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12 June 2018

SRNL-L3100-2017-00033, Rev. 1

TO: E. J. FREED

FROM: C. L. CRAWFORD

Results for the First Quarter Calendar Year 2017 Tank 50H Salt Solution Sample

Approved by:

S. H. Reboul, Technical Reviewer, per E7, 2.60

Date

B. J. Wiedenman, Manager, Advanced Characterization & Processing Date

REVISION LOG

Revision 1: This Memorandum is revised to reflect changes in the reported monomethyl mercury data and the ammonium values in Table 1 and Table 5

SUMMARY

In this memorandum, the chemical and radionuclide contaminant results from the First Quarter Calendar Year 2017 (CY17) sample of Tank 50H salt solution are presented in tabulated form. The First Quarter CY17 Tank 50H samples [a 200 mL sample obtained 6" below the surface (HTF-50-17-7) and a 1 L sample obtained 66" from the tank bottom (HTF-50-17-8)] were obtained on January 15, 2017 and received at Savannah River National Laboratory (SRNL) on January 16, 2017.¹ Prior to obtaining the samples from Tank 50H, a single pump was run at least 4.4 hours and the samples were pulled immediately after pump shut down.¹ All volatile organic analysis (VOA) and semi-volatile organic analysis (SVOA) were performed on the surface sample and all other analyses were performed on the variable depth sample. The information from this characterization will be used by Savannah River Remediation (SRR) for the transfer of aqueous waste from Tank 50H to the Saltstone Production Facility, where the waste will be treated and disposed of in the Saltstone Disposal Facility. This memorandum compares results, where applicable, to Saltstone Waste Acceptance Criteria (WAC) limits and targets.² The chemical and radionuclide contaminant results from the characterization of the First Quarter CY17 sampling of Tank 50H were requested by SRR personnel³ and details of the testing are presented in the SRNL Task Technical and Quality Assurance Plan (TTQAP).⁴ This memorandum is part of Deliverable 2 from SRR

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request.³ Data pertaining to the regulatory limits for Resource Conservation and Recovery Act (RCRA) metals will be documented at a later time per the TTQAP for the Tank 50H saltstone task.⁴

The following facts pertaining to the WAC are drawn from the analytical results provided in this memorandum:

- WAC targets or limits were met for all analyzed chemical and radioactive contaminants for which the detection limits are below the WAC targets or limits.
- Isopar L has a higher detection limit⁵ compared with the current Saltstone WAC² value of 11 ppm that has been in effect since revision 12 of the WAC dating back to July of 2013.⁶
- Minimum detection limits are reported for ⁹⁴Nb, ²⁴⁷Cm, ²⁴⁹Cf, and ²⁵¹Cf as determined from the minimum detectable activity associated with the radiochemical methods used for these radionuclides. The reported detection limits are above the requested SRR target minimum detection limit concentrations.⁷ However, the reported minimum detection limits reported for the First Quarter CY17 Tank 50H sample for these four radionuclides are all lower than the estimated detection limits initially established by SRNL in 2009.⁸ Thus per guidance from SRR,⁷ SRNL continues to achieve as low as practical detection limits for these radionuclides.

TABLES CONTAINING RESULTS

Unless otherwise stated, all of the concentrations presented in the tables (except upper limits) are averages based on triplicate analyses of the First Quarter CY17 Tank 50H samples. A memorandum reporting the average Cs-137 value has been previously issued.⁹ The standard deviation of each average is also presented. Several of the contaminants were either not detected in the slurry samples or detected at values below the method reporting limit (MRL). For contaminants not detected or detected below the MRL, the result is preceded by a "<", which indicates the result is an upper limit based on the sensitivity of the method used to analyze the individual analyte. If only one value out of the triplicate analysis is above the detection limit, then that single value is reported and noted in the tables. Also, if only two values out of the triplicate analyses are above the detection limit, then the average of those two values is reported and noted in the tables. Data reported for atomic absorption (AA), cold-vapor atomic absorption (CVAA), inductively coupled plasma atomic emission spectroscopy (ICP-AES) and inductively coupled plasma mass spectroscopy (ICP-MS) are derived from the digested Tank 50H supernate by the aqua regia method. All analytical methods shown by the acronyms in the tables for this memorandum have been previously defined in the TTQAP.⁴ Radionuclides reported in Table 3 and Table 4 using the ICP-MS method are converted from a reported mass per volume basis to activity per volume units using the specific activities (Ci/g) reported from the Department of Energy 1996 Integrated Data Base Report.¹⁰

Total mercury was analyzed by SRNL using the Cold Vapor Atomic Absorption (CVAA) method. Other mercury (Hg) speciation data shown in Table 1, Table 2, and Table 5 are taken from previous work.¹¹ These species include elemental mercury (Hg(0)), monomethyl mercury, ethyl mercury, and dimethyl

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mercury. Monomethyl, ethyl, and dimethyl mercury are organomercury species. The concentration values for the organomercury species are calculated from the Hg speciation data on a mg Hg/L basis.¹¹ As a sample calculation for monomethyl mercury, information from Table 1 of reference 11 shows that the reported monomethyl concentration on a mg Hg/L basis is 31.7 mg Hg/L. This value is then multiplied by the formula weight of monomethyl mercury from the WAC² (215.62 g monomethyl mercury/mol) divided by the molecular weight of Hg (200.6 g Hg/mol). Thus the calculated concentration of the species monomethyl mercury is 31.7 mg Hg/L x (215.62 g monomethyl mercury/mol / 200.6 g Hg/mol) = 34.1 mg monomethyl mercury/L.

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<u>Chemical Name (Formula)</u>	<u>Method</u>	<u>Average Concentration</u> (mg/L)	<u>Std. Dev.</u>	<u>WAC Limit</u> (mg/L)
Aluminate (Al(OH)4 ⁻)	ICP-ES	1.82E+04 ^a	1.76E+02	4.08E+05
Ammonium (NH4 ⁺)	IC	< 1.00E+02	NA	2.12E+02
Carbonate (CO ₃ ²⁻)	TIC	1.53E+04 ^b	5.00E+01	1.20E+05
Chloride (Cl ⁻)	IC	4.96E+02	2.01E+01	7.95E+03
Fluoride (F ⁻)	IC	< 1.00E+02	NA	4.07E+03
Free Hydroxide (OH ⁻)	Total Base	3.67E+04 ^b	2.60E+02	1.58E+05
Nitrate (NO3 ⁻)	IC	1.11E+05	3.51E+03	4.37E+05
Nitrite (NO2 ⁻)	IC	2.79E+04	1.12E+03	2.14E+05
Oxalate (C ₂ O ₄ ²⁻)	IC	3.94E+02	1.47E+01	2.72E+04
Phosphate (PO4 ³⁻)	IC	3.44E+02	1.64E+01	2.94E+04
Sulfate (SO4 ²⁻)	IC	4.72E+03	1.88E+02	5.69E+04
Arsenic (As)	ICP-MS	<1.59E-01	NA	2.30E+01
Barium (Ba)	ICP-ES	< 1.85E-01	NA	6.19E+02
Cadmium (Cd)	ICP-ES	< 2.38E+00	NA	3.10E+02
Chromium (Cr)	ICP-ES	5.24E+01	4.34E-01	1.24E+03
Lead (Pb)	ICP-MS	< 3.31E+01	NA	6.19E+02
Total Mercury (Hg)	CVAA	8.05E+01	7.51E-01	3.25E+02
Elemental Mercury (Hg(0))	CVAFS	2.51E+00 °	3.26E-01	1.82E+01
Monomethyl Mercury (CH ₃ Hg)	CVAFS w/ Distillation	3.41E+01 °	1.40E+00	3.50E+02
Ethyl Mercury (C ₂ H ₅ Hg)	CVAFS w/ Distillation	< 1.95E-02 °	NA	3.73E+02
Selenium (Se)	ICP-MS	<5.98E-01	NA	4.46E+02
Silver (Ag)	ICP-ES	<2.43E+00	NA	6.19E+02
Aluminum (Al)	ICP-ES	5.16E+03	4.99E+01	1.16E+05
Potassium (K)	AA	4.41E+02	1.68E+01	3.03E+04
Butanol (C4H9OH)	VOA	< 5.00E-01 °	NA	7.73E+00
Propanol (C ₃ H ₇ OH)	VOA	< 2.50E-01 °	NA	1.88E+00
Phenol (C ₆ H ₅ OH)	SVOA	< 1.00E+01 °	NA	7.50E+02
Isopar L ()	SVOA	< 2.67E+01 ppm ^{c,d}	NA	1.10E+01 ppm
Total Organic Carbon ()	TOC	2.17E+02 ^b	2.31E+00	5.00E+03
Tetraphenylborate [TPB] (B(C ₆ H ₅) ₄)	HPLC	< 5.00E+00	NA	5.00E+00

Table 1.	Chemical Contaminants from	First Quarter	CY17 Tank 5	50H Samples a	and Saltstone	WAC, Revision	16,
Attachme	ent 8.1 Limits ²			_			

a. Result is calculated from the measured Al concentration assuming all the Al is present as the OH compound.

b. Measurement performed on filtered supernate samples.

c. Measurement performed on duplicate samples rather than triplicate samples.
d. Result is calculated from the reported concentration of < 33 mg/L and the density of the slurry sample listed in Table 8.

e. Mercury species calculated from data presented in Reference 11.

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<u>Chemical Name (Formula)</u>	<u>Method</u>	<u>Average Concentration</u> (mg/L)	<u>Std. Dev.</u>	<u>WAC Target</u> (mg/L)
Boron (B)	ICP-ES	5.21E+01	1.57E+00	7.43E+02
Cobalt (Co)	ICP-MS ^a	<1.59E-02	NA	1.75E+02
Copper (Cu)	ICP-ES	< 8.70E+00	NA	7.43E+02
Iron (Fe)	ICP-ES	7.88E+00	8.67E-02	4.95E+03
Lithium (Li)	ICP-ES	7.36E+00	1.42E-01	7.43E+02
Manganese (Mn)	ICP-ES	3.00E+00	4.28E-02	7.43E+02
Molybdenum (Mo)	ICP-ES	1.95E+01	8.93E-01	7.43E+02
Nickel (Ni)	ICP-ES	< 4.07E+00	NA	7.43E+02
Silicon (Si)	ICP-ES	<2.74E+01	NA	1.07E+04
Strontium (Sr)	ICP-ES	< 6.72E-02	NA	7.43E+02
Zinc (Zn)	ICP-ES	4.37E+00	1.46E-01	8.03E+02
Benzene (C ₆ H ₆)	VOA	< 1.50E-01 ^b	NA	3.10E+02
Methanol (CH ₃ OH)	VOA	с	NA	1.88E+00
Dibutylphosphate [DBP] (C ₈ H ₁₉ O ₄ P)	IC	< 2.50E+02	NA	3.47E+02
Tributylphosphate [TBP] ((C4H9O)3PO)	SVOA	< 7.50E-01 ^b	NA	7.50E+00
Toluene (C ₆ H ₅ CH ₃)	VOA	< 1.50E-01 ^b	NA	3.10E+02
EDTA (C ₁₀ H ₁₂ N ₂ O ₈ ⁴⁻)	HPLC	< 1.00E+02	NA	3.10E+02
NORPAR 13 (CnH2.n)	SVOA	< 7.50E-01 ^b	NA	7.50E-01
Dimethyl Mercury ((CH ₃) ₂ Hg)	CVAFS	1.85E-02 ^d	2.04E-03	1.00E+00

 Table 2. Chemical Contaminants from First Quarter CY17 Tank 50H Samples and Saltstone WAC, Revision 16,

 Attachment 8.2 Targets²

a. Cobalt based on the stable Co-59 isotope.

b. Measurement performed on duplicate samples rather than triplicate samples.

c. Currently, a routine method for detecting this species does not exist in Analytical Development (AD).

d. Mercury species calculated from data presented in Reference 11.

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Radionuclide	<u>Method</u>	<u>Average Concentration</u> (pCi/mL)	<u>Std. Dev.</u>	<u>WAC Limit</u> (pCi/mL)
Tritium (³ H)	Tritium counting	1.36E+03	6.03E+01	5.63E+05
Carbon-14 (¹⁴ C)	C-14 Liquid scintillation	6.01E+02	2.13E+01	1.13E+05
Nickel-63 (⁶³ Ni)	Ni-59/63	< 8.92E+00	NA	1.13E+05
Strontium-90 (⁹⁰ Sr)	Sr-90 Liquid scintillation	1.36E+05	3.83E+03	3.15E+06
Technetium-99 (⁹⁹ Tc)	Tc-99 Liquid scintillation	4.70E+04	3.36E+03	2.11E+05
Iodine-129 (¹²⁹ I)	I-129 (w/ separation) Liquid scintillation	4.04E+01	3.68E+00	6.30E+01
Cesium-137 (¹³⁷ Cs)	Gamma Scan	2.65E+05	6.88E+02	3.96E+06
Uranium-233 (²³³ U)	ICP-MS	< 1.54E+02	NA	1.13E+04
Uranium-235 (²³⁵ U)	ICP-MS	2.22E-01	1.43E-03	1.13E+02
Plutonium-241 (²⁴¹ Pu)	Pu238/241 Liquid scintillation	1.05E+04	3.93E+02	8.38E+05
Total Alpha	Liquid Scintillation Counting	<2.46E+04	NA	2.13E+05

 Table 3. Radionuclide Contaminants from First Quarter CY17 Tank 50H Samples and Saltstone WAC, Revision Table

 3. 16, Attachment 8.3 Limits²

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Table 4.	Radionuclide Contaminants f	rom First Quarter	CY17 Tank 50H S	Samples and Saltston	e WAC, Revision 16,
Attachn	ent 8.4 Targets ²				

Radionuclide	Method	<u>Average</u> <u>Concentration</u> (pCi/mL)	<u>Std. Dev.</u>	<u>WAC Target</u> (pCi/mL)
Aluminum-26 (²⁶ Al)	Gamma scan (Cs removed)	7.34E-02 ^a	N/A	2.88E+03
Cobalt-60 (⁶⁰ Co)	Gamma scan (Cs removed)	3.49E-01 ^b	5.22E-02	9.75E+02
Potassium-40 (⁴⁰ K)	Gamma scan (Cs removed)	<2.45E+00	NA	1.00E+02
Nickel-59 (⁵⁹ Ni)	Ni-59/63	< 7.93E+00	NA	1.13E+03
Selenium-79 (⁷⁹ Se)	Se-79	3.05E+01	4.76E+00	1.90E+04
Yttrium-90 (⁹⁰ Y)	Secular Equilibrium w/ 100% of Sr-90	1.36E+05	3.83E+03	3.15E+06
Zirconium-93 (⁹³ Zr)	ICP-MS	< 8.01E+01	NA	1.00E+05
Niobium-94 (⁹⁴ Nb)	Nb-94	< 3.85E-01	NA	1.53E+02
Rhodium-106 (¹⁰⁶ Rh)	Secular Equilibrium w/ 100% of Ru-106	< 2.64E+00	NA	1.13E+06
Ruthenium-106 (¹⁰⁶ Ru)	Gamma scan (Cs removed)	< 2.64E+00	NA	1.13E+06
Antimony-125 (¹²⁵ Sb)	Gamma scan (Cs removed)	1.14E+01	9.46E-01	7.99E+03
Tellurium-125m (^{125m} Te)	Secular Equilibrium w/ 100% of Sb-125	1.14E+01	9.46E-01	1.83E+03
Tin-126 (¹²⁶ Sn)	Gamma scan (Cs removed)	4.86E+02	4.50E+00	1.80E+04
Cesium-134 (¹³⁴ Cs)	Gamma Scan	< 7.12E+01	NA	1.82E+04
Cesium-135 (¹³⁵ Cs)	ICP-MS	< 1.84E+01	NA	2.50E+02
Barium-137m (^{137m} Ba)	Calculation (Secular Equilibrium w/ 94.6% of Cs-137)	2.51E+05	6.51E+02	3.75E+06
Cerium-144 (¹⁴⁴ Ce)	Gamma scan (Cs removed)	< 5.90E+00	NA	1.13E+05
Promethium-147 (¹⁴⁷ Pm)	Pm-147/Sm-151 Liquid scintillation	<4.06E+01	NA	5.63E+06
Samarium-151 (¹⁵¹ Sm)	Pm-147/Sm-151 Liquid scintillation	<3.56E+01	NA	2.25E+04
Europium-154 (¹⁵⁴ Eu)	Gamma scan (Cs removed)	< 6.08E-01	NA	1.62E+03
Europium-155 (¹⁵⁵ Eu)	Gamma scan (Cs removed)	< 2.30E+00	NA	1.13E+04
Radium-226 (²²⁶ Ra)	Ra-226	<1.55E+00	NA	1.00E+03
Radium-228 (²²⁸ Ra)	Gamma scan (Cs removed)	< 1.41E+00	NA	1.00E+04
Actinium-227 (²²⁷ Ac)	Th-229/230	<9.55E-03	NA	1.00E+04
Thorium-229 (^{229Th})	Th-229/230	<1.80E-02	NA	1.63E+05
Thorium-230 (^{230Th})	Th-229/230	9.77E-03 ^a	NA	6.26E+03
Thorium-232 (²³² Th)	ICP-MS	< 1.75E-03	NA	2.88E+03
Protactinium-231 (²³¹ Pa)	Pa-231	< 5.99E-01	NA	1.00E+03
Uranium-232 (²³² U)	U-232	3.70E+00	2.07E+00	9.06E+03
Uranium-234 (²³⁴ U)	ICP-MS	< 9.96E+01	NA	1.13E+04
Uranium-236 (²³⁶ U)	ICP-MS	1.22E+00	2.48E-02	1.13E+04
Uranium-238 (²³⁸ U)	ICP-MS	4.19E+00	2.89E-02	1.13E+04

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Table 4. Radionuclide Contaminants from	First Quarter	CY17 Tank 50	0H Samples ai	nd Saltstone WA	AC, Revision 16,
Attachment 8.4 Targets ² , continued					

Radionuclide	<u>Method</u>	<u>Average</u> <u>Concentration</u> (pCi/mL)	<u>Std. Dev.</u>	<u>WAC Target</u> (pCi/mL)
Neptunium-237 (²³⁷ Np)	ICP-MS	<1.12E+01	NA	1.00E+04
Plutonium-238 (²³⁸ Pu)	Pu238/241 Pu alpha PHA	2.80E+04	1.22E+03	2.13E+05
Plutonium-239 (²³⁹ Pu)	Pu238/241 Pu alpha PHA	6.80E+02	4.30E+01	2.13E+05
Plutonium-240 (²⁴⁰ Pu)	Pu238/241 Pu alpha PHA	6.80E+02	4.30E+01	2.13E+05
Plutonium-242 (²⁴² Pu)	ICP-MS	<6.09E+01	NA	2.13E+05
Plutonium-244 (²⁴⁴ Pu)	ICP-MS	<2.83E-01	NA	7.02E+04
Americium-241 (²⁴¹ Am)	Am/Cm	4.52E+00	4.99E-01	2.13E+05
Americium-242m (^{242m} Am)	Am/Cm	<2.18E-02	NA	4.50E+05
Americium-243 (²⁴³ Am)	Am/Cm	<5.09E-01	NA	2.13E+05
Curium-242 (²⁴² Cm)	Am/Cm	<1.80E-02	NA	1.13E+04
Curium-244 (²⁴⁴ Cm)	Am/Cm	1.48E+00	3.94E-02	2.13E+05
Curium-245 (²⁴⁵ Cm)	Am/Cm	<1.98E+00	NA	2.25E+05

a. Only one detectable value from the analyzed triplicate set.

b. Measurement from average of two detectable values from the analyzed triplicate set.

Table 5.	Chemical Contaminants Impacting Saltstone Disposal Unit (SDU) Flammability from First Quarter C	Y17
Tank 50H	H Samples and Saltstone WAC, Revision 16, Table 2 Limits and Targets ²	

Chemical Name (Formula)	<u>Method</u>	<u>Average Concentration</u> (mg/L)	<u>Std. Dev.</u>	<u>WAC</u> Limit/Target
Isopar L ()	SVOA	< 2.67E+01 ppm ^{a,b}	NA	1.10E+01 ppm (Limit)
Tetraphenylborate [TPB] (B(C6H5)4 ⁻)	HPLC	< 5.00E+00	NA	5.00E+00 mg/L (Limit)
Ammonium (NH ₄ ⁺)	IC	< 1.00E+02	NA	2.12E+02 mg/L (Limit)
Total Mercury (Hg)	CVAA	8.05E+01	7.51E-01	3.25E+02 mg/L (Limit)
Monomethyl Mercury (CH ₃ Hg)	CVAFS w/ Distillation	3.41E+01 °	1.40E+00	3.50E+02 mg/L (Limit)
Dimethyl Mercury ((CH3)2Hg)	CVAFS	1.85E-02 °	2.04E-03	1.00E+00 mg/L (Target)

a. Measurement performed on duplicate samples rather than triplicate samples.

b. Result is calculated from the reported concentration of < 33 mg/L and the density of the slurry sample listed in Table 8.

c. Mercury species calculated from data presented in Reference 11.

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<u>Chemical Name (Formula)</u>	<u>Method</u>	Average Concentration (mg/L)	<u>Std. Dev.</u>	WAC Concentrations (mg/L)
Butanol (C4H9OH)	VOA	< 5.00E-01	NA	0.75
Tributylphosphate[TBP] ((C4H9O)3PO)	SVOA	< 7.50E-01	NA	1.0
Isopropanol (C3H7OH)	VOA	< 2.50E-01	NA	0.25
Methanol (CH ₃ OH)	а	NA	NA	0.05
NORPAR 13 (C _n H _{2.n})	SVOA	< 7.50E-01	NA	0.75

Table 6. Other Organics Impacting SDU Flammability from First Quarter CY17 Tank 50H Samples and Saltstone WAC, Revision 16, Table 3 Concentrations²

a. Currently, a routine method for detecting this species does not exist in AD.

Table 7.	Processing Co	onstituents from	First Quarter	· CY17 Tank !	50H Samples an	nd Saltstone '	WAC, I	Revision	16, 1	Fable
4 Limits	2									

Processing Constituents	Method	Value	<u>Std. Dev.</u>	WAC Limit
рН	Calculated	> 13	NA	> 10
Sodium Concentration	ICP-ES	5.44 M	6.20E-02	2.5 M < [Na ⁺] < 7.0 M
Total Insoluble Solids	Calculated	~0 wt%	NA	< 15 wt%

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<u>Constituent</u>	<u>Method</u>	<u>Average Value</u>	<u>Std. Dev.</u>
Density (slurry)	Measured (21.6°C)	1.2347 g/mL	0.0003
Specific Gravity	а	1.2376	0.0003
Total Solids	Measured	27.16 wt%	0.19
Total Beta	LSC	7.16E+05 pCi/mL	7.80E+03
Total Gamma	b	2.52E+05 pCi/mL	3.76E+02 °
Thorium-228 (²²⁸ Th)	Gamma scan (Cs removed)	< 1.34E+01 pCi/mL	NA
Curium-247 (²⁴⁷ Cm) ^d	Am/Cm	<3.12E+00 pCi/mL	NA
Californium-249 (²⁴⁹ Cf) ^d	Am/Cm	<3.28E+00 pCi/mL	NA
Californium-251 (²⁵¹ Cf) ^d	Am/Cm	<2.26E+00 pCi/mL	NA
Beryllium (Be)	ICP-ES	< 7.70E-02 mg/L	NA
Formate ^e	IC	2.13E+02 mg/L	9.71E+00
Total Organic Carbon (minus formate & oxalate) ^f	Calculated	5.29E+01 mg/L	NA

Table 8. Additional Measured Constituents⁴

a. Calculated from the measured density of slurry and density of water at 22.5 $^\circ$ C.¹²

b. Calculated from the sum of gamma emitters (Sb-126, Sn-126, Sb-125, Am-241, Co-60 and Ba-137m).

c. Value is the "standard error of the mean" rather than the standard deviation of the measurements since its calculation involves multiple radionuclides.

d. Reported values are all below the estimated detection limits of 90.1 pCi/mL established by SRNL.⁸

e. Formate is not required by the WAC but is used in the Total Organic Carbon (minus formate & oxalate) calculation.²

f. Total Organic Carbon (minus formate & oxalate) as shown in Appendix 1 of the WAC.²

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REFERENCES

- ¹ Crawford, C. L, "1Q CY17 Tank 50 WAC Characterization", B9108-00026-37, SRNL E-Notebook (Production), Savannah River National Laboratory, December 2016.
- ² Ray, J. W., "Waste Acceptance Criteria for Aqueous Waste Sent to the Z-Area Saltstone Production Facility", Savannah River Remediation, X-SD-Z-00001, Rev. 16, April 2016.
- ³ Ray, J.W., "Routine Saltstone Support for Salt Solution Analyses FY2017", Savannah River Remediation, X-TTR-Z-00010, Rev. 0, September 2016.
- ⁴ Hill, K.A. and Miller, D.H, "Task Technical and Quality Assurance Plan for SRNL Support of Salt Solution Analyses and Grout Sample Preparation and Analyses – FY2017", Savannah River National Laboratory, SRNL-RP-2016-00654, Rev. 1, November 2016.
- ⁵ Crump, S. L., "Determination of Method Reporting Limits for Select Analytes by GC/MS", Savannah River National Laboratory, SRNL-TR-2010-00206, Rev. 0, October 2010.
- ⁶ Potvin, M. M., "Waste Acceptance Criteria for Aqueous Waste Sent to the Z-Area Saltstone Production Facility", Savannah River Remediation, X-SD-Z-00001, Rev. 12, July 2013.
- ⁷ Dixon, D. B., "Minimum Detection Limits for Saltstone Quarterly WAC Analyses", Savannah River Remediation, SRR-WSE-2013-00005, Rev. 1, January 2013.
- ⁸ DiPrete, C. C., "Overview of Capability to Measure Radionuclides of Interest for Saltstone", Savannah River National Laboratory, SRNL-L4000-2009-00028, Rev. 0, June 2009.
- ⁹ Crawford, C. L., "Results for the First Quarter Calendar Year 2017 Tank 50 WAC Sample: Cs-137", Savannah River National Laboratory, SRNL-L3100-2017-00012, Rev.0, January 2017.
- ¹⁰ Integrated Data Base Report 1996: U.S. Spent Nuclear Fuel and Radioactive Waste Inventories, Projections, and Characteristics, DOE/RW-0006, Rev. 13, December 1997. <u>https://www.nrc.gov/docs/ML1028/ML102850100.pdf</u> (accessed March 24, 2017)
- ¹¹ Bannochie, C. J, "Results of Hg Speciation Testing on 1Q17 Tank 50 WAC Sample", Savannah River National Laboratory, SRNL-L3300-2017-00005, Rev. 1, May 2018.
- ¹² *CRC Handbook of Chemistry and Physics*, 97th ed.; Section 6: Fluid Properties. Edited by Haynes, W. M., CRC Press Taylor and Francis Group, Boca Raton, FL, Internet Version 2017.

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