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Rev 1

# Quarterly Sampling of the Wetlands Along the Old F-Area Effluent Ditch: May 1994<sup>(U)</sup>

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12/27/94

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# Quarterly Sampling of the Wetlands Along the Old F-Area Effluent Ditch: May 1994 (U)

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## Executive Summary

In May 1994, well point water and near-surface water samples (i.e. bucket samples) were collected to further characterize tritium and volatile organic compounds (VOC) in the wetlands along the old F-Area effluent ditch south of 643-E (old burial ground). The May sampling event was the second in a series of eight events. Groundwater flow paths suggest that compounds detected in water table wells around 643-E would migrate towards the old F-Area effluent ditch and Fourmile Branch. Recent analytical results from well point and near-surface water sampling in the wetlands that comprise the old F-Area effluent ditch have shown that tritium and small quantities of VOCs are outcropping in the area. For this study, seven locations along the old F-Area effluent ditch were selected to be sampled. Well point samples were collected from all seven locations and near-surface water samples were collected at four locations.

Results of the May 1994 sampling event further support findings that tritium and volatile organic compounds originating from 643-E are outcropping in the wetlands near the old F-Area effluent ditch. Four different analytes were detected in the well points at least once at concentrations greater than the method detection limit. These analytes included 1,2-dichloroethylene, tetrachloroethylene, trichloroethylene, and tritium. Each analyte was detected at least once at concentrations above the Primary Drinking Water Standards (PDWS; EPA, 1991) or the Maximum Contaminant Levels (MCL; EPA, 1993a and b) list.

Five analytes, chloromethane, cis-1,2-dichloroethylene, methyl ethyl ketone, toluene, and tritium, were detected in the buckets, at least once, at concentrations greater than the method detection limit. Tritium and cis-1,2-dichloroethylene were detected at concentrations above the limits established by the PDWS or MCL at one or more bucket locations.

A secondary objective of this project was to compare VOC concentrations between the well points installed to depths of 6 to 8 ft and the near-surface water sampling devices (buckets) installed to depths of 1 to 2 ft. Based on differences in tritium concentrations at each location, it was determined that the sampling devices intercepted different groundwater flow paths. This negated direct comparison of analytical results between devices. However, when VOC concentrations measured at each well point and bucket location were normalized, based on the percent differences observed in tritium concentrations at that location, resulting well point and bucket VOC concentrations were comparable in most cases. These results suggest that volatilization losses of VOCs from the buckets may be negligible.

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## Introduction

The Radioactive Waste Burial Grounds are central waste storage sites used for disposing of onsite and offsite radioactive and nonradioactive wastes (Jaegge et al., 1985). The burial grounds are divided into three areas: Building 643-E (old burial ground), which received wastes from 1952 through 1972; Building 643-7E, which has been receiving wastes since 1969; and Building 643-28E which is a closure area within Building 643-7E and is termed the Mixed Waste Management Facility (MWMF) (Jaegge et al., 1985). The sampling zone along the old F-Area effluent ditch receives discharge from shallow groundwater flowing under the southwest corner of 643-E.

Detectable quantities of organic compounds have been measured in the groundwater beneath 643-E (Hoeffner et al., 1985; Hoeffner, 1984). Potential sources of the organic compounds in 643-E were identified as liquid scintillation wastes, spent solvents, decontamination reagents, and waste oils (Hoeffner et al., 1985) as well as decomposition of humic substances. Hoeffner et al. (1985) performed in-depth analysis of semi-volatile organics (using GC/MS) on ten wells in 643-E that consistently contained measurable amounts of total organic carbon. One other well included in the study was located in close proximity to 643-E near the site of previous decontamination operations. From these 11 wells, more than 50 organic compounds were detected. Detected aromatic volatile organic compounds included benzene, toluene, phenol, and naphthalene, among others. Hoeffner et al. (1985) indicated that the primary source of these four compounds was liquid scintillation wastes.

Many water table wells (BGO series, Group D) near 643-E are sampled quarterly as part of the SRS groundwater monitoring program. These wells are sampled for Resource Conservation and Recovery Act (RCRA), Appendix IX analyses. Chlorinated VOCs are routinely detected in several water table wells, particularly in the southwest corner of 643-E. These compounds, which include 1,1-dichloroethane, 1,2-dichloroethylene, 1,1-dichloroethylene, tetrachloroethylene, trichloroethylene, and vinyl chloride (EMS, 1993), originate from or are biodegradation products of wastes buried in 643-E (i.e., organic degreasers). Tritium is also routinely detected in the wells and originates primarily from spent melt crucibles in 643-E.

Groundwater flow paths for Aquifer Zone IIB2 (WSRC, 1991) suggest that compounds detected in the water table wells around 643-E would migrate towards the old F-Area effluent ditch and Fourmile Branch (FMB) (Figure 1). As the flow paths approach FMB, shallow groundwater outcrops into FMB and its wetlands including those along the old F-Area effluent ditch. Recent analytical results from near-surface water sampling in the wetlands that comprise the old F-Area effluent ditch have shown that tritium and small quantities of VOCs are outcropping in the area. Detected organic compounds include toluene, total 1,2-dichloroethylene, vinyl chloride, and trichloroethylene (Dixon, 1993a and b). Based on the regional groundwater flow patterns for the water table aquifer in this area, it is believed that these constituents are migrating from beneath 643-E and outcropping into the wetlands. This hypothesis is further corroborated by tritium and conductivity results from the quarterly tritium survey (Dixon and Rogers, 1993a) conducted in the wetlands of FMB down gradient of the F- and H-Area seepage basins. Waste discharged into the F- and H-Area seepage basins, in which 643-E is located between, contained high levels of salt ions compared to waste stored in 643-E. Using a Spearman rank correlation, Dixon and Rogers (1993a) found a strong correlation between tritium activities and conductivity values at locations where wetland water was being influenced by the contaminant plumes migrating from the F- and H-Area seepage basins. However, conductivity values measured in the wetlands south of 643-E were at or near background, even at locations where tritium activities were in excess of 50,000 pCi/ml (Dixon and Rogers, 1993a); no correlation exists between tritium activities and conductivity values for this wetland. These results are consistent with previous studies (Haselow et al, 1990; Dixon and Rogers, 1993b) and suggest that tritium and VOC concentrations measured in the wetlands south of 643-E originate from wastes buried in 643-E.

The purpose of this study is to further characterize tritium and VOC outcrops in the wetlands south of 643-E along the old F-Area effluent ditch. This project differs from previous sampling events in that well points (less than 10 ft below the soil surface) are used to sample the shallow groundwater. In the past, a relatively simple near-surface water sampling methodology has been employed to collect samples from the wetlands. Using the near-surface water sampling methodology, a small hole is excavated using a shovel at each sampling location. A five gallon plastic bucket with the sides cut out is then installed in the hole to a depth of about 1.5 ft. After installation, these buckets are covered with a plastic lid. Twenty-four hours later, after water has seeped into the buckets, the water

samples are collected. This methodology, while simple, provides good analytical results for a wide range of parameters at a reasonable cost for many sampling locations. Using the well point sampling methodology, the shallow well points are installed to a depth of 6 to 8 ft and are completely sealed to prevent VOCs from escaping. This should provide additional assurance that representative VOC samples are collected. As a secondary objective to characterizing VOC and tritium contamination in the wetlands, the two sampling methodologies will be compared to determine any differences. If it can be shown that the results from both methodologies are comparable, it may not be necessary to use the more expensive well point sampling methodology when VOCs are of interest on future projects of this type (i.e., either the near-surface water or shallow well point samples may be used to evaluate outcropping VOC plumes).

### Old F-Area Effluent Ditch

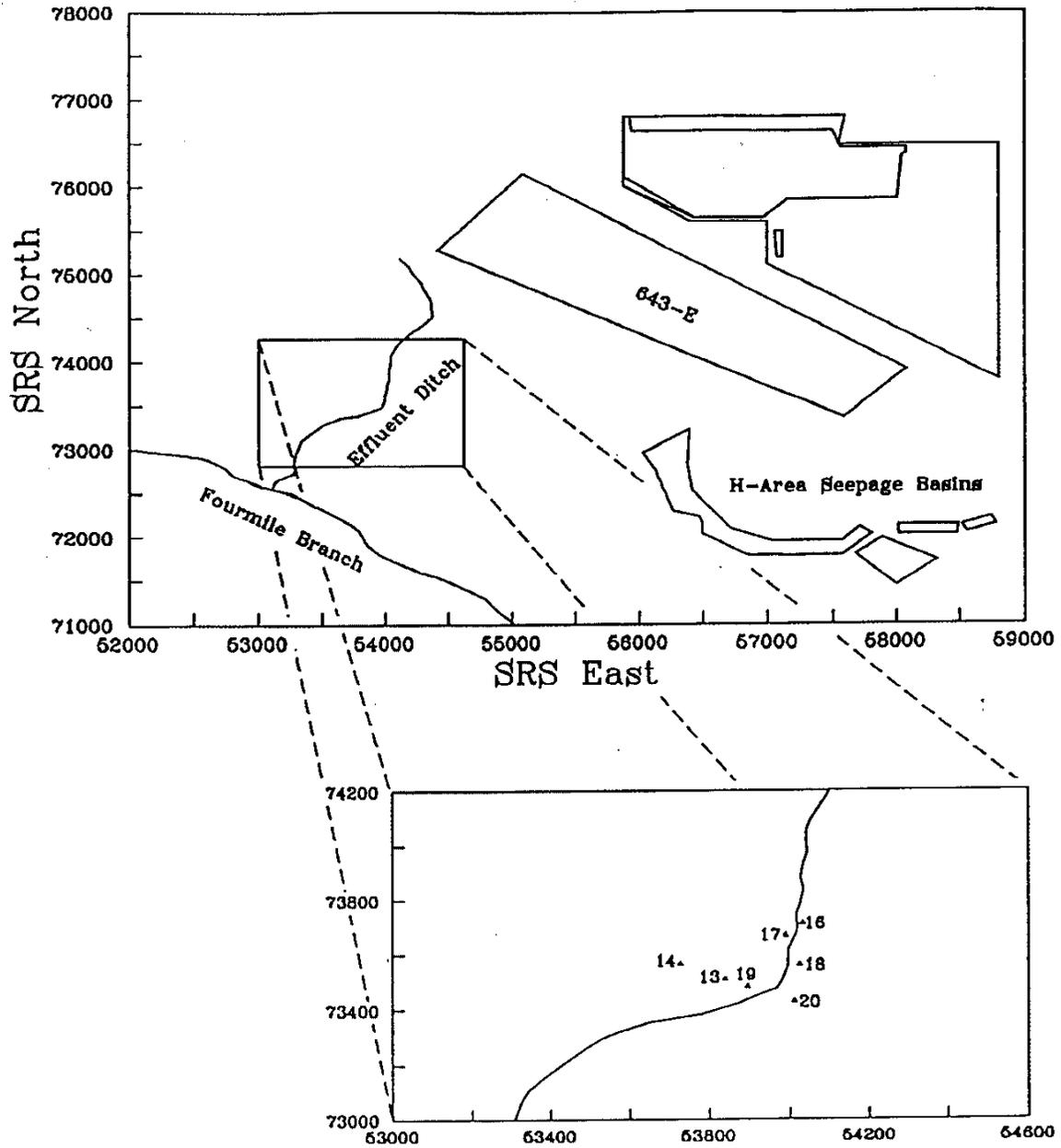


Figure 1 Sampling Locations Along the Old F-Area Effluent Ditch. (All locations are prefixed by "FHB0").



## Objectives

The objectives of this report are to:

- present the results of the May 1994 sampling event in the wetlands south of 643-E along the old F-Area effluent ditch
- compare, where possible, these results to related standards (Table 1)
- compare analytical results from well points and adjacent near-surface water sampling (i.e. buckets) locations

## Methods

The Environmental Sciences Section (ESS) of the Savannah River Technology Center (SRTC) conducted the second round of well point volatile organic compounds (VOC) and tritium sampling in the wetlands at the old F-Area effluent ditch on May 24, 1994. Analytical services for the well point samples were obtained via task order under Contract AA01244N with Weston Analytics (WA). WA analyzed the samples for compounds on the Appendix IX Volatiles list. Analytical services for the bucket samples were arranged onsite through the Analytical Development Section (ADS) of SRTC. ADS analyzed the bucket samples for compounds on the Target Compound List (volatile organics only) following Contract Laboratory Program (CLP) and those samples were considered the reference samples for the bucket locations. ADS also analyzed two split samples for the volatile organics listed on the Target Compound List. A tentatively identified compounds (TIC) scan was performed on each sample analyzed by ADS. All samples were submitted to the Environmental Monitoring Section (EMS) for tritium analysis. For the well point samples, those analyzed by WA and EMS were considered to be the reference samples on which all analyses and interpretations were based. Well point samples analyzed by AES were considered quality assurance samples and were not used in the analyses and interpretations. These considerations should prevent interlab variability of constituent concentrations from biasing the analyses and interpretations. A list of the constituents analyzed by each laboratory, along with the methods used and method detection limits, is given on Tables 2 and 3.

### *Sampling Locations*

Samples were collected from seven locations in the wetlands along the old F-Area effluent ditch (Table 4 and Figure 1). Based on data from previous studies, locations exhibiting high tritium and volatile organic compound (VOC) concentrations were selected. Well point water samples were collected at all locations and bucket samples were collected at four locations. Bucket samples and well point samples share the same identification numbers, except the identification numbers for the bucket samples are suffixed with a "B". Background well point and bucket water samples were collected at two locations in wetlands along Upper Three Runs (UTR).

A blind duplicate sample was collected at location FHB014 and submitted to WA and EMS as a quality assurance/quality control (QA/QC) sample. Split samples were collected at locations FHB013 and FHB018 and submitted to ADS. One trip blank sample, QA2, was carried throughout all sampling activities. The trip blank consisted of two 40 ml vials of deionized water. The trip blank was carried to each sampling location, handled like the collected samples, and submitted to WA for analyses. A field blank, QA3, was collected at location FHB018. It consisted of two 40 ml vials filled with deionized water. At location FHB018, the caps were removed from the vials and the samples were exposed to the atmosphere during collection activities. The field blank was then submitted to WA for analyses.

### *Well Point Sample Collection*

Well points were installed according to WSRC-3Q5 (Chapter 7) procedures at each sampling location in February 1994. The well points were constructed of a 2-inch diameter PVC TriLok® material to depths ranging from approximately 6 to 8 ft below ground level. Each well point was screened over the bottom 2.5 ft with TriLok® well screen (0.010 inch slot size). Figure 2 illustrates a schematic diagram of the well points. Boreholes were hand augured to depths ranging from approximately 6 to 8 ft or until unconsolidated material was reached. The well

points were then inserted into the boreholes and driven to the final depth. After the well points were inserted into the boreholes, a plastic lid with a 2-inch diameter hole was slipped onto the casing to a depth of 6 to 8 inches below the soil surface. To create a surface seal, bentonite was placed on top of the lid and the bentonite was covered with soil from the surrounding area. No grout or filter pack material was used in the construction of the well points.

One-inch diameter, 12 VDC submersible pumps were installed in each well and connected to the well cap using stainless steel tubing and swagelok connections. Each well cap was fitted with connections to facilitate measuring water levels and making headspace measurements.

Well point samples were collected according to procedure ESSOP 2-115 (WSRC-L-14.1, 1993a). A brief description of the sampling procedure is described in this report; however, for a detailed discussion of the sampling procedure, ESSOP 2-115 should be reviewed.

The day before sample collection began, a plastic cap was placed over the pump discharge tube and well vent to prevent the loss of VOCs to the atmosphere. Prior to sample collection, air samples were collected for analysis of chlorinated volatile organic compounds (CVOCs) by gas chromatography. The air samples (300  $\mu$ l) were collected from the headspace of each well (via the well vent) using a 1 ml Hamilton gastight syringe (glass with teflon plunger). The samples were stored in the collection syringe by sealing the needle using a thermogreen (Supelco, Inc.) septum. The syringes were stored in a cooler containing blue ice prior to analysis.

All samples were analyzed within 24 hours on a Hewlett Packard 5890 gas chromatograph equipped with an electron capture detector (providing sensitive analysis for halogen containing compounds). The GC was calibrated with 1 and 10 ppmv trichloroethylene and tetrachloroethylene standards and 10 ppmv 1,1-dichloroethylene, 1,1,1-trichloroethane, carbon tetrachloride, and trichloromethane standards. Individual CVOCs were separated using the following conditions:

Column:	60m silicone coated glass membrane capillary column
Temperature:	40° C for 2 minutes, + 10° C/min. up to 200° C
Total Flow Rate:	85 cm <sup>3</sup> /min
Split Ratio:	10:1

After collecting the air samples, water level measurements were taken to determine the volume of water contained within the well point. Once the volume of water in each well point was determined, pumping was started and the flowrate was determined. Attempts were made at each well point to purge two well volumes; however, several of the well points pumped dry prior to purging two well volumes. In these cases, sample collection began when the well points recovered.

Conductivity and pH measurements were taken initially at the start of purging, and subsequently thereafter, for at least three sets of measurements until successive readings differed by less than ten percent. When three sets of successive measurements were within ten percent, stabilization requirements had been met and sample collection began. In cases where well points pumped dry, initial conductivity and pH measurements were taken and sample collection began when the well point recovered. All samples were obtained by collecting discharge directly from the pump. All containers were filled to the shoulder except VOC samples, which were completely filled such that no headspace remained. Total activity samples for shipping clearance were collected at each well point except at the background locations.

### *Near-Surface Water Sample Collection*

Near-surface water samples (bucket samples) were collected according to procedure ESSOP 2-104 (WSRC-L-14.1, 1993d). A brief description of the sampling procedure is given described in this report; however, for a detailed discussion of the sampling procedure, ESSOP 2-104 should be reviewed.

Near-surface water sampling locations were already established as part of a previous project. At each location, a hole was excavated (approximately 18 inches deep) using a decontaminated stainless steel shovel. A five-gallon

plastic bucket screened on the sides and bottom was placed in the hole and covered with a lid. All buckets were bailed the day before sample collection began to ensure that a representative sample of the local shallow groundwater was being collected. Conductivity and pH measurements were taken at each bucket sampling location. All samples were collected by dipping the containers directly into the water in the bucket. VOC samples were completely filled such that no headspace remained. Other sample containers were filled to the shoulder.

### *Sample Processing*

When the sample containers for a location were filled, they were stored in a cooler containing blue ice for transport to the processing area. At the processing area, the samples were stored at 4°C until they were packaged for shipment. Samples were packed and shipped according to "Packaging Nonradioactive Water Samples for Offsite Shipment" and "Packaging Radioactive Water Samples for Offsite Shipment" (WSRC-3Q5, Ch. 15) procedures. The particular procedure used was determined by the total activity results. Chain of custody procedures were followed for all samples.

### *Data Validation*

Weston Analytics (WA) and their subcontractors submitted hard copies and electronic data diskettes (EDD) containing the analytical data to ESS. The hard copies were considered the laboratory-validated data packages and consisted of a case narrative for each batch, the result of the analyses, and QA/QC information (i.e., spike recoveries). The EDDs consisted of one flat ASCII computer file for each laboratory batch. The data were formatted on the EDDs into the AN92 data format to provide the information in Appendix B. ADS and EMS submitted hard copies of the analytical data, which were manually entered into a spreadsheet and verified by ESS. Each EDD file was uploaded onto an IBM mainframe computer at SRS. The files were then merged to create one file to accommodate data review and validation activities.

The names of the blind replicate samples were changed to the correct location number and marked as a replicate sample. Statistical Application Software (SAS) computer programs were run on each file to verify the data. The following data verification checks were made using the SAS programs:

- verifying the location names and sample dates
- identifying data entry errors (incorrect methods or units)
- verifying that samples were analyzed within hold times
- verifying that all result and analysis qualifiers were valid.

Any suspected data entry errors were documented and submitted to WA for verification or correction. Changes to the database were recorded in a logbook.

After all data entry errors were resolved, the data were downloaded into a spreadsheet, where they were screened to identify missing analyses and duplicate records. The laboratory was notified of any data that were missing and responded by submitting an EDD for the analysis. Duplicate records were deleted from the database after the laboratory determined they had, in error, submitted the data twice.

The results from samples that were used as laboratory replicates were averaged when both analyses were above the detection limit. Data below the detection limit and data presented with result or analysis qualifiers were not averaged and are noted in the comprehensive data tables (Appendix A) by "NA" in the "Averaged" column. Data that were averaged are noted with an "A" in the "Averaged" column.

Other data validation checks included:

- verifying that all surrogates and spikes were within acceptable recovery ranges
- determining lab blanks with results above detection limits and verifying that these results were qualified accordingly.

Table 1. Primary Drinking Water Standards and Maximum Contaminant Levels Used in Data Comparisons.

Analyte	Standard	Units	Source
1,1,1,2-Tetrachloroethane	NS <sup>1</sup>		
1,1,1-Trichloroethane	200	µg/L	EPA, 1993a
1,1,2,2-Tetrachloroethane	NS		
1,1,2-Trichloroethane	5	µg/L	EPA, 1993b
1,1-Dichloroethane	NS		
1,1-Dichloroethylene	7	µg/L	EPA, 1993a
1,2,3-Trichloropropane	NS		
1,2-Dibromo-3-chloropropane	0.2	µg/L	EPA, 1993a
1,2-Dibromoethane	NS		
1,2-Dichloroethane	5	µg/L	EPA, 1993a
1,2-Dichloroethylene (total)	70 <sup>2</sup>	µg/L	EPA, 1993a
1,2-Dichloropropane	5	µg/L	EPA, 1993a
2-Hexanone	NS		
4-Methyl-2-pentanone	NS		
Acetone	NS		
Acetonitrile (Methyl cyanide)	NS		
Acrolein	NS		
Acrylonitrile	NS		
Allyl chloride	NS		
Benzene	5	µg/L	EPA, 1993a
Bromodichloromethane	NS		
Bromoform	NS		
Carbon disulfide	NS		
Carbon tetrachloride	5	µg/L	EPA, 1993a
Chlorobenzene	100	µg/L	EPA, 1993a
Chloroethane	NS		
Chloroform	100	µg/L	EPA, 1991a
Chloroprene	NS		
cis-1,2-Dichloroethylene	70	µg/L	EPA, 1993a
cis-1,3-Dichloropropene	NS		
Dibromochloromethane	NS		
Dichlorodifluoromethane	NS		
Ethyl methacrylate	NS		
Ethylbenzene	700	µg/L	EPA, 1993a
Isobutyl alcohol	NS		
Methacrylonitrile	NS		
Methyl bromide	NS		
Methyl chloride	NS		
Methyl ethyl ketone	NS		
Methyl iodide	NS		
Methyl methacrylate	NS		
Methylene bromide	NS		
Methylene chloride	5	µg/L	EPA, 1993b
Pentachloroethane	NS		
Propionitrile	NS		
Styrene	100	µg/L	EPA, 1993a
Tetrachloroethylene	5	µg/L	EPA, 1993a
Toluene	1000	µg/L	EPA, 1993a

**Table 1. Primary Drinking Water Standards and Maximum Contaminant Levels Used in Data Comparisons.  
(Cont.)**

Analyte	Standard	Units	Source
trans-1,2-Dichloroethylene	100	µg/L	EPA, 1993a
trans-1,3-Dichloropropene	NS		
trans-1,4-Dichloro-2-butene	NS		
Trichloroethylene	5	µg/L	EPA, 1991a
Trichlorofluoromethane	NS		
Trichlorofluoromethane	NS		
Tritium	20	pCi/ml	EPA, 1993a
Vinyl acetate	NS		
Vinyl chloride	2	µg/L	EPA, 1993a
Xylenes (total)	10000	µg/L	EPA, 1993a

<sup>1</sup>No standard

<sup>2</sup>MCL listed is for cis-1,2-dichloroethylene.

Table 2. Methods and Method Detection Limits for Weston Analytics and Environmental Monitoring Section.

Analyte	Lab	Method	MDL <sup>1</sup>	Units
1,1,1,2-Tetrachloroethane	WA	EPA8240	10	µg/L
1,1,1-Trichloroethane	WA	EPA8240	5	µg/L
1,1,2,2-Tetrachloroethane	WA	EPA8240	5	µg/L
1,1,2-Trichloroethane	WA	EPA8240	5	µg/L
1,1-Dichloroethane	WA	EPA8240	5	µg/L
1,1-Dichloroethylene	WA	EPA8240	5	µg/L
1,2,3-Trichloropropane	WA	EPA8240	10	µg/L
1,2-Dibromo-3-chloropropane	WA	EPA8240	20	µg/L
1,2-Dibromoethane	WA	EPA8240	20	µg/L
1,2-Dichloroethane	WA	EPA8240	5	µg/L
1,2-Dichloroethylene (total)	WA	EPA8240	5	µg/L
1,2-Dichloropropane	WA	EPA8240	5	µg/L
2-Hexanone	WA	EPA8240	10	µg/L
4-Methyl-2-pentanone	WA	EPA8240	10	µg/L
Acetone	WA	EPA8240	10	µg/L
Acetonitrile (Methyl cyanide)	WA	EPA8240	20	µg/L
Acrolein	WA	EPA8240	10	µg/L
Acrylonitrile	WA	EPA8240	10	µg/L
Allyl chloride	WA	EPA8240	100	µg/L
Benzene	WA	EPA8240	5	µg/L
Bromodichloromethane	WA	EPA8240	5	µg/L
Bromoform	WA	EPA8240	5	µg/L
Carbon disulfide	WA	EPA8240	5	µg/L
Carbon tetrachloride	WA	EPA8240	5	µg/L
Chlorobenzene	WA	EPA8240	5	µg/L
Chloroethane	WA	EPA8240	10	µg/L
Chloroform	WA	EPA8240	5	µg/L
Chloroprene	WA	EPA8240	20	µg/L
cis-1,3-Dichloropropene	WA	EPA8240	5	µg/L
Dibromochloromethane	WA	EPA8240	5	µg/L
Dichlorodifluoromethane	WA	EPA8240	10	µg/L
Ethylbenzene	WA	EPA8240	5	µg/L
Isobutyl alcohol	WA	EPA8240	20	µg/L
Methacrylonitrile	WA	EPA8240	20	µg/L
Methyl bromide	WA	EPA8240	10	µg/L
Methyl chloride	WA	EPA8240	10	µg/L
Methyl ethyl ketone	WA	EPA8240	10	µg/L
Methyl iodide	WA	EPA8240	10	µg/L
Methylene bromide	WA	EPA8240	10	µg/L
Methylene chloride	WA	EPA8240	5	µg/L
Propionitrile	WA	EPA8240	50	µg/L
Styrene	WA	EPA8240	5	µg/L
Tetrachloroethylene	WA	EPA8240	5	µg/L
Toluene	WA	EPA8240	5	µg/L
trans-1,3-Dichloropropene	WA	EPA8240	5	µg/L
trans-1,4-Dichloro-2-butene	WA	EPA8240	100	µg/L

Table 2. Methods and Method Detection Limits for Weston Analytics and Environmental Monitoring Section.  
(Cont.)

Analyte	Lab	Method	MDL <sup>1</sup>	Units
Trichloroethylene	WA	EPA8240	5	µg/L
Trichlorofluoromethane	WA	EPA8240	5	µg/L
Tritium	EMS	EMS2760	1	pCi/ml
Vinyl acetate	WA	EPA8240	10	µg/L
Vinyl chloride	WA	EPA8240	10	µg/L
Xylenes (total)	WA	EPA8240	5	µg/L

Table 3. Methods and Practical Quantitation Limits for Analytical Development Section.

Analyte	Lab	Method	PQL <sup>1</sup>	Units
1,1,1-Trichloroethane	ADS	ADS2656	50	µg/L
1,1,2,2-Tetrachloroethane	ADS	ADS2656	50	µg/L
1,1,2-Trichloroethane	ADS	ADS2656	50	µg/L
1,1-Dichloroethane	ADS	ADS2656	50	µg/L
1,1-Dichloroethylene	ADS	ADS2656	50	µg/L
1,2-Dichloroethane	ADS	ADS2656	50	µg/L
1,2-Dichloropropane	ADS	ADS2656	50	µg/L
2-Hexanone	ADS	ADS2656	50	µg/L
4-Methyl-2-pentanone	ADS	ADS2656	50	µg/L
Acetone	ADS	ADS2656	50	µg/L
Benzene	ADS	ADS2656	50	µg/L
Bromodichloromethane	ADS	ADS2656	50	µg/L
Bromoform	ADS	ADS2656	50	µg/L
Carbon disulfide	ADS	ADS2656	50	µg/L
Carbon tetrachloride	ADS	ADS2656	50	µg/L
Chlorobenzene	ADS	ADS2656	50	µg/L
Chloroethane	ADS	ADS2656	50	µg/L
Chloroform	ADS	ADS2656	50	µg/L
cis-1,2-Dichloroethylene	ADS	ADS2656	50	µg/L
cis-1,3-Dichloropropene	ADS	ADS2656	50	µg/L
Dibromochloromethane	ADS	ADS2656	50	µg/L
Ethylbenzene	ADS	ADS2656	50	µg/L
Methyl ethyl ketone	ADS	ADS2656	50	µg/L
Methylene chloride	ADS	ADS2656	50	µg/L
Styrene	ADS	ADS2656	50	µg/L
Tetrachloroethylene	ADS	ADS2656	50	µg/L
Toluene	ADS	ADS2656	50	µg/L
trans-1,2-Dichloroethylene	ADS	ADS2656	50	µg/L
trans-1,3-Dichloropropene	ADS	ADS2656	50	µg/L
Trichloroethylene	ADS	ADS2656	50	µg/L
Vinyl Acetate	ADS	ADS2656	50	µg/L
Vinyl chloride	ADS	ADS2656	50	µg/L
Xylenes (total)	ADS	ADS2656	50	µg/L

<sup>1</sup>Practical Quantitation Limit. The PQL is sample specific and is dependent upon the number of dilutions made to the sample. The PQL for all samples was 50 µg/l, except FHB013, for which the PQL was 10 µg/L.

Table 4. Well Point and Near-Surface Water (bucket) Sampling Locations.

Sampling Location	SRS Coordinates		Elevation <sup>1</sup> (msl)
	North	East	
FHB013	73514	53840	211.2
FHB013B <sup>2</sup>	73514	53840	211.2
FHB014	73571	53729	212.0
FHB014B	73571	53729	212.0
FHB016	73718	54032	217.4
FHB017	73672	53990	217.5
FHB018	73567	54024	214.4
FHB018B	73567	54024	214.4
FHB019	73489	53896	211.6
FHB020	73431	54011	212.4
FHB020B	73431	54011	212.4
BGS002	93839	87018	177.7
BGS002B	93839	87018	177.7
BGS003	94252	81794	178.6
BGS003B	94252	81794	178.6

<sup>1</sup>Ground elevation above mean sea level.

<sup>2</sup>Location identification numbers ending with the suffix "B" designate near-surface water, or bucket, sampling locations.

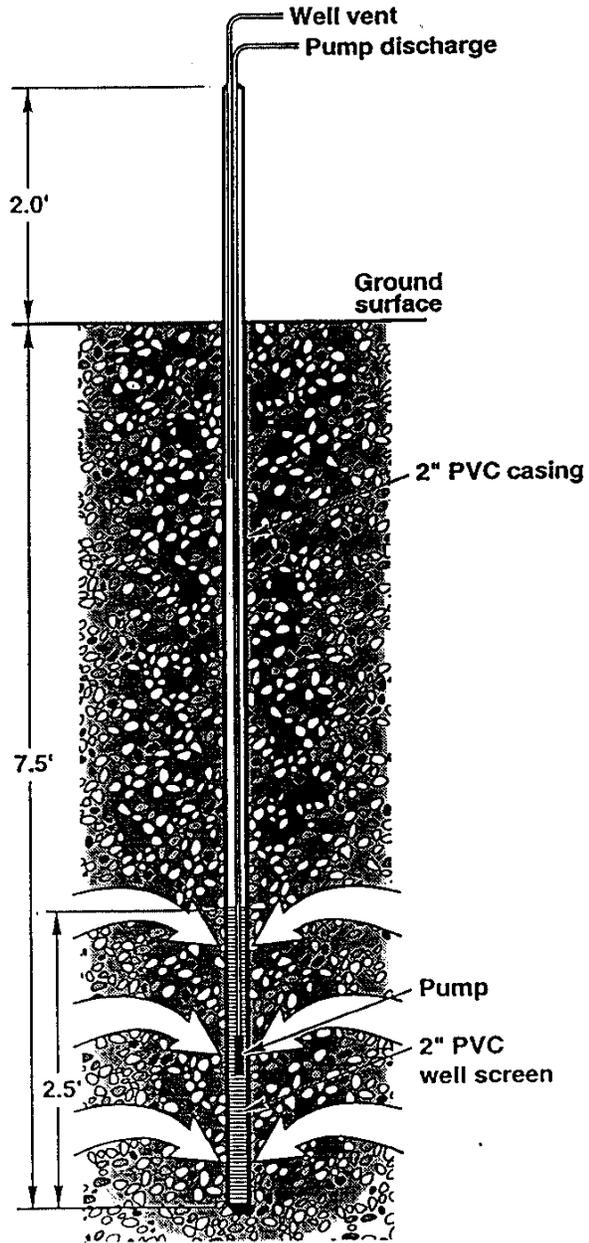


Figure 2. Schematic Diagram of Well Points

CVOC detected in the water for a particular well point and the resulting air concentrations were summed to give a total CVOC concentration in the headspace for the well point. Calculated CVOC concentrations were all less than 2 ppmv and were primarily influenced by 1,2-dichloroethylene (total). The GC was not calibrated with a 1,2-dichloroethylene (total) standard. Excluding 1,2-dichloroethylene (total) from the calculations yielded results similar to those measured with the GC.

### *Comparison of Well and Bucket Sampling Locations to Established Standards*

Since most of the analytical data from the well point and bucket sampling locations in the wetlands along the old F-Area effluent ditch were reported as below the method detection limit (MDL) and virtually all of the background data was reported as below the MDL, interpretation and analysis of the data are based completely on observed trends rather than statistical analyses. The small sample size and numerous data reported below the MDL make statistical analysis of the data impractical.

Four different analytes were detected at concentrations above the MDL for the well points (including background locations). These analytes were reported without any type of result or analysis qualifiers and represented eight percent of the analyses performed for each well point. Table 5 lists the detected analytes and the number of well points at which each analyte was detected.

Acetone, which is a common laboratory contaminant that is routinely detected in samples and method blanks, was detected at three well point locations during the March 1994 sampling event. Only one of the results was flagged by the lab due to method blank contamination. The results were considered suspect since one of the locations was a background location. Acetone was not detected in any of the well point samples collected during the May 1994 sampling event. This suggests that the March 1994 results were likely due to laboratory contamination, as suspected; however, before final conclusions are drawn, results from future sampling events should be assessed.

Figures 3, 4, and 5 show maps of the 1,2-dichloroethylene (total), trichloroethylene, and tritium plumes as measured in the well points at the old F-Area effluent ditch. For each figure, symbol size for the analyte presented is linearly proportional to the concentration measured. These figures should give the reader a feel for the general location where each of these analytes were measured as well as the concentration of each analyte. The highest 1,2-dichloroethylene (total) and tritium concentrations (280 µg/L and 51,100 pCi/ml, respectively) were measured at well point FHB018, whereas the highest trichloroethylene concentration (32.4 µg/L) was measured at well point FHB013.

A list of analytes, along with the corresponding well point locations where concentrations were at levels greater than those established by the PDWS (EPA, 1991) or the MCL (EPA, 1993a and b), is given in Table 6. Four analytes, 1,2-dichloroethylene (total), tetrachloroethylene, trichloroethylene, and tritium, were found to be above these standards at one or more well point locations.

Analytical Development Section (ADS) performed all VOC analyses on the bucket samples for the May 1994 sampling event. The practical quantitation limit (PQL) for all the bucket samples was 50 µg/L. The PQL is sample specific and is dependent upon the number of dilutions made to the sample prior to analysis. For the bucket samples, dilutions were made to all samples based on the cis-1,2-dichloroethylene concentrations at well point location FHB018. Well point location FHB018 routinely yields cis-1,2-dichloroethylene results an order of magnitude greater than any VOC measured at any of the other well point or bucket locations. Thus, the dilutions made based on this result, raised the PQL for the remaining samples (50 µg/L) above the range of expected concentrations. The PQL differs from the method detection limit (MDL) and the PQL is usually several times higher than the MDL for a particular analyte. This was confirmed with ADS and they said results reported below the PQL of 50 µg/L and flagged only with a "J" qualifier should be considered as detects. Had the samples not been diluted, the PQL would have been 10 µg/L. Therefore, for the purpose of this discussion, any results greater than 10 µg/L and flagged only with the "J" qualifier were considered as positive results.

Five different analytes were detected at concentrations above 10 µg/L for the bucket locations (including background locations). These analytes were reported without result or analysis qualifiers (except the "J" qualifier)

## Discussion

Analytical services were provided by several laboratories. Weston Analytics (WA) was used as the reference laboratory for well point volatile organic compound (VOC) analyses and Environmental Protection Department's (EPD) Environmental Monitoring Section (EMS) was used as the reference laboratory for well point and bucket tritium analyses. SRTC's Analytical Development Section (ADS) analyzed split samples and VOC bucket samples. For the well point samples, those analyzed by WA and EMS were considered the reference samples on which all analyses and interpretations were based. Well point samples analyzed by ADS were considered the quality assurance samples and were not used in the analyses and interpretations. ADS analyzed all bucket samples and those samples were considered the reference samples for the bucket locations. These considerations should prevent inter-lab variability of constituent concentrations from biasing the analyses and interpretation.

Analytical results for the May 1994 sampling event in the wetlands along the old F-Area effluent ditch south of 643-E (old burial ground) are presented in Appendix A. Location identification numbers ending with the "B" suffix designate bucket water sampling locations, whereas, all other identification numbers, excluding quality assurance samples, are well point locations. The results are presented in a format similar to that used in the quarterly groundwater monitoring reports issued by EMS (EMS, 1993). For each sampling location, the sampling date, field measurements, including water level elevation (well points only), pH, conductivity, etc., and all laboratory analyses, are presented (excluding lab blanks and standards). Included in the laboratory analysis tables for each location and analyte is the lab that performed the analyses, result or analytical modifiers, the analytical results, and standards (PDWS and MCL). The legend at the beginning of Appendix A explains the abbreviations and modifiers used to present the data. For locations where split samples were collected (well points FHB013 and FHB018), results from each lab are presented for each analyte. Laboratory duplicates were averaged with their corresponding reference samples, except in cases where the duplicate or reference analysis was reported at or below detection limit or in cases where results were qualified with a result or analysis qualifier.

In addition to the split samples, other standard quality assurance/quality control (QA/QC) samples were collected. These included a blind replicate sample collected at FHB014 (FHB014 R), a field blank sample collected at FHB018 (QA3), a trip blank sample carried throughout sampling activities (QA2), and a volatile trip blank shipped with the nonradioactive samples (FVA001). All samples deemed nonradioactive, as defined by Department of Transportation (DOT) regulations, were shipped in the same shipping container on May 26, 1994. By the same definition, samples considered to be limited quantities of exempted radioactive material were shipped in the same shipping container on May 26, 1994. To minimize volume for the radioactive shipment, a volatile trip blank was not shipped with these samples.

### *Field Measurements*

Conductivity and pH were measured at each well point and bucket location and the results are presented in Appendix A. Conductivity and pH results for well point and bucket locations are measurements taken as the samples were collected. Conductivity and pH measurements taken during well purging are given in the field data sheets. The pH of the well point and bucket locations ranged from 3.6 to 5.7 and 4.8 to 5.7, respectively. The average pH of the well point and bucket locations was 5.0 and 5.1, respectively. The conductivity of the well point and bucket locations ranged from 31 to 50  $\mu\text{S}/\text{cm}$  and 33 to 128  $\mu\text{S}/\text{cm}$ , respectively. The pH values measured for the well point and bucket locations are comparable to the pH measured at the background locations, while conductivity values suggest some locations are being influenced by elevated ion concentrations. Conductivity and pH measurements for well points and buckets are within the range commonly observed in the water table wells around 643-E.

At each well point, air samples were collected for analysis of chlorinated volatile organic compounds (CVOCs) by gas chromatography (GC). Trichloroethylene and tetrachloroethylene were detected in trace amounts (below 1 ppmv) in most of the samples. None of the other CVOC's were observed in any of the air samples. Calculations based on Henry's law constants were made to determine headspace CVOC concentrations (ppmv) given the concentration of CVOC's in the water for a particular well point (Appendix C). Calculations were made for each

and represented approximately 15 percent of the analyses performed for each bucket location. Table 7 lists the detected analytes and the number of buckets at which each analyte was detected.

A list of analytes, along with the corresponding bucket sampling locations where concentrations were at levels greater than those established by the PDWS or the MCL, is given in Table 8. Two analytes, cis-1,2-dichloroethylene and tritium, were above these standards at one or more bucket locations.

#### *Comparison of Well and Bucket Sampling Locations*

As mentioned earlier, a secondary objective of this project is to compare analytical results from the well and bucket sampling methodologies to determine differences. Originally it was planned to collect bucket samples during the first and third sampling events. However, to provide better resolution, samples were collected during the second event and analyzed by ADS. Since bucket samples were analyzed by ADS and well point samples were analyzed by Weston Analytics, there is potential interlab error; however, this error is expected to be small since the labs use the same analytical methods and both are certified in the contract laboratory program (CLP). Further, results of well point samples analyzed by both labs (FHB013 and FHB018) are very comparable.

The only VOC common to both well point and bucket samples was 1,2-dichloroethylene. Weston Analytics reports cis and trans-1,2-dichloroethylene together as total 1,2-dichloroethylene whereas ADS reports each compound separately. Past analysis has shown that cis-1,2-dichloroethylene, rather than trans-1,2-dichloroethylene, is the primary component of the 1,2-dichloroethylene (total) result reported by Weston Analytics. Therefore, cis-1,2-dichloroethylene results from the bucket locations should be compared to 1,2-dichloroethylene (total) from the well point locations. Trichloroethylene and tetrachloroethylene were detected at least once in the well points but not at any of the bucket locations. Toluene, chloromethane, and methyl ethyl ketone were detected at least once in the buckets but not in the well points

Local flow lines in areas of shallow groundwater discharge can be complex. Therefore, it would not be uncommon for sampling locations in close proximity to intercept different flow lines and yield water with different chemical characteristics, as indicated in Figure 6. For this project, well points and buckets were within a few feet of each other, but installed at different depths. Figure 6 shows a schematic diagram of the water table flow lines near 643-E. This figure clearly shows that well points and buckets (or near-surface samplers) installed at different depths can intercept different groundwater flow paths and plume concentration zones. Since plume concentrations can vary substantially across different flow lines, direct comparison of the results obtained from the two sampling devices would be inappropriate.

Volatilization losses of tritium should be negligible and if the sampling devices intercept the same flow lines, tritium results should be comparable. Percent differences between tritium concentrations measured in the well points and those measured in the buckets are shown in Table 9. Observed differences are substantial for those locations where significant concentrations were measured. These differences suggest that the well points and buckets intercept different flow lines. As a result, differences would be expected and were observed between VOC concentrations measured in the well points and those measured in the buckets.

For the March 1994 sampling event, VOC concentrations measured at each well point and bucket location were normalized based on the percent differences observed in tritium concentrations at that location. Resulting well point and bucket VOC concentrations were comparable in most cases. As mentioned earlier, for the May 1994 sampling event, the only VOC common to both the well points and buckets was 1,2-dichloroethylene. Percent differences in tritium concentrations were used to normalize the 1,2-dichloroethylene concentrations measured in the buckets. Again it was found that resulting well point and bucket VOC concentrations were comparable. These results suggested that volatilization losses of organic compounds from the buckets may be negligible. For direct comparison of results in the future, it will be necessary to install well points and buckets to about the same depth to ensure that the devices intercept the same groundwater flow paths.

### 1,2-Dichloroethylene Plume at Old F-Area Effluent Ditch

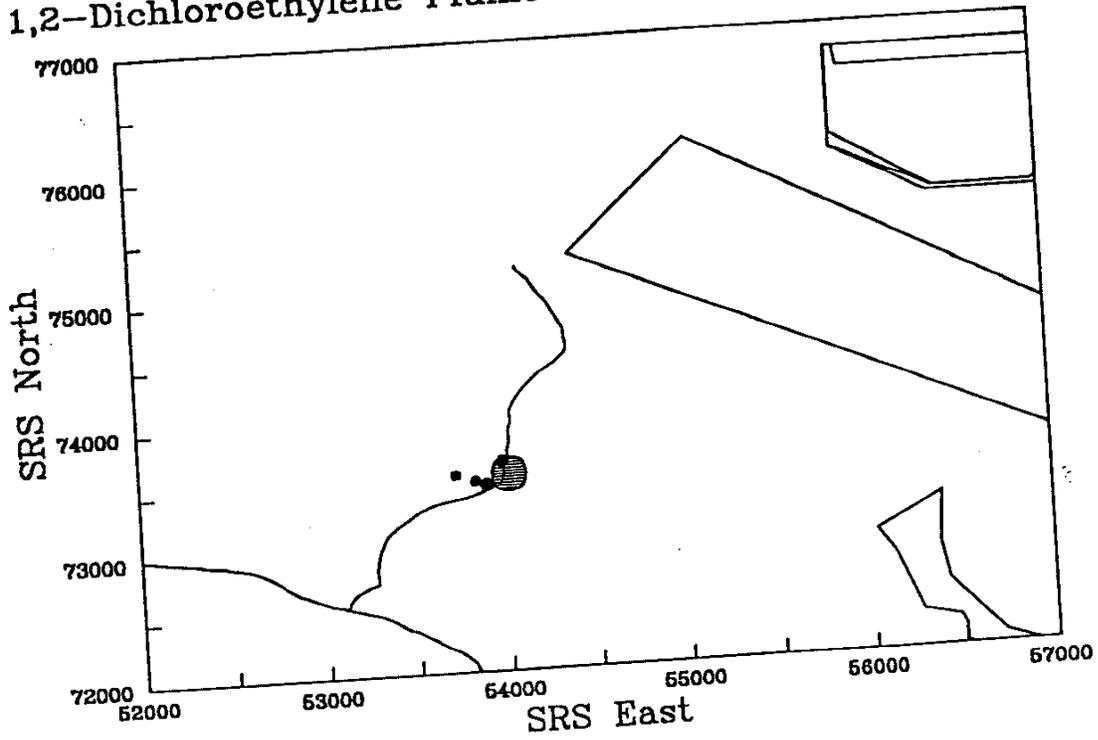


Figure 3. 1,2-Dichloroethylene (total) Plume at the Old F-Area Effluent Ditch.

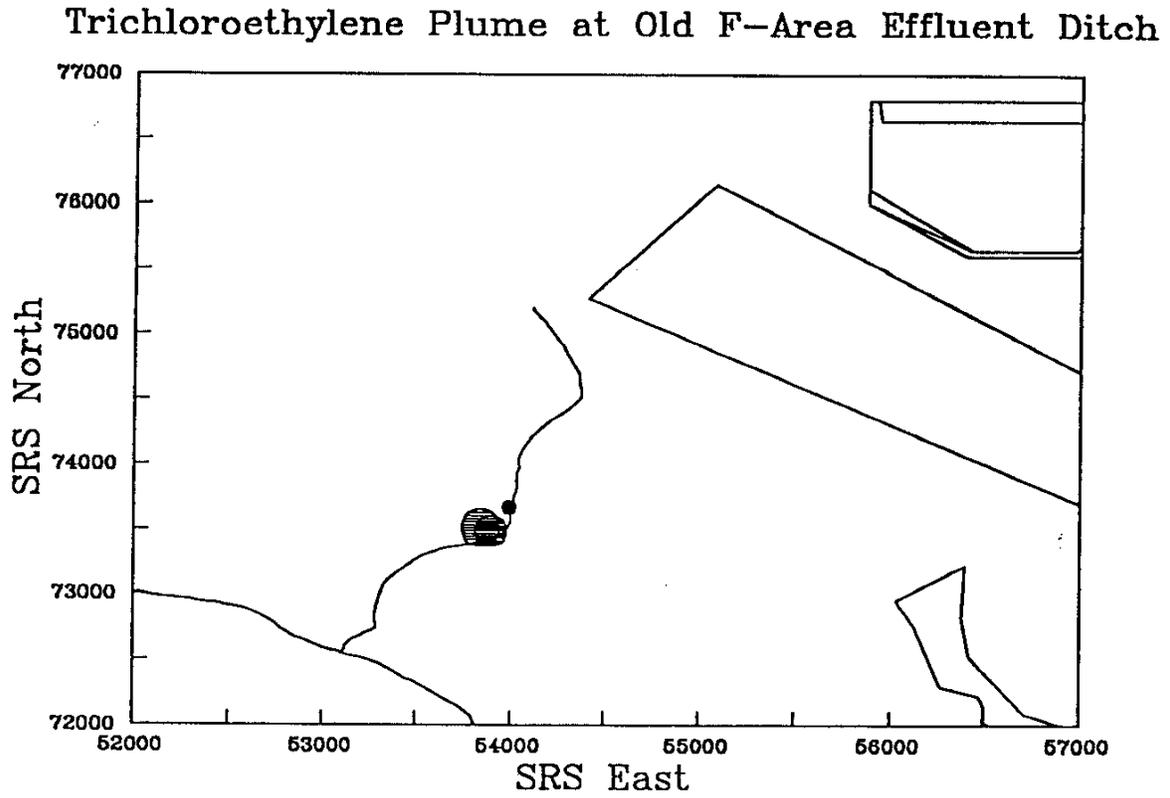


Figure 4. Trichloroethylene Plume at the Old F-Area Effluent Ditch.

### Tritium Plume at Old F-Area Effluent Ditch

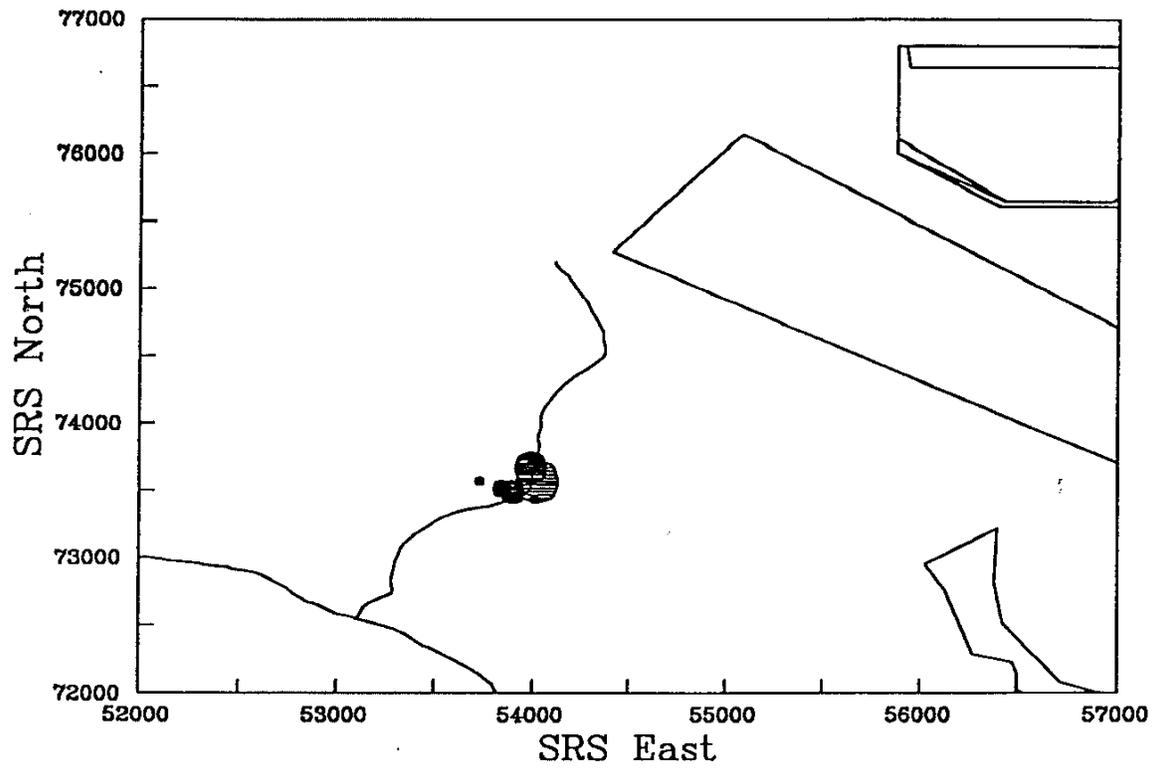


Figure 5. Tritium Plume at the Old F-Area Effluent Ditch.

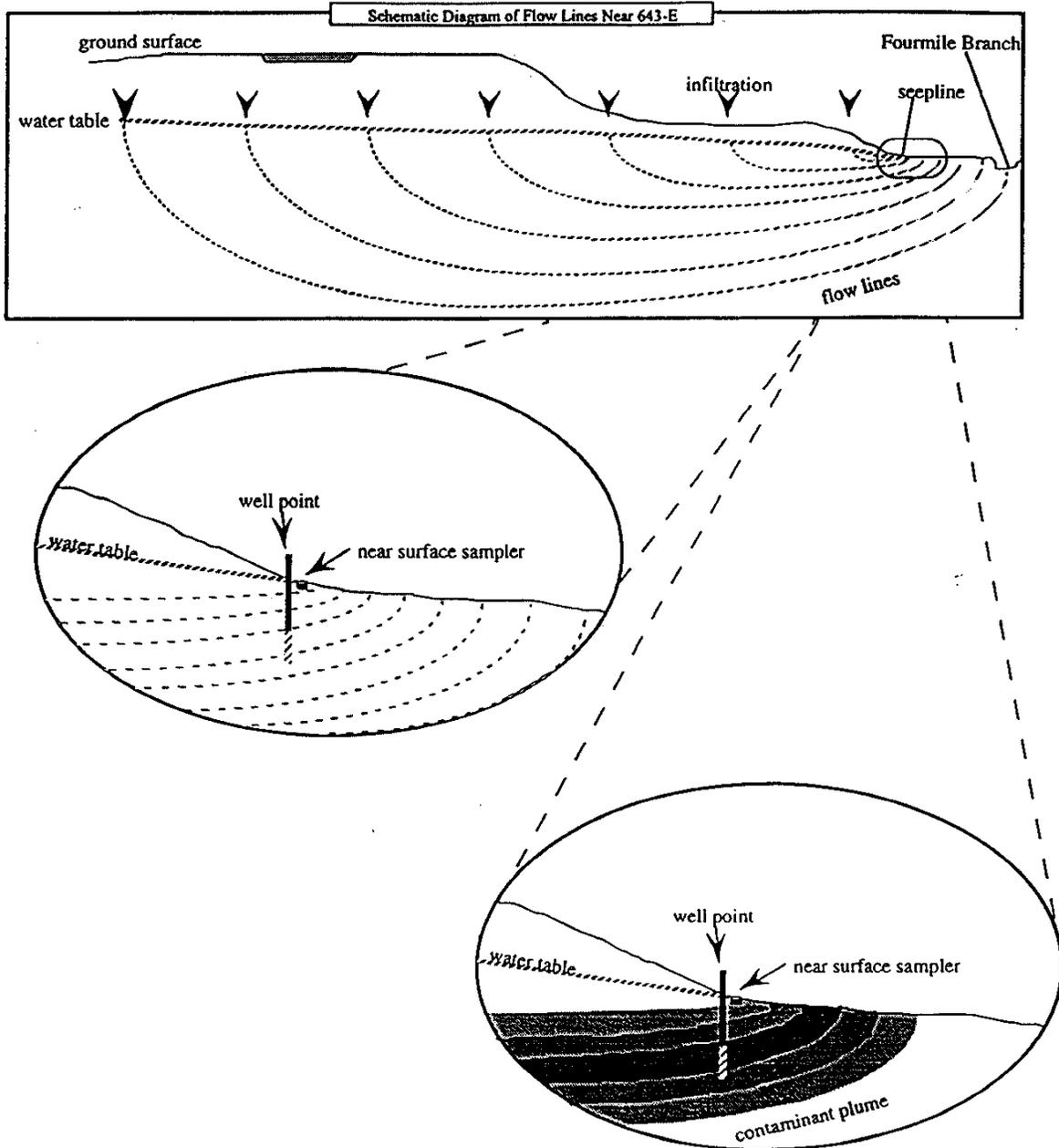


Figure 6. Schematic Diagram of Flow Lines Near 643-E with the Seepage Intercept Enlarged to Show How Well Points and Near-Surface Water Samplers (buckets) Intercept Different Flow Lines and Plume Concentrations Zones.

Table 5. Analytes Detected at Well Point Locations.

Analyte	Number of Well Points	Number of Detects
1,2-Dichloroethylene (total)	9	5
Tetrachloroethylene	9	2
Trichloroethylene	9	3
Tritium	9	8

Table 6. Analytes and Corresponding Well Point Locations Where May 1994 Concentrations Exceeded the Primary Drinking Water Standards or Maximum Contaminant Levels.

Analyte	Lab	Result	Standard	Units	Location
1,2-Dichloroethylene (total)	WA	280	70	µg/L	FHB018
Tetrachloroethylene	WA	8.6	5	µg/L	FHB019
Trichloroethylene	WA	28.7	5	µg/L	FHB013
Trichloroethylene	WA	9.9	5	µg/L	FHB017
Trichloroethylene	WA	23.8	5	µg/L	FHB019
Tritium	EMS	14100	20	pCi/ml	FHB013
Tritium	EMS	1060	20	pCi/ml	FHB014
Tritium	EMS	1440	20	pCi/ml	FHB016
Tritium	EMS	35000	20	pCi/ml	FHB017
Tritium	EMS	51100	20	pCi/ml	FHB018
Tritium	EMS	23500	20	pCi/ml	FHB019

Table 7. Analytes Detected at Bucket Locations.

Analyte	Number of Buckets	Number of Detects
Chloromethane	6	2
cis-1,2-Dichloroethylene	6	3
Methyl ethyl ketone	6	1
Toluene	6	1
Tritium	6	5

Table 8. Analytes and Corresponding Bucket Locations Where May 1994 Concentrations Exceeded the Primary Drinking Water Standards or Maximum Contaminant Levels.

Analyte	Lab	Result	Standard	Units	Location
cis-1,2-Dichloroethylene	ADS	96	70	µg/L	FHB018B
Tritium	EMS	16400	20	pCi/ml	FHB013B
Tritium	EMS	1190	20	pCi/ml	FHB014B
Tritium	EMS	35400	20	pCi/ml	FHB018B
Tritium	EMS	26.8	20	pCi/ml	FHB020B

Table 9. Comparison of Tritium Results from Well and Bucket Locations.

Location		Result	Percent Difference <sup>1</sup>	Units
BGS003	Tritium	2.02	2.5	pCi/ml
BGS003B	Tritium	1.97		pCi/ml
FHB013	Tritium	14100	-16.3	pCi/ml
FHB013B	Tritium	16400		pCi/ml
FHB014	Tritium	1060	-12.3	pCi/ml
FHB014B	Tritium	1190		pCi/ml
FHB018	Tritium	51100	30.7	pCi/ml
FHB018B	Tritium	35400		pCi/ml
FHB020	Tritium	1.7	-1476.5	pCi/ml
FHB020B	Tritium	26.8		pCi/ml

<sup>1</sup>Differences observed in tritium concentrations suggest that the wells and buckets intercept different groundwater flow paths.



## Summary

Results of the May 1994 sampling event suggest that tritium and volatile organic compounds originating from 643-E are outcropping in the wetlands near the old F-Area effluent ditch. Samples collected from well point and bucket sampling locations contained measurable quantities of organics and tritium. These findings are expected because organics are routinely detected in water table wells in the vicinity of 643-E and shallow groundwater flowing beneath the southwest corner of 643-E outcrops in the wetlands along the old F-Area effluent ditch.

Four different analytes were detected in the well points at least once at concentrations greater than the method detection limit. These analytes included 1,2-dichloroethylene, tetrachloroethylene, trichloroethylene, and tritium. All four analytes were detected at levels above the PDWS or the MCL at one or more locations.

Five analytes, chloromethane, cis-1,2-dichloroethylene, methyl ethyl ketone, toluene, and tritium, were detected in the buckets, at least once, at concentrations greater than the method detection limit. Tritium and cis-1,2-dichloroethylene were detected at concentrations above the limits established by the PDWS or MCL at one or more bucket locations.

An attempt was made to compare analytical results between well point and bucket sampling locations. Differences in tritium concentrations at each location indicated that the sampling devices, which were installed at different depths, intercepted different groundwater flow paths. This negated direct comparison of analytical results. However, results suggest that volatilization losses of VOCs from the buckets were negligible. In the future, well points and buckets should be installed to about the same depth for adequate comparisons of the two sampling methodologies.

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*Appendix A*  
*May 1994 Analytical Results*



**ANALYTICAL RESULTS**

Analytical results from the May 1994 sampling event in the wetlands along the old F-Area effluent ditch are presented in this section. The analytical data for each location (well point and bucket) are presented in tabular format with the background samples presented first followed by the well point samples, bucket samples, and QA/QC samples. Locations ending with the suffix "B" designate bucket sampling locations.

For each sampling location, the results of all field measurements are presented including depth to water (well points only), water level elevation (well points only), pH, and specific conductance. This information is followed by all laboratory analyses for each location. For each analyte in the laboratory analyses tables, the following information is presented: the lab that performed the analysis, any result and analysis qualifiers, the analytical results, any appropriate standard for comparison, and the units in which the results are presented.

The following is a list of abbreviations used in the tables:

- Lab    Laboratory performing analysis
- Mod    Result or analysis qualifier
- Rep    Replication sample (denoted by "R")
- Avg    Average code
  - "A" - reference sample averaged with lab duplicates for that sample
  - "NA" - reference sample not averaged with lab duplicates for that sample
- WA    Weston Analytics
- ADS    Analytical Development Section
- EMS    Environmental Monitoring Section

**Result and Analysis Qualifiers**

Result Qualifiers	
CODE	DESCRIPTION
(BLANK)	Data not remarked. Number should be interpreted exactly as reported.
J	The analytical result is an estimated quantity.
R	Rejected because of performance requirements in the sample or associated quality control analyses were not met. The analyte may or may not be present.
U	Material analyzed for but not detected. Value reported is the sample quantification limit.

Analysis Qualifiers	
CODE	DESCRIPTION
Q	Sample held beyond normal holding time. If holding time is exceeded by less than 30 days, used with "J". If holding time is exceeded by more than 30 days, used with "R".
V	Indicates the analyte was detected in both the sample and associated method blank.



Quarterly Sampling of the Wetlands Along the Old F-Area Effluent Ditch: May 1994

BGS002

Field Measurements:

Sample Date: 05/24/94

pH: 5.0

Conductivity: 26  $\mu$ S/cm

Analyte	Lab	Rep	Avg	Qualifier	Result	Stand	Units
1,1,1,2-Tetrachloroethane	WA			U	10	NS	$\mu$ g/L
1,1,1-Trichloroethane	WA			U	5	200	$\mu$ g/L
1,1,2,2-Tetrachloroethane	WA			U	5	NS	$\mu$ g/L
1,1,2-Trichloroethane	WA			U	5	5	$\mu$ g/L
1,1-Dichloroethane	WA			U	5	NS	$\mu$ g/L
1,1-Dichloroethylene	WA			U	5	7	$\mu$ g/L
1,2,3-Trichloropropane	WA			U	10	NS	$\mu$ g/L
1,2-Dibromo-3-chloropropane	WA			U	20	0.2	$\mu$ g/L
1,2-Dibromoethane	WA			U	20	NS	$\mu$ g/L
1,2-Dichloroethane	WA			U	5	5	$\mu$ g/L
1,2-Dichloroethylene	WA			U	5	70	$\mu$ g/L
1,2-Dichloropropane	WA			U	5	5	$\mu$ g/L
2-Hexanone	WA			U	10	NS	$\mu$ g/L
4-Methyl-2-pentanone	WA			U	10	NS	$\mu$ g/L
Acetone	WA			U	10	NS	$\mu$ g/L
Acetonitrile (Methyl cyanide)	WA			U	20	NS	$\mu$ g/L
Acrolein	WA			U	10	NS	$\mu$ g/L
Acrylonitrile	WA			U	10	NS	$\mu$ g/L
Allyl chloride	WA			U	100	NS	$\mu$ g/L
Benzene	WA			U	5	5	$\mu$ g/L
Bromodichloromethane	WA			U	5	NS	$\mu$ g/L
Bromoform	WA			U	5	NS	$\mu$ g/L
Carbon disulfide	WA			U	5	NS	$\mu$ g/L
Carbon tetrachloride	WA			U	5	5	$\mu$ g/L
Chlorobenzene	WA			U	5	100	$\mu$ g/L
Chloroethane	WA			U	10	NS	$\mu$ g/L
Chloroform	WA			U	5	100	$\mu$ g/L
Chloroprene	WA			U	20	NS	$\mu$ g/L
cis-1,3-Dichloropropene	WA			U	5	NS	$\mu$ g/L
Dibromochloromethane	WA			U	5	NS	$\mu$ g/L
Dichlorodifluoromethane	WA			U	10	NS	$\mu$ g/L
Ethylbenzene	WA			U	5	700	$\mu$ g/L
Isobutyl alcohol	WA			U	20	NS	$\mu$ g/L
Methacrylonitrile	WA			U	20	NS	$\mu$ g/L
Methyl bromide	WA			U	10	NS	$\mu$ g/L
Methyl chloride	WA			U	10	NS	$\mu$ g/L
Methyl ethyl ketone	WA			U	10	NS	$\mu$ g/L
Methyl iodide	WA			U	10	NS	$\mu$ g/L
Methylene bromide	WA			U	10	NS	$\mu$ g/L
Methylene chloride	WA			U	5	5	$\mu$ g/L
Propionitrile	WA			U	50	NS	$\mu$ g/L
Styrene	WA			U	5	100	$\mu$ g/L
Tetrachloroethylene	WA			U	5	5	$\mu$ g/L

BGS002 Laboratory Analyses (Cont'd)

Analyte	Lab	Rep	Avg	Qualifier	Result	Stand	Units
Toluene	WA			U	5	1000	µg/L
trans-1,3-Dichloropropene	WA			U	5	NS	µg/L
trans-1,4-Dichloro-2-butene	WA			U	100	NS	µg/L
Trichloroethylene	WA			U	5	5	µg/L
Trichlorofluoromethane	WA			U	5	NS	µg/L
Tritium	EMS			U	2	20	pCi/ml
Vinyl acetate	WA			U	10	NS	µg/L
Vinyl chloride	WA			U	10	2	µg/L
Xylenes (total)	WA			U	5	10000	µg/L

*Quarterly Sampling of the Wetlands Along the Old F-Area Effluent Ditch: May 1994*

BGS002B

Field Measurements:

Sample Date: 5/24/94

pH: 5.2

Conductivity: 54  $\mu$ S/cm

Analyte	Lab	Rep	Avg	Qualifier	Result	Stand	Units
1,1,1-Trichloroethane	ADS			U	50	200	$\mu$ g/L
1,1,2,2-Tetrachloroethane	ADS			U	50	NS	$\mu$ g/L
1,1,2-Trichloroethane	ADS			U	50	5	$\mu$ g/L
1,1-Dichloroethane	ADS			U	50	NS	$\mu$ g/L
1,1-Dichloroethylene	ADS			U	50	7	$\mu$ g/L
1,2-Dichloroethane	ADS			U	50	5	$\mu$ g/L
1,2-Dichloropropane	ADS			J	1	5	$\mu$ g/L
2-Hexanone	ADS			U	50	NS	$\mu$ g/L
4-Methyl-2-pentanone	ADS			U	50	NS	$\mu$ g/L
Acetone	ADS			U	50	NS	$\mu$ g/L
Benzene	ADS			J	1	5	$\mu$ g/L
Bromodichloromethane	ADS			U	50	NS	$\mu$ g/L
Bromoform	ADS			U	50	NS	$\mu$ g/L
Carbon disulfide	ADS			J	8	NS	$\mu$ g/L
Carbon tetrachloride	ADS			U	50	5	$\mu$ g/L
Chlorobenzene	ADS			U	50	100	$\mu$ g/L
Chloroethane	ADS			U	50	NS	$\mu$ g/L
Chloroform	ADS			J	1	100	$\mu$ g/L
cis-1,2-Dichloroethylene	ADS			U	50	70	$\mu$ g/L
cis-1,3-Dichloropropene	ADS			U	50	NS	$\mu$ g/L
Dibromochloromethane	ADS			U	50	NS	$\mu$ g/L
Ethylbenzene	ADS			U	50	700	$\mu$ g/L
Methyl chloride	ADS			U	50	NS	$\mu$ g/L
Methyl ethyl ketone	ADS			U	50	NS	$\mu$ g/L
Methylene chloride	ADS			U	50	5	$\mu$ g/L
Styrene	ADS			U	50	100	$\mu$ g/L
Tetrachloroethylene	ADS			J	7	5	$\mu$ g/L
Toluene	ADS			J	11	1000	$\mu$ g/L
trans-1,2-Dichloroethylene	ADS			U	50	100	$\mu$ g/L
trans-1,3-Dichloropropene	ADS			U	50	NS	$\mu$ g/L
Trichloroethylene	ADS			U	50	5	$\mu$ g/L
Tritium	EMS			U	2	20	pCi/ml
Vinyl Acetate	ADS			U	50	NS	$\mu$ g/L
Vinyl chloride	ADS			U	50	2	$\mu$ g/L
Xylenes (total)	ADS			U	50	10000	$\mu$ g/L

BGS003

## Field Measurements:

Sample Date: 05/24/94

pH: 5.3

Conductivity: 23  $\mu$ S/cm

Analyte	Lab	Rep	Avg	Qualifier	Result	Stand	Units
1,1,1,2-Tetrachloroethane	WA			U	10	NS	$\mu$ g/L
1,1,1-Trichloroethane	WA			U	5	200	$\mu$ g/L
1,1,2,2-Tetrachloroethane	WA			U	5	NS	$\mu$ g/L
1,1,2-Trichloroethane	WA			U	5	5	$\mu$ g/L
1,1-Dichloroethane	WA			U	5	NS	$\mu$ g/L
1,1-Dichloroethylene	WA			U	5	7	$\mu$ g/L
1,2,3-Trichloropropane	WA			U	10	NS	$\mu$ g/L
1,2-Dibromo-3-chloropropane	WA			U	20	0.2	$\mu$ g/L
1,2-Dibromoethane	WA			U	20	NS	$\mu$ g/L
1,2-Dichloroethane	WA			U	5	5	$\mu$ g/L
1,2-Dichloroethylene	WA			U	5	70	$\mu$ g/L
1,2-Dichloropropane	WA			U	5	5	$\mu$ g/L
2-Hexanone	WA			U	10	NS	$\mu$ g/L
4-Methyl-2-pentanone	WA			U	10	NS	$\mu$ g/L
Acetone	WA			V	15.1	NS	$\mu$ g/L
Acetonitrile (Methyl cyanide)	WA			U	20	NS	$\mu$ g/L
Acrolein	WA			U	10	NS	$\mu$ g/L
Acrylonitrile	WA			U	10	NS	$\mu$ g/L
Allyl chloride	WA			U	100	NS	$\mu$ g/L
Benzene	WA			U	5	5	$\mu$ g/L
Bromodichloromethane	WA			U	5	NS	$\mu$ g/L
Bromoform	WA			U	5	NS	$\mu$ g/L
Carbon disulfide	WA			U	5	NS	$\mu$ g/L
Carbon tetrachloride	WA			U	5	5	$\mu$ g/L
Chlorobenzene	WA			U	5	100	$\mu$ g/L
Chloroethane	WA			U	10	NS	$\mu$ g/L
Chloroform	WA			U	5	100	$\mu$ g/L
Chloroprene	WA			U	20	NS	$\mu$ g/L
cis-1,3-Dichloropropene	WA			U	5	NS	$\mu$ g/L
Dibromochloromethane	WA			U	5	NS	$\mu$ g/L
Dichlorodifluoromethane	WA			U	10	NS	$\mu$ g/L
Ethylbenzene	WA			J	1.51	700	$\mu$ g/L
Isobutyl alcohol	WA			U	20	NS	$\mu$ g/L
Methacrylonitrile	WA			U	20	NS	$\mu$ g/L
Methyl bromide	WA			U	10	NS	$\mu$ g/L
Methyl chloride	WA			U	10	NS	$\mu$ g/L
Methyl ethyl ketone	WA			U	10	NS	$\mu$ g/L
Methyl iodide	WA			U	10	NS	$\mu$ g/L
Methylene bromide	WA			U	10	NS	$\mu$ g/L
Methylene chloride	WA			JV	1.05	5	$\mu$ g/L
Propionitrile	WA			U	50	NS	$\mu$ g/L
Styrene	WA			U	5	100	$\mu$ g/L
Tetrachloroethylene	WA			U	5	5	$\mu$ g/L

BGS003 Laboratory Analyses (Cont'd)

Analyte	Lab	Rep	Avg	Qualifier	Result	Stand	Units
Toluene	WA			U	5	1000	µg/L
trans-1,3-Dichloropropene	WA			U	5	NS	µg/L
trans-1,4-Dichloro-2-butene	WA			U	100	NS	µg/L
Trichloroethylene	WA			U	5	5	µg/L
Trichlorofluoromethane	WA			U	5	NS	µg/L
Tritium	EMS			U	2	20	pCi/ml
Vinyl acetate	WA			U	10	NS	µg/L
Vinyl chloride	WA			U	10	2	µg/L
Xylenes (total)	WA			U	5	10000	µg/L

BGS003B

## Field Measurements:

Sample Date: 5/24/94

pH: 4.6

Conductivity: 27  $\mu$ S/cm

Analyte	Lab	Rep	Avg	Qualifier	Result	Stand	Units
1,1,1-Trichloroethane	ADS			U	50	200	$\mu$ g/L
1,1,2,2-Tetrachloroethane	ADS			U	50	NS	$\mu$ g/L
1,1,2-Trichloroethane	ADS			U	50	5	$\mu$ g/L
1,1-Dichloroethane	ADS			U	50	NS	$\mu$ g/L
1,1-Dichloroethylene	ADS			U	50	7	$\mu$ g/L
1,2-Dichloroethane	ADS			U	50	5	$\mu$ g/L
1,2-Dichloropropane	ADS			J	1	5	$\mu$ g/L
2-Hexanone	ADS			U	50	NS	$\mu$ g/L
4-Methyl-2-pentanone	ADS			U	50	NS	$\mu$ g/L
Acetone	ADS			U	50	NS	$\mu$ g/L
Benzene	ADS			J	0	5	$\mu$ g/L
Bromodichloromethane	ADS			U	50	NS	$\mu$ g/L
Bromoform	ADS			U	50	NS	$\mu$ g/L
Carbon disulfide	ADS			J	5	NS	$\mu$ g/L
Carbon tetrachloride	ADS			U	50	5	$\mu$ g/L
Chlorobenzene	ADS			U	50	100	$\mu$ g/L
Chloroethane	ADS			U	50	NS	$\mu$ g/L
Chloroform	ADS			U	50	100	$\mu$ g/L
cis-1,2-Dichloroethylene	ADS			U	50	70	$\mu$ g/L
cis-1,3-Dichloropropene	ADS			U	50	NS	$\mu$ g/L
Dibromochloromethane	ADS			U	50	NS	$\mu$ g/L
Ethylbenzene	ADS			U	50	700	$\mu$ g/L
Methyl chloride	ADS			J	31	NS	$\mu$ g/L
Methyl ethyl ketone	ADS			U	50	NS	$\mu$ g/L
Methylene chloride	ADS			U	50	5	$\mu$ g/L
Styrene	ADS			U	50	100	$\mu$ g/L
Tetrachloroethylene	ADS			J	6	5	$\mu$ g/L
Toluene	ADS			U	50	1000	$\mu$ g/L
trans-1,2-Dichloroethylene	ADS			U	50	100	$\mu$ g/L
trans-1,3-Dichloropropene	ADS			U	50	NS	$\mu$ g/L
Trichloroethylene	ADS			U	50	5	$\mu$ g/L
Tritium	EMS			U	2	20	pCi/ml
Vinyl Acetate	ADS			U	50	NS	$\mu$ g/L
Vinyl chloride	ADS			U	50	2	$\mu$ g/L
Xylenes (total)	ADS			U	50	10000	$\mu$ g/L

*Quarterly Sampling of the Wetlands Along the Old F-Area Effluent Ditch: May 1994*

FHB013

Field Measurements:

Sample Date: 05/24/94

pH: 5.1

Conductivity: 44 µS/cm

Analyte	Lab	Rep	Avg	Qualifier	Result	Stand	Units
1,1,1,2-Tetrachloroethane	WA			U	10	NS	µg/L
1,1,1-Trichloroethane	ADS			J	1	200	µg/L
1,1,1-Trichloroethane	WA			U	5	200	µg/L
1,1,2,2-Tetrachloroethane	ADS			U	10	NS	µg/L
1,1,2,2-Tetrachloroethane	WA			U	5	NS	µg/L
1,1,2-Trichloroethane	ADS			U	10	5	µg/L
1,1,2-Trichloroethane	WA			U	5	5	µg/L
1,1-Dichloroethane	ADS			U	10	NS	µg/L
1,1-Dichloroethane	WA			U	5	NS	µg/L
1,1-Dichloroethylene	ADS			U	10	7	µg/L
1,1-Dichloroethylene	WA			U	5	7	µg/L
1,2,3-Trichloropropane	WA			U	10	NS	µg/L
1,2-Dibromo-3-chloropropane	WA			U	20	0.2	µg/L
1,2-Dibromoethane	WA			U	20	NS	µg/L
1,2-Dichloroethane	ADS			U	10	5	µg/L
1,2-Dichloroethane	WA			U	5	5	µg/L
1,2-Dichloroethylene	WA				29.4	70	µg/L
1,2-Dichloropropane	ADS			U	10	5	µg/L
1,2-Dichloropropane	WA			U	5	5	µg/L
2-Hexanone	ADS			U	10	NS	µg/L
2-Hexanone	WA			U	10	NS	µg/L
4-Methyl-2-pentanone	ADS			U	10	NS	µg/L
4-Methyl-2-pentanone	WA			U	10	NS	µg/L
Acetone	ADS			U	10	NS	µg/L
Acetone	WA			U	10	NS	µg/L
Acetonitrile (Methyl cyanide)	WA			U	20	NS	µg/L
Acrolein	WA			U	10	NS	µg/L
Acrylonitrile	WA			U	10	NS	µg/L
Allyl chloride	WA			U	100	NS	µg/L
Benzene	ADS			J	1	5	µg/L
Benzene	WA			U	5	5	µg/L
Bromodichloromethane	ADS			U	10	NS	µg/L
Bromodichloromethane	WA			U	5	NS	µg/L
Bromoform	ADS			U	10	NS	µg/L
Bromoform	WA			U	5	NS	µg/L
Carbon disulfide	ADS			J	0	NS	µg/L
Carbon disulfide	WA			U	5	NS	µg/L
Carbon tetrachloride	ADS			J	1	5	µg/L
Carbon tetrachloride	WA			J	1.08	5	µg/L
Chlorobenzene	ADS			U	10	100	µg/L
Chlorobenzene	WA			U	5	100	µg/L
Chloroethane	ADS			U	10	NS	µg/L
Chloroethane	WA			U	10	NS	µg/L

## FHB013 Laboratory Analyses (Cont'd)

Analyte	Lab	Rep	Avg	Qualifier	Result	Stand	Units
Chloroform	ADS			J	2	100	µg/L
Chloroform	WA			J	1.92	100	µg/L
Chloroprene	WA			U	20	NS	µg/L
cis-1,2-Dichloroethylene	ADS				37	70	µg/L
cis-1,3-Dichloropropene	ADS			U	10	NS	µg/L
cis-1,3-Dichloropropene	WA			U	5	NS	µg/L
Dibromochloromethane	ADS			U	10	NS	µg/L
Dibromochloromethane	WA			U	5	NS	µg/L
Dichlorodifluoromethane	WA			U	10	NS	µg/L
Ethylbenzene	ADS			U	10	700	µg/L
Ethylbenzene	WA			U	5	700	µg/L
Isobutyl alcohol	WA			U	20	NS	µg/L
Methacrylonitrile	WA			U	20	NS	µg/L
Methyl bromide	WA			U	10	NS	µg/L
Methyl chloride	ADS			U	10	NS	µg/L
Methyl chloride	WA			U	10	NS	µg/L
Methyl ethyl ketone	ADS			U	10	NS	µg/L
Methyl ethyl ketone	WA			U	10	NS	µg/L
Methyl iodide	WA			U	10	NS	µg/L
Methylene bromide	WA			U	10	NS	µg/L
Methylene chloride	ADS			U	10	5	µg/L
Methylene chloride	WA			V	6.91	5	µg/L
Propionitrile	WA			U	50	NS	µg/L
Styrene	ADS			U	10	100	µg/L
Styrene	WA			U	5	100	µg/L
Tetrachloroethylene	ADS			J	6	5	µg/L
Tetrachloroethylene	WA				5	5	µg/L
Toluene	ADS			J	0	1000	µg/L
Toluene	WA			U	5	1000	µg/L
trans-1,2-Dichloroethylene	ADS			J	0	100	µg/L
trans-1,3-Dichloropropene	ADS			U	10	NS	µg/L
trans-1,3-Dichloropropene	WA			U	5	NS	µg/L
trans-1,4-Dichloro-2-butene	WA			U	100	NS	µg/L
Trichloroethylene	ADS				29	5	µg/L
Trichloroethylene	WA				28.7	5	µg/L
Trichlorofluoromethane	WA			U	5	NS	µg/L
Tritium	EMS				14100	20	pCi/ml
Vinyl Acetate	ADS			J	1	NS	µg/L
Vinyl acetate	WA			U	10	NS	µg/L
Vinyl chloride	ADS			U	10	2	µg/L
Vinyl chloride	WA			U	10	2	µg/L
Xylenes (total)	ADS			U	10	10000	µg/L
Xylenes (total)	WA			U	.5	10000	µg/L

*Quarterly Sampling of the Wetlands Along the Old F-Area Effluent Ditch: May 1994*

FHB013B

Field Measurements:

Sample Date: 5/24/94

pH: 4.8

Conductivity: 38  $\mu$ S/cm

Analyte	Lab	Rep	Avg	Qualifier	Result	Stand	Units
1,1,1-Trichloroethane	ADS			U	50	200	$\mu$ g/L
1,1,2,2-Tetrachloroethane	ADS			U	50	NS	$\mu$ g/L
1,1,2-Trichloroethane	ADS			U	50	5	$\mu$ g/L
1,1-Dichloroethane	ADS			U	50	NS	$\mu$ g/L
1,1-Dichloroethylene	ADS			U	50	7	$\mu$ g/L
1,2-Dichloroethane	ADS			U	50	5	$\mu$ g/L
1,2-Dichloropropane	ADS			U	50	5	$\mu$ g/L
2-Hexanone	ADS			U	50	NS	$\mu$ g/L
4-Methyl-2-pentanone	ADS			U	50	NS	$\mu$ g/L
Acetone	ADS			U	50	NS	$\mu$ g/L
Benzene	ADS			J	0	5	$\mu$ g/L
Bromodichloromethane	ADS			U	50	NS	$\mu$ g/L
Bromoform	ADS			U	50	NS	$\mu$ g/L
Carbon disulfide	ADS			J	2	NS	$\mu$ g/L
Carbon tetrachloride	ADS			U	50	5	$\mu$ g/L
Chlorobenzene	ADS			U	50	100	$\mu$ g/L
Chloroethane	ADS			U	50	NS	$\mu$ g/L
Chloroform	ADS			J	1	100	$\mu$ g/L
cis-1,2-Dichloroethylene	ADS			J	31	70	$\mu$ g/L
cis-1,3-Dichloropropene	ADS			U	50	NS	$\mu$ g/L
Dibromochloromethane	ADS			U	50	NS	$\mu$ g/L
Ethylbenzene	ADS			U	50	700	$\mu$ g/L
Methyl chloride	ADS			U	50	NS	$\mu$ g/L
Methyl ethyl ketone	ADS			U	50	NS	$\mu$ g/L
Methylene chloride	ADS			U	50	5	$\mu$ g/L
Styrene	ADS			U	50	100	$\mu$ g/L
Tetrachloroethylene	ADS			J	6	5	$\mu$ g/L
Toluene	ADS			J	0	1000	$\mu$ g/L
trans-1,2-Dichloroethylene	ADS			U	50	100	$\mu$ g/L
trans-1,3-Dichloropropene	ADS			U	50	NS	$\mu$ g/L
Trichloroethylene	ADS			U	50	5	$\mu$ g/L
Tritium	EMS				16400	20	pCi/ml
Vinyl Acetate	ADS			U	50	NS	$\mu$ g/L
Vinyl chloride	ADS			U	50	2	$\mu$ g/L
Xylenes (total)	ADS			U	50	10000	$\mu$ g/L

FHB014

Field Measurements:

Sample Date: 05/24/94

pH: 5.3

Conductivity: 34  $\mu$ S/cm

Analyte	Lab	Rep	Avg	Qualifier	Result	Stand	Units
1,1,1,2-Tetrachloroethane	WA			U	10	NS	$\mu$ g/L
1,1,1,2-Tetrachloroethane	WA	R		U	10	NS	$\mu$ g/L
1,1,1-Trichloroethane	WA			U	5	200	$\mu$ g/L
1,1,1-Trichloroethane	WA	R		U	5	200	$\mu$ g/L
1,1,2,2-Tetrachloroethane	WA			U	5	NS	$\mu$ g/L
1,1,2,2-Tetrachloroethane	WA	R		U	5	NS	$\mu$ g/L
1,1,2-Trichloroethane	WA			U	5	5	$\mu$ g/L
1,1,2-Trichloroethane	WA	R		U	5	5	$\mu$ g/L
1,1-Dichloroethane	WA			U	5	NS	$\mu$ g/L
1,1-Dichloroethane	WA	R		U	5	NS	$\mu$ g/L
1,1-Dichloroethylene	WA			U	5	7	$\mu$ g/L
1,1-Dichloroethylene	WA	R		U	5	7	$\mu$ g/L
1,2,3-Trichloropropane	WA			U	10	NS	$\mu$ g/L
1,2,3-Trichloropropane	WA	R		U	10	NS	$\mu$ g/L
1,2-Dibromo-3-chloropropane	WA			U	20	0.2	$\mu$ g/L
1,2-Dibromo-3-chloropropane	WA	R		U	20	0.2	$\mu$ g/L
1,2-Dibromoethane	WA			U	20	NS	$\mu$ g/L
1,2-Dibromoethane	WA	R		U	20	NS	$\mu$ g/L
1,2-Dichloroethane	WA			U	5	5	$\mu$ g/L
1,2-Dichloroethane	WA	R		U	5	5	$\mu$ g/L
1,2-Dichloroethylene	WA				22.3	70	$\mu$ g/L
1,2-Dichloroethylene	WA	R			22.1	70	$\mu$ g/L
1,2-Dichloropropane	WA			U	5	5	$\mu$ g/L
1,2-Dichloropropane	WA	R		U	5	5	$\mu$ g/L
2-Hexanone	WA			U	10	NS	$\mu$ g/L
2-Hexanone	WA	R		U	10	NS	$\mu$ g/L
4-Methyl-2-pentanone	WA			U	10	NS	$\mu$ g/L
4-Methyl-2-pentanone	WA	R		U	10	NS	$\mu$ g/L
Acetone	WA			U	10	NS	$\mu$ g/L
Acetone	WA	R		U	10	NS	$\mu$ g/L
Acetonitrile (Methyl cyanide)	WA			U	20	NS	$\mu$ g/L
Acetonitrile (Methyl cyanide)	WA	R		U	20	NS	$\mu$ g/L
Acrolein	WA			U	10	NS	$\mu$ g/L
Acrolein	WA	R		U	10	NS	$\mu$ g/L
Acrylonitrile	WA			U	10	NS	$\mu$ g/L
Acrylonitrile	WA	R		U	10	NS	$\mu$ g/L
Allyl chloride	WA			U	100	NS	$\mu$ g/L
Allyl chloride	WA	R		U	100	NS	$\mu$ g/L
Benzene	WA			U	5	5	$\mu$ g/L
Benzene	WA	R		U	5	5	$\mu$ g/L
Bromodichloromethane	WA			U	5	NS	$\mu$ g/L
Bromodichloromethane	WA	R		U	5	NS	$\mu$ g/L
Bromoform	WA			U	5	NS	$\mu$ g/L
Bromoform	WA	R		U	5	NS	$\mu$ g/L

Quarterly Sampling of the Wetlands Along the Old F-Area Effluent Ditch: May 1994

FHB013B

Field Measurements:

Sample Date: 5/24/94

pH: 4.8

Conductivity: 38  $\mu$ S/cm

Analyte	Lab	Rep	Avg	Qualifier	Result	Stand	Units
1,1,1-Trichloroethane	ADS			U	50	200	$\mu$ g/L
1,1,2,2-Tetrachloroethane	ADS			U	50	NS	$\mu$ g/L
1,1,2-Trichloroethane	ADS			U	50	5	$\mu$ g/L
1,1-Dichloroethane	ADS			U	50	NS	$\mu$ g/L
1,1-Dichloroethylene	ADS			U	50	7	$\mu$ g/L
1,2-Dichloroethane	ADS			U	50	5	$\mu$ g/L
1,2-Dichloropropane	ADS			U	50	5	$\mu$ g/L
2-Hexanone	ADS			U	50	NS	$\mu$ g/L
4-Methyl-2-pentanone	ADS			U	50	NS	$\mu$ g/L
Acetone	ADS			U	50	NS	$\mu$ g/L
Benzene	ADS			J	0	5	$\mu$ g/L
Bromodichloromethane	ADS			U	50	NS	$\mu$ g/L
Bromoform	ADS			U	50	NS	$\mu$ g/L
Carbon disulfide	ADS			J	2	NS	$\mu$ g/L
Carbon tetrachloride	ADS			U	50	5	$\mu$ g/L
Chlorobenzene	ADS			U	50	100	$\mu$ g/L
Chloroethane	ADS			U	50	NS	$\mu$ g/L
Chloroform	ADS			J	1	100	$\mu$ g/L
cis-1,2-Dichloroethylene	ADS			J	31	70	$\mu$ g/L
cis-1,3-Dichloropropene	ADS			U	50	NS	$\mu$ g/L
Dibromochloromethane	ADS			U	50	NS	$\mu$ g/L
Ethylbenzene	ADS			U	50	700	$\mu$ g/L
Methyl chloride	ADS			U	50	NS	$\mu$ g/L
Methyl ethyl ketone	ADS			U	50	NS	$\mu$ g/L
Methylene chloride	ADS			U	50	5	$\mu$ g/L
Styrene	ADS			U	50	100	$\mu$ g/L
Tetrachloroethylene	ADS			J	6	5	$\mu$ g/L
Toluene	ADS			J	0	1000	$\mu$ g/L
trans-1,2-Dichloroethylene	ADS			U	50	100	$\mu$ g/L
trans-1,3-Dichloropropene	ADS			U	50	NS	$\mu$ g/L
Trichloroethylene	ADS			U	50	5	$\mu$ g/L
Tritium	EMS				16400	20	pCi/ml
Vinyl Acetate	ADS			U	50	NS	$\mu$ g/L
Vinyl chloride	ADS			U	50	2	$\mu$ g/L
Xylenes (total)	ADS			U	50	10000	$\mu$ g/L

FHB014

## Field Measurements:

Sample Date: 05/24/94

pH: 5.3

Conductivity: 34  $\mu$ S/cm

Analyte	Lab	Rep	Avg	Qualifier	Result	Stand	Units
1,1,1,2-Tetrachloroethane	WA			U	10	NS	$\mu$ g/L
1,1,1,2-Tetrachloroethane	WA	R		U	10	NS	$\mu$ g/L
1,1,1-Trichloroethane	WA			U	5	200	$\mu$ g/L
1,1,1-Trichloroethane	WA	R		U	5	200	$\mu$ g/L
1,1,2,2-Tetrachloroethane	WA			U	5	NS	$\mu$ g/L
1,1,2,2-Tetrachloroethane	WA	R		U	5	NS	$\mu$ g/L
1,1,2-Trichloroethane	WA			U	5	5	$\mu$ g/L
1,1,2-Trichloroethane	WA	R		U	5	5	$\mu$ g/L
1,1-Dichloroethane	WA			U	5	NS	$\mu$ g/L
1,1-Dichloroethane	WA	R		U	5	NS	$\mu$ g/L
1,1-Dichloroethylene	WA			U	5	7	$\mu$ g/L
1,1-Dichloroethylene	WA	R		U	5	7	$\mu$ g/L
1,2,3-Trichloropropane	WA			U	10	NS	$\mu$ g/L
1,2,3-Trichloropropane	WA	R		U	10	NS	$\mu$ g/L
1,2-Dibromo-3-chloropropane	WA			U	20	0.2	$\mu$ g/L
1,2-Dibromo-3-chloropropane	WA	R		U	20	0.2	$\mu$ g/L
1,2-Dibromoethane	WA			U	20	NS	$\mu$ g/L
1,2-Dibromoethane	WA	R		U	20	NS	$\mu$ g/L
1,2-Dichloroethane	WA			U	5	5	$\mu$ g/L
1,2-Dichloroethane	WA	R		U	5	5	$\mu$ g/L
1,2-Dichloroethylene	WA				22.3	70	$\mu$ g/L
1,2-Dichloroethylene	WA	R			22.1	70	$\mu$ g/L
1,2-Dichloropropane	WA			U	5	5	$\mu$ g/L
1,2-Dichloropropane	WA	R		U	5	5	$\mu$ g/L
2-Hexanone	WA			U	10	NS	$\mu$ g/L
2-Hexanone	WA	R		U	10	NS	$\mu$ g/L
4-Methyl-2-pentanone	WA			U	10	NS	$\mu$ g/L
4-Methyl-2-pentanone	WA	R		U	10	NS	$\mu$ g/L
Acetone	WA			U	10	NS	$\mu$ g/L
Acetone	WA	R		U	10	NS	$\mu$ g/L
Acetonitrile (Methyl cyanide)	WA			U	20	NS	$\mu$ g/L
Acetonitrile (Methyl cyanide)	WA	R		U	20	NS	$\mu$ g/L
Acrolein	WA			U	10	NS	$\mu$ g/L
Acrolein	WA	R		U	10	NS	$\mu$ g/L
Acrylonitrile	WA			U	10	NS	$\mu$ g/L
Acrylonitrile	WA	R		U	10	NS	$\mu$ g/L
Allyl chloride	WA			U	100	NS	$\mu$ g/L
Allyl chloride	WA	R		U	100	NS	$\mu$ g/L
Benzene	WA			U	5	5	$\mu$ g/L
Benzene	WA	R		U	5	5	$\mu$ g/L
Bromodichloromethane	WA			U	5	NS	$\mu$ g/L
Bromodichloromethane	WA	R		U	5	NS	$\mu$ g/L
Bromoform	WA			U	5	NS	$\mu$ g/L
Bromoform	WA	R		U	5	NS	$\mu$ g/L

*Quarterly Sampling of the Wetlands Along the Old F-Area Effluent Ditch: May 1994*

FHB014 Laboratory Analyses (Cont'd)

Analyte	Lab	Rep	Avg	Qualifier	Result	Stand	Units
Carbon disulfide	WA			U	5	NS	µg/L
Carbon disulfide	WA	R		U	5	NS	µg/L
Carbon tetrachloride	WA			U	5	5	µg/L
Carbon tetrachloride	WA	R		U	5	5	µg/L
Chlorobenzene	WA			U	5	100	µg/L
Chlorobenzene	WA	R		U	5	100	µg/L
Chloroethane	WA			U	10	NS	µg/L
Chloroethane	WA	R		U	10	NS	µg/L
Chloroform	WA			U	5	100	µg/L
Chloroform	WA	R		U	5	100	µg/L
Chloroprene	WA			U	20	NS	µg/L
Chloroprene	WA	R		U	20	NS	µg/L
cis-1,3-Dichloropropene	WA			U	5	NS	µg/L
cis-1,3-Dichloropropene	WA	R		U	5	NS	µg/L
Dibromochloromethane	WA			U	5	NS	µg/L
Dibromochloromethane	WA	R		U	5	NS	µg/L
Dichlorodifluoromethane	WA			U	10	NS	µg/L
Dichlorodifluoromethane	WA	R		U	10	NS	µg/L
Ethylbenzene	WA			U	5	700	µg/L
Ethylbenzene	WA	R		U	5	700	µg/L
Isobutyl alcohol	WA			U	20	NS	µg/L
Isobutyl alcohol	WA	R		U	20	NS	µg/L
Methacrylonitrile	WA			U	20	NS	µg/L
Methacrylonitrile	WA	R		U	20	NS	µg/L
Methyl bromide	WA			U	10	NS	µg/L
Methyl bromide	WA	R		U	10	NS	µg/L
Methyl chloride	WA			U	10	NS	µg/L
Methyl chloride	WA	R		U	10	NS	µg/L
Methyl ethyl ketone	WA			U	10	NS	µg/L
Methyl ethyl ketone	WA	R		U	10	NS	µg/L
Methyl iodide	WA			U	10	NS	µg/L
Methyl iodide	WA	R		U	10	NS	µg/L
Methylene bromide	WA			U	10	NS	µg/L
Methylene bromide	WA	R		U	10	NS	µg/L
Methylene chloride	WA			U	5	5	µg/L
Methylene chloride	WA	R		JV	1.75	5	µg/L
Propionitrile	WA			U	50	NS	µg/L
Propionitrile	WA	R		U	50	NS	µg/L
Styrene	WA			U	5	100	µg/L
Styrene	WA	R		U	5	100	µg/L
Tetrachloroethylene	WA			U	5	5	µg/L
Tetrachloroethylene	WA	R		U	5	5	µg/L
Toluene	WA			U	5	1000	µg/L
Toluene	WA	R		U	5	1000	µg/L
trans-1,3-Dichloropropene	WA			U	5	NS	µg/L
trans-1,3-Dichloropropene	WA	R		U	5	NS	µg/L
trans-1,4-Dichloro-2-butene	WA			U	100	NS	µg/L
trans-1,4-Dichloro-2-butene	WA	R		U	100	NS	µg/L
Trichloroethylene	WA			J	2.22	5	µg/L
Trichloroethylene	WA	R		J	1.91	5	µg/L

## FHB014 Laboratory Analyses (Cont'd)

Analyte	Lab	Rep	Avg	Qualifier	Result	Stand	Units
Trichlorofluoromethane	WA			U	5	NS	µg/L
Trichlorofluoromethane	WA	R		U	5	NS	µg/L
Tritium	EMS				1060	20	pCi/ml
Vinyl acetate	WA			U	10	NS	µg/L
Vinyl acetate	WA	R		U	10	NS	µg/L
Vinyl chloride	WA			U	10	2	µg/L
Vinyl chloride	WA	R		U	10	2	µg/L
Xylenes (total)	WA			U	5	10000	µg/L
Xylenes (total)	WA	R		U	5	10000	µg/L

*Quarterly Sampling of the Wetlands Along the Old F-Area Effluent Ditch: May 1994*

FHB014B

Field Measurements:

Sample Date: 5/24/94

pH: 5.1

Conductivity: 33  $\mu$ S/cm

Analyte	Lab	Rep	Avg	Qualifier	Result	Stand	Units
1,1,1-Trichloroethane	ADS			U	50	200	$\mu$ g/L
1,1,2,2-Tetrachloroethane	ADS			U	50	NS	$\mu$ g/L
1,1,2-Trichloroethane	ADS			U	50	5	$\mu$ g/L
1,1-Dichloroethane	ADS			U	50	NS	$\mu$ g/L
1,1-Dichloroethylene	ADS			U	50	7	$\mu$ g/L
1,2-Dichloroethane	ADS			U	50	5	$\mu$ g/L
1,2-Dichloropropane	ADS			U	50	5	$\mu$ g/L
2-Hexanone	ADS			U	50	NS	$\mu$ g/L
4-Methyl-2-pentanone	ADS			U	50	NS	$\mu$ g/L
Acetone	ADS			U	50	NS	$\mu$ g/L
Benzene	ADS			J	3	5	$\mu$ g/L
Bromodichloromethane	ADS			U	50	NS	$\mu$ g/L
Bromoform	ADS			U	50	NS	$\mu$ g/L
Carbon disulfide	ADS			J	3	NS	$\mu$ g/L
Carbon tetrachloride	ADS			U	50	5	$\mu$ g/L
Chlorobenzene	ADS			U	50	100	$\mu$ g/L
Chloroethane	ADS			U	50	NS	$\mu$ g/L
Chloroform	ADS			U	50	100	$\mu$ g/L
cis-1,2-Dichloroethylene	ADS			J	34	70	$\mu$ g/L
cis-1,3-Dichloropropene	ADS			U	50	NS	$\mu$ g/L
Dibromochloromethane	ADS			U	50	NS	$\mu$ g/L
Ethylbenzene	ADS			U	50	700	$\mu$ g/L
Methyl chloride	ADS			J	29	NS	$\mu$ g/L
Methyl ethyl ketone	ADS			U	50	NS	$\mu$ g/L
Methylene chloride	ADS			U	50	5	$\mu$ g/L
Styrene	ADS			U	50	100	$\mu$ g/L
Tetrachloroethylene	ADS			J	7	5	$\mu$ g/L
Toluene	ADS			J	4	1000	$\mu$ g/L
trans-1,2-Dichloroethylene	ADS			U	50	100	$\mu$ g/L
trans-1,3-Dichloropropene	ADS			U	50	NS	$\mu$ g/L
Trichloroethylene	ADS			U	50	5	$\mu$ g/L
Tritium	EMS				1190	20	pCi/ml
Vinyl Acetate	ADS			J	2	NS	$\mu$ g/L
Vinyl chloride	ADS			U	50	2	$\mu$ g/L
Xylenes (total)	ADS			U	50	10000	$\mu$ g/L

FHB016

Field Measurements:

Sample Date: 05/24/94

pH: 5.5

Conductivity: 43  $\mu\text{S}/\text{cm}$ 

Analyte	Lab	Rep	Avg	Qualifier	Result	Stand	Units
1,1,1,2-Tetrachloroethane	WA			U	10	NS	$\mu\text{g}/\text{L}$
1,1,1-Trichloroethane	WA			U	5	200	$\mu\text{g}/\text{L}$
1,1,2,2-Tetrachloroethane	WA			U	5	NS	$\mu\text{g}/\text{L}$
1,1,2-Trichloroethane	WA			U	5	5	$\mu\text{g}/\text{L}$
1,1-Dichloroethane	WA			U	5	NS	$\mu\text{g}/\text{L}$
1,1-Dichloroethylene	WA			U	5	7	$\mu\text{g}/\text{L}$
1,2,3-Trichloropropane	WA			U	10	NS	$\mu\text{g}/\text{L}$
1,2-Dibromo-3-chloropropane	WA			U	20	0.2	$\mu\text{g}/\text{L}$
1,2-Dibromoethane	WA			U	20	NS	$\mu\text{g}/\text{L}$
1,2-Dichloroethane	WA			U	5	5	$\mu\text{g}/\text{L}$
1,2-Dichloroethylene	WA			J	3.33	70	$\mu\text{g}/\text{L}$
1,2-Dichloropropane	WA			U	5	5	$\mu\text{g}/\text{L}$
2-Hexanone	WA			U	10	NS	$\mu\text{g}/\text{L}$
4-Methyl-2-pentanone	WA			U	10	NS	$\mu\text{g}/\text{L}$
Acetone	WA			U	10	NS	$\mu\text{g}/\text{L}$
Acetonitrile (Methyl cyanide)	WA			U	20	NS	$\mu\text{g}/\text{L}$
Acrolein	WA			U	10	NS	$\mu\text{g}/\text{L}$
Acrylonitrile	WA			U	10	NS	$\mu\text{g}/\text{L}$
Allyl chloride	WA			U	100	NS	$\mu\text{g}/\text{L}$
Benzene	WA			U	5	5	$\mu\text{g}/\text{L}$
Bromodichloromethane	WA			U	5	NS	$\mu\text{g}/\text{L}$
Bromoform	WA			U	5	NS	$\mu\text{g}/\text{L}$
Carbon disulfide	WA			J	1.69	NS	$\mu\text{g}/\text{L}$
Carbon tetrachloride	WA			U	5	5	$\mu\text{g}/\text{L}$
Chlorobenzene	WA			U	5	100	$\mu\text{g}/\text{L}$
Chloroethane	WA			U	10	NS	$\mu\text{g}/\text{L}$
Chloroform	WA			U	5	100	$\mu\text{g}/\text{L}$
Chloroprene	WA			U	20	NS	$\mu\text{g}/\text{L}$
cis-1,3-Dichloropropene	WA			U	5	NS	$\mu\text{g}/\text{L}$
Dibromochloromethane	WA			U	5	NS	$\mu\text{g}/\text{L}$
Dichlorodifluoromethane	WA			U	10	NS	$\mu\text{g}/\text{L}$
Ethylbenzene	WA			J	2.93	700	$\mu\text{g}/\text{L}$
Isobutyl alcohol	WA			U	20	NS	$\mu\text{g}/\text{L}$
Methacrylonitrile	WA			U	20	NS	$\mu\text{g}/\text{L}$
Methyl bromide	WA			U	10	NS	$\mu\text{g}/\text{L}$
Methyl chloride	WA			U	10	NS	$\mu\text{g}/\text{L}$
Methyl ethyl ketone	WA			U	10	NS	$\mu\text{g}/\text{L}$
Methyl iodide	WA			U	10	NS	$\mu\text{g}/\text{L}$
Methylene bromide	WA			U	10	NS	$\mu\text{g}/\text{L}$
Methylene chloride	WA			U	5	5	$\mu\text{g}/\text{L}$
Propionitrile	WA			U	50	NS	$\mu\text{g}/\text{L}$
Styrene	WA			U	5	100	$\mu\text{g}/\text{L}$
Tetrachloroethylene	WA			U	5	5	$\mu\text{g}/\text{L}$

Quarterly Sampling of the Wetlands Along the Old F-Area Effluent Ditch: May 1994

FHB016 Laboratory Analyses (Cont'd)

Analyte	Lab	Rep	Avg	Qualifier	Result	Stand	Units
Toluene	WA			U	5	1000	µg/L
trans-1,3-Dichloropropene	WA			U	5	NS	µg/L
trans-1,4-Dichloro-2-butene	WA			U	100	NS	µg/L
Trichloroethylene	WA			U	5	5	µg/L
Trichlorofluoromethane	WA			U	5	NS	µg/L
Tritium	EMS				1440	20	pCi/ml
Vinyl acetate	WA			U	10	NS	µg/L
Vinyl chloride	WA			U	10	2	µg/L
Xylenes (total)	WA			U	5	10000	µg/L

FHB017

## Field Measurements:

Sample Date: 05/24/94

pH: 5.4

Conductivity: 44  $\mu$ S/cm

Analyte	Lab	Rep	Avg	Qualifier	Result	Stand	Units
1,1,1,2-Tetrachloroethane	WA			U	10	NS	$\mu$ g/L
1,1,1-Trichloroethane	WA			U	5	200	$\mu$ g/L
1,1,2,2-Tetrachloroethane	WA			U	5	NS	$\mu$ g/L
1,1,2-Trichloroethane	WA			U	5	5	$\mu$ g/L
1,1-Dichloroethane	WA			U	5	NS	$\mu$ g/L
1,1-Dichloroethylene	WA			U	5	7	$\mu$ g/L
1,2,3-Trichloropropane	WA			U	10	NS	$\mu$ g/L
1,2-Dibromo-3-chloropropane	WA			U	20	0.2	$\mu$ g/L
1,2-Dibromoethane	WA			U	20	NS	$\mu$ g/L
1,2-Dichloroethane	WA			U	5	5	$\mu$ g/L
1,2-Dichloroethylene	WA				36.8	70	$\mu$ g/L
1,2-Dichloropropane	WA			U	5	5	$\mu$ g/L
2-Hexanone	WA			U	10	NS	$\mu$ g/L
4-Methyl-2-pentanone	WA			U	10	NS	$\mu$ g/L
Acetone	WA			U	10	NS	$\mu$ g/L
Acetonitrile (Methyl cyanide)	WA			U	20	NS	$\mu$ g/L
Acrolein	WA			U	10	NS	$\mu$ g/L
Acrylonitrile	WA			U	10	NS	$\mu$ g/L
Allyl chloride	WA			U	100	NS	$\mu$ g/L
Benzene	WA			U	5	5	$\mu$ g/L
Bromodichloromethane	WA			U	5	NS	$\mu$ g/L
Bromoform	WA			U	5	NS	$\mu$ g/L
Carbon disulfide	WA			U	5	NS	$\mu$ g/L
Carbon tetrachloride	WA			U	5	5	$\mu$ g/L
Chlorobenzene	WA			U	5	100	$\mu$ g/L
Chloroethane	WA			U	10	NS	$\mu$ g/L
Chloroform	WA			J	1.68	100	$\mu$ g/L
Chloroprene	WA			U	20	NS	$\mu$ g/L
cis-1,3-Dichloropropene	WA			U	5	NS	$\mu$ g/L
Dibromochloromethane	WA			U	5	NS	$\mu$ g/L
Dichlorodifluoromethane	WA			U	10	NS	$\mu$ g/L
Ethylbenzene	WA			U	5	700	$\mu$ g/L
Isobutyl alcohol	WA			U	20	NS	$\mu$ g/L
Methacrylonitrile	WA			U	20	NS	$\mu$ g/L
Methyl bromide	WA			U	10	NS	$\mu$ g/L
Methyl chloride	WA			U	10	NS	$\mu$ g/L
Methyl ethyl ketone	WA			U	10	NS	$\mu$ g/L
Methyl iodide	WA			U	10	NS	$\mu$ g/L
Methylene bromide	WA			U	10	NS	$\mu$ g/L
Methylene chloride	WA			JV	1.1	5	$\mu$ g/L
Propionitrile	WA			U	50	NS	$\mu$ g/L
Styrene	WA			U	5	100	$\mu$ g/L
Tetrachloroethylene	WA			J	2.49	5	$\mu$ g/L

*Quarterly Sampling of the Wetlands Along the Old F-Area Effluent Ditch: May 1994*

FHB017 Laboratory Analyses (Cont'd)

Analyte	Lab	Rep	Avg	Qualifier	Result	Stand	Units
Toluene	WA			U	5	1000	µg/L
trans-1,3-Dichloropropene	WA			U	5	NS	µg/L
trans-1,4-Dichloro-2-butene	WA			U	100	NS	µg/L
Trichloroethylene	WA				9.9	5	µg/L
Trichlorofluoromethane	WA			U	5	NS	µg/L
Tritium	EMS				35000	20	pCi/ml
Vinyl acetate	WA			U	10	NS	µg/L
Vinyl chloride	WA			U	10	2	µg/L
Xylenes (total)	WA			U	5	10000	µg/L

FHB018

## Field Measurements:

Sample Date: 05/24/94

pH: 5.7

Conductivity: 50  $\mu\text{S}/\text{cm}$ 

Analyte	Lab	Rep	Avg	Qualifier	Result	Stand	Units
1,1,1,2-Tetrachloroethane	WA			U	10	NS	$\mu\text{g}/\text{L}$
1,1,1-Trichloroethane	ADS			U	50	200	$\mu\text{g}/\text{L}$
1,1,1-Trichloroethane	WA			U	5	200	$\mu\text{g}/\text{L}$
1,1,2,2-Tetrachloroethane	ADS			U	50	NS	$\mu\text{g}/\text{L}$
1,1,2,2-Tetrachloroethane	WA			U	5	NS	$\mu\text{g}/\text{L}$
1,1,2-Trichloroethane	ADS			U	50	5	$\mu\text{g}/\text{L}$
1,1,2-Trichloroethane	WA			U	5	5	$\mu\text{g}/\text{L}$
1,1-Dichloroethane	ADS			U	50	NS	$\mu\text{g}/\text{L}$
1,1-Dichloroethane	WA			U	5	NS	$\mu\text{g}/\text{L}$
1,1-Dichloroethylene	ADS			J	2	7	$\mu\text{g}/\text{L}$
1,1-Dichloroethylene	WA			U	5	7	$\mu\text{g}/\text{L}$
1,2,3-Trichloropropane	WA			U	10	NS	$\mu\text{g}/\text{L}$
1,2-Dibromo-3-chloropropane	WA			U	20	0.2	$\mu\text{g}/\text{L}$
1,2-Dibromoethane	WA			U	20	NS	$\mu\text{g}/\text{L}$
1,2-Dichloroethane	ADS			U	50	5	$\mu\text{g}/\text{L}$
1,2-Dichloroethane	WA			U	5	5	$\mu\text{g}/\text{L}$
1,2-Dichloroethylene	WA		A		280	70	$\mu\text{g}/\text{L}$
1,2-Dichloropropane	ADS			U	50	5	$\mu\text{g}/\text{L}$
1,2-Dichloropropane	WA			U	5	5	$\mu\text{g}/\text{L}$
2-Hexanone	ADS			U	50	NS	$\mu\text{g}/\text{L}$
2-Hexanone	WA			U	10	NS	$\mu\text{g}/\text{L}$
4-Methyl-2-pentanone	ADS			U	50	NS	$\mu\text{g}/\text{L}$
4-Methyl-2-pentanone	WA			U	10	NS	$\mu\text{g}/\text{L}$
Acetone	ADS			U	50	NS	$\mu\text{g}/\text{L}$
Acetone	WA			U	10	NS	$\mu\text{g}/\text{L}$
Acetonitrile (Methyl cyanide)	WA			U	20	NS	$\mu\text{g}/\text{L}$
Acrolein	WA			U	10	NS	$\mu\text{g}/\text{L}$
Acrylonitrile	WA			U	10	NS	$\mu\text{g}/\text{L}$
Allyl chloride	WA			U	100	NS	$\mu\text{g}/\text{L}$
Benzene	ADS			J	4	5	$\mu\text{g}/\text{L}$
Benzene	WA			U	5	5	$\mu\text{g}/\text{L}$
Bromodichloromethane	ADS			U	50	NS	$\mu\text{g}/\text{L}$
Bromodichloromethane	WA			U	5	NS	$\mu\text{g}/\text{L}$
Bromoform	ADS			U	50	NS	$\mu\text{g}/\text{L}$
Bromoform	WA			U	5	NS	$\mu\text{g}/\text{L}$
Carbon disulfide	ADS			J	2	NS	$\mu\text{g}/\text{L}$
Carbon disulfide	WA			U	5	NS	$\mu\text{g}/\text{L}$
Carbon tetrachloride	ADS			U	50	5	$\mu\text{g}/\text{L}$
Carbon tetrachloride	WA			U	5	5	$\mu\text{g}/\text{L}$
Chlorobenzene	ADS			U	50	100	$\mu\text{g}/\text{L}$
Chlorobenzene	WA			U	5	100	$\mu\text{g}/\text{L}$
Chloroethane	ADS			U	50	NS	$\mu\text{g}/\text{L}$
Chloroethane	WA			U	10	NS	$\mu\text{g}/\text{L}$

*Quarterly Sampling of the Wetlands Along the Old F-Area Effluent Ditch: May 1994*

FHB018 Laboratory Analyses (Cont'd)

Analyte	Lab	Rep	Avg	Qualifier	Result	Stand	Units
Chloroform	ADS			J	2	100	µg/L
Chloroform	WA			U	5	100	µg/L
Chloroprene	WA			U	20	NS	µg/L
cis-1,2-Dichloroethylene	ADS				492	70	µg/L
cis-1,3-Dichloropropene	ADS			U	50	NS	µg/L
cis-1,3-Dichloropropene	WA			U	5	NS	µg/L
Dibromochloromethane	ADS			U	50	NS	µg/L
Dibromochloromethane	WA			U	5	NS	µg/L
Dichlorodifluoromethane	WA			J	3.08	NS	µg/L
Ethylbenzene	ADS			U	50	700	µg/L
Ethylbenzene	WA			U	5	700	µg/L
Isobutyl alcohol	WA			U	20	NS	µg/L
Methacrylonitrile	WA			U	20	NS	µg/L
Methyl bromide	WA			U	10	NS	µg/L
Methyl chloride	ADS			U	50	NS	µg/L
Methyl chloride	WA			U	10	NS	µg/L
Methyl ethyl ketone	ADS			U	50	NS	µg/L
Methyl ethyl ketone	WA			U	10	NS	µg/L
Methyl iodide	WA			U	10	NS	µg/L
Methylene bromide	WA			U	10	NS	µg/L
Methylene chloride	ADS			U	50	5	µg/L
Methylene chloride	WA			U	5	5	µg/L
Propionitrile	WA			U	50	NS	µg/L
Styrene	ADS			U	50	100	µg/L
Styrene	WA			U	5	100	µg/L
Tetrachloroethylene	ADS			J	4	5	µg/L
Tetrachloroethylene	WA			U	5	5	µg/L
Toluene	ADS			U	50	1000	µg/L
Toluene	WA			U	5	1000	µg/L
trans-1,2-Dichloroethylene	ADS			J	6	100	µg/L
trans-1,3-Dichloropropene	ADS			U	50	NS	µg/L
trans-1,3-Dichloropropene	WA			U	5	NS	µg/L
trans-1,4-Dichloro-2-butene	WA			U	100	NS	µg/L
Trichloroethylene	ADS			U	50	5	µg/L
Trichloroethylene	WA			J	2.95	5	µg/L
Trichlorofluoromethane	WA			U	5	NS	µg/L
Tritium	EMS				51100	20	pCi/ml
Vinyl Acetate	ADS			U	50	NS	µg/L
Vinyl acetate	WA			U	10	NS	µg/L
Vinyl chloride	ADS			U	50	2	µg/L
Vinyl chloride	WA			U	10	2	µg/L
Xylenes (total)	ADS			U	50	10000	µg/L
Xylenes (total)	WA			U	5	10000	µg/L

FHB018B

Field Measurements:

Sample Date: 5/24/94

pH: 5.7

Conductivity: 98  $\mu$ S/cm

Analyte	Lab	Rep	Avg	Qualifier	Result	Stand	Units
1,1,1-Trichloroethane	ADS			U	50	200	$\mu$ g/L
1,1,2,2-Tetrachloroethane	ADS			U	50	NS	$\mu$ g/L
1,1,2-Trichloroethane	ADS			U	50	5	$\mu$ g/L
1,1-Dichloroethane	ADS			U	50	NS	$\mu$ g/L
1,1-Dichloroethylene	ADS			U	50	7	$\mu$ g/L
1,2-Dichloroethane	ADS			U	50	5	$\mu$ g/L
1,2-Dichloropropane	ADS			U	50	5	$\mu$ g/L
2-Hexanone	ADS			U	50	NS	$\mu$ g/L
4-Methyl-2-pentanone	ADS			U	50	NS	$\mu$ g/L
Acetone	ADS			U	50	NS	$\mu$ g/L
Benzene	ADS			J	4	5	$\mu$ g/L
Bromodichloromethane	ADS			U	50	NS	$\mu$ g/L
Bromoform	ADS			U	50	NS	$\mu$ g/L
Carbon disulfide	ADS			J	4	NS	$\mu$ g/L
Carbon tetrachloride	ADS			U	50	5	$\mu$ g/L
Chlorobenzene	ADS			U	50	100	$\mu$ g/L
Chloroethane	ADS			U	50	NS	$\mu$ g/L
Chloroform	ADS			J	2	100	$\mu$ g/L
cis-1,2-Dichloroethylene	ADS				96	70	$\mu$ g/L
cis-1,3-Dichloropropene	ADS			U	50	NS	$\mu$ g/L
Dibromochloromethane	ADS			U	50	NS	$\mu$ g/L
Ethylbenzene	ADS			U	50	700	$\mu$ g/L
Methyl chloride	ADS			U	50	NS	$\mu$ g/L
Methyl ethyl ketone	ADS			J	13	NS	$\mu$ g/L
Methylene chloride	ADS			J	2	5	$\mu$ g/L
Styrene	ADS			U	50	100	$\mu$ g/L
Tetrachloroethylene	ADS			J	8	5	$\mu$ g/L
Toluene	ADS			J	5	1000	$\mu$ g/L
trans-1,2-Dichloroethylene	ADS			U	50	100	$\mu$ g/L
trans-1,3-Dichloropropene	ADS			U	50	NS	$\mu$ g/L
Trichloroethylene	ADS			U	50	5	$\mu$ g/L
Tritium	EMS				35400	20	pCi/ml
Vinyl Acetate	ADS			J	0	NS	$\mu$ g/L
Vinyl chloride	ADS			U	50	2	$\mu$ g/L
Xylenes (total)	ADS			U	50	10000	$\mu$ g/L

Quarterly Sampling of the Wetlands Along the Old F-Area Effluent Ditch: May 1994

FHB019

Field Measurements:

Sample Date: 05/24/94

pH: 4.6

Conductivity: 37  $\mu$ S/cm

Analyte	Lab	Rep	Avg	Qualifier	Result	Stand	Units
1,1,1,2-Tetrachloroethane	WA			U	10	NS	$\mu$ g/L
1,1,1,2-Tetrachloroethane	WA		NA	U	10	NS	$\mu$ g/L
1,1,1,2-Tetrachloroethane	WA		NA	U	10	NS	$\mu$ g/L
1,1,1-Trichloroethane	WA			U	5	200	$\mu$ g/L
1,1,1-Trichloroethane	WA		NA	U	5	200	$\mu$ g/L
1,1,1-Trichloroethane	WA		NA	U	5	200	$\mu$ g/L
1,1,2,2-Tetrachloroethane	WA			U	5	NS	$\mu$ g/L
1,1,2,2-Tetrachloroethane	WA		NA	U	5	NS	$\mu$ g/L
1,1,2,2-Tetrachloroethane	WA		NA	U	5	NS	$\mu$ g/L
1,1,2-Trichloroethane	WA			U	5	5	$\mu$ g/L
1,1,2-Trichloroethane	WA		NA	U	5	5	$\mu$ g/L
1,1,2-Trichloroethane	WA		NA	U	5	5	$\mu$ g/L
1,1-Dichloroethane	WA			U	5	NS	$\mu$ g/L
1,1-Dichloroethane	WA		NA	U	5	NS	$\mu$ g/L
1,1-Dichloroethane	WA		NA	U	5	NS	$\mu$ g/L
1,1-Dichloroethylene	WA			U	5	7	$\mu$ g/L
1,2,3-Trichloropropane	WA			U	10	NS	$\mu$ g/L
1,2,3-Trichloropropane	WA		NA	U	10	NS	$\mu$ g/L
1,2,3-Trichloropropane	WA		NA	U	10	NS	$\mu$ g/L
1,2-Dibromo-3-chloropropane	WA			U	20	0.2	$\mu$ g/L
1,2-Dibromo-3-chloropropane	WA		NA	U	20	0.2	$\mu$ g/L
1,2-Dibromo-3-chloropropane	WA		NA	U	20	0.2	$\mu$ g/L
1,2-Dibromoethane	WA			U	20	NS	$\mu$ g/L
1,2-Dibromoethane	WA		NA	U	20	NS	$\mu$ g/L
1,2-Dibromoethane	WA		NA	U	20	NS	$\mu$ g/L
1,2-Dichloroethane	WA			U	5	5	$\mu$ g/L
1,2-Dichloroethane	WA		NA	U	5	5	$\mu$ g/L
1,2-Dichloroethane	WA		NA	U	5	5	$\mu$ g/L
1,2-Dichloroethylene	WA		A		47.5	70	$\mu$ g/L
1,2-Dichloropropane	WA			U	5	5	$\mu$ g/L
1,2-Dichloropropane	WA		NA	U	5	5	$\mu$ g/L
1,2-Dichloropropane	WA		NA	U	5	5	$\mu$ g/L
2-Hexanone	WA			U	10	NS	$\mu$ g/L
2-Hexanone	WA		NA	U	10	NS	$\mu$ g/L
2-Hexanone	WA		NA	U	10	NS	$\mu$ g/L
4-Methyl-2-pentanone	WA			U	10	NS	$\mu$ g/L
4-Methyl-2-pentanone	WA		NA	U	10	NS	$\mu$ g/L
4-Methyl-2-pentanone	WA		NA	U	10	NS	$\mu$ g/L
Acetone	WA			U	10	NS	$\mu$ g/L
Acetone	WA		NA	U	10	NS	$\mu$ g/L
Acetone	WA		NA	U	10	NS	$\mu$ g/L
Acetonitrile (Methyl cyanide)	WA			U	20	NS	$\mu$ g/L
Acetonitrile (Methyl cyanide)	WA		NA	U	20	NS	$\mu$ g/L
Acetonitrile (Methyl cyanide)	WA		NA	U	20	NS	$\mu$ g/L

## FHB019 Laboratory Analyses (Cont'd)

Analyte	Lab	Rep	Avg	Qualifier	Result	Stand	Units
Acrolein	WA			U	10	NS	µg/L
Acrolein	WA		NA	U	10	NS	µg/L
Acrolein	WA		NA	U	10	NS	µg/L
Acrylonitrile	WA			U	10	NS	µg/L
Acrylonitrile	WA		NA	U	10	NS	µg/L
Acrylonitrile	WA		NA	U	10	NS	µg/L
Allyl chloride	WA			U	100	NS	µg/L
Allyl chloride	WA		NA	U	100	NS	µg/L
Allyl chloride	WA		NA	U	100	NS	µg/L
Benzene	WA			U	5	5	µg/L
Bromodichloromethane	WA			U	5	NS	µg/L
Bromodichloromethane	WA		NA	U	5	NS	µg/L
Bromodichloromethane	WA		NA	U	5	NS	µg/L
Bromoform	WA			U	5	NS	µg/L
Bromoform	WA		NA	U	5	NS	µg/L
Bromoform	WA		NA	U	5	NS	µg/L
Carbon disulfide	WA			U	5	NS	µg/L
Carbon disulfide	WA		NA	U	5	NS	µg/L
Carbon disulfide	WA		NA	U	5	NS	µg/L
Carbon tetrachloride	WA			U	5	5	µg/L
Carbon tetrachloride	WA		NA	U	5	5	µg/L
Carbon tetrachloride	WA		NA	U	5	5	µg/L
Chlorobenzene	WA			U	5	100	µg/L
Chloroethane	WA			U	10	NS	µg/L
Chloroethane	WA		NA	U	10	NS	µg/L
Chloroethane	WA		NA	U	10	NS	µg/L
Chloroform	WA			J	1.97	100	µg/L
Chloroform	WA		NA	J	2.14	100	µg/L
Chloroform	WA		NA	J	2.09	100	µg/L
Chloroprene	WA			U	20	NS	µg/L
Chloroprene	WA		NA	U	20	NS	µg/L
Chloroprene	WA		NA	U	20	NS	µg/L
cis-1,3-Dichloropropene	WA			U	5	NS	µg/L
cis-1,3-Dichloropropene	WA		NA	U	5	NS	µg/L
cis-1,3-Dichloropropene	WA		NA	U	5	NS	µg/L
Dibromochloromethane	WA			U	5	NS	µg/L
Dibromochloromethane	WA		NA	U	5	NS	µg/L
Dibromochloromethane	WA		NA	U	5	NS	µg/L
Dichlorodifluoromethane	WA			J	1.02	NS	µg/L
Dichlorodifluoromethane	WA		NA	J	1.67	NS	µg/L
Dichlorodifluoromethane	WA		NA	J	1.51	NS	µg/L
Ethylbenzene	WA			U	5	700	µg/L
Ethylbenzene	WA		NA	U	5	700	µg/L
Ethylbenzene	WA		NA	U	5	700	µg/L
Isobutyl alcohol	WA			U	20	NS	µg/L
Isobutyl alcohol	WA		NA	U	20	NS	µg/L
Isobutyl alcohol	WA		NA	U	20	NS	µg/L
Methacrylonitrile	WA			U	20	NS	µg/L
Methacrylonitrile	WA		NA	U	20	NS	µg/L
Methacrylonitrile	WA		NA	U	20	NS	µg/L

Quarterly Sampling of the Wetlands Along the Old F-Area Effluent Ditch: May 1994

FHB019 Laboratory Analyses (Cont'd)

Analyte	Lab	Rep	Avg	Qualifier	Result	Stand	Units
Methyl bromide	WA			U	10	NS	µg/L
Methyl bromide	WA		NA	U	10	NS	µg/L
Methyl bromide	WA		NA	U	10	NS	µg/L
Methyl chloride	WA			U	10	NS	µg/L
Methyl chloride	WA		NA	U	10	NS	µg/L
Methyl chloride	WA		NA	U	10	NS	µg/L
Methyl ethyl ketone	WA			U	10	NS	µg/L
Methyl ethyl ketone	WA		NA	U	10	NS	µg/L
Methyl ethyl ketone	WA		NA	U	10	NS	µg/L
Methyl iodide	WA			U	10	NS	µg/L
Methyl iodide	WA		NA	U	10	NS	µg/L
Methyl iodide	WA		NA	U	10	NS	µg/L
Methylene bromide	WA			U	10	NS	µg/L
Methylene bromide	WA		NA	U	10	NS	µg/L
Methylene bromide	WA		NA	U	10	NS	µg/L
Methylene chloride	WA			U	5	5	µg/L
Methylene chloride	WA		NA	U	5	5	µg/L
Methylene chloride	WA		NA	U	5	5	µg/L
Propionitrile	WA			U	50	NS	µg/L
Propionitrile	WA		NA	U	50	NS	µg/L
Propionitrile	WA		NA	U	50	NS	µg/L
Styrene	WA			U	5	100	µg/L
Styrene	WA		NA	U	5	100	µg/L
Styrene	WA		NA	U	5	100	µg/L
Tetrachloroethylene	WA		A		8.60	5	µg/L
Toluene	WA			U	5	1000	µg/L
trans-1,3-Dichloropropene	WA			U	5	NS	µg/L
trans-1,3-Dichloropropene	WA		NA	U	5	NS	µg/L
trans-1,3-Dichloropropene	WA		NA	U	5	NS	µg/L
trans-1,4-Dichloro-2-butene	WA			U	100	NS	µg/L
trans-1,4-Dichloro-2-butene	WA		NA	U	100	NS	µg/L
trans-1,4-Dichloro-2-butene	WA		NA	U	100	NS	µg/L
Trichloroethylene	WA				23.8	5	µg/L
Trichlorofluoromethane	WA			U	5	NS	µg/L
Trichlorofluoromethane	WA		NA	U	5	NS	µg/L
Trichlorofluoromethane	WA		NA	U	5	NS	µg/L
Tritium	EMS				23500	20	pCi/ml
Vinyl acetate	WA			U	10	NS	µg/L
Vinyl acetate	WA		NA	U	10	NS	µg/L
Vinyl acetate	WA		NA	U	10	NS	µg/L
Vinyl chloride	WA			U	10	2	µg/L
Vinyl chloride	WA		NA	U	10	2	µg/L
Vinyl chloride	WA		NA	U	10	2	µg/L
Xylenes (total)	WA			U	5	10000	µg/L
Xylenes (total)	WA		NA	U	5	10000	µg/L
Xylenes (total)	WA		NA	U	5	10000	µg/L

FHB020

## Field Measurements:

Sample Date: 05/24/94

pH: 3.6

Conductivity: 31  $\mu$ S/cm

Analyte	Lab	Rep	Avg	Qualifier	Result	Stand	Units
1,1,1,2-Tetrachloroethane	WA			U	10	NS	$\mu$ g/L
1,1,1-Trichloroethane	WA			U	5	200	$\mu$ g/L
1,1,1,2-Tetrachloroethane	WA			U	5	NS	$\mu$ g/L
1,1,2-Trichloroethane	WA			U	5	5	$\mu$ g/L
1,1-Dichloroethane	WA			U	5	NS	$\mu$ g/L
1,1-Dichloroethylene	WA			U	5	7	$\mu$ g/L
1,2,3-Trichloropropane	WA			U	10	NS	$\mu$ g/L
1,2-Dibromo-3-chloropropane	WA			U	20	0.2	$\mu$ g/L
1,2-Dibromoethane	WA			U	20	NS	$\mu$ g/L
1,2-Dichloroethane	WA			U	5	5	$\mu$ g/L
1,2-Dichloroethylene	WA			U	5	70	$\mu$ g/L
1,2-Dichloropropane	WA			U	5	5	$\mu$ g/L
2-Hexanone	WA			U	10	NS	$\mu$ g/L
4-Methyl-2-pentanone	WA			U	10	NS	$\mu$ g/L
Acetone	WA			U	10	NS	$\mu$ g/L
Acetonitrile (Methyl cyanide)	WA			U	20	NS	$\mu$ g/L
Acrolein	WA			U	10	NS	$\mu$ g/L
Acrylonitrile	WA			U	10	NS	$\mu$ g/L
Allyl chloride	WA			U	100	NS	$\mu$ g/L
Benzene	WA			U	5	5	$\mu$ g/L
Bromodichloromethane	WA			U	5	NS	$\mu$ g/L
Bromoform	WA			U	5	NS	$\mu$ g/L
Carbon disulfide	WA			U	5	NS	$\mu$ g/L
Carbon tetrachloride	WA			U	5	5	$\mu$ g/L
Chlorobenzene	WA			U	5	100	$\mu$ g/L
Chloroethane	WA			U	10	NS	$\mu$ g/L
Chloroform	WA			U	5	100	$\mu$ g/L
Chloroprene	WA			U	20	NS	$\mu$ g/L
cis-1,3-Dichloropropene	WA			U	5	NS	$\mu$ g/L
Dibromochloromethane	WA			U	5	NS	$\mu$ g/L
Dichlorodifluoromethane	WA			U	10	NS	$\mu$ g/L
Ethylbenzene	WA			U	5	700	$\mu$ g/L
Isobutyl alcohol	WA			U	20	NS	$\mu$ g/L
Methacrylonitrile	WA			U	20	NS	$\mu$ g/L
Methyl bromide	WA			U	10	NS	$\mu$ g/L
Methyl chloride	WA			U	10	NS	$\mu$ g/L
Methyl ethyl ketone	WA			U	10	NS	$\mu$ g/L
Methyl iodide	WA			U	10	NS	$\mu$ g/L
Methylene bromide	WA			U	10	NS	$\mu$ g/L
Methylene chloride	WA			U	5	5	$\mu$ g/L
Propionitrile	WA			U	50	NS	$\mu$ g/L
Styrene	WA			U	5	100	$\mu$ g/L
Tetrachloroethylene	WA			U	5	5	$\mu$ g/L

Quarterly Sampling of the Wetlands Along the Old F-Area Effluent Ditch: May 1994

FHB020 Laboratory Analyses (Cont'd)

Analyte	Lab	Rep	Avg	Qualifier	Result	Stand	Units
Toluene	WA			U	5	1000	µg/L
trans-1,3-Dichloropropene	WA			U	5	NS	µg/L
trans-1,4-Dichloro-2-butene	WA			U	100	NS	µg/L
Trichloroethylene	WA			U	5	5	µg/L
Trichlorofluoromethane	WA			U	5	NS	µg/L
Tritium	EMS			U	2	20	pCi/ml
Vinyl acetate	WA			U	10	NS	µg/L
Vinyl chloride	WA			U	10	2	µg/L
Xylenes (total)	WA			U	5	10000	µg/L

FHB020B

## Field Measurements:

Sample Date: 5/24/94

pH: 5.0

Conductivity: 128  $\mu$ S/cm

Analyte	Lab	Rep	Avg	Qualifier	Result	Stand	Units
1,1,1-Trichloroethane	ADS			U	50	200	$\mu$ g/L
1,1,2,2-Tetrachloroethane	ADS			U	50	NS	$\mu$ g/L
1,1,2-Trichloroethane	ADS			U	50	5	$\mu$ g/L
1,1-Dichloroethane	ADS			U	50	NS	$\mu$ g/L
1,1-Dichloroethylene	ADS			U	50	7	$\mu$ g/L
1,2-Dichloroethane	ADS			U	50	5	$\mu$ g/L
1,2-Dichloropropane	ADS			U	50	5	$\mu$ g/L
2-Hexanone	ADS			U	50	NS	$\mu$ g/L
4-Methyl-2-pentanone	ADS			U	50	NS	$\mu$ g/L
Acetone	ADS			U	50	NS	$\mu$ g/L
Benzene	ADS			J	3	5	$\mu$ g/L
Bromodichloromethane	ADS			U	50	NS	$\mu$ g/L
Bromoform	ADS			U	50	NS	$\mu$ g/L
Carbon disulfide	ADS			J	0	NS	$\mu$ g/L
Carbon tetrachloride	ADS			U	50	5	$\mu$ g/L
Chlorobenzene	ADS			U	50	100	$\mu$ g/L
Chloroethane	ADS			U	50	NS	$\mu$ g/L
Chloroform	ADS			U	50	100	$\mu$ g/L
cis-1,2-Dichloroethylene	ADS			J	4	70	$\mu$ g/L
cis-1,3-Dichloropropene	ADS			U	50	NS	$\mu$ g/L
Dibromochloromethane	ADS			U	50	NS	$\mu$ g/L
Ethylbenzene	ADS			U	50	700	$\mu$ g/L
Methyl chloride	ADS			U	50	NS	$\mu$ g/L
Methyl ethyl ketone	ADS			U	50	NS	$\mu$ g/L
Methylene chloride	ADS			U	50	5	$\mu$ g/L
Styrene	ADS			U	50	100	$\mu$ g/L
Tetrachloroethylene	ADS			J	6	5	$\mu$ g/L
Toluene	ADS			U	50	1000	$\mu$ g/L
trans-1,2-Dichloroethylene	ADS			U	50	100	$\mu$ g/L
trans-1,3-Dichloropropene	ADS			U	50	NS	$\mu$ g/L
Trichloroethylene	ADS			U	50	5	$\mu$ g/L
Tritium	EMS				27	20	pCi/ml
Vinyl Acetate	ADS			U	50	NS	$\mu$ g/L
Vinyl chloride	ADS			U	50	2	$\mu$ g/L
Xylenes (total)	ADS			U	50	10000	$\mu$ g/L

*Quarterly Sampling of the Wetlands Along the Old F-Area Effluent Ditch: May 1994*

FVA001  
(Volatile Trip Blank)

Field Measurements:  
Sample Date: 05/26/94

Analyte	Lab	Rep	Avg	Qualifier	Result	Stand	Units
1,1,1,2-Tetrachloroethane	WA			U	10	NS	µg/L
1,1,1-Trichloroethane	WA			U	5	200	µg/L
1,1,2,2-Tetrachloroethane	WA			U	5	NS	µg/L
1,1,2-Trichloroethane	WA			U	5	5	µg/L
1,1-Dichloroethane	WA			U	5	NS	µg/L
1,1-Dichloroethylene	WA			U	5	7	µg/L
1,2,3-Trichloropropane	WA			U	10	NS	µg/L
1,2-Dibromo-3-chloropropane	WA			U	20	0.2	µg/L
1,2-Dibromoethane	WA			U	20	NS	µg/L
1,2-Dichloroethane	WA			U	5	5	µg/L
1,2-Dichloroethylene	WA			U	5	70	µg/L
1,2-Dichloropropane	WA			U	5	5	µg/L
2-Hexanone	WA			U	10	NS	µg/L
4-Methyl-2-pentanone	WA			U	10	NS	µg/L
Acetone	WA			U	10	NS	µg/L
Acetonitrile (Methyl cyanide)	WA			U	20	NS	µg/L
Acrolein	WA			U	10	NS	µg/L
Acrylonitrile	WA			U	10	NS	µg/L
Allyl chloride	WA			U	100	NS	µg/L
Benzene	WA			U	5	5	µg/L
Bromodichloromethane	WA			U	5	NS	µg/L
Bromoform	WA			U	5	NS	µg/L
Carbon disulfide	WA			U	5	NS	µg/L
Carbon tetrachloride	WA			U	5	5	µg/L
Chlorobenzene	WA			U	5	100	µg/L
Chloroethane	WA			U	10	NS	µg/L
Chloroform	WA			U	5	100	µg/L
Chloroprene	WA			U	20	NS	µg/L
cis-1,3-Dichloropropene	WA			U	5	NS	µg/L
Dibromochloromethane	WA			U	5	NS	µg/L
Dichlorodifluoromethane	WA			U	10	NS	µg/L
Ethylbenzene	WA			U	5	700	µg/L
Isobutyl alcohol	WA			U	20	NS	µg/L
Methacrylonitrile	WA			U	20	NS	µg/L
Methyl bromide	WA			U	10	NS	µg/L
Methyl chloride	WA			U	10	NS	µg/L
Methyl ethyl ketone	WA			U	10	NS	µg/L
Methyl iodide	WA			U	10	NS	µg/L
Methylene bromide	WA			U	10	NS	µg/L
Methylene chloride	WA			JV	1.78	5	µg/L
Propionitrile	WA			U	50	NS	µg/L
Styrene	WA			U	5	100	µg/L
Tetrachloroethylene	WA			U	5	5	µg/L

## FVA001 Laboratory Analyses (Cont'd)

Analyte	Lab	Rep	Avg	Qualifier	Result	Stand	Units
Toluene	WA			U	5	1000	µg/L
trans-1,3-Dichloropropene	WA			U	5	NS	µg/L
trans-1,4-Dichloro-2-butene	WA			U	100	NS	µg/L
Trichloroethylene	WA			U	5	5	µg/L
Trichlorofluoromethane	WA			U	5	NS	µg/L
Vinyl acetate	WA			U	10	NS	µg/L
Vinyl chloride	WA			U	10	2	µg/L
Xylenes (total)	WA			U	5	10000	µg/L

*Quarterly Sampling of the Wetlands Along the Old F-Area Effluent Ditch: May 1994*

QA2  
(Trip Blank)

Field Measurements:  
Sample Date: 05/24/94

Analyte	Lab	Rep	Avg	Qualifier	Result	Stand	Units
1,1,1,2-Tetrachloroethane	WA			U	10	NS	µg/L
1,1,1-Trichloroethane	WA			U	5	200	µg/L
1,1,2,2-Tetrachloroethane	WA			U	5	NS	µg/L
1,1,2-Trichloroethane	WA			U	5	5	µg/L
1,1-Dichloroethane	WA			U	5	NS	µg/L
1,1-Dichloroethylene	WA			U	5	7	µg/L
1,2,3-Trichloropropane	WA			U	10	NS	µg/L
1,2-Dibromo-3-chloropropane	WA			U	20	0.2	µg/L
1,2-Dibromoethane	WA			U	20	NS	µg/L
1,2-Dichloroethane	WA			U	5	5	µg/L
1,2-Dichloroethylene	WA			U	5	70	µg/L
1,2-Dichloropropane	WA			U	5	5	µg/L
2-Hexanone	WA			U	10	NS	µg/L
4-Methyl-2-pentanone	WA			U	10	NS	µg/L
Acetone	WA			U	10	NS	µg/L
Acetonitrile (Methyl cyanide)	WA			U	20	NS	µg/L
Acrolein	WA			U	10	NS	µg/L
Acrylonitrile	WA			U	10	NS	µg/L
Allyl chloride	WA			U	100	NS	µg/L
Benzene	WA			U	5	5	µg/L
Bromodichloromethane	WA			U	5	NS	µg/L
Bromoform	WA			U	5	NS	µg/L
Carbon disulfide	WA			U	5	NS	µg/L
Carbon tetrachloride	WA			U	5	5	µg/L
Chlorobenzene	WA			U	5	100	µg/L
Chloroethane	WA			U	10	NS	µg/L
Chloroform	WA			U	5	100	µg/L
Chloroprene	WA			U	20	NS	µg/L
cis-1,3-Dichloropropene	WA			U	5	NS	µg/L
Dibromochloromethane	WA			U	5	NS	µg/L
Dichlorodifluoromethane	WA			U	10	NS	µg/L
Ethylbenzene	WA			U	5	700	µg/L
Isobutyl alcohol	WA			U	20	NS	µg/L
Methacrylonitrile	WA			U	20	NS	µg/L
Methyl bromide	WA			U	10	NS	µg/L
Methyl chloride	WA			U	10	NS	µg/L
Methyl ethyl ketone	WA			U	10	NS	µg/L
Methyl iodide	WA			U	10	NS	µg/L
Methylene bromide	WA			U	10	NS	µg/L
Methylene chloride	WA			JV	1.91	5	µg/L
Propionitrile	WA			U	50	NS	µg/L
Styrene	WA			U	5	100	µg/L
Tetrachloroethylene	WA			U	5	5	µg/L

QA2 Laboratory Analyses (Cont'd)

Analyte	Lab	Rep	Avg	Qualifier	Result	Stand	Units
Toluene	WA			U	5	1000	µg/L
trans-1,3-Dichloropropene	WA			U	5	NS	µg/L
trans-1,4-Dichloro-2-butene	WA			U	100	NS	µg/L
Trichloroethylene	WA			U	5	5	µg/L
Trichlorofluoromethane	WA			U	5	NS	µg/L
Vinyl acetate	WA			U	10	NS	µg/L
Vinyl chloride	WA			U	10	2	µg/L
Xylenes (total)	WA			U	5	10000	µg/L

Quarterly Sampling of the Wetlands Along the Old F-Area Effluent Ditch: May 1994

QA  
(Trip Blank)

Field Measurements:  
Sample Date: 05/24/94

Analyte	Lab	Rep	Avg	Qualifier	Result	Stand	Units
1,1,1,2-Tetrachloroethane	WA			U	10	NS	µg/L
1,1,1-Trichloroethane	WA			U	5	200	µg/L
1,1,2,2-Tetrachloroethane	WA			U	5	NS	µg/L
1,1,2-Trichloroethane	WA			U	5	5	µg/L
1,1-Dichloroethane	WA			U	5	NS	µg/L
1,1-Dichloroethylene	WA			U	5	7	µg/L
1,2,3-Trichloropropane	WA			U	10	NS	µg/L
1,2-Dibromo-3-chloropropane	WA			U	20	0.2	µg/L
1,2-Dibromoethane	WA			U	20	NS	µg/L
1,2-Dichloroethane	WA			U	5	5	µg/L
1,2-Dichloroethylene	WA			U	5	70	µg/L
1,2-Dichloropropane	WA			U	5	5	µg/L
2-Hexanone	WA			U	10	NS	µg/L
4-Methyl-2-pentanone	WA			U	10	NS	µg/L
Acetone	WA			U	10	NS	µg/L
Acetonitrile (Methyl cyanide)	WA			U	20	NS	µg/L
Acrolein	WA			U	10	NS	µg/L
Acrylonitrile	WA			U	10	NS	µg/L
Allyl chloride	WA			U	100	NS	µg/L
Benzene	WA			U	5	5	µg/L
Bromodichloromethane	WA			U	5	NS	µg/L
Bromoform	WA			U	5	NS	µg/L
Carbon disulfide	WA			U	5	NS	µg/L
Carbon tetrachloride	WA			U	5	5	µg/L
Chlorobenzene	WA			U	5	100	µg/L
Chloroethane	WA			U	10	NS	µg/L
Chloroform	WA			U	5	100	µg/L
Chloroprene	WA			U	20	NS	µg/L
cis-1,3-Dichloropropene	WA			U	5	NS	µg/L
Dibromochloromethane	WA			U	5	NS	µg/L
Dichlorodifluoromethane	WA			U	10	NS	µg/L
Ethylbenzene	WA			U	5	700	µg/L
Isobutyl alcohol	WA			U	20	NS	µg/L
Methacrylonitrile	WA			U	20	NS	µg/L
Methyl bromide	WA			U	10	NS	µg/L
Methyl chloride	WA			U	10	NS	µg/L
Methyl ethyl ketone	WA			U	10	NS	µg/L
Methyl iodide	WA			U	10	NS	µg/L
Methylene bromide	WA			U	10	NS	µg/L
Methylene chloride	WA			JV	1.96	5	µg/L
Propionitrile	WA			U	50	NS	µg/L
Styrene	WA			U	5	100	µg/L
Tetrachloroethylene	WA			U	5	5	µg/L

## QA Laboratory Analyses (Cont'd)

Analyte	Lab	Rep	Avg	Qualifier	Result	Stand	Units
Toluene	WA			U	5	1000	µg/L
trans-1,3-Dichloropropene	WA			U	5	NS	µg/L
trans-1,4-Dichloro-2-butene	WA			U	100	NS	µg/L
Trichloroethylene	WA			U	5	5	µg/L
Trichlorofluoromethane	WA			U	5	NS	µg/L
Vinyl acetate	WA			U	10	NS	µg/L
Vinyl chloride	WA			U	10	2	µg/L
Xylenes (total)	WA			U	5	10000	µg/L

*Appendix B*

*Information on Electronic Data Diskettes*



**INFORMATION PROVIDED TO ESS ON ELECTRONIC DATA DISKETTES**

Sample name	Laboratory sample number	Accuracy
Chain of custody number	Laboratory replicate	Number of dilutions
Sample date	Testname	Dilution factor
Received date	Extraction/digestion method	Instrument
Extraction date	Analysis method	Analyst
Extraction time	Detection limit	Nominal concentration
Analysis date	Result qualifier	
Analysis time	Analysis qualifier	
Analysis batch	Bias	
Laboratory code	Analytical result	

WSRC-TR-94-0225

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*Appendix C*

*Calculation of VOC Concentration  
in the Headspace of a Well*

WSRC-TR-94-0225

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## CALCULATION OF VOC CONCENTRATION IN THE HEADSPACE OF A WELL

### Symbols:

H	=	Henry's Law constant
S	=	solubility in water
R	=	gas constant
T	=	temperature (K)
$P_{vp}$	=	vapor pressure
$Y_1$	=	mole fraction in gas
$X_1$	=	mole fraction in water
$P_T$	=	total pressure
$P_i$	=	partial pressure of gas

### EXAMPLE

Given: TCE concentration of 5000 ug/L in water at 20 degrees C (293K)

$$\text{Molecular weight of water} = \frac{1 \text{ mol}}{18 \text{ g}}$$

$$H_{TCE} = 1 \times 10^{-2} \left( \frac{\text{atm} \cdot \text{m}^3}{\text{mol}} \right)$$

$$\text{Molecular weight of TCE} = 131.5 \frac{\text{g}}{\text{mol}}$$

Find: TCE concentration in air.

### SOLUTION:

$$H = \frac{P_{vp}}{S}$$

$$\frac{Y_1}{X_1} = \frac{H}{P_T} \rightarrow Y_1 = \left( \frac{H}{P_T} \right) X_1 = \text{mole fraction within gas}$$

$$H_{TCE} = \left( 1 \times 10^{-2} \frac{\text{atm} \cdot \text{m}^3}{\text{mol}_{TCE}} \right) \left( 1000 \frac{\text{g}}{\text{L}} \right) \left( \frac{1 \text{ mol}_{\text{water}}}{18 \text{ g}} \right) \left( \frac{1000 \text{ L}}{\text{m}^3} \right)$$

$$H_{TCE} = 556 \frac{\text{atm} \cdot \text{mol}_{\text{water}}}{\text{mol}_{TCE}}$$

Calculate moles of TCE per liter of water for the given concentration.

$$\left( \frac{5000 \mu\text{g}}{\text{L}} \right) (1 \text{ L}) \left( \frac{1 \text{ mol}}{131.5 \text{ g}} \right) \left( \frac{10^{-6} \text{ g}}{\mu\text{g}} \right) = 3.8 \times 10^{-5} \text{ mol}_{TCE}$$

Calculate moles of water per liter of water

$$(1 L) \left( \frac{1000 g}{L} \right) \left( \frac{1 mol_{water}}{18 g} \right) = 55.6 mol_{water}$$

Calculate mole fraction in water

$$X_i = \frac{mol_{TCE}}{mol_{water}} = \frac{3.8 \times 10^{-5} mol_{TCE}}{55.6 mol_{water}} = 6.84 \times 10^{-7} \frac{mol_{TCE}}{mol_{water}}$$

Calculate mole fraction in gas

$$Y_i = \left( \frac{H}{P_T} \right) X_i$$

$$Y_i = \left( \frac{556 atm \cdot mol_{water}}{mol_{TCE}} \right) \left( \frac{1}{atm} \right) \left( 6.84 \times 10^{-7} \frac{mol_{TCE}}{mol_{water}} \right)$$

$$Y_i = 3.81 \times 10^{-4}$$

The partial pressure of gas at 1 atm =  $3.81 \times 10^{-4}$  atm

$$P_i V = n_i R T$$

$$n_i = \frac{M}{m}$$

Where: M = molecular weight of gas  
m = mass of gas confined to volume V.

$$\frac{P_i}{RT} = \frac{n_i}{V}$$

$$\frac{n_i}{V} = \frac{3.81 \times 10^{-4} atm}{(0.08285 \frac{L \cdot atm}{mol \cdot K}) (293K)}$$

$$\frac{n_i}{V} = (1.57 \times 10^{-5} \frac{mol}{L}) (131.5 \frac{g}{mol}) = 2.06 \times 10^{-3} \frac{g}{L}$$

Convert g/L to ug/m<sup>3</sup>

$$(2.06 \times 10^{-3} \frac{g}{L}) (10^6 \frac{\mu g}{g}) = 2.06 \times 10^3 \frac{\mu g}{L}$$

$$(2.06 \times 10^3 \frac{\mu g}{L}) (\frac{1 L}{1000 \text{ cm}^3}) (1 \times 10^6 \frac{\text{cm}^3}{\text{m}^3}) = 2.06 \times 10^6 \frac{\mu g}{\text{m}^3}$$

Convert to ppmv

$$\frac{m_g}{\text{m}^3} = \frac{(\text{ppmv}) (\text{mol wt})}{24.45}$$

$$\text{ppmv} = \frac{\frac{m_g}{\text{m}^3} (24.45)}{\text{mol wt}}$$

$$\text{ppmv} = 2.06 \times 10^6 \frac{\mu g}{\text{m}^3} (24.45) (\frac{1 \text{ mol}}{131.5 \text{ g}}) (\frac{1 \text{ mg}}{1000 \mu g}) = 383 \text{ ppmv TCE}$$