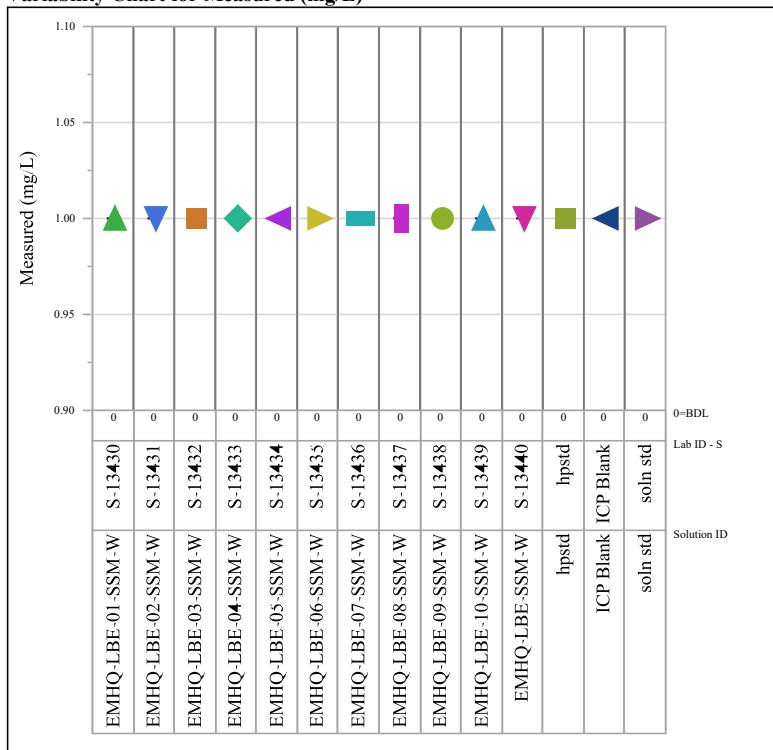
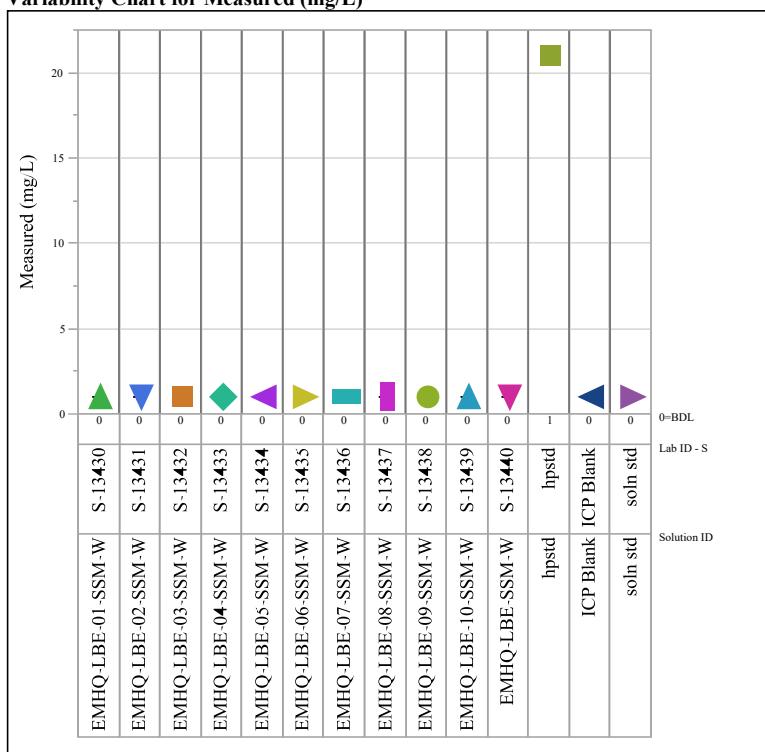


Exhibit B-2. Average Measurements of the SSM Wash Solutions (continued)

Analyte=Mn, Analysis=ICP

Variability Chart for Measured (mg/L)



Analyte=Na, Analysis=ICP

Variability Chart for Measured (mg/L)

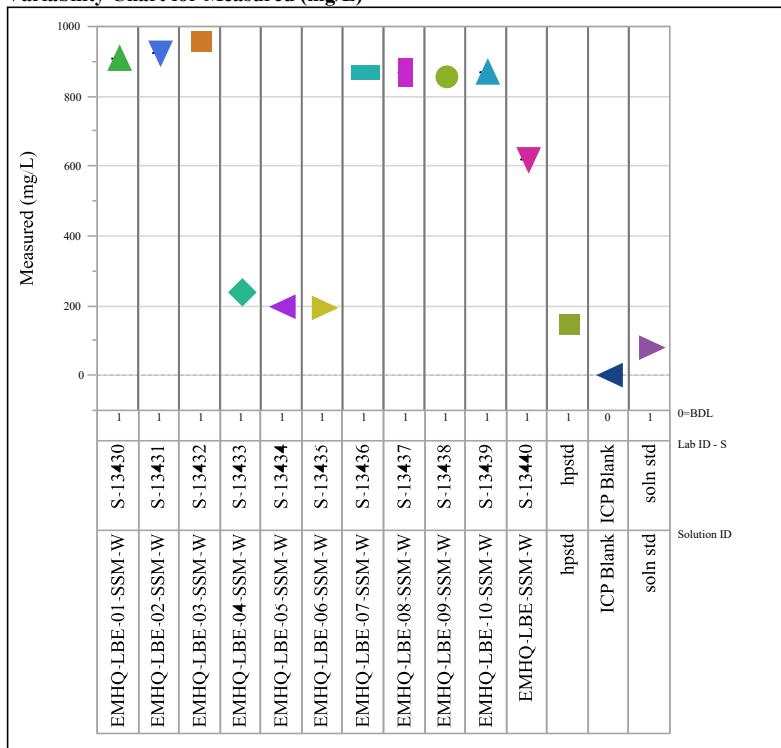
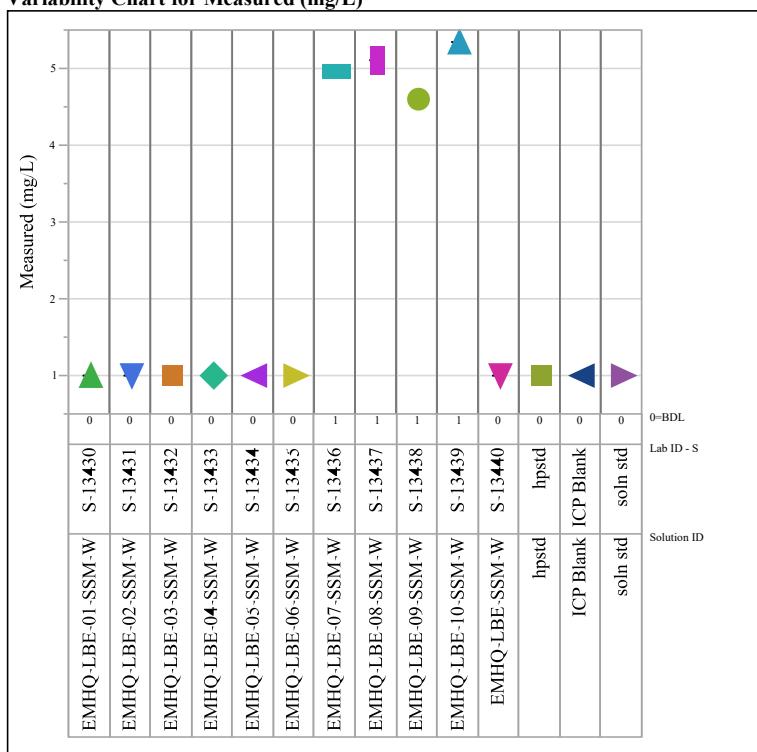


Exhibit B-2. Average Measurements of the SSM Wash Solutions (continued)

Analyte=P, Analysis=ICP

Variability Chart for Measured (mg/L)

Analyte=PO₄, Analysis=IC

Variability Chart for Measured (mg/L)

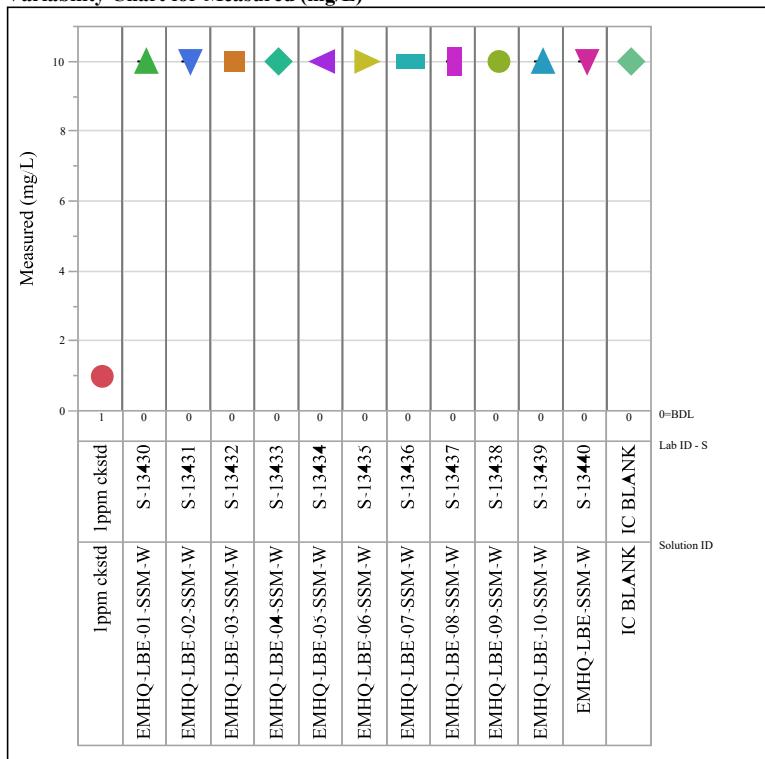
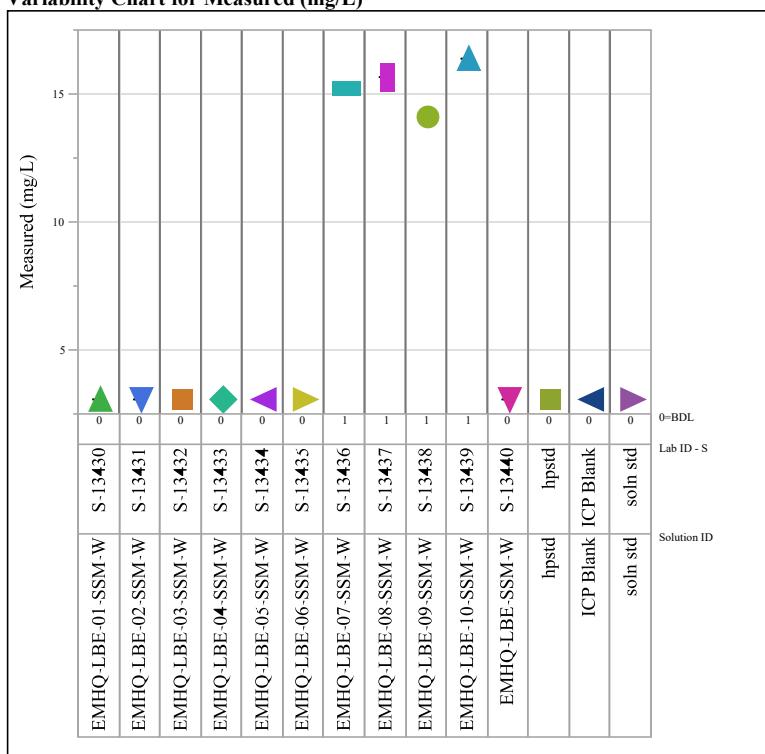


Exhibit B-2. Average Measurements of the SSM Wash Solutions (continued)Analyte=PO₄, Analysis=ICP

Variability Chart for Measured (mg/L)



Analyte=S, Analysis=ICP

Variability Chart for Measured (mg/L)

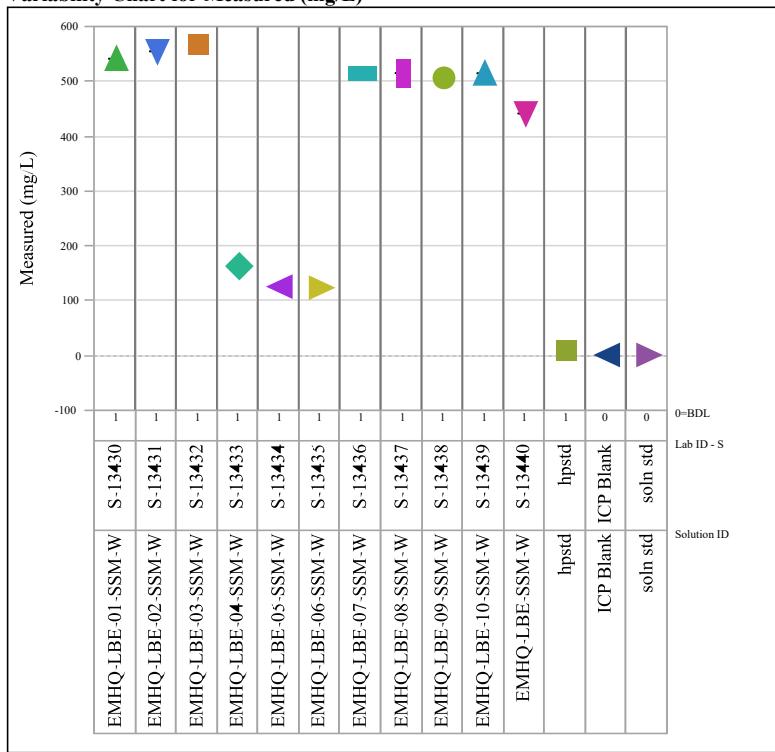
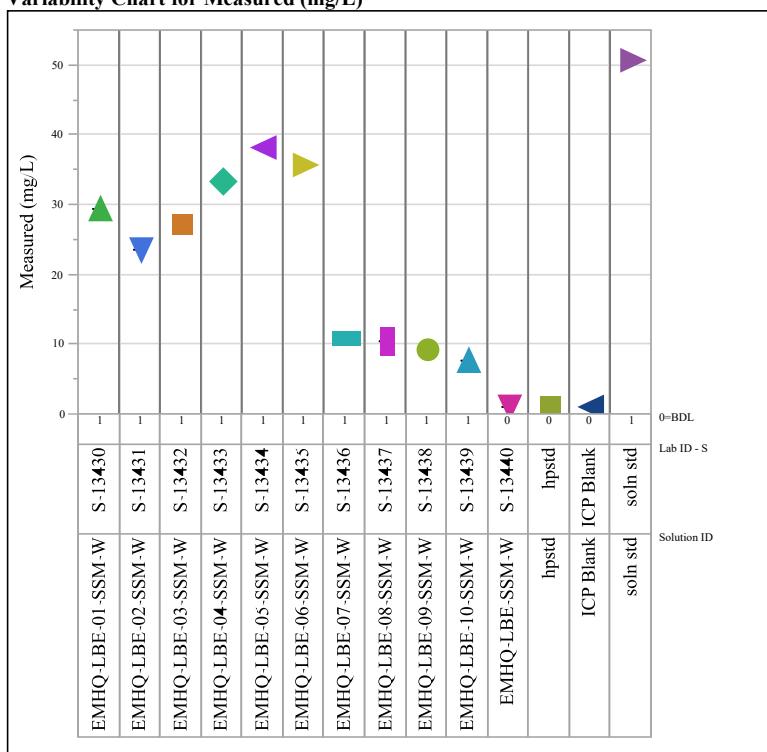


Exhibit B-2. Average Measurements of the SSM Wash Solutions (continued)

Analyte=Si, Analysis=ICP

Variability Chart for Measured (mg/L)



Analyte=Sn, Analysis=ICP

Variability Chart for Measured (mg/L)

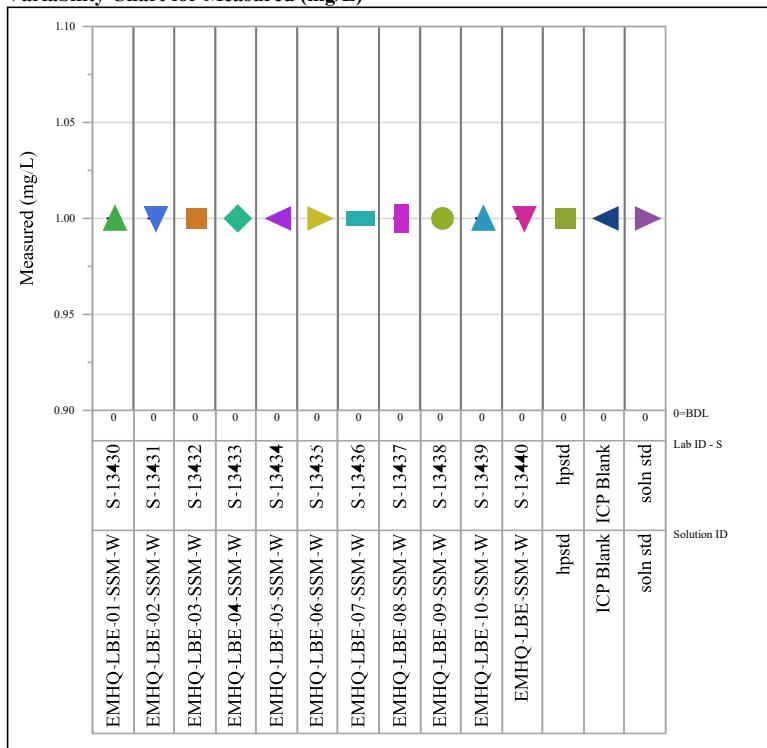
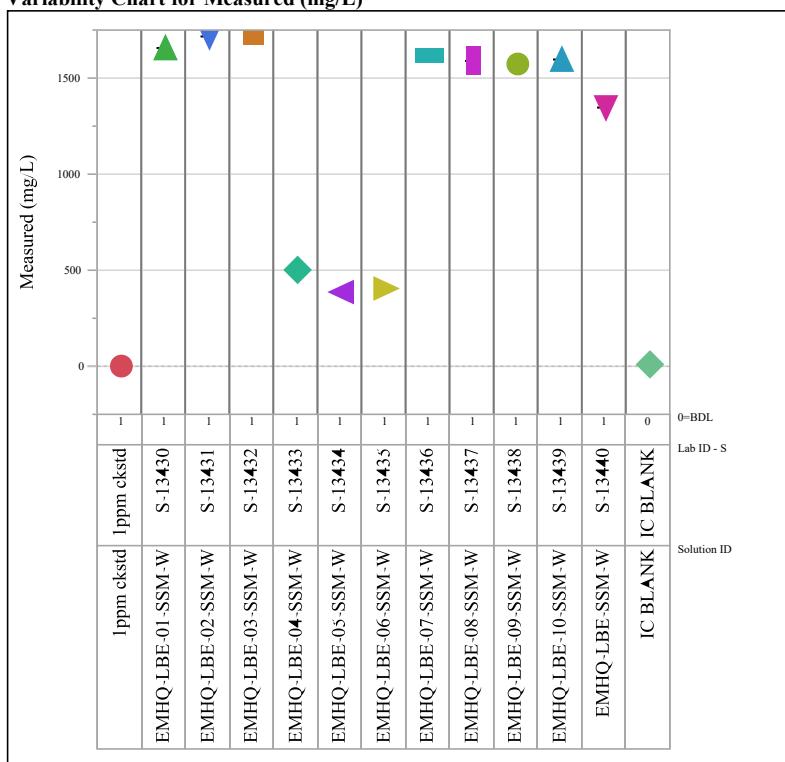


Exhibit B-2. Average Measurements of the SSM Wash Solutions (continued)Analyte=SO₄, Analysis=IC

Variability Chart for Measured (mg/L)

Analyte=SO₄, Analysis=ICP

Variability Chart for Measured (mg/L)

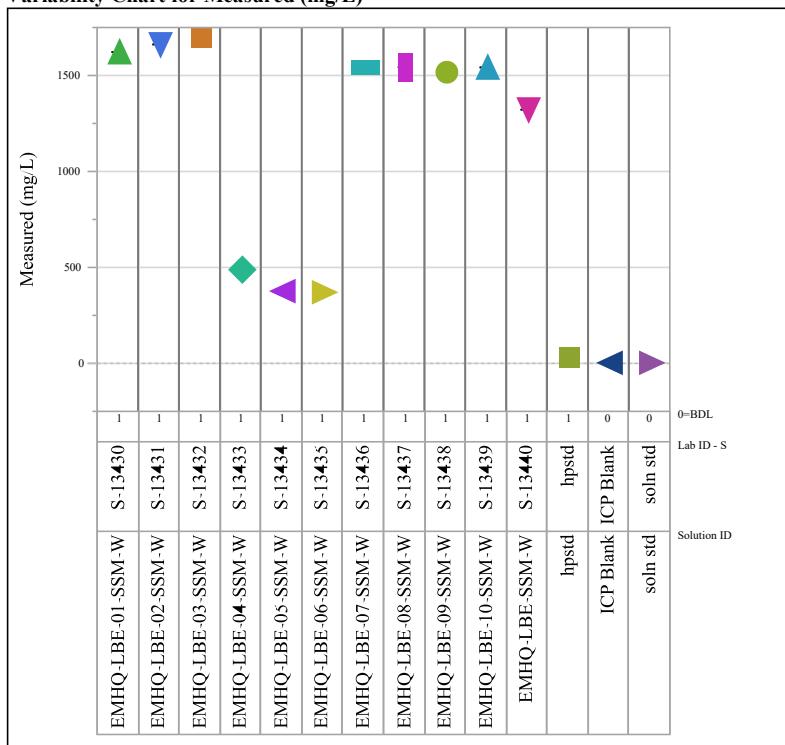
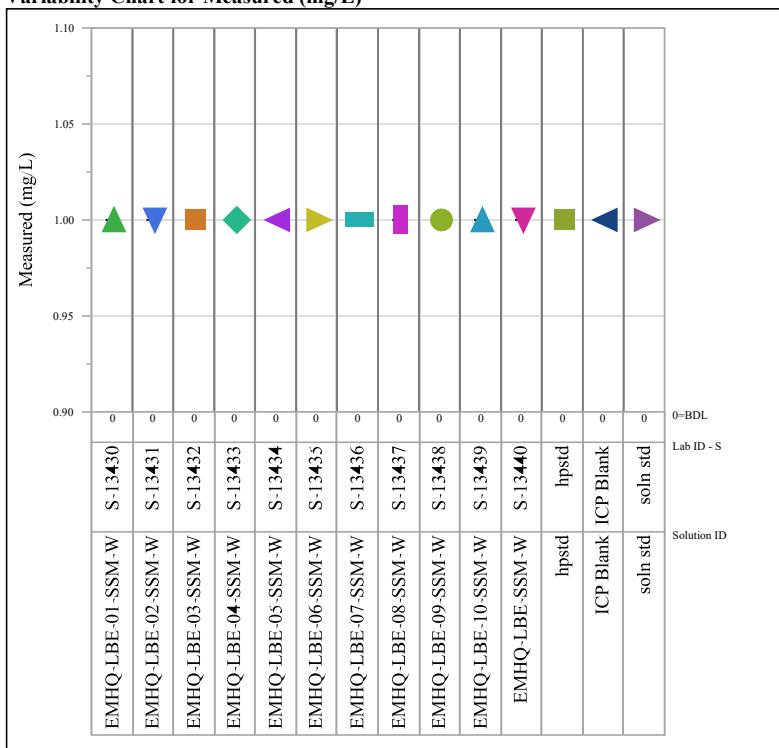


Exhibit B-2. Average Measurements of the SSM Wash Solutions (continued)

Analyte=Ti, Analysis=ICP

Variability Chart for Measured (mg/L)



Analyte=V, Analysis=ICP

Variability Chart for Measured (mg/L)

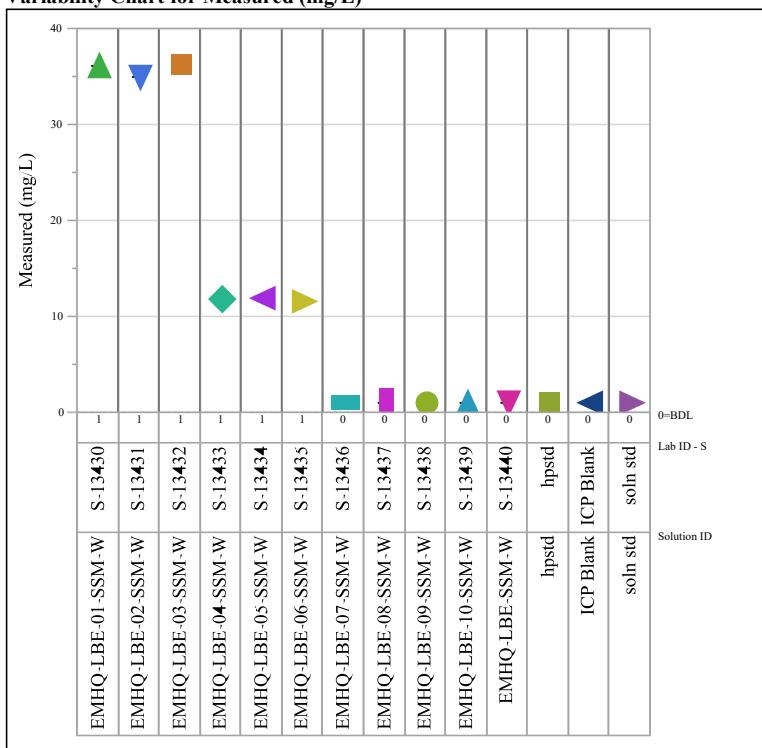
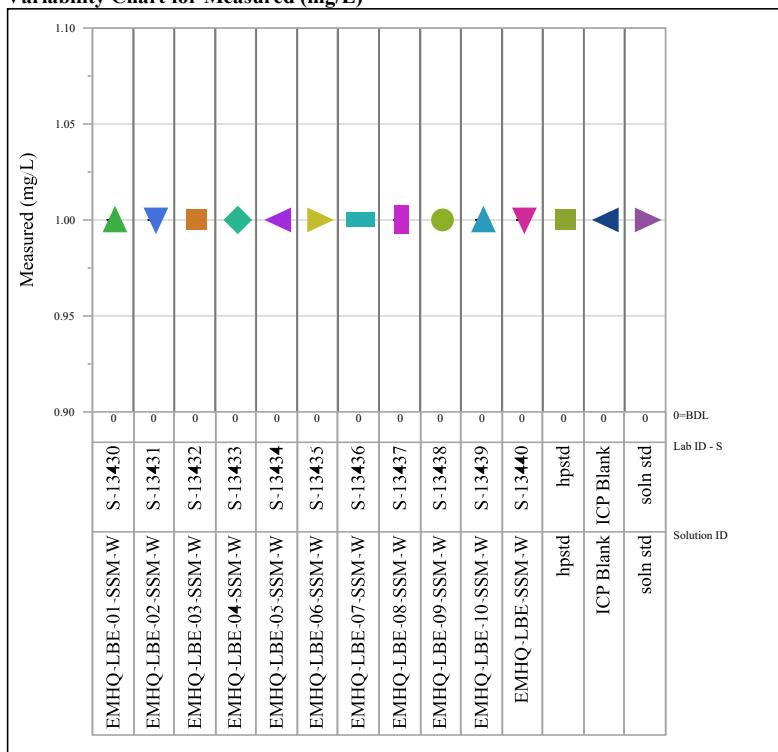


Exhibit B-2. Average Measurements of the SSM Wash Solutions (continued)

Analyte=Zr, Analysis=ICP

Variability Chart for Measured (mg/L)



Distribution:

Jake.Amoroso@srln.doe.gov
CJ.Bannochie@srln.doe.gov
Alex.Cozzi@srln.doe.gov
Charles.Crawford@srln.doe.gov
Elaine_N_Diaz@orp.doe.gov
William.C.Eaton@pnnl.gov
A.Fellinger@srln.doe.gov
Holly.Hall@srln.doe.gov
Erich.Hansen@srln.doe.gov
Connie.Herman@srln.doe.gov
Anthony.Howe@srln.doe.gov
Madison.Hsieh@srln.doe.gov
Dennis.Jackson@srln.doe.gov
Fabienne.Johnson@srln.doe.gov
Albert_A_Kruger@orp.doe.gov
Christine.Langton@srln.doe.gov
Brady.Lee@srln.doe.gov
Charmayne.Lonergan@pnnl.gov
Joseph.Manna@srln.doe.gov
Daniel.McCabe@srln.doe.gov
Kandice.Miles@srln.doe.gov
Gregg.Morgan@srln.doe.gov
Frank.Pennebaker@srln.doe.gov
William.Ramsey@srln.doe.gov
Whitney.Riley@srln.doe.gov
Renee.Russell@pnnl.gov
Eric.Skidmore@srln.doe.gov
Anna.Stanfield@srln.doe.gov
Michael.Stone@srln.doe.gov
John.Vienna@pnnl.gov
Boyd.Wiedenman@srln.doe.gov
Richard.Wyrwas@srln.doe.gov
Records Administration (EDWS)